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**博士論文**

**Monazite as a tracer of crustal evolution**  
(モナサイトからたどる地殻進化)

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## Abstract

Orogeny has played an important role in the evolution of continental crust. Orogeny is divided into two types (Dewey and Bird, 1970; Matsuda and Uyeda, 1971): Pacific and collisional types. Pacific type orogeny, associated with the subduction of oceanic plate, is vital for the formation of juvenile continental crust. In contrast, collision type orogeny, causing buoyant continental collision, is significant for the reworking of continental crust. Constraints on the nature and timing of orogeny are essential for a better understanding of crustal evolution over the Earth's history.

Monazite, a light rare earth element (REE) phosphate, has a potential to be a powerful tracer for crustal processes. Monazite has been used as a geochronometer, geochemical tracer, and Nd isotopic tracer. The wide occurrence of monazite enables its various applications to the petrogenesis of granitic and metamorphic rocks. Monazite is also suited as a detrital high-resolution archive of orogenic events. The objective of this study is to advance our understanding of monazite geochemistry in order to use monazite as a tracer of crustal evolution.

It has been demonstrated that detrital monazite potentially provides more detailed insights in various scale orogenic events as compared to detrital zircon (Hietpas et al., 2010). Unfortunately, the various origins of monazite bring the difficulty in an interpretation of detrital monazite age peak. This difficulty can be overcome by linking geochemical fingerprints of monazite and its source rock types. Trace element composition is the key to understand the linkage due to its high dependence on co-existing minerals via element partitioning. However, there are two obstacles in monazite trace element geochemistry: (i) analytical problem in REE determination by laser ablation-inductively coupled plasma mass spectrometry (LA-ICP-MS) and (ii) shortage of trace element data especially for igneous monazite. To overcome these obstacles, I attempted to develop a method of accurate REE analysis. Furthermore, trace element systematics of igneous monazite has been studied by the comprehensive analyses of monazites from various granitic rocks. The synthesis of reliable datasets of monazite chemical compositions demonstrates the utility of monazite REE-Th-U systematics as an indicator of source rock type. The improved analytical method and geochemical provenance indicator were applied to detrital monazites from major African rivers to obtain the constraints on the timing and nature of Pan-African orogenic events.

Accurate REE determination of monazite by LA-ICP-MS can be inhibited by oxide interferences. Yet, the oxide production mechanism has been poorly understood. Here I determined the oxide production rates ( $\text{MO}^+/\text{M}^+$  ratios) of 15 REEs, Th, and U using synthetic and natural phosphates. I

found two contrasting oxide production behaviors against the Ar sample gas flow rate, depending on the  $\text{MO}^+$  dissociation energy ( $D_0(\text{MO}^+)$ ): the  $\text{MO}^+/\text{M}^+$  ratios of elements with  $D_0(\text{MO}^+) \geq 7$  eV increased with the gas flow rates from 0.85 to 1.00 L min<sup>-1</sup>, whereas those with  $D_0(\text{MO}^+) < 7$  eV were nearly constant. Moreover, while there is a generally positive relationship between  $\log(\text{MO}^+/\text{M}^+)$  and  $D_0(\text{MO}^+)$  among all elements, only the former showed a linear correlation whose slope increased with the gas flow rate. These results indicate that oxide production is a combined result of the following two processes: (i) the equilibrium reaction of  $\text{MO}^+ \leftrightarrow \text{M}^+ + \text{O}$  within the plasma and (ii) the bonding of  $\text{M}^+$  and O through collisions followed by radiative transition in the interface region. Oxide production through the first process was significant relative to the second process only when the elements had  $D_0(\text{MO}^+) \geq 7$  eV and the gas flow rate was high enough to decrease the plasma temperature to  $\sim 8000$  K. We further demonstrate the significance of middle-REE oxide interference on heavy-REEs during the analyses of monazites from pegmatites and garnet-bearing metamorphic rocks, which should be corrected using the determined oxide production rates.

Trace element systematics of igneous monazite has not been well understood so far. I conducted trace element and Nd isotope analyses of monazite in granitic rocks from the Busetsu pluton of the Japan arc to investigate the change of monazite geochemical signatures with granitic magma evolution. Furthermore, by integrating of monazite Nd-isotope and plagioclase Sr-isotope microanalyses, I revisited the magma sources, which were inferred from conventional whole-rock chemistry. The studied monazite grains show variations in the degree of negative Eu anomaly, Ce/Gd, and Th/U ratios. These variations correlate with each other, which can be accounted for by the fractionation of plagioclase and monazite considering the strong partitioning of Eu to plagioclase and of light-REE and Th to monazite, respectively. In the context of Rayleigh fractionation model, the degrees of plagioclase and monazite fractionation were quantitatively evaluated and found to be compatible with the geologic and petrologic observations. The isotopic data revealed significant variations in monazite  $^{143}\text{Nd}/^{144}\text{Nd}$  ratio and plagioclase  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios, which are obviously larger than those of whole-rock samples. This finding reflects that the multiple sources were involved in the generation of the granitic magma. It is demonstrated that an integration of geochemical and isotope microanalyses of monazite enables tracking open system magmatic processes.

To link monazite composition with igneous petrogenesis and differentiation, we carried out comprehensive LA-ICP-MS measurements of REE-Th-U abundances in monazites from magnetite-series and ilmenite-series granitic rocks across the Japan arc. The data revealed systematic

differences in monazite composition between magnetite-series and ilmenite-series samples, and between pegmatites and granites. In ilmenite-series granitic rocks, monazites from pegmatites showed larger negative Eu anomalies, lower light-REE/middle-REE ratios, and higher middle-REE/heavy-REE ratios than those from granites. These geochemical variations were attributed to significant fractional crystallization of feldspars, monazite, xenotime, and garnet during differentiation in relatively reduced peraluminous granitic magmas. In contrast, there was no remarkable difference in the REE fractionation pattern between magnetite-series pegmatites and granites. Furthermore, the magnitudes of the negative Eu anomalies in the magnetite-series monazites were smaller than those observed in the ilmenite-series samples. These features were interpreted to reflect the suppression of monazite, xenotime, and garnet fractionation and limited Eu incorporation into fractionating feldspars in relatively oxidized and non-peraluminous magmas. A comparison of the present data with previously reported data indicates that igneous monazites from granitic rocks are distinct from metamorphic monazites in REE composition, especially due to their larger negative Eu anomalies. Metamorphic monazites are further grouped into garnet-rich metamorphic with high middle-REE/heavy-REE and garnet-poor metamorphic origin with the low ratio. This finding highlights the potential utility of monazite REE-Th-U systematics as a provenance indicator for detrital monazites.

The geochemical indicator for source rock type was applied to detrital monazite from large rivers in Africa. We carried out U–Pb dating and geochemical analyses of approximately 500 detrital monazites from the Nile, Niger, Congo, Zambezi and Orange Rivers. The U–Pb dating defined age peaks at ~600 Ma for the Nile, ~580 Ma for the Niger, ~630, ~610 and ~550 Ma for the Congo, ~500 Ma for the Zambezi and 1200–1000 Ma for the Orange. All but the Orange age peaks correspond to the period of the Pan-African Orogeny. In the Niger, Congo and Zambezi, the Pan-African age peaks are 20–40 Myr younger than those defined by the detrital zircons from the same rivers. Based on the geochemical provenance indicator, the transition between igneous, garnet-rich, and garnet-poor metamorphic events were described in the age spectra of detrital monazite. The interpreted age peaks, together with the discrepancy between the detrital monazite and zircon ages, suggest that the Pan-African monazite age peaks essentially reflect the timing of syn- to post-collisional metamorphic events in the source terranes, whereas the detrital zircons ages provide a more representative record of magmatic events. The results imply that the Pan-African Orogeny was promoted by pre- to syn-collisional felsic magmatism during the early stages and, 20–40 Myr later, it was dominated by syn- and/or post-collisional metamorphism.

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## Table of Contents

<b>1. General Introduction .....</b>	<b>10</b>
<b>1.1 Crustal evolution and Orogeny .....</b>	<b>10</b>
<b>1.2 Monazite.....</b>	<b>13</b>
<b>1.2.1 Occurrence of monazite .....</b>	<b>13</b>
<b>1.2.2 Monazite as a geochronometer.....</b>	<b>15</b>
<b>1.2.3 Monazite as a geochemical tracer.....</b>	<b>17</b>
<b>1.2.4 Monazite as an isotope tracer.....</b>	<b>18</b>
<b>1.2.5 Obstacles for monazite as a tracer of crustal evolution .....</b>	<b>18</b>
<b>1.3 Overview of this thesis .....</b>	<b>20</b>
<b>1.3.1 Motivation and Objective.....</b>	<b>20</b>
<b>1.3.2 Outline.....</b>	<b>21</b>
<b>2. Unraveling the mechanism and impact of oxide production in LA-ICP-MS by comprehensive analysis of REE-Th-U phosphates .....</b>	<b>24</b>
<b>2.1 Introduction.....</b>	<b>24</b>
<b>2.2 Experimental .....</b>	<b>27</b>
<b>2.3 Results and Discussion.....</b>	<b>31</b>
<b>2.3.1 Oxide production rates .....</b>	<b>31</b>
<b>2.3.2 Mechanism of oxide production .....</b>	<b>36</b>
<b>2.3.3 Matrix effect .....</b>	<b>40</b>
<b>2.3.4 Impact of oxide interference .....</b>	<b>41</b>
<b>3. Granitic magma evolution deciphered by trace element and Nd isotope microanalysis of monazite.....</b>	<b>46</b>
<b>3.1 Introduction.....</b>	<b>46</b>
<b>3.2 The Busetsu granite in the Mikawa area, Japan.....</b>	<b>48</b>
<b>3.3 Analytical methods .....</b>	<b>51</b>
<b>3.5 Results .....</b>	<b>60</b>
<b>3.5.1 U–Pb age of monazite.....</b>	<b>60</b>
<b>3.5.2 Geochemical and Nd isotope compositions of monazite.....</b>	<b>62</b>
<b>3.5.3 Geochemical and Sr isotope compositions of plagioclase .....</b>	<b>65</b>
<b>3.5.4 Whole-rock analyses .....</b>	<b>68</b>

<b>3.6 Discussion .....</b>	<b>72</b>
<b>3.6.1 Geochronology of the Busetsu granite .....</b>	<b>72</b>
<b>3.6.2 Rare earth element behavior recorded in monazite .....</b>	<b>75</b>
<b>3.6.3 Geochemical evolution of the Busetsu granitic magma: integration of whole-rock and mineral geochemistry .....</b>	<b>81</b>
<b>3.7 Conclusion .....</b>	<b>85</b>
<b>4. Trace element systematics of igneous monazite from granitic rocks in the Japan arc: Implication for its use as an indicator of source rock type .....</b>	<b>86</b>
<b>4.1 Introduction.....</b>	<b>86</b>
<b>4.2 Geologic outline and samples .....</b>	<b>89</b>
<b>4.3. Analytical method .....</b>	<b>93</b>
<b>4.4 Results .....</b>	<b>98</b>
<b>4.3.1 Monazite from granitic rocks .....</b>	<b>98</b>
<b>4.4.1 Differentiation of granitic magmas recorded by monazite geochemistry .....</b>	<b>106</b>
<b>4.4.2 Intra-grain compositional variations in the Masaki pegmatite monazite.....</b>	<b>110</b>
<b>4.4.3 Geochemical fingerprints of monazites in various rock types for a provenance indicator.....</b>	<b>112</b>
<b>4.5 Conclusion .....</b>	<b>118</b>
<b>5. Application to detrital monazites from major African rives: Constraints on the timing and nature of the Pan-African Orogeny.....</b>	<b>119</b>
<b>5.1 Introduction.....</b>	<b>119</b>
<b>5.4 Results .....</b>	<b>136</b>
.....	<b>137</b>
<b>5.5 Discussion .....</b>	<b>144</b>
<b>5.5.1 Geochemistry and genesis of the detrital monazites .....</b>	<b>144</b>
<b>5.5.2 Comparison with detrital zircon .....</b>	<b>147</b>
<b>5.5.3 Constraints on the timing and nature of the Pan-African Orogeny .....</b>	<b>153</b>
<b>5.6 Conclusion .....</b>	<b>158</b>
<b>6. Synthesis.....</b>	<b>160</b>
<b>References.....</b>	<b>163</b>
<b>Appendix.....</b>	<b>189</b>



## **1. General Introduction**

### **1.1 Crustal evolution and Orogeny**

The Earth is unique among the terrestrial planets in a bimodal topography due to the existence of continental and oceanic crust (Taylor and McLennan, 1996). The oceanic crust is a ~7 km layer and dominantly made up of basaltic rocks, whereas the continental crust is a ~40 km layer and has an andesite composition on average (Christensen and Mooney, 1995; Rudnick and Fountain, 1995). The continental crust is subdivided into upper, middle, and lower layers on the basis of seismological observations (Holbrook et al., 1992; Christensen and Mooney, 1995; Rudnick and Fountain, 1995). The upper part has a granitic composition ( $\text{SiO}_2$  66.2% on average), leading to its preservation and long-term stability (Rudnick and Gao, 2003; Hawkesworth et al., 2010; Dhuime et al., 2011). Granitic melts are not in equilibrium with mineral assemblages of any mantle rocks, indicating that granitic crust cannot be directly derived from mantle melting (Rudnick, 1995). Hence, the generation of continental crust requires crust differentiation such as re-melting of mantle-derived materials and/or crystal fractionation (Anderson 1987, Rudnick, 1995; Taylor and McLennan, 1995). Although how continental crust has been formed is still in debate (e.g., Taira et al., 1998; Jagoutz and Schmidt, 2012; Arndt, 2013), the continental crusts with chemically and structurally mature characteristics indicate its generation and evolution through plate tectonics over the Earth's history (Korenaga, 2013; Hawkesworth et al. 2017, Iizuka et al., 2017).

Plate tectonics has driven the interaction of continental crust with not only mantle but also hydrosphere, atmosphere, and biosphere over the Earth's history. It is widely accepted that the continental crust plays an important role in carbon cycle via silicate weathering, which controls the long-term climate change (e.g., Tajika and Matsui, 1992). The secular change in elemental flux from eroded continental crust largely affects the chemical composition of sea water (Large et al., 2018), and it might be a trigger of the evolution of life around the Precambrian-Cambrian boundary (Shimura et al., 2014). Hence, how the continental crust has evolved is one of the most important topics in Earth science.

Orogeny is the results of plate convergence and essential for the evolution of continental crust. Dewey and Bird (1970) originally distinguished the Cordilleran-type orogeny and collision-type orogeny in terms of the geometric relation of thrusts and the development of a paired metamorphic belt (Miyashiro, 1961), and subsequently these types were refined as the Pacific and collision types to emphasize the generality (Matsuda and Uyeda, 1971). The Pacific type, associated with the subduction of oceanic plate, is a continuous process; whereas the collision type, caused by buoyant continental collision, has episodically occurred during the supercontinent cycle (Nance and Murphy, 2013). These two type orogenic events play contrasting roles in the evolution of continental crust. In convergent margins where the Pacific-type orogeny is active, the production of voluminous granitic rocks is thought to have been caused by the differentiation of basaltic and andesitic magma generated by a flux melting of wedge

mantle (Jagoutz, 2010; Annen et al., 2006; Jagoutz et al., 2011; Nandedkar et al., 2014).

In contrast, continent-continent collisions have produced granitic magmas through remelting of pre-existing rocks including meta-sedimentary rocks (Collins, 2002; Clemens, 2003; Chappell et al., 2000; Miller et al., 2003). Although the lower layer of continental crust is widely believed as basaltic material, Hacker et al. (2011) proposed that the decompression melting of subducted continental crust resulted in the formation of lower crust consisting of felsic rocks rather than mafic rocks. The Pacific type orogeny is significant for the formation of juvenile continental crust (Santosh et al., 2010), whereas the collision type orogeny is important for the reworking of continental crust (Cawood et al., 2009).

A high-resolution archive of orogenic events has been constructed over several decades through detailed geological, structural, petrological, seismological, and paleomagnetic studies. One approach to sample the continental crust is using detrital minerals derived by weathering and erosion of various exposed rocks. The application of in-situ dating techniques of zircons to detrital samples showed the correspondence between zircon age peaks and the timing of supercontinent assembly accompanying global Pacific- and collision-type orogeneses (Condie et al., 2005; Campbell and Allen, 2008; Rino et al., 2008; Belousova et al., 2010; Iizuka et al., 2010a). Although it is controversial whether the zircon age peaks reflect the major timing of crustal production (Arndt and Davaille, 2013; Condie et al., 2017) or the preservation bias during supercontinent assembly (Hawkesworth et al., 2009; Spencer et al., 2015, 2017), detrital

zircon study has provided insights regarding the timing of orogenic events. The zircon age data basically reflect the timing of voluminous felsic magmatism including both of juvenile magma input (Pacific-type orogeny) and reworking (collision-type orogeny). The Hf and O isotope analyses of zircon enable an estimate of relative contributions of reworking to juvenile magma input (e.g., Valley et al., 1994; Griffin et al., 2000). However, the approach using detrital zircon has limitations in deconvolution between the two-type orogenic events during supercontinent assembly and precise constraints on the timing of small-scale orogenic events.

Monazite [(La, Nd, Ce, Th) PO<sub>4</sub>] has the potential to provide more robust constraints on the timing and nature of orogenic events. Due to its wide occurrence, monazite geochemistry, geochronology, and isotope chemistry have been used in the pursuit of understanding various processes in crustal evolution, including magmatic and metamorphic processes and diagenesis (Overstreet, 1967; Parrish, 1990). Recently, it has been shown that detrital monazites derived from igneous and metamorphic rocks record orogenic events more accurately than detrital zircons do (Hietpas et al., 2010, 2011). There are, however, still tasks to maximize the potential of monazite. The prospects and current limitations of monazite as a tool in crustal evolution studies are reviewed in the next section.

## 1.2 Monazite

### 1.2.1 Occurrence of monazite

Igneous monazite commonly occurs in low-Ca peraluminous granitoids (e.g., Xie et al., 2006; Kelts et al., 2008; Chattopadhyay et al., 2015) and less frequently in peralkaline igneous rocks (Bea, 1996; Hoshino et al., 2012). Igneous monazite is also found as a primary accessory phase in granitic pegmatites that are highly enriched in rare earth element (REE) (Hoshino et al., 2012; Chen et al., 2017). While monazite geochronology might be complementary to zircon one in many cases, Th–U–Pb dating of monazite is essential for the petrogenesis of rocks where zircon is absent or unstable, such as carbonatites and other highly alkaline systems (Roberts et al., 2010). Monazite plays a critical role in REE behaviors in granitic magma as a host phase of light-REE (LREE). The 80~55 wt% of whole-rock LREE contents reside within monazite in peraluminous granites (Bea, 1996).

Metamorphic monazite is frequently found in pelitic and psamitic low- to high-grade metamorphic rocks (e.g., Rasmussen et al., 2001, 2005; Rubatto et al., 2001, 2006; Buick et al., 2010; Holder et al., 2015), though less commonly reported from intermediate metamorphic rocks (Bingen and van Breemen, 1998). Monazite can grow during not only high- but also low-grade metamorphic events, which is unique among accessory minerals suitable for U–Pb dating. These metamorphic monazites are characterized by complicated chemical zoning (Williams et al., 2007). The change in monazite trace element composition has been interpreted to reflect the change in crystallization/growth condition such as co-existing mineral assemblages. Thus, monazite geochemistry might allow us to link the multiple monazite-forming events

with specific points within P-T-t path (Mottram et al., 2014; Štípská et al., 2015).

Hydrothermal monazite typically shows differences in texture and chemical composition from other origins. Its texture is characterized by a cluster of minute grains or highly irregular morphology (e.g., Wing et al., 2003; Grew et al., 2008). Many studies reported that recrystallization *via* dissolution-reprecipitation processes fully or partially reset the Th–U–Pb isotopic system even when temperature is enough low to regard the effect of Pb diffusion as negligible, leading to the potential for dating hydrothermal activity (Ayers et al., 2006; Kelly et al., 2012). Hydrothermal alteration of monazite has been mainly investigated based on experimental studies (Seydoux-Guillaume et al. 2002; Harlov and Hetherington, 2010, 2011; Hetherington et al., 2010; Budzyń et al., 2011, 2017), revealing high dependence of monazite stability on the fluid composition.

### 1.2.2 Monazite as a geochronometer

The value of monazite as a geochronometer comes from its crystal structure that accommodates Th<sup>4+</sup> and U<sup>4+</sup> (Ni et al. 1995; Huminicki and Hawthorne 2002; Clavier et al. 2011). Importantly, monazite contains little common Pb and the diffusion of the isotope system in monazite is slow enough to retain the original isotopic information (Williams et al., 2007). The diffusion experiment reported by Cherniak et al. (2004) demonstrated that the Pb closure temperature of a 10-μm-monazite grain would be >900 °C, given a cooling rate of 10 °C/Ma. Similar estimates of Pb closure temperatures have

been reported by many previous studies using natural samples (Rubatto et al., 2001; Bosch et al., 2002; McFarlane and Harrison, 2006). Besides, it is widely reported that metamictization caused by radiation damage is limited for monazite; whereas Pb-loss enhanced by metamictization often occurs for zircon (Black et al., 1984; Meldrum et al., 1998; Seydoux-Guillaume et al., 2007; Nasdala et al., 2010).

Different *in-situ* monazite Th–U–Pb dating methods have different advantages. Radiometric dating is basically conducted by mass spectrometry to obtain isotopic ratios. Monazite, however, can be dated by EPMA thanks to enough high concentrations of Th, U, and Pb. The ages are calculated from the collected Th–U–total Pb data by assuming no common Pb and no Pb-loss or by constructing a chemical isochron (CHIME method) (Suzuki and Adachi, 1998, 1991; Jercinovic and Williams, 2005; Suzuki and Kato, 2008; Montel et al., 1996; Cocherie et al., 1998; Pyle et al., 2005;). The EPMA is advantageous in terms of the high spatial resolution of  $\sim 1 \mu\text{m}$  and the capability of major element mapping. In contrast, the SIMS and LA-ICP-MS techniques provide isotopic information with higher sensitivity although the spatial resolutions are lower than the resolutions possible with EPMA. The concordance of the U–Th–Pb system is essential for robust interpretation of the obtained ages. U–Th–Pb dating using SIMS was pioneered by DeWolf et al. (1993) and has been developed and applied by many workers (Ireland and Gibson, 1998; Stern and Berman, 2001; Vry et al., 1996; Zhu et al., 1997; Rasmussen et al., 2001; Sano et al., 2006; Jantos et al., 2012; Kusiak et al., 2014). Regarding LA-ICP-MS, solid-state 213 nm Nd:YAG or 193 nm ArF excimer

and quadrupole-ICP-MS (Q-ICP-MS) and sector filed-ICP-MS (SF-ICP-MS) are widely deployed (Machado and Gauthier, 1996; Kosler et al., 2001; Foster et al., 2002; Horstwood et al., 2003). LA-ICP-MS is capable of a rapid measurement (e.g., <1 min per a measurement) and, therefore, collecting large statistically meaningful datasets for detrital monazite study.

### **1.2.3 Monazite as a geochemical tracer**

The REE composition of monazite has been used as a geochemical tracer for crustal processes (e.g., Suzuki et al., 1992; Spear and Pyle 2002). Monazite has abundant LREE contents in the order of weight percentage and various heavy-REE (HREE) contents ranging from several thousands to <1 ppm. The variability of monazite REE composition could be linked to its genesis, or crystallization condition, because the REE composition is controlled by co-existing mineral assemblages, whole-rock composition, melt composition, temperature, pressure, and fluid activity (e.g., Williams et al., 2007; Hetherington et al., 2017). The high contents of REEs allow us to carry out spot analysis with high spatial resolution by electron probe microanalyzer (EPMA), laser ablation-inductively coupled plasma-mass spectrometry (LA-ICP-MS), and secondary ion mass spectrometry (SIMS). The trace element behavior in monazite has been investigated mainly for metamorphic monazite, and the geochemical variation described by microanalysis is connected to the change in a stable mineral assemblage of metamorphic rocks (e.g., Mottram et al., 2014; Holder et al., 2015). On the other hand, reports of trace element geochemistry on igneous monazite are limited.

#### **1.2.4 Monazite as an isotope tracer**

The Sm–Nd isotope system of monazite is also used as a powerful tracer for crustal processes. The  $^{143}\text{Nd}/^{144}\text{Nd}$  and  $^{147}\text{Sm}/^{144}\text{Nd}$  ratios of monazite can be obtained by the LA-MC-ICP-MS technique with a spatial resolution of 20–30  $\mu\text{m}$  due to high concentrations of Nd and Sm ( $\sim 10^4$  to  $10^5$  ppm) (Mcfarlane and Mcculloch, 2007; Yang et al., 2008; Iizuka et al., 2011a). Fisher et al. (2017) demonstrated that the Nd isotope variations within single monazite grains in a granite reflect the open-system magmatic processes (Fisher et al., 2017). The Nd isotope analysis has been also applied to monazites in metasedimentary rocks and detrital samples to track the timing of crustal evolutionary events (Iizuka et al., 2011b; Liu et al., 2017). As compared to the Lu–Hf isotope analysis of zircon, the Sm–Nd isotope analysis of monazite is advantageous in minor inheritance of older core and smaller analytical volumes required.

#### **1.2.5 Obstacles for monazite as a tracer of crustal evolution**

The typical occurrence of igneous monazites in peraluminous granites, of which voluminous production is associated with the collision-type orogeny, in addition to metamorphic origins is in contrast to common occurrence of zircon in metaluminous to peraluminous granite. This difference in occurrence between monazite and zircon can be a clue to capture the transition from Pacific-type to collision-type orogenic events. However, the various origins of monazite bring the difficulty in interpretations of monazite age spectra. This difficulty can be overcome by linking geochemical

fingerprints of monazite and its source rock types. Trace element composition of monazite is the key to understand the linkage due to its high dependence on co-existing minerals via element partitioning. However, there are two obstacles: (i) analytical problems of REE determination and (ii) limited data of monazite trace element compositions.

Microanalysis techniques are essential for determining reliable geochemical composition of minute monazite grains or individual sub-domains within single grains. Traditionally, the REE systematics has been investigated using the EPMA technique because of its utility for elemental mapping (e.g., Franz et al., 1996; Broska et al., 2005). For instance, the Th- and Y-zoning of monazite grains have been linked to multiple growth in prograde/retrograde metamorphism (e.g., Pyle and Spear 2003; Kohn et al. 2005). However, the EPMA technique has a difficulty in determining accurate HREE and Eu contents in monazite because of their low abundances and serious interferences in the complicated REE L-line series (Suzuki et al., 1990; Holder et al., 2015). As a result, the systematics of trace element in monazite, especially HREE, is still unknown.

Recent advances in LA-ICP-MS have enabled the quantification of trace HREE with high spatial resolution (<20 $\mu$ m). Spectral interferences, however, can be problematic during the REE analysis of monazite by LA-ICP-MS as well. The contrasting concentrations for weight percent-order LREE and ppm-order HREE in monazite result in high signal intensities of LREE-oxides comparable to those of HREEs even when oxide production rates are decreased to less than 1%. Unfortunately, the productions of

REE oxides are inevitable due to the presence of oxygen derived from samples. This impact on the quantification of REE concentrations needs to be assessed and corrected for an accurate measurement.

Monazites in pelitic and psamitic metamorphic rocks have been relatively well studied, and the effects of co-existing minerals, such as feldspars, garnet and zircon on the trace element composition of monazite were inferred (Rubatto et al., 2006; Buick et al., 2010; Iizuka et al., 2010b; Mottram et al., 2014; Štípská et al., 2015). In contrast, the available trace element data of monazites in granitic rocks, carbonatites, and thermal fluid deposits are significantly scarce. Accurate trace element data of various monazite, especially igneous monazite, are required for better understanding of monazite trace element systematics. Comparing the trace element compositions of monazites from various rock types unravels their geochemical characteristics, which leads to its application as a provenance indicator.

### **1.3 Overview of this thesis**

#### **1.3.1 Motivation and Objective**

The evolution of continental crust is closely related to orogeny in convergent margins. The questions to be addressed are when orogenic events have occurred through the Earth's history and what is the nature of orogenic events (the nature of orogeny: crustal growth vs. reworking). The objective of this work is to advance our understanding of monazite geochemistry in order to use monazite as a tracer of crustal evolution.

Particular emphasis is placed on a trace element microanalysis of monazite using the LA-ICP-MS technique. Better understanding of the trace element systematics is essential for tracing complex magmatic and metamorphic processes during monazite crystallization/growth. The approach using detrital monazite has a potential to be a breakthrough for more accurate constraints on the timing of orogenic events and tracking the transition from the Pacific-type to collision-type orogeny during supercontinent assembly.

There are some obstacles to be overcome for this objective, which includes an analytical problem, deficit in monazite trace element data, and ambiguity in the relation between monazite origin and its trace element characteristics, as discussed in the previous section. First, more accurate REE analysis by LA-ICP-MS was developed using synthetic REE phosphates. Second, the trace elements systematics of igneous monazite was revealed by the geochemical study of monazite in granitic rocks. The synthesis of reliable datasets of monazite trace element composition clarified the linkage between the geochemical fingerprints and source rock types. Finally, the improved REE analysis method and the developed geochemical indicator of source rock type were applied to detrital monazites to obtain constraints on the timing and nature of orogenic events.

### **1.3.2 Outline**

The following body of this thesis is divided up into five chapters. Each of the

subsequent chapters has been individually published or prepared for submission. Each chapter contains its own introduction, results, discussion, and conclusion. In the final chapter, I summarize these works and provide future perspectives on the utility of monazite.

For reference, Chapter 2 is about the improvement of REE analysis, which was published as “Unraveling the mechanism and impact of oxide production in LA-ICP-MS by comprehensive analysis of REE-Th-U phosphates” in *Journal of Analytical Atomic Spectrometry*, **32**, 2003–2010, 2017. In chapter 3, the REE systematics of igneous monazite is mainly discussed and the contents of this chapter is in preparation for submission. Chapter 4 is based on the published paper entitled “REE-Th-U and Nd isotope systematics of monazites in magnetite-and ilmenite-series granitic rocks of the Japan arc: Implications for its use as a tracer of magma evolution and detrital provenance” in *Chemical Geology*, **484**, 69–80, 2018, in addition to new data obtained in this work. In this chapter, the geochemical indicator of source rock type is proposed based on the comparison of compositions between monazites of various origins. Chapter 5 has been revised from the paper entitled “U–Pb chronology and geochemistry of detrital monazites from major African rivers: Constraints on the timing and nature of the Pan-African Orogeny” in *Precambrian Research*, **282**, 139–156, 2016. Geological interpretation of detrital monazite age peaks is given based on the geochemical compositions for the first time and the comparison between the detrital monazite and zircon provides the insights on orogenic events around the Gondwana

supercontinent assembly.

## **2. Unraveling the mechanism and impact of oxide production in LA-ICP-MS by comprehensive analysis of REE-Th-U phosphates**

### **2.1 Introduction**

Laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) has been widely used for trace element and isotopic determinations of geological and biological samples because it offers a means of *in-situ* analysis of solid samples with high spatial resolution and high analytical throughput (Russo et al., 2002, 2013). However, isobaric and polyatomic spectral interferences remain a major challenge in accurate LA-ICP-MS analysis (Hattendorf et al., 2016). One of the most problematic interference is from oxide ions of elements that are abundantly included in the samples because the introduction of oxygen from the sample matrix and entrained air in the plasma is inevitable during ICP-MS analysis. For instance, although rare earth elements (REE) in rock and mineral samples can provide clues to their origins (Condie and Hunter, 1976; Rapp et al., 1991; Plank and Langmuir, 1998; Belousova et al., 2002; Itano et al., 2016), oxide ions of light- and middle-REE (LREE and MREE) interfere with M-REE and heavy-REE (HREE) ions, and these interferences can cause inaccuracies for REE analyses of samples that are depleted in HREE relative to LREE.

Oxide production during ICP-MS measurements has been traditionally studied by analyzing solution samples under wet plasma conditions (Douglas and French, 1986; Vaughan and Horlick, 1990; Niu and Houk, 1996; Longerich et al., 1987; Poussel et al.,

1994; Romero et al., 1997; Aries et al., 2000). The results showed that the oxide production rate ( $\text{MO}^+/\text{M}^+$  ratio) strongly depends on plasma and interface conditions, such as the radio frequency (RF) power applied to the plasma, sample gas flow rate, sampling depth and geometry of the sampling and skimmer cones (Douglas and French, 1986; Longerich et al., 1987; Vaughan and Horlick, 1990; Poussel et al., 1994; Niu and Houk, 1996; Romero et al., 1997; Aries et al., 2000). Specifically, a higher plasma power or a lower sample gas flow rate results in a lower  $\text{MO}^+/\text{M}^+$  ratio. These observations suggest that oxide production occurs within the plasma and/or interface regions (Gray, 1986; Gray and Williams, 1987; Vaughan and Horlick, 1990; Niu and Houk, 1996; Lichte et al., 1987; Longerich et al., 1987; Poussel et al., 1994; Payne et al., 2013). Moreover, a positive correlation has been demonstrated between the log  $(\text{MO}^+/\text{M}^+)$  and the  $\text{MO}^+$  dissociation energy ( $D_0(\text{MO}^+)$ ). This relationship may reflect the thermal equilibrium between  $\text{M}^+$  and  $\text{MO}^+$  species. However, equilibrium temperatures, estimated by fitting to a Boltzmann distribution, range from  $\sim 5000$  to  $\sim 21000$  K (Douglas and French, 1986; Vaughan and Horlick, 1990; Houk and Praphairaksit, 2001; Longerich et al., 1987; Kubota et al., 1989; Shibata et al., 1993; Evans et al., 2002), many of which are apparently higher than the plasma temperature of  $\sim 8000$  K. This led some authors to argue that significant oxide production within the plasma is unlikely (Douglas and French, 1986; Vaughan and Horlick, 1990; Longerich et al., 1987). More recently, oxide production under dry plasma conditions was studied using the desolvating nebulizer technique (Newman et al., 2009; Newman, 2012;

Kimura et al., 2013). The results revealed that oxide production is dependent on the cone geometry and, for some elements such as Nd and Ce, it is associated with mass-independent isotope fractionation. Because the isotope fractionation can be well explained by the nuclear field shift effect, it has been proposed that the oxide production occurs in the interface region through collisional reactions under an excited state of the outermost electron orbital spin that enhance a nuclear volume shift and result in bonding/dissociation between metal and oxygen ions under a metastable state (Newman et al., 2009).

There are a few comprehensive researches on oxide production during LA-ICP-MS analysis (Kent and Ungerer, 2005; Witte and Houk, 2012; Gilbert et al., 2017). Kent and Unger (2005) investigated oxide production of Ba and LREE (La, Ce, Pr, and Nd) during LA-ICP-MS analyses under variable samples gas flow rates and found a dependence of the  $\text{MO}^+/\text{M}^+$  ratio on the gas flow rate and  $D_0(\text{MO})$ , generally consistent with the results of solution-ICP-MS measurements. Unlike solution-ICP-MS, however, the relative ratios of  $\text{MO}^+/\text{M}^+$  between each of L-REE showed a non-monotonic change with the sample gas flow rate (Kent and Ungerer, 2005). Witte and Houk (2012) found that the addition of N<sub>2</sub> and He to Ar sample gas flow reduces  $\text{MO}^+/\text{M}^+$  ratios of some REE (Y, Ce, Tb, and Tm), Cs, and Ta as a result of temperature increase at the optimum sampling position within the plasma. More recently, Gilbert et al. (2017) showed that  $\text{MO}^+/\text{M}^+$  ratios of some elements such as Th depend on the matrix of the sample being ablated (e.g., glass, metal, silicate, and phosphate), indicating the requirement of a

matrix-match reference material for a robust correction of interferences by the oxide ions.

To gain a better understanding of oxide production in LA-ICP-MS and to assess the impact of oxide interferences, we analyzed synthetic and natural REE phosphates as well as a synthetic silicate glass using the LA-ICP-MS technique. The synthetic REE phosphates enabled us, for the first time, to directly determine the  $\text{MO}^+/\text{M}^+$  ratios of all REE during LA-ICP-MS analysis. Since the REE are f-block elements and have similar chemical properties but highly variable  $D_0(\text{MO}^+)$ , our results allow a rigorous evaluation of the dependence of the  $\text{MO}^+/\text{M}^+$  on  $D_0(\text{MO}^+)$ , which, in turn, provides new insights into the mechanism of oxide production in ICP-MS. Furthermore, the results highlight the importance of oxide interference correction for the accurate determination of REE concentrations in natural monazite, a phosphate mineral that has been widely used for U–Pb dating and REE geochemistry (Rubatto et al., 2006; Mottram et al., 2014; Holder et al., 2015; Stipska et al., 2015).

## 2.2 Experimental

For determination of REE oxide production rates ( $\text{REEO}^+/\text{REE}^+$ ) during LA-ICP-MS analysis, we analyzed 15 synthetic phosphates of single REE (Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, and Lu) that were originally prepared as standards for electron probe microanalysis (EPMA) following the method of Cherniak et al. (2004) (courtesy of K. Ichimura). To evaluate the oxide production rates of U and Th and the

effect of the sample matrix on oxide production, we used the synthetic silicate glass NIST SRM 610 standard (Pearce et al., 1997) and a natural monazite 16-F-6 from the Pilbara Craton, Western Australia, which is used as an in-house standard for U–Pb dating and Sm–Nd isotopic analysis (Black et al., 2004; Simonetti et al., 2006; Iizuka et al., 2011a). In addition, the significance of REE oxide interferences were assessed by analyzing igneous monazites in a granite and pegmatite from Japan (Busetsu and Shichironai granitic bodies, respectively) (Hoshino et al., 2012) and a metamorphic monazite in a meta-conglomerate from Mt. Narryer in the Yilgarn Craton of Western Australia (Iizuka et al., 2010b).

All analyses were performed on a Thermo Fisher Scientific iCAP Qc ICP-MS coupled with a CETAC LSX-213 G2+ Nd:YAG LA system at the Department of Earth and Planetary Science, The University of Tokyo. Operational conditions of the instruments are listed in Table 2-1. The iCAP Qc is equipped with a collision/reaction cell where oxide ions can be attenuated through the collision/reaction with an additional gas. We did not utilize the collision/reaction cell technique, however, because the aim of this study is to better understand the mechanism of oxide production in ICP-MS. The LA system is equipped with the active two-volume HelEx™ Cell, in which He gas was used to flush the ablated aerosol particles (Eggins et al., 1998). Before sample analysis, the sample cell was thoroughly purged with He gas so that the remaining atmospheric oxygen in the cell had less effect on the measured  $\text{MO}^+/\text{M}^+$  ratios. When the purge was insufficient, the  $\text{MO}^+/\text{M}^+$  ratios decreased markedly with time during each analytical

session (Kent and Ungerer, 2005). The Ar sample gas flow rate was varied from 0.75 to 1.00 L min<sup>-1</sup> in a 0.05 L min<sup>-1</sup> steps to investigate its effect on the oxide production, while other operating conditions thought to control the MO<sup>+</sup>/M<sup>+</sup> ratio, such as sampling depth, plasma power, and cone geometry (Douglas and French, 1986; Longerich et al., 1987; Vaughan and Horlick, 1990; Poussel et al., 1994; Niu and Houk, 1996; Romero et al., 1997; Aries et al., 2000) were not changed. Measured MO<sup>+</sup>/M<sup>+</sup> ratios were corrected for mass bias using an exponential law. The mass bias factor was obtained for each analytical condition by normalizing the measured <sup>160</sup>Gd/<sup>155</sup>Gd ratio to the synthetic Gd phosphate of 1.477 (Berglund and Engi, 2011). Note that the choice of linear or power law for the mass bias correction would not alter the arguments below.

To evaluate the significance of the REE oxide interference correction during LA-ICP-MS analyses of natural monazites, the gas flow rate was set at 0.77 L min<sup>-1</sup> to reduce the MO<sup>+</sup>/M<sup>+</sup> ratio to the same level as in previous studies (Janots et al., 2008; d'Abzac et al., 2010). For the determination of REE concentrations, NIST SRM 610 standard glass was used as an external standard. Furthermore, independently determined Ce concentrations were used as an internal standard. The Ce concentrations were measured by EPMA using a JEOL JXA-8530F at the Department of Earth and Planetary Science, The University of Tokyo.

Table 2-1: Instrumental operational parameters of the LA-ICP-MS

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<b>Nd:YAG Laser</b>	CETAC LSX-213 G2+ (CETAC, USA) equipped with the active two-volume HelEx™ Cell
Laser source	213 nm
Pulse width	5 ns
Beam diameter	20 $\mu\text{m}$
Repetition rate	4 Hz
Laser fluence	20.54 J $\text{cm}^{-2}$
Laser He carrier gas	0.95 L $\text{min}^{-1}$
<b>ICP-MS</b>	iCAP Qc (Thermo Fisher Scientific)
RF-power	1550 W
Sampling depth	5.1 mm
Cool gas (Ar)	14 L $\text{min}^{-1}$
Auxiliary gas (Ar)	0.78 L $\text{min}^{-1}$
Sample gas (Ar)	0.75–1.00 L $\text{min}^{-1}$
Measured isotopes	$^{89}\text{Y}$ , $^{105}\text{Pd}$ , $^{139, 138}\text{La}$ , $^{140, 136, 138, 142}\text{Ce}$ , $^{141}\text{Pr}$ , $^{143, 142, 146}\text{Nd}$ , $^{147, 149, 152, 154}\text{Sm}$ , $^{151, 153}\text{Eu}$ , $^{157, 154, 155, 156, 158, 160}\text{Gd}$ , $^{159}\text{Tb}$ , $^{163, 160, 161, 162}\text{Dy}$ , $^{165}\text{Ho}$ , $^{167, 168, 170}\text{Er}$ , $^{169}\text{Tm}$ , $^{173, 171, 172, 174, 176}\text{Yb}$ , $^{175, 176}\text{Lu}$ , $^{179}\text{Hf}$ , $^{195}\text{Pt}$ , $^{232}\text{Th}$ , $^{238}\text{U}$ , $^{232}\text{Th}^{16}\text{O}$ , $^{238}\text{U}^{16}\text{O}$

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## 2.3 Results and Discussion

### 2.3.1 Oxide production rates

The mass-bias-corrected  $\text{REEO}^+/\text{REE}^+$  ratios of the synthetic REE phosphates, 16-F-6 standard monazite, and NIST SRM 610 glass are presented in Appendix Table 2-1. The  $\text{ThO}^+/\text{Th}^+$  and  $\text{UO}^+/\text{U}^+$  ratios of the 16-F-6 monazite and NIST SRM 610 glass are listed in Table 2-2, showing that the two different matrix samples yielded indistinguishable  $\text{MO}^+/\text{M}^+$  ratios within analytical uncertainty at all studied sample gas flow rates except for  $0.85 \text{ L min}^{-1}$ , where the monazite yielded a  $\sim 20\%$  lower  $\text{ThO}^+/\text{Th}^+$  ratio. The  $\text{MO}^+/\text{M}^+$  ratios are plotted against sample gas flow rate in Fig. 2-1, together with signal intensities of  $\text{M}^+$  and  $\text{MO}^+$ . Signal intensities of  $\text{M}^+$  decreased with the increase of the sample gas flow rate for all analyte elements (Fig. 2-1). By contrast, the response of  $\text{MO}^+$  signals and  $\text{MO}^+/\text{M}^+$  ratios to the change in sample gas flow rate varied and can be divided into two groups: (1)  $\text{MO}^+/\text{M}^+$  ratios of Sm, Eu, Dy, Ho, Tm, Er, Yb, and Lu were relatively constant over the range of studied sample gas flow rates, whereas (2) those of Y, La, Ce, Pr, Nd, Gd, Tb, Th and U were reasonably constant between the sample gas flow rates of  $0.75$  and  $0.85 \text{ L min}^{-1}$  but exponentially increased from  $0.85$  to  $1.00 \text{ L min}^{-1}$ . Notably, the first group has low  $D_0(\text{MO}^+)$  ( $\leq 6 \text{ eV}$ , hereinafter referred to as weak-oxide formers), whereas the second group has higher  $D_0(\text{MO}^+)$  ( $\geq 7 \text{ eV}$ , referred to as strong-oxide formers) (Ackermann, 1976; Haynes, 2013). In addition, detailed examination of the data shows that the  $\text{YbO}^+/\text{Yb}^+$  and  $\text{EuO}^+/\text{Eu}^+$  ratios decreased slightly with an increase in the sample gas flow rate (Figs. 2-1 and 2-2).

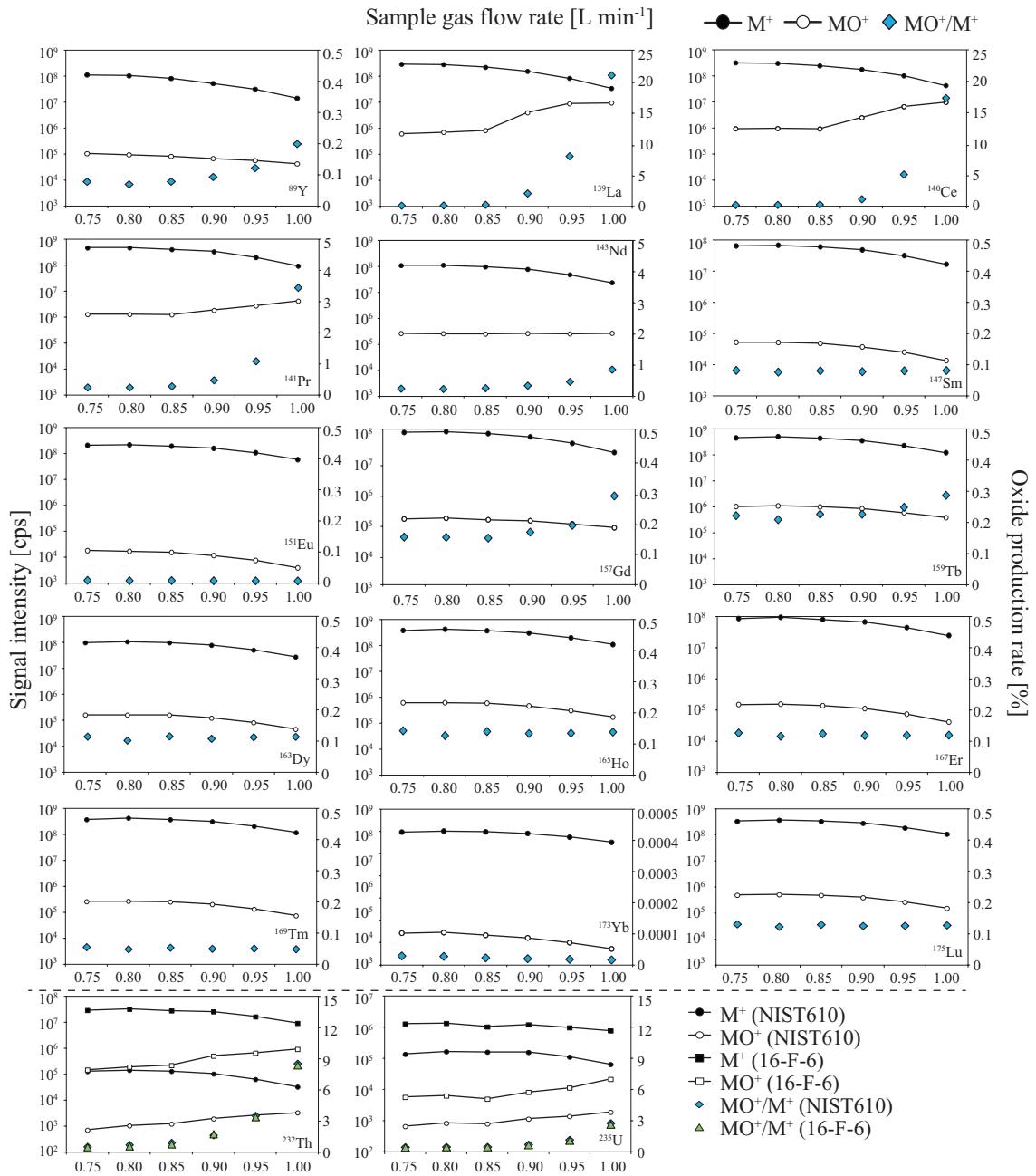


Fig. 2-1. Signal intensities of analyte elements and their oxides and the oxide production rates as a function of sample gas flow rates ranging from 0.75 to 1.00 L min<sup>-1</sup>. The data were obtained from synthetic REE phosphates for REE and from the NIST SRM 610 and the 16-F-6 monazite for Th and U.

The log ( $\text{MO}^+/\text{M}^+$ ) values are plotted as a function of  $D_0(\text{MO}^+)$  in Fig. 2-2a. The log ( $\text{MO}^+/\text{M}^+$ ) values are also plotted against  $D_0(\text{MO})$  in the Appendix Fig. S2-1, following several previous studies (Aries et al., 2000; Kent and Ungerer, 2005; Ackermann, 1976). There is a generally positive relationship between the log ( $\text{MO}^+/\text{M}^+$ ) and  $D_0(\text{MO}^+)$  as well as  $D_0(\text{MO})$ , consistent with the previous studies (Douglas and French, 1986; Longerich et al., 1987; Aries et al., 2000; Litche et al., 1987; Kent and Ungerer, 2005). In detail, however, our data reveal that the positive relationship systematically changes with the ranges of the  $D_0(\text{MO}^+)$  and sample gas flow rate (Fig. 2-2a). At gas flow rates of 0.75 to 0.85 L min<sup>-1</sup>, the log ( $\text{MO}^+/\text{M}^+$ ) increases markedly with  $D_0(\text{MO}^+)$  until ~5 eV, but only moderately increases after this point. At higher gas flow rates (0.90 to 1.00 L min<sup>-1</sup>), while a similar trend is observed among the weak-oxide formers, the log ( $\text{MO}^+/\text{M}^+$ ) of the strong-oxide formers increases significantly and is linearly correlated with  $D_0(\text{MO}^+)$ . The rate of increase in the oxide production rates is larger at higher sample gas flow rates. Moreover, at high sample gas flow rates, the linear correlations of log ( $\text{MO}^+/\text{M}^+$ ) against  $D_0(\text{MO}^+)$  (correlation coefficient of 0.96–0.99) are stronger than those of log ( $\text{MO}^+/\text{M}^+$ ) against  $D_0(\text{MO})$  (correlation coefficient of 0.81–0.87) (Figs. 2-2a and Appendix Fig. 2-1).

Table 2-2. Oxide production rates of Th and U for the NIST SRM 610 standard and the 16-F-6 monazite

	NIST SRM 610		Monazite 16-F-6	
Sample gas flow rate [L min <sup>-1</sup> ]	ThO <sup>+</sup> /Th <sup>+</sup> [%]	UO <sup>+</sup> /U <sup>+</sup> [%]	ThO <sup>+</sup> /Th <sup>+</sup> [%]	UO <sup>+</sup> /U <sup>+</sup> [%]
0.75	0.50 ± 0.08	0.44 ± 0.07	0.47 ± 0.04	0.41 ± 0.04
0.80	0.64 ± 0.15	0.44 ± 0.07	0.53 ± 0.03	0.43 ± 0.07
0.85	0.83 ± 0.10	0.44 ± 0.09	0.69 ± 0.03	0.43 ± 0.03
0.90	1.6 ± 0.3	0.63 ± 0.08	1.7 ± 0.1	0.59 ± 0.01
0.95	3.5 ± 0.4	1.1 ± 0.2	3.4 ± 0.1	1.0 ± 0.1
1.00	8.6 ± 1.4	2.5 ± 0.3	8.5 ± 0.8	2.4 ± 0.2

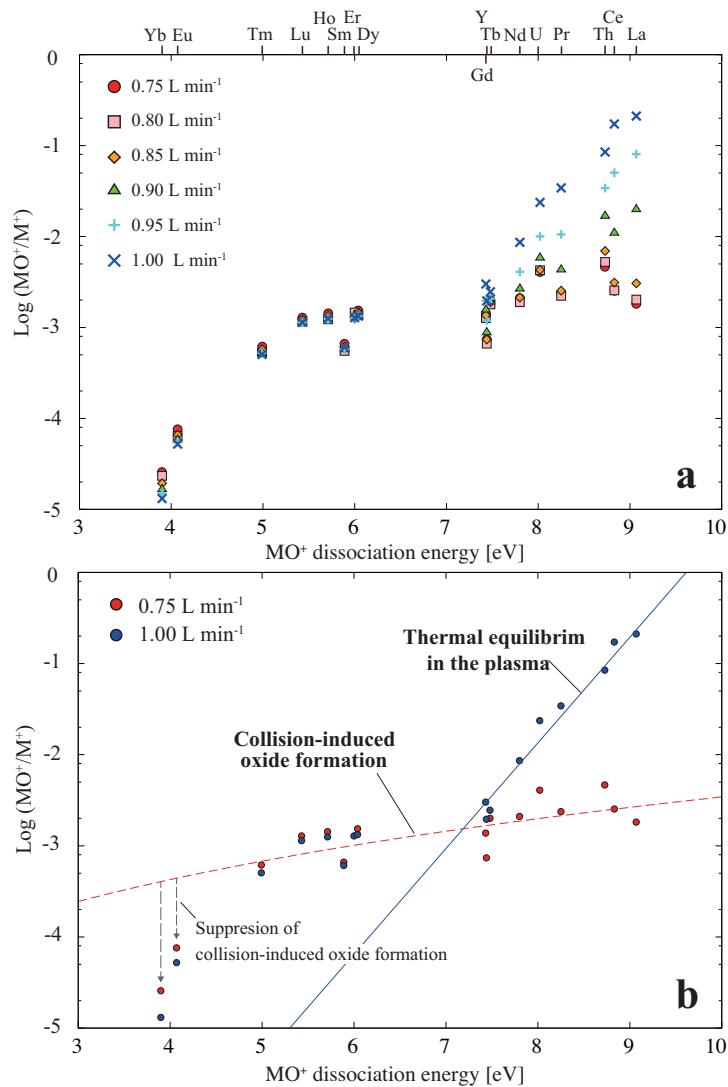


Fig. 2-2: (a) Plot of log (MO<sup>+</sup>/M<sup>+</sup>) of phosphates versus MO<sup>+</sup> dissociation energy for REE, Th and U with various sample gas flow rates. (b) Schematic diagram showing the effects of oxide production and dissociation processes on the MO<sup>+</sup>/M<sup>+</sup> ratio. The solid line represents the best fit with the strong-oxide former data at the gas flow rate of 1.00 L min<sup>-1</sup>, which is considered to reflect the equilibrium reaction of  $\text{MO}^+ \leftrightarrow \text{M}^+ + \text{O}$  within the plasma at 4300 K. The dashed curve represents the relationship between the MO<sup>+</sup>/M<sup>+</sup> ratio and MO<sup>+</sup> dissociation energy obtained from Eq. 2 with  $c\tau = 2 \times 10^{-6}$ , likely reflecting collision-induced oxide formation. The vertical arrows indicate the suppression of collision-induced oxide formation for Yb and Eu especially at a higher gas flow rate.

### 2.3.2 Mechanism of oxide production

The distinct behaviors of the  $\text{MO}^+/\text{M}^+$  ratio between the strong- and weak-oxide formers strongly suggest that the main oxide formation process is different between these two groups, especially at high sample gas flow rates. The strong-oxide formers exhibit a linear correlation between the  $\log(\text{MO}^+/\text{M}^+)$  and  $D_0(\text{MO}^+)$  for each sample gas flow rate ranging from 0.90 to 1.00  $\text{L min}^{-1}$ , in which the slope increases with the sample gas flow rate. As a higher sample gas flow rate results in a lower plasma temperature (Lindner and Bogaerts, 2011; Raut et al., 2003), the dependence of the slope of the linear correlation on the sample gas flow rate can be interpreted as reflecting the thermal equilibrium between  $\text{MO}^+$  and  $\text{M}^+ + \text{O}$  within the plasma. By fitting the strong-oxide former data to a Boltzmann distribution expressed as (Douglas and French, 1986; Longerich et al., 1987):

$$\frac{\text{MO}^+}{\text{M}^+} \propto e^{-D_0/kT} \quad (1)$$

where  $k$  is Boltzmann's constant and  $T$  is the temperature, we derived equilibrium temperatures of  $\sim 7900$ ,  $\sim 5000$ , and  $\sim 4300$  K at sample gas flow rates of 0.90, 0.95, and  $1.00 \text{ L min}^{-1}$ , respectively (Fig. 2-2b). These equilibrium temperatures are within the range of the plasma temperatures typically sampled in ICP-MS (Houk and Praphairaksit, 2001; Raut et al., 2003; Lindner and Bogaerts, 2011; Bogaerts and Aghaei, 2017), supporting the inference of the oxide production under equilibrium within the plasma.

By contrast, at sample gas flow rates of  $\leq 0.85 \text{ L min}^{-1}$ , the strong-oxide former data can be fitted by a line with a slope corresponding to a temperature of  $> 20000 \text{ K}$ , significantly higher than the maximum plasma temperature. This unreasonably high temperature, together with the independence of their  $\text{MO}^+/\text{M}^+$  ratio on the sample gas flow rate change, indicates that a process other than the thermal equilibrium is responsible for oxide production at the low gas flow rates. Note that fitting the data for both strong- and weak-oxide formers at a gas flow rate of  $\geq 0.90 \text{ L min}^{-1}$  also results in an apparent high temperature, which might lead some to argue that oxide production within the plasma is insignificant (Douglas and French, 1986; Longerich et al., 1987; Vaughan and Horlick, 1990).

Oxide production through collisions in the interface region has been suggested based on variable  $\text{MO}^+/\text{M}^+$  ratios depending on the geometry of the sampling and skimmer cones (Douglas and French, 1986; Longerich et al., 1987; Vaughan and Horlick, 1990; Poussel et al., 1994; Shibata et al., 1993). For a stable  $\text{MO}^+$  to be formed thorough collisions, the excess energy liberated by oxide bonding needs to be removed by either a collision to the third body or by radiative decay of  $\text{MO}^+$  from an excited state to the ground state (Niu and Houk, 1996; Newman et al., 2009). Given that three-body collision is trivial in the low-pressure interface region (Douglas and French, 1986), the latter mechanism is more likely responsible for the oxide production. Indeed, oxide formation involving the collision of  $\text{M}^+$  and O followed by radiative transition can account well for the isotope-dependent  $\text{MO}^+/\text{M}^+$  ratio and concomitant

mass-independent isotope fractionation observed for Nd and Ce (Newman et al., 2009).

In this case, the  $\text{MO}^+/\text{M}^+$  ratio can be expressed as Newman et al., 2009:

$$\ln\left(\frac{\text{MO}^+}{\text{M}^+}\right) = \ln(1 - e^{-cv^3\tau}) + cv^3\tau \quad (2)$$

where  $c$  is a proportional constant reflecting the transition probability,  $v$  is the transition frequency that is proportional to the sum of  $D_0(\text{MO}^+)$  and the potential energy of the metastable O (1.97 eV), and  $\tau$  is the life time of the collision. Remarkably, the relationship between  $\log(\text{MO}^+/\text{M}^+)$  and  $D_0(\text{MO}^+)$  calculated from this equation with  $c\tau = 2 \times 10^{-6}$  fits the data for both the strong-oxide formers at the gas flow rates of  $\leq 0.85 \text{ L min}^{-1}$  and for the weak-oxide formers, except Yb and Eu (Fig. 2-2b). Although the calculation of  $\log(\text{MO}^+/\text{M}^+)$  depends highly on the values chosen for  $c$  and  $\tau$ , the shape of the relation curve is insensitive to the choice. Thus, the observed moderate increase of  $\log(\text{MO}^+/\text{M}^+)$  with the  $D_0(\text{MO}^+)$  from 5 to 9 eV is well explained by oxide formation through collisions of  $\text{M}^+$  and O followed by radiative energy loss to stabilize the  $\text{MO}^+$  ions in the interface region.

The data for Yb and Eu, that have the lowest  $D_0(\text{MO}^+)$  among the studied elements, deviate from the expected trend for the oxide production in the interface region, even though their  $\text{MO}^+/\text{M}^+$  ratios are still higher than the values expected for the equilibrium reactions of  $\text{MO}^+ \leftrightarrow \text{M}^+ + \text{O}$  within the plasma (Fig. 2-2b). Moreover, the measured  $\text{YbO}^+/\text{Yb}^+$  and  $\text{EuO}^+/\text{Eu}^+$  ratios decreased with increased Ar sample gas flow rate.

These observations suggest that Ar suppressed the oxide formation in the interface region only for elements having  $D_0(\text{MO}^+) \leq 4$  eV. Such energy-threshold behavior may imply that the suppression results from the dissociation of weakly-bound oxides through collisions with Ar species in the interface region.

Overall, the relationship of the  $\text{MO}^+/\text{M}^+$  ratio with the  $D_0(\text{MO}^+)$  and the sample gas flow rate can be attributed to a combination of the following two oxide formation processes (Fig. 2-2b): (i) the equilibrium reactions of  $\text{MO}^+ \leftrightarrow \text{M}^+ + \text{O}$  within the plasma and (ii) bonding of  $\text{M}^+$  and  $\text{O}$  by collisions and subsequent radiative transition in the interface region. The first process was significant only for oxide production of the elements with  $D_0(\text{MO}^+) \geq 7$  eV, under low-temperature plasma conditions ( $\leq 8000$  K). Otherwise, the second process dominated the oxide production. Furthermore, the oxide formation *via* the second process was suppressed for elements having  $D_0(\text{MO}^+) \leq 4$  eV especially at a higher sample gas flow rate, possibly due to collision-induced dissociation in the interface region. The relative significance of the two oxide production processes may change with the cone geometry and oxygen availability that is controlled mainly by the methods of sample introduction. However, the finding that the  $\text{MO}^+/\text{M}^+$  ratio reflects a combined result of the two processes should persist regardless of various analytical conditions and be taken into account when applying a correction for oxide interferences.

### 2.3.3 Matrix effect

Our data for the 16-F-6 monazite and NIST SRM 610 glass (Table 2-2) indicate restricted effect of sample matrix on  $\text{UO}^+/\text{U}^+$  and  $\text{ThO}^+/\text{Th}^+$  ratios. This differs from the results of Gilbert et al. showing that  $\text{ThO}^+/\text{Th}^+$  ratio of monazite can be down to half of that of NIST SRM 610 glass, while  $\text{UO}^+/\text{U}^+$  ratio was more consistent between the two samples. In detail, the matrix effect was more significant under laser ablation conditions where larger aerosols were generated *i.e.* longer laser wavelength, lower laser fluence, and more opaque samples (Hanselman et al., 1994; Kuhn and Günter, 2004). Because ionization energies of oxides of strong-oxide formers, including Th and U, are generally lower than their  $D_0(\text{MO})$  and  $D_0(\text{MO}^+)$ , atomization and ionization of the oxides within the plasma should essentially proceed *via* reactions of (i)  $\text{MO} \leftrightarrow \text{MO}^+ + e^-$  and (ii)  $\text{MO}^+ \leftrightarrow \text{M}^+ + \text{O}$ . The introduction of large aerosols into the plasma would prevent efficient ionization and atomization of the oxides, leading to the matrix dependency of  $\text{MO}^+/\text{M}^+$  for strong-oxide formers. The restricted matrix effect in this study would be attributed to a higher laser fluence ( $20 \text{ J cm}^{-2}$ ) compared to the previous study ( $3.5\text{--}11 \text{ J cm}^{-2}$ ) (Gilbert et al., 2017), which resulted in the generation of fine aerosols and allowed the reactions of  $\text{MO}^+ \leftrightarrow \text{M}^+ + \text{O}$  to reach equilibrium within the plasma. This inference is consistent with the good linear correlation between the  $\log (\text{MO}^+/\text{M}^+)$  and  $D_0(\text{MO}^+)$  among the strong-oxide formers (Fig. 2-2a).

### 2.3.4 Impact of oxide interference

The results of the REE analyses of the igneous and metamorphic monazite from the Japan island arc and Yilgarn Craton are presented in Appendix Table 2-2 and the chondrite-normalized REE patterns are shown in Fig. 2-3. The monitored  $\text{ThO}^+/\text{Th}^+$  and  $\text{UO}^+/\text{U}^+$  were  $\sim 0.6\%$  and  $\sim 0.5\%$ , respectively and nearly constant during each analytical session. These oxide production rates are comparable to those during LA-ICP-MS REE analyses in previous studies (Berglund and Wieser, 2011; Janots et al., 2008). The measured  $^{139}\text{La}/^{173}(\text{Yb} + \text{GdO})$  and  $^{139}\text{La}/^{175}(\text{Lu} + \text{TbO})$  ratios range from  $\sim 2100$  to  $\sim 26000$  and from  $\sim 4300$  to  $\sim 46000$ , respectively.

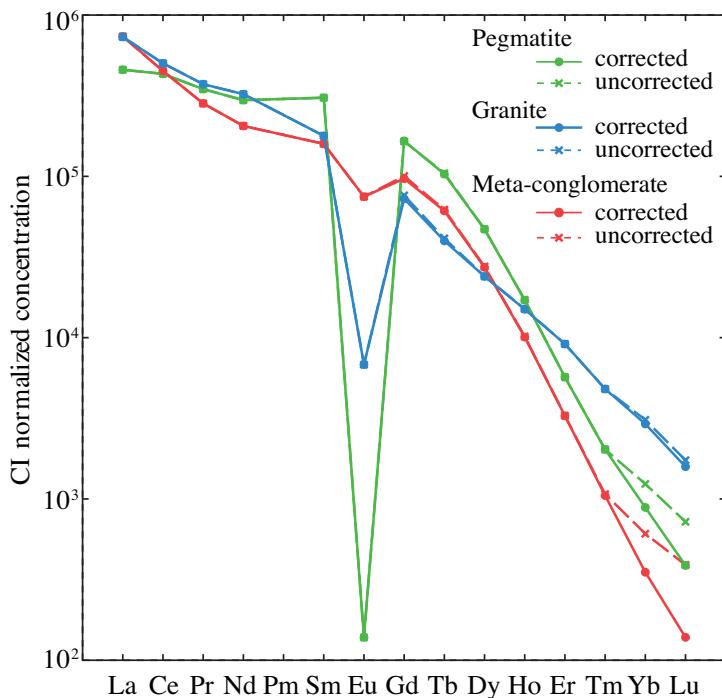


Fig. 2-3: Chondrite-normalized REE plots for monazites in pegmatite, granite, and meta-conglomerate. The solid and broken lines represent the results with and without the oxide interference correction, respectively.

The REEO<sup>+</sup>/REE<sup>+</sup> ratio during the analyses can be estimated from the measurements of the synthetic REE phosphates within the same analytical session, because the ThO<sup>+</sup>/Th<sup>+</sup> and UO<sup>+</sup>/U<sup>+</sup> ratios monitored for the analyzed natural monazites indicate insignificant changes of the MO<sup>+</sup>/M<sup>+</sup> ratio during the analytical sessions. Moreover, because of the possible matrix dependency for oxide production rates by LA-ICP-MS (Gilbert et al., 2017), the synthetic phosphates, rather than synthetic glass or silicate, provide more robust estimation of the oxide production rate. Applying the REEO<sup>+</sup>/REE<sup>+</sup> ratios measured for the synthetic phosphates, we have calculated the signal intensities of MO<sup>+</sup> of LREE and MREE that interfere with those of MREE and HREE ions (Fig. 2-4) and compared the REE concentrations obtained with and without the interference correction (Fig. 2-3 and Appendix Table S2-2). The results demonstrate that oxide interferences are most critical for the two heaviest REE, Yb and Lu (Fig. 2-4). The estimated  $^{173}\text{GdO}/^{173}(\text{Yb} + \text{GdO})$  and  $^{175}\text{TbO}/^{175}(\text{Lu} + \text{TbO})$  range from 6% to 42% and from 9% to 64%, respectively. The highest  $^{173}\text{GdO}/^{173}(\text{Yb} + \text{GdO})$  and  $^{175}\text{TbO}/^{175}(\text{Lu} + \text{TbO})$  ratios were obtained for the monazite from the meta-conglomerate that shows a REE pattern characterized by significant HREE depletion (Fig. 2-3). The depletion of HREE is attributed to their strong partitioning into co-existing garnet (Hauri et al., 1994; Johnson, 1998). Importantly, monazite commonly occurs with garnet in high-grade metamorphic rocks, particularly in peraluminous metasedimentary rocks (Rubatto et al., 2006; Mottram et al., 2014; Holder et al., 2015; Stipska et al., 2015). The REE pattern of the monazite from the pegmatite also exhibits

significant H-REE depletion, together with a highly negative Eu anomaly (Fig. 2-3). The REE characteristics likely result from significant fractional crystallization of feldspars and zircon/xenotime that strongly partition Eu and HREE, respectively. Note that although Eu ions are interfered by Ba oxide ions, the oxide interferences should be negligible because of the low dissociation energy of  $\text{BaO}^+$  (~4.57 eV) (Fig. 2-2). Specifically,  $\text{BaO}^+/\text{BaO}^+$  is estimated from Eq.2 to be at most 0.03%, which results in  $^{151}\text{BaO}/^{151}(\text{Eu} + \text{BaO})$  of <3% for the pegmatite monazite even if the monazite contains 1wt% Ba. These results demonstrate that the oxide interference correction is required for accurate LA-ICP-MS determination of HREE in monazites from high-grade metamorphic rocks and pegmatites, unless the analytical setting is optimized to reduce  $\text{ThO}^+/\text{Th}^+$  and  $\text{UO}^+/\text{U}^+$  ratios to < 0.1%. Thus, we propose that the oxide interference correction should be made for accurate HREE determination of monazite using synthetic MREE phosphates.

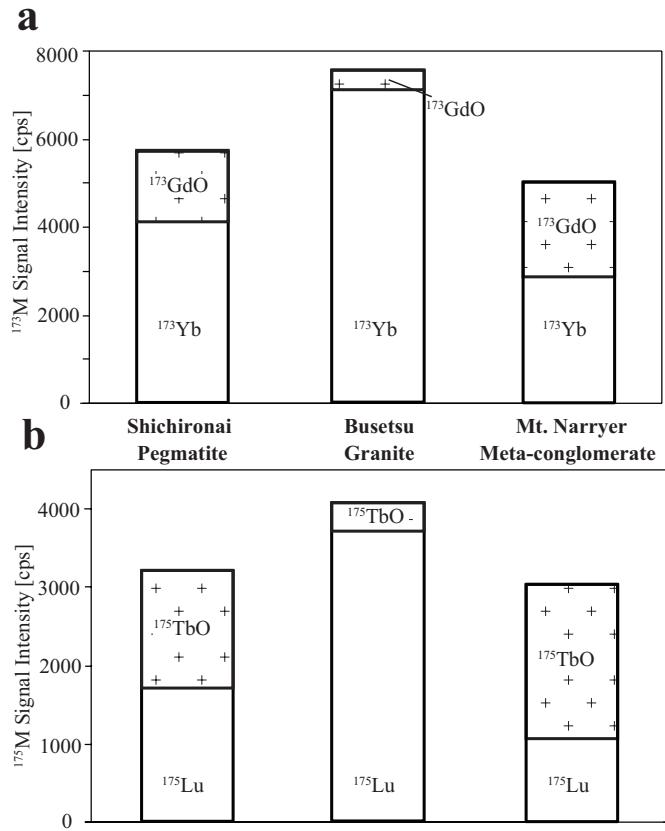


Fig. 2-4: Comparison of signal intensities between (a)  $^{173}\text{Yb}$  and  $^{173}\text{GdO}$ , and between (b)  $^{175}\text{Lu}$  and  $^{175}\text{TbO}$  for natural monazites in pegmatite, granite, and meta-conglomerate.

## 2.4 Conclusion

Our comprehensive analysis of REE-Th-U phosphates revealed systematic changes of the  $\text{MO}^+/\text{M}^+$  ratio with  $D_0(\text{MO}^+)$  and sample gas flow rate, which in turn unraveled the two oxide formation processes in ICP-MS: (i) the equilibrium reaction of  $\text{MO}^+ \leftrightarrow \text{M}^+ + \text{O}$  within the plasma and (ii) bonding of  $\text{M}^+$  and  $\text{O}$  through collisions and following radiative decay in the interface region. The oxide production *via* the equilibrium reaction in the plasma was significant only when the elements had a  $D_0(\text{MO}^+)$  higher than 7 eV and the plasma temperature was lower than ~8000 K. Otherwise, the collisional oxide production in the interface region was dominant. In addition, the collisional oxide production was suppressed for the elements with  $D_0(\text{MO}^+)$  lower than 4 eV especially at a higher Ar sample gas flow. This study further highlighted the impacts of MREE oxide interferences on accurate HREE determination of monazites from pegmatites and garnet-bearing metamorphic rocks, which can be corrected by using synthetic MREE phosphates.

### **3. Granitic magma evolution deciphered by trace element and Nd isotope microanalysis of monazite**

#### **3.1 Introduction**

Granitic magma plays an important role in differentiation of continental crust. It is widely accepted that the generation of felsic melts, which is not in equilibrium with the ultramafic mantle mineral assemblage, requires partial melting of crustal rocks as well as differentiation through fractional crystallization (e.g., Clemens, 1990; Patino-Douce and Beard, 1995; Johannes and Holtz, 1996; Montel and Vielzeuf, 1997; Clemens and Watkins, 2001). Experimental study can provide direct constraints on the partial melting in closed systems, which has revealed that the composition of granitic melt is dominantly controlled by the composition of source rock (e.g., Gao et al., 2016). In contrast, the evolution of granitic magma in natural systems involves open-system processes: hybridization of magmas derived from multiple sources, fractional crystallization, and assimilation of country rocks at emplacement levels. Indeed, granitic plutons commonly display geochemical and radiogenic isotopic zonings (Bateman et al., 1979; Barbey et al., 2001; Antunes et al., 2008). Consequently, it is fundamental for understanding granitic magma evolution to identify multiple sources and track subsequent processes to its solidification such as differentiation and assimilation.

Monazite is a well-suited tracer for granitic magma evolution. The Nd isotope composition of monazite can be determined by laser ablation-multiple

collector-inductively coupled plasma mass spectrometry (LA-MC-ICP-MS) (McFarlane and McCulloch, 2007; Gregory et al., 2009; Wu et al., 2010; Iizuka et al., 2011b). Fisher et al. (2017) demonstrated that the inputs of magma from isotopically variable sources are recorded in the heterogeneity of Nd isotope composition within single monazite grains. Moreover, monazite has a potential to track the differentiation of granitic magma through the change of light-, middle-, and heavy-REE (LREE, MREE, HREE) concentrations with the differentiation (Fisher et al., 2017). Most of studies on igneous monazite, however, were conducted by the EPMA technique which has a difficulty in precise and accurate determination of trace element concentration including Eu and HREE (e.g., Suzuki et al., 1990, 1992; Harlov et al., 2008). As a result, trace element systematics of monazite has not been quantitatively explored so far.

The Rb–Sr isotope system is complementary to the Sm–Nd isotope system and a strong tool for the petrogenesis of granitic rocks (e.g., Kagami, 1973). The Sr isotope microanalysis of plagioclase has been widely performed to investigate the heterogeneity in isotope compositions at the mineral or sub-mineral scales (Davidson et al., 1990, 2007, 2008; Gagnevin et al., 2005; Charlier et al., 2006). While the extraction of small sample using mechanical devices such as Micromill™ manufactured by New Wave are suitable to obtain high precision isotopic data, the *in situ* laser ablation technique is capable of enough precise analysis to detect the change in  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios (Christensen et al., 1995; Davidson et al., 2001; Ramos et al., 2004; Tatsumi et al., 2008). Plagioclase also plays an important role in Eu behavior during magma differentiation because

plagioclase as well as K-feldspar is the largest Eu reservoir, accounting for > 80 wt% of total Eu in granitic rocks (Bea, 1996).

In this study, the trace element systematics of monazite in granitic rocks from a single pluton was investigated to assess the utility of monazite as a tracer for granitic magma evolution. Furthermore, I evaluated the utility of the combined monazite Nd and plagioclase Sr isotope analyses to constrain magma sources, which were conventionally investigated by whole-rock isotope analysis. The Busetsu granite is suited for this objective because the geochemical and isotopic heterogeneities within the Busetsu granite were suggested based on whole-rock chemistry but what caused the heterogeneity remains ambiguous (Ishihara and Chappell, 2017).

### **3.2 The Busetsu granite in the Mikawa area, Japan**

The Busetsu pluton is located in Mikwa area, Japan (Fig. 3-1) and a part of the Ryoke granitic rocks. The voluminous Ryoke granitic rocks are associated with low-P/high-T metamorphic rocks including metachert, metasandstone and metamudstone, extending for ~800 km along the Inner Zone of SW Japan (Asami et al., 1982; Miyake et al., 1992, 2014, 2016, 2017; Makimoto et al., 2004; Adachi and Wallis, 2008; Miyazaki, 2010). The Busetsu granite is known as the youngest pluton in this area based on the intrusive relationship and K-Ar biotite ages (Ryoke Research Group, 1972; Yamazaki, 2013). It is noteworthy that the intrusion of the Busetsu granite was obviously subsequent to metamorphic peak of ca. 97–89 Ma (Takatsuka et al., 2018).

The Busetsu granite is characterized by S-type granite affinities, whereas other Ryoke granites in this area are classified into I-type granites (Ishihara and Chappell, 2007). Early isotopic studies on the Rb–Sr isotope system reported different isochron ages of ~71.7 and 82.5 Ma, resulting from different and scattered datasets (Shibata and Ishihara, 1979; Nakai et al., 1982). Besides, the  $\delta^{18}\text{O}$  values of Busetsu granite ranges from 10.5 to 13.5, which are higher than most of granitic rocks in the Japan arc (Ishihara and Matsuhisa, 2002; Ishihara and Chappell, 2007). These geochemical and isotopic signatures of the Busetsu granite indicate the contributions of sedimentary rocks to its petrogenesis (Ishihara and Chappell, 2007). However, the details of what processes caused the isotopic heterogeneity and the melting of sedimentary component are still unknown.

Suzuki et al. (1992) investigated the geochemistry of accessory minerals including monazite. The intra-grain variation in LREE composition between monazites within biotite and interstitial space was interpreted to reflect the onset and cessation of crystallization of other accessory minerals. The CHIME dating of the monazite in the Busetsu granite yielded ages of ~79–75 Ma (Suzuki et al., 1994; Nakai and Suzuki, 2003), which are inconsistent with the results of a recent zircon study on the Busetsu pluton (Takatsuka et al., 2018). Zircons from the Busetsu granites yielded various ages that are divided into four groups of ~100 Ma, 95–84 Ma, ~80 Ma, and 71–69 Ma (Takatsuka et al., 2018). The youngest ages are interpreted as the crystallization age of the granites, whereas the older ages are considered as reflecting ages of inherited zircon.

The monazite CHIME ages do not correspond to neither magmatic nor inherited zircon ages. This age discrepancy remains enigmatic and requires a re-examination of monazite crystallization by LA-ICP-MS U–Pb isotopic dating.

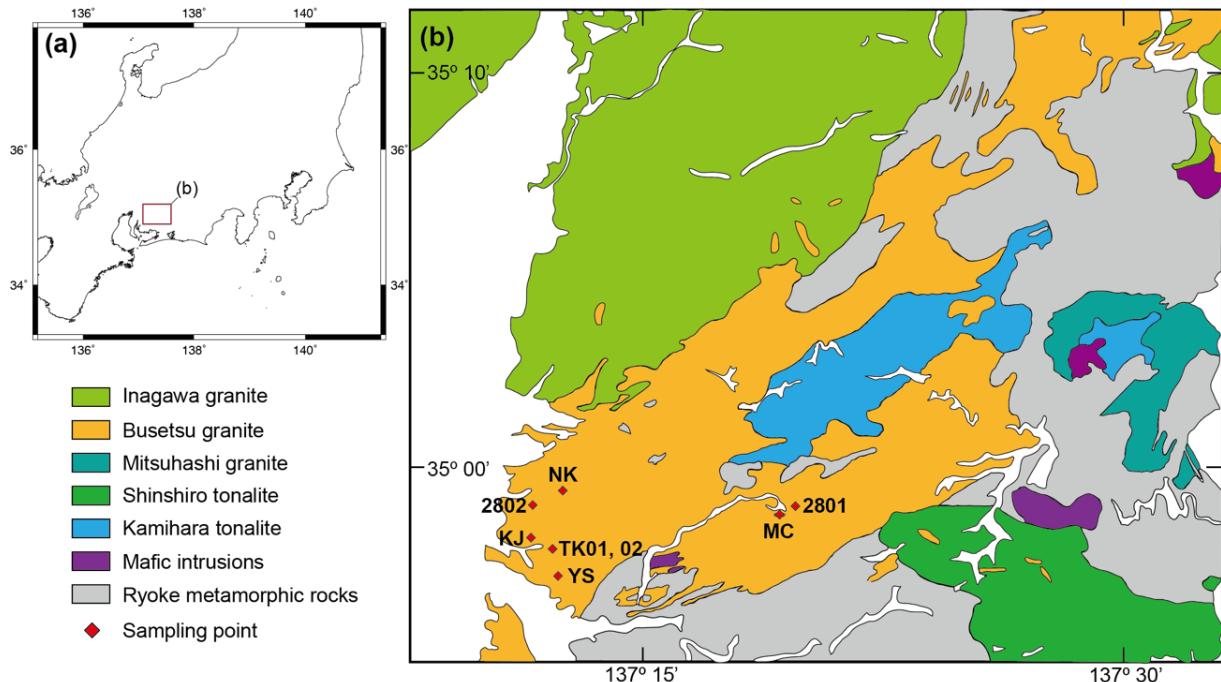


Fig. 3-1. (a) Location of the Busetsu granite pluton in the central area of Japan. (b) Geological map around the Busetsu granite associated with Ryoke metamorphic complex. Sample points of granitic rocks and a metamorphic rock (MC) are presented. The geological map is compiled and modified after Ishihara and Chappell (2017), Miyazaki (2010), and Takatsuka et al. (2018).

### 3.3 Analytical methods

The whole-rock major element analysis was performed with a PANalytical Axios X-ray fluorescence spectrometer (XRF) using a fused glass bead of silicate rock powder diluted with lithium-tetraborate at the University of Tokyo. After the major element analysis, the trace element compositions were determined for the glass beads using a iCAP Q quadrupole-ICP-MS (ICP-Q-MS) attached to a CETAC LSX-213 G2+ Nd:YAG LA system. Unknown samples were measured in runs of several samples, bracketed by three measurements of a standard NIST SRM 612 glass (Pearce et al., 1997). The ICP source operated with an Ar plasma with a forward RF power of 1550 W. The laser sampling was performed in a raster mode at  $\sim$ 50  $\mu\text{m}$  crater size, a repetition rate of 10 Hz, and  $20.54 \text{ J m}^{-1}$  fluence. The Mg contents, determined by XRF, were used for internal correction. Analytical uncertainty based on replicate analyses of USGS reference material BCR-2 was at most 20% for trace elements.

For whole-rock isotope analyses, the dissolved samples with a mixture of concentrated HF + HNO<sub>3</sub> in a Savillex® vial were placed in Teflon bombs and heated to 210 °C for 48 h. After digestion, the solution was evaporated to dryness. The samples were re-digested in HNO<sub>3</sub> and heated in the same way for 12 h and this process was repeated one more time. The sample was then evaporated and dissolved in 6M HCl at 210 °C using the bomb for 12 h. To separate and purify Sr and Nd from the decomposed whole rock sample solutions, two-step column chromatography was used following the modified recipe of Takahashi et al. (2009) and Hirahara et al. (2012), respectively. A

flow diagram of separation is illustrated in Fig. 3-2. The Sr and Nd isotope analyses were carried out using a Neptune plus multiple collector ICP-MS (MC-ICP-MS). The operating conditions of the instruments are listed in Tables 3-1, 2.

*In-situ* U–Pb dating, trace element, and Nd isotope analyses for monazite were done on separated monazite grains in addition to grains on thin sections to obtain a better dataset for describing the inter-grain variations. Before analysis, each grain was investigated by back-scattered electron (BSE) imaging to check internal zoning structures and to avoid sampling inclusions during laser ablation. The BSE images were obtained using a JEOL JSM-7000F. The U–Pb ages were measured using the LA-ICP-Q-MS. We measured  $^{202}\text{Hg}$ ,  $^{204}(\text{Pb} + \text{Hg})$ ,  $^{206}\text{Pb}$ ,  $^{207}\text{Pb}$ , and  $^{238}\text{U}$ , of which signal intensities fell within the maximum dynamic range of the counting system. The LA conditions were set at  $\sim 20\text{ }\mu\text{m}$  crater size, a repetition rate of 3 Hz, and a fluence of  $20.54\text{ J cm}^{-1}$ . The 44069 monazite (Aleinikoff et al., 2006) was used as a primary reference material for Pb/U ratios. The accuracy of our monazite U–Pb dating was evaluated by analyzing the monazite Manangotry standard, for which the U–Pb ages were previously determined by the isotope dilution-thermal ionization mass spectrometry ( $555 \pm 2\text{ Ma}$ , Horstwood et al., 2003). The obtained U–Pb ages for the Manangotry monazite are consistent with the previously reported values within the analytical uncertainty:  $^{206}\text{Pb}/^{238}\text{U}$  age of  $568 \pm 18\text{ Ma}$  and  $^{207}\text{Pb}/^{206}\text{Pb}$  age of  $571 \pm 23\text{ Ma}$  for Manangotry ( $n=22$ ). Isoplot R (Vermeesch, 2018) was used for visualization of the obtained age data. The concentrations of REEs, Th, and U in the studied monazite

were determined using the same LA-ICP-Q-MS as the U–Pb dating. The LA conditions were set at ~15 µm crater size and a repetition rate of 3 Hz. The Sm contents, measured in advance with a Field Emission-EPMA JEOL JXA 8530F HyperProbe, were used for internal correction. The NIST SRM 612 glass was used as an external standard. Interferences from LREE-oxides on HREE were corrected by determined oxide production rates using the synesthetic REE phosphates during each measurement session (Itano and Iizuka, 2017). The Nd isotopic composition were determined by LA-MC-ICP-MS. The operating parameters for this analysis are summarized in Table 3-3. Faraday amplifier calibration, correction for isobaric interference from  $^{144}\text{Sm}$  on  $^{144}\text{Nd}$ , and mass bias correction were carried out using a La–Ce–Nd synthetic monazite (Synth-LCN) standard and Nd–Sm–Pr synthetic monazite (Synth-NSP) standard (Iizuka et al., 2011a).

Major element compositions of plagioclase were analyzed by the EPMA. ZAF correction procedures were employed. The operating conditions of the EPMA analysis were: a beam diameter of 2 µm, a probe current of 20 nA, and an accelerating voltage of 15 kV. Natural materials were used as standards for seven elements (Na, K, Ca, Mg, Al, Si, Fe). The trace element and Sr isotope microanalyses of plagioclase were conducted by LA-MC-ICP-MS and LA-ICP-Q-MS. Analytical scheme for plagioclase trace element is almost same as that for monazite. The Ca contents were used as internal standard to determine the concentrations of Li, Mg, Sc, Ti, V, Mn, Zn, Rb, Sr, Y, Ba, La, Ce, Pr, Nd, Sm, Eu, Gd, and Pb. The LA conditions were set at ~30 µm crater size,

repetition rate at 5 Hz, and a fluence of  $20.54 \text{ J cm}^{-1}$  due to its lower concentration of REEs as compared to monazite. The Sr isotope analysis was performed using a solution-aerosol-dual intake system (Tatsumi et al., 2008; Kimura et al., 2013). Raw data were corrected for by a conventional protocol: blank subtraction, correction for Rb and Kr isobaric interferences, mass bias correction using  $^{86}\text{Sr}/^{88}\text{Sr}$ , and normalization to bracketing NIST SRM987 standard values. The operating parameters for this analysis are shown in Table 3-4.

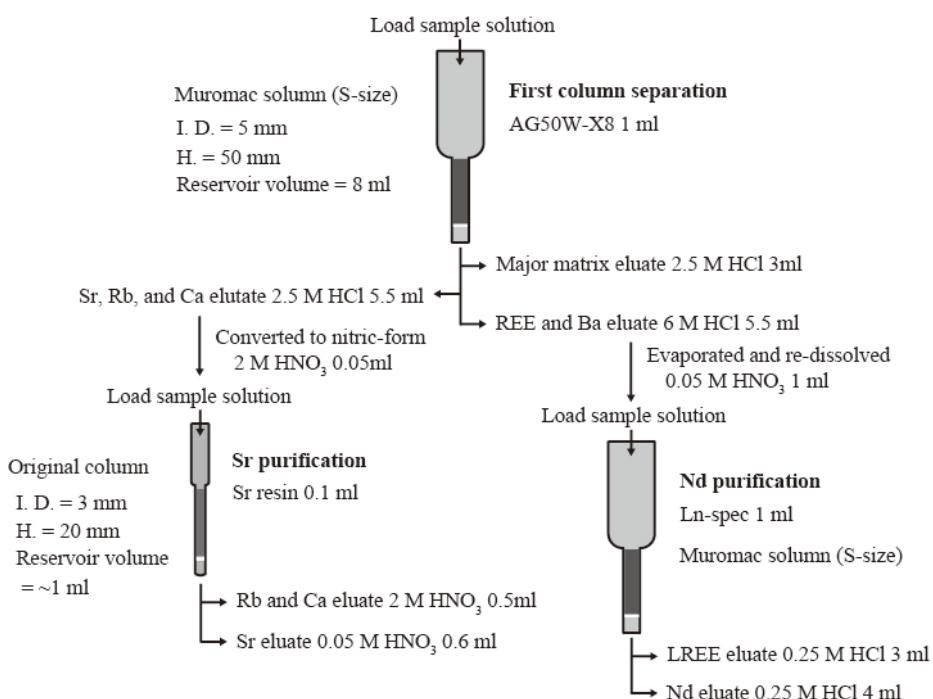


Fig. 3-2. Schematic diagram of Sr-Nd separation. First, major elements are removed using a conventional cation-exchange chromatography. After changing solvents, Sr and Nd are separated using a Sr resin and a Ln resin column, respectively.

**Table. 3-1** Operational parameters of the MC-ICP-MS for whole-rock Nd isotope analysis

<b>Sample introduction</b>	ARIDUS (CETAC, USA)
Sweep gas flow rates (Ar)	6-7 L min <sup>-1</sup>
N <sub>2</sub> Gas flow rates	5 L min <sup>-1</sup>
<b>MC-ICP-MS</b>	Neptune plus (Thermo Fisher Scientific)
RF-power	1200 W
Interface cones	Ni sample cone      Ni skimmer cone
Gas flow rates	Ar cooling: 14 L min <sup>-1</sup> Ar sample: 0.8–0.9 L min <sup>-1</sup>
Mass resolution	Low
Integration time	8.389 sec
Analysis mode	static
Collector arrangement	
Faraday cups	L3      L2      L1      Axial      H1      H2      H3
Analyte isotopes	<sup>142</sup> (Nd+Ce) <sup>143</sup> Nd <sup>144</sup> (Sm+Nd) <sup>145</sup> Nd <sup>146</sup> Nd <sup>147</sup> Sm <sup>149</sup> Sm

**Table. 3-2** Operational parameters of the MC-ICP-MS for whole-rock Sr isotope analysis

<b>Sample introduction</b>	ARIDUS (CETAC, USA)
Sweep gas flow rates (Ar)	5-5.5 L min <sup>-1</sup>
N <sub>2</sub> Gas flow rates	3–4 L min <sup>-1</sup>
<b>MC-ICP-MS</b>	Neptune plus (Thermo Fisher Scientific)
RF-power	1200 W
Interface cones	Ni sample cone      Ni skimmer cone
Gas flow rates	Ar cooling: 14 L min <sup>-1</sup> Ar sample: 0.9–1.1 L min <sup>-1</sup>
Mass resolution	Low
Integration time	8.389 sec
Analysis mode	static
Collector arrangement	
Faraday cups	L3      L2      L1      Axial      H1      H2      H3      H4
Analyte isotopes	<sup>82</sup> Kr <sup>83</sup> Kr <sup>84</sup> (Kr + Sr) <sup>85</sup> Rb <sup>87</sup> (Rb + Sr) <sup>88</sup> Sr <sup>89</sup> Y

**Table. 3-3** Operational parameters of the LA-MC-ICP-MS for Sm-Nd isotope analysis for monazite

<b>Laser ablation system</b>	CETAC LSX-213 G2+ (CETAC, USA) equipped with the active two-volume HeEx™ Cell												
Laser source	213 nm												
Pulse width	5 ns												
Beam diameter	30 $\mu\text{m}$												
Repetition rate	4 Hz												
Laser fluence	20.54 J $\text{cm}^{-2}$												
<b>MC-ICP-MS</b>	Neptune plus (Thermo Fisher Scientific)												
RF-power	1200 W												
Interface cones	Ni sample cone	Ni skimmer cone											
Gas flow rates	Ar cooling: 14 L $\text{min}^{-1}$	Ar auxiliary: 1.0 L $\text{min}^{-1}$											
	Ar sample: 1.2–1.3 L $\text{min}^{-1}$	$\text{N}_2$ carrier: ~4 mL $\text{min}^{-1}$		He carrier: 0.95 L $\text{min}^{-1}$									
Mass resolution	Low												
Data acquisition mode	Time resolved analysis												
Integration time	1.02 sec												
Analysis mode	static												
Collector arrangement													
Faraday cups	L3	L2	L1	Axial	H1	H2	H3						
Amplifier calibration factor*	0.0021	-0.00097	0.0021		-0.0013	-0.0011	0.0041						
Analyte isotopes	$^{142}\text{(Nd+Ce)}$	$^{143}\text{Nd}$	$^{144}\text{(Sm+Nd)}$	$^{145}\text{Nd}$	$^{146}\text{Nd}$	$^{147}\text{Sm}$	$^{149}\text{Sm}$						

\*The calibration factors were determined by analyzing the synthetic monazite Synth-LCN for Nd isotopes using LA-MC-ICPMS, following Iizuka et al. (2011a).

**Table. 3-4** Operational parameters of the dual intake system-MC-ICP-MS for Rb-Sr isotope analysis for plagioclase

<b>Laser ablation system</b>	CETAC LSX-213 G2+ (CETAC, USA) equipped with the active two-volume HeEx™ Cell												
Laser source	Pulse width												
Beam diameter	100 $\mu\text{m}$	Laser fluence											
Repetition rate	8 Hz	20.54 J $\text{cm}^{-2}$											
<b>Dissolvent system</b>	Aridus (CETAC, USA)												
Sweep gas flow rates (Ar)	6–7 L $\text{min}^{-1}$	$\text{N}_2$ Gas flow rates											
		5 L $\text{min}^{-1}$											
<b>MC-ICP-MS</b>	Neptune plus (Thermo Fisher Scientific)												
RF-power	1200 W												
Interface cones	Ni sample cone	Ni skimmer cone											
Gas flow rates	Ar cooling: 14 L $\text{min}^{-1}$	Ar auxiliary: 1.0 L $\text{min}^{-1}$											
	Ar sample: 1.0–1.1 L $\text{min}^{-1}$	He carrier: 1.2 L $\text{min}^{-1}$											
Mass resolution	Low												
Data acquisition mode	Time resolved analysis												
Integration time	1.02 sec												
Analysis mode	static												
Collector arrangement													
Faraday cups	L3	L2	L1	Axial	H1	H2	H3						
Analyte isotopes	$^{82}\text{Kr}$	$^{83}\text{Kr}$	$^{84}\text{(Kr + Sr)}$	$^{85}\text{Rb}$	$^{87}\text{(Rb + Sr)}$	$^{88}\text{Sr}$	$^{89}\text{Y}$						

### 3.4 Sample Descriptions

Seven granitic rock samples were collected at six locations in addition to a sampling of metapsamite in the Mikawa area (Fig. 3-1).

Samples TK01 and TK02 were collected at the same quarry (the Tokai quarry, 34°58'28.69"N, 137°11'55.16"E). In the outcrop, their clear boundaries were observed (Fig. 3-3a, b). Sample TK01 is a biotite granite and sample TK02 is a two-mica granite. Sample TK01 mainly consists of Pl, Qtz, Kfs, and Bt with accessory Ilm, Zrn, Ap, and Xen; whereas sample TK02 is composed of Pl, Qtz, Kfs, Bt, and Ms with accessory Ilm, Grt, Mnz, Zrn, Ap, and Xen (Mineral abbreviations are from Kretz, 1983). Note that monazite could not be identified within sample TK01, whereas the other studied granite samples contain monazite. Based on their mode compositions, sample TK01 and TK02 are classified into granodiorite and monzogranite, respectively (Fig., 3-3g). Metasomatic myrmekitic texture was rarely observed.

Sample NK was collected at the Nakane quarry (34°59'52.76"N, 137°12'17.35"E). Sample NK consists of Pl, Qtz, Kfs, Bt, and Ms with accessory Ilm, Grt, Mnz, Zrn, Ap, Xen, Aln, Py. This sample is classified into monzogranite (Fig., 3-3c, g) and coarsest among the studied samples (>1–2 mm). The co-existence of monazite and allanite indicates the significant change in the composition of granitic magma because relative phase relations of the two minerals depend on bulk CaO contents (Spear, 2010; Budzyń et al., 2017), though the allanite grains are anhedral with minute grain size (~10 µm) and much scarcer than monazite and xenotime.

Sample YS was collected at the Yamasen quarry ( $34^{\circ}58'28.69''N$ ,  $137^{\circ}11'55.16''E$ ).

Sample YS consists of Pl, Qtz, Kfs, and Bt with accessory ilmenite, Grt, Mnz, Zrn, Ap, Xen, and Aln. This sample is classified into granodiorite (Fig. 3-3d, g).

Sample 2802 was collected at the Nakayama quarry ( $34^{\circ}59'36.16''N$ ,  $137^{\circ}11'56.86''E$ ). Sample 2802 comprises Pl, Qtz, Kfs, Bt, and Ms with accessory Ilm, Grt, Mnz, Zrn, Ap, Aln, and Xen. This sample is classified into monzogranite (Fig. 3-3g).

Sample KJ was collected at the Kissho-seki quarry ( $34^{\circ}58'37.05''N$ ,  $137^{\circ}11'23.35''E$ ). The major minerals are Pl, Qtz, Kfs, Bt, and Ms, while the accessory phases are Ilm, Grt, Mnz, Zrn, Ap, and Xen. This sample is classified into granodiorite (Fig. 3-3g).

Sample 2801 was collected from a riverbed ( $34^{\circ}59'00.18''N$ ,  $137^{\circ}19'20.21''E$ ) and, therefore, is slightly weathered as compared to the other granite samples from the quarries. Sample 2801 consists of Pl, Qtz, Kfs, and Bt with accessory Ilm, Grt, Mnz, Zrn, Ap, and Xen. This sample is classified into granodiorite (Fig. 3-3g).

Sample MC [ $34^{\circ}58'58.92''N$ ,  $137^{\circ}18'44.12''E$ ] is a metapsamite, which was collected as an example of country rock (Fig.3-3f). The observed foliation structure is defined by alternating finer and coarser grain bands and preferred orientation of clay minerals.

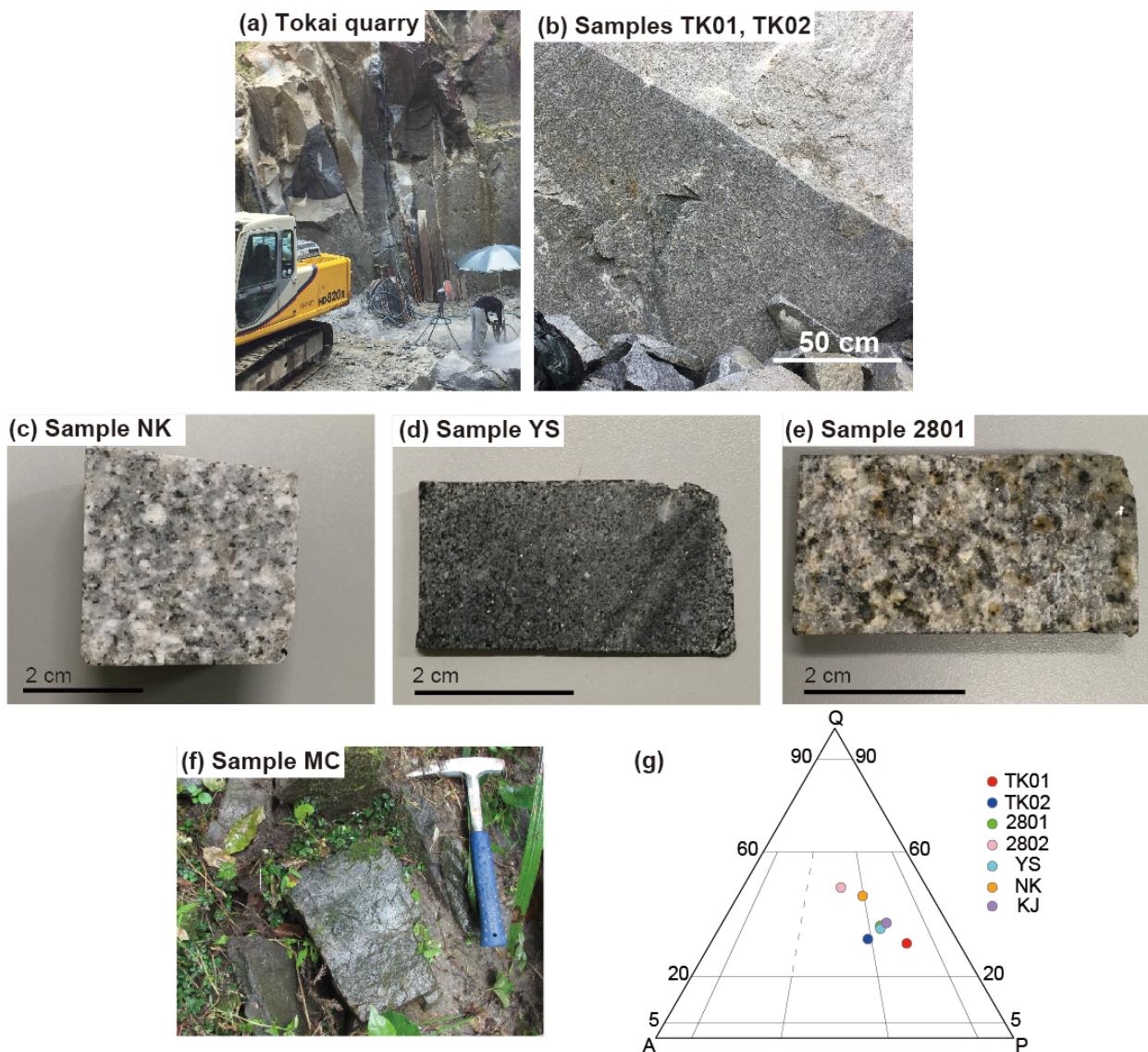


Fig. 3-3. (a) Outcrop in the Tokai quarry, showing a Bt-granite (granodiorite) enclosed by monzogranite (monzogranite). (b) Sharp boundary between monzogranite and granodiorite. (c) Slab photograph of two-mica granite: Sample NK. (d) Slab photograph of Bt-mica granite: Sample YS. (e) Slab photograph of Bt-mica granite: Sample 2801. This sample slightly appears weathered. (f) Field occurrence of sample MC. (g) QAP diagram showing modal abundances of quartz, plagioclase, and K-feldspar in the studied granitic samples. Samples 2802, NK, and TK02 are classified as monzogranites and the others are granodiorites.

### **3.5 Results**

#### **3.5.1 U–Pb age of monazite**

The U–Pb ages of 177 monazite grains of monazite from all samples other than TK01 were measured (54 grains for sample 2801; 42 grains for sample 2802; 28 grains for sample NK; 20 grains for sample YS; 18 grains for sample KJ; 15 grains for sample TK02) and summarized in appendix Table 3-1. Most of the data show discordant ages as shown in Fig. 3-4a. Incorporation of even a small amount of non-radiogenic Pb can affect the concordance in the case of young monazite. Assuming an initial  $^{206}\text{Pb}/^{204}\text{Pb}$  of 18.56 and  $^{207}\text{Pb}/^{204}\text{Pb}$  of 15.62 at 68 million years ago based on the two-stage evolution model (Stacey and Kramer, 1975), the highly discordant data are plotted along the line formed by the various  $\mu$  values ( $^{238}\text{U}/^{204}\text{Pb}$ ). Therefore, the initial Pb is likely to be a cause of the discordant data. The data yielding concordant U–Pb ages within analytical uncertainty are used for the following discussion. The filtered concordant data yielded  $68.4 \text{ Ma} \pm 3.7 \text{ Ma}$  (2SD, Fig. 3-4b, c).

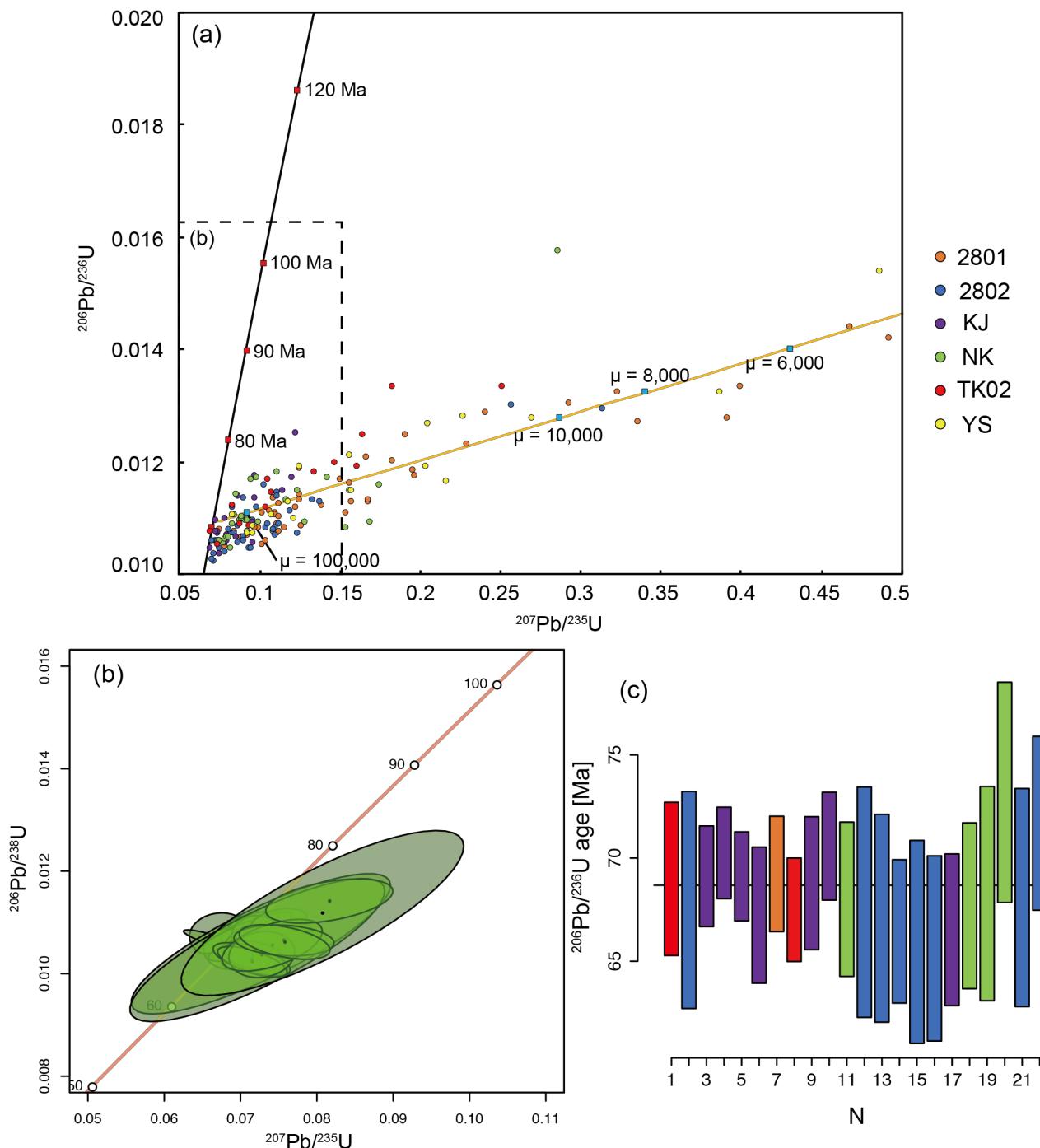


Fig. 3-4. (a) Concordia plot of all U-Pb data for monazite from studied samples. Yellow line represents the mixing between no-common Pb isotope composition yielding 68 Ma and Pb isotope composition of common Pb at 68 Ma, which is calculated using the two-stage evolution model (Stacey and Kramer, 1975).  $\mu$  is  $^{238}\text{U}/^{204}\text{Pb}$  ratio at the timing of crystallization. (b) Concordia plot for only concordant U-Pb data. Rejected data are not shown. (c) Plot of adopted  $^{206}\text{Pb}/^{238}\text{U}$  ages for the studied monazites. Color represents the samples as with the legend of (a).

### 3.5.2 Geochemical and Nd isotope compositions of monazite

Monazite grains from the studied samples are basically subhedral and included in biotite or plagioclase (Fig. 3-5). Some grains show homogeneous distribution of Th and other displays oscillatory zoning, suggesting crystallization from melts. The trace element data of 115 monazite grains are listed in Appendix Table 3-2. All monazite grains show typical REE patterns with abundant LREEs relative to HREEs and Eu negative anomalies. However, their slopes and the magnitude of the Eu anomaly are various, which are visualized in Figure 3-6. The variations are quantified by the chondrite-normalized values of  $[Eu/Eu^*]_N$ ,  $[Ce/Gd]_N$ , and  $[Th/U]_N$ . These parameters are correlated with each other (Fig. 3-6). The monazite grains from different rock samples yielded the different ranges of these parameters, which cannot be simply explained by the differences in rock types or whole-rock chemical compositions.

The analytical results of monazite Nd isotope compositions are provided in Appendix Table 3-3. The initial Nd isotope compositions of the studied monazites at 68 Ma are shown in Figure 3-7. The  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$  ratios range from  $0.51193 \pm 0.00003$  to  $0.51228 \pm 0.00003$ . The  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$  of analyzed grains are identical within analytical errors for samples 2801, 2802, NK, YS, and KJ. In contrast, a significant isotopic variation was observed for sample TK02, implying the involvement of isotopically different sources during the stage of monazite crystallization. No correlation was observed between the  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$  and trace element features such as  $[Ce/Gd]_N$ .

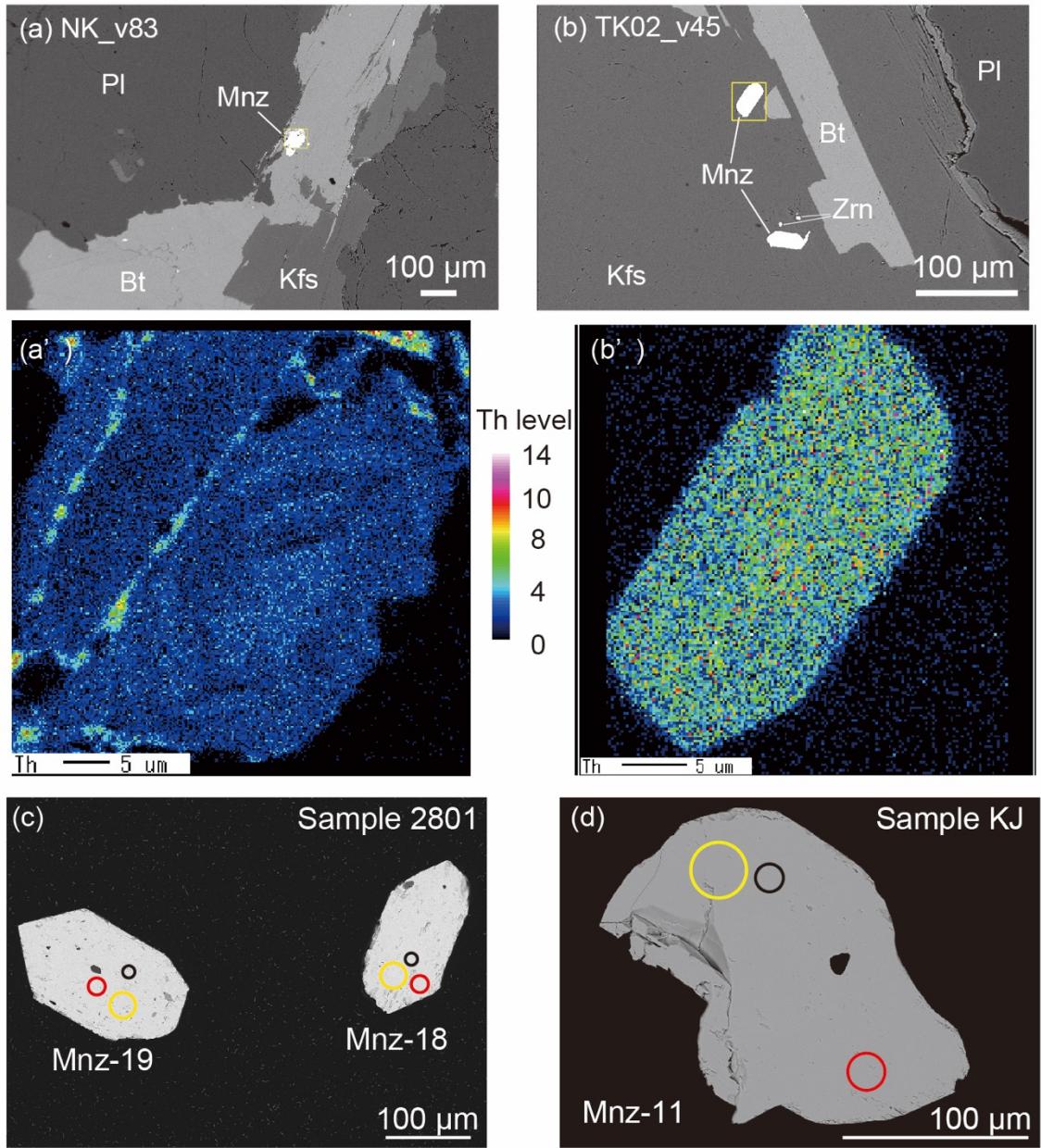


Fig. 3-5. BSE images and elemental mappings of representative monazite grains from Samples NK and TK02. (a) Monazite grain is included by biotite and (a') its Th distribution shows oscillatory zoning. (b) Euhedral monazite grains are next to biotite and included in K-feldspars and (b') these grains show homogenous Th distributions. (c) and (d) are representative BSE image of analyzed grains from samples 2801 and KJ, respectively. Yellow, red, and black circles represent analytical spots for Nd isotope composition, U-Pb age, and trace element composition, respectively.

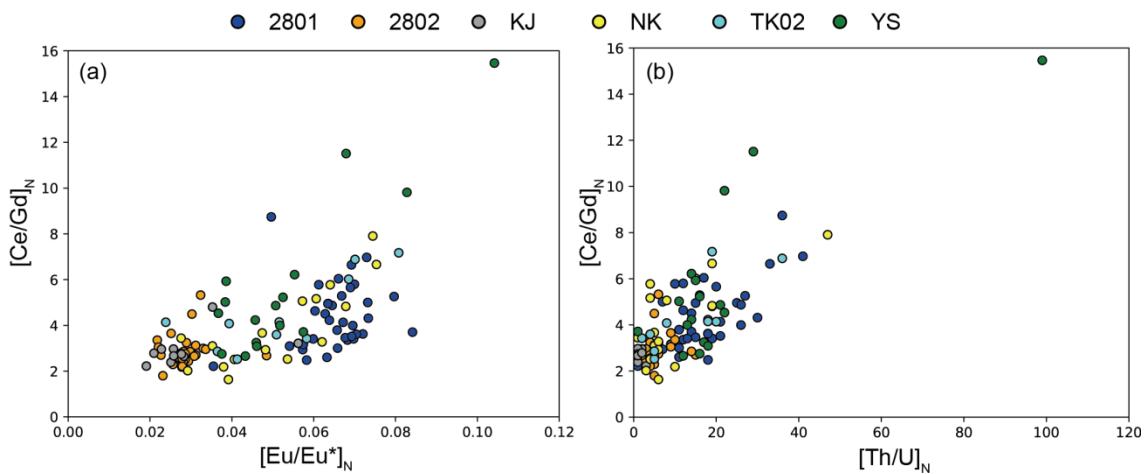


Fig. 3-6. Plots of (a) [Ce/Gd]<sub>N</sub> vs. [Eu/Eu\*]<sub>N</sub> and (b) [Ce/Gd]<sub>N</sub> vs. [Th/U]<sub>N</sub> for the measured monazite grains.

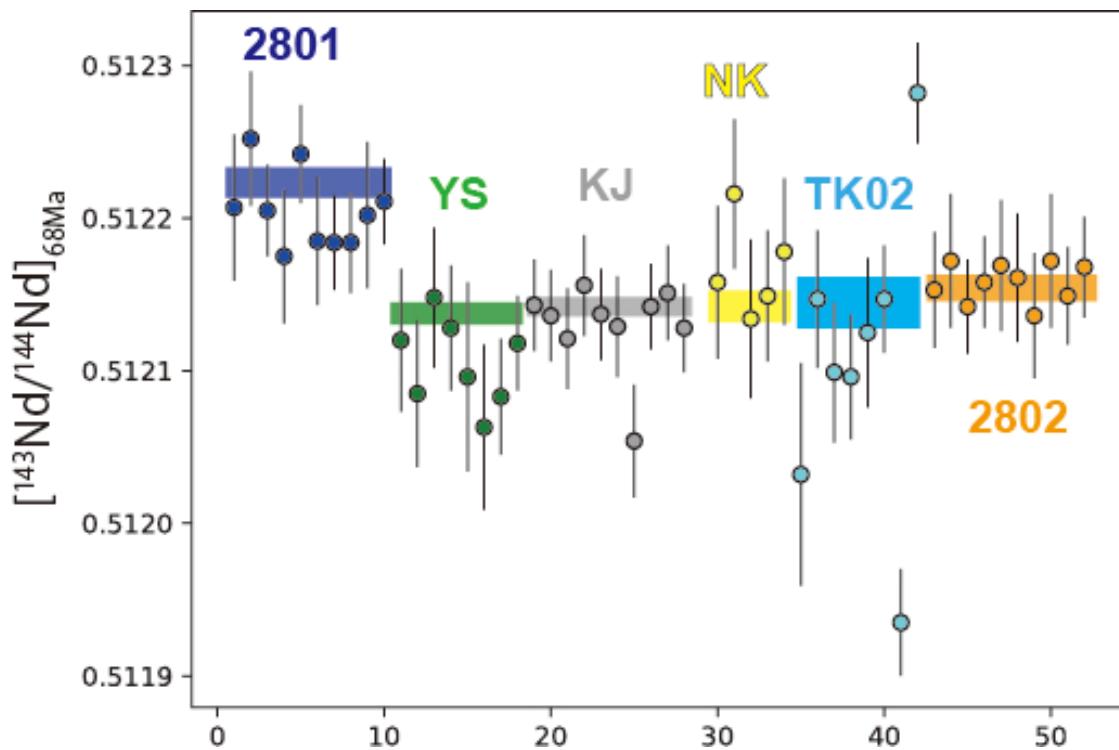


Fig. 3-7.  $^{143}\text{Nd}/^{144}\text{Nd}$  ratios at 68 million years ago of monazites. Bands overlapped plots represent the whole-rock Nd isotope ratios of each granite sample.

### 3.5.3 Geochemical and Sr isotope compositions of plagioclase

The plagioclase grains for samples NK, YS, TK01, TK02 were analyzed by EPMA, LA-ICP-Q-MS, and LA-MC-ICP-MS. Representative BSE images are presented in Figure 3-8. The analyzed grains have cores and homogeneous rims (Fig. 3-8b, c) or cores, oscillatory-zoned mantles, and rims (Fig. 3-8a, d). In this study, I defined all domains surrounding cores as rim domains. The measured trace element compositions and the averaged anorthite contents (An content: molar Ca / Ca + Na + K) within the measured domains for Sr isotope analysis are summarized in appendix Table 3-4. Figure 3-9 shows the examples of trace element data. The measured Sr contents of plagioclase are positively correlated with REEs, Ti, Mn, and anorthite contents. The correlation coefficient between Sr and Eu contents is ~0.76, which is the strongest correlation between trace elements. The measured  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios are plotted as a function of the An contents (Fig. 3-10). All of the isotope data are given in appendix Table 3-5. The  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios of rim domains are relatively homogenous for each rock sample; in contrast, those of core domains show large variations. The  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios range from  $0.7079 \pm 0.0002$  to  $0.7136 \pm 0.0002$ . The  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios of rim domains are deviated from the averaged values, irrespective of whether the rock sample is monzogranite or granodiorite. It is suggested that more than two isotopically different sources were involved for the formation of the Busetsu granites.

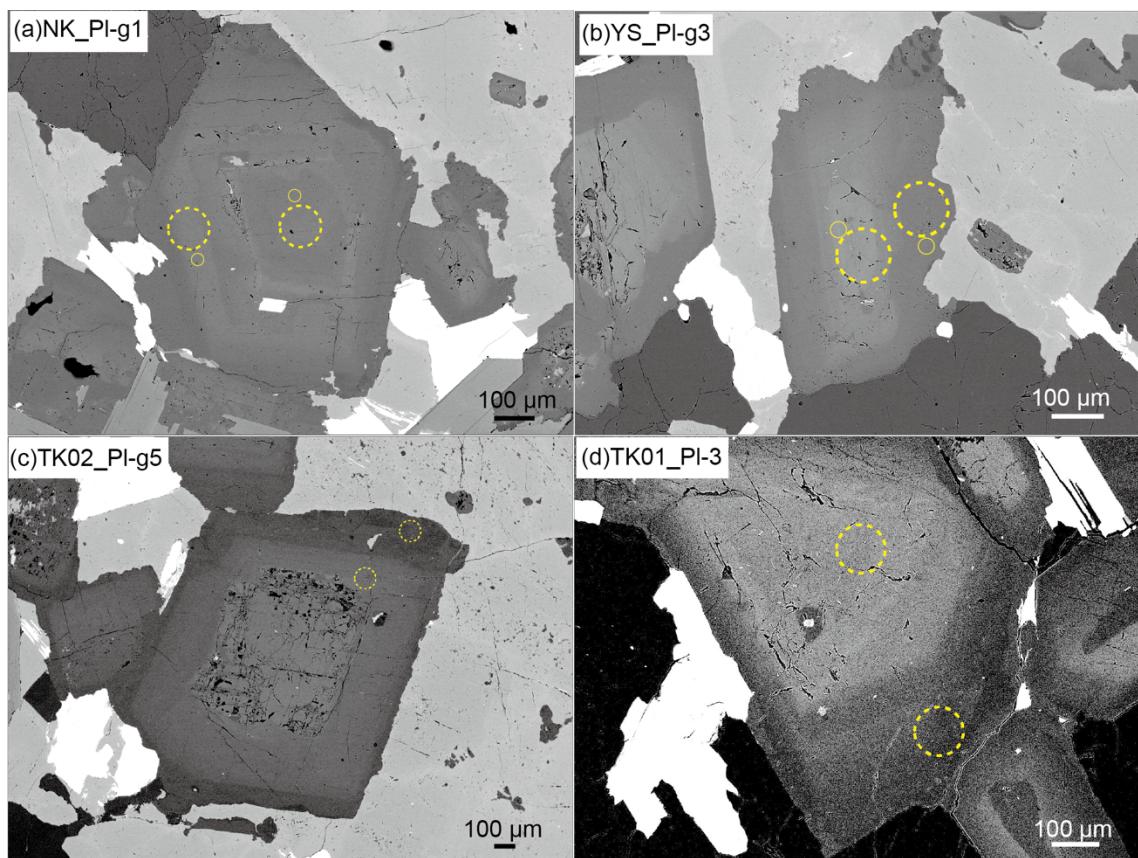


Fig. 3-8. Representative BSE images of plagioclase grains from samples (a)NK, (b) YS, (c) TK02, and (d) TK01. Solid- and dotted-circles represent analytical spots for REE and Sr isotope analyses, respectively.

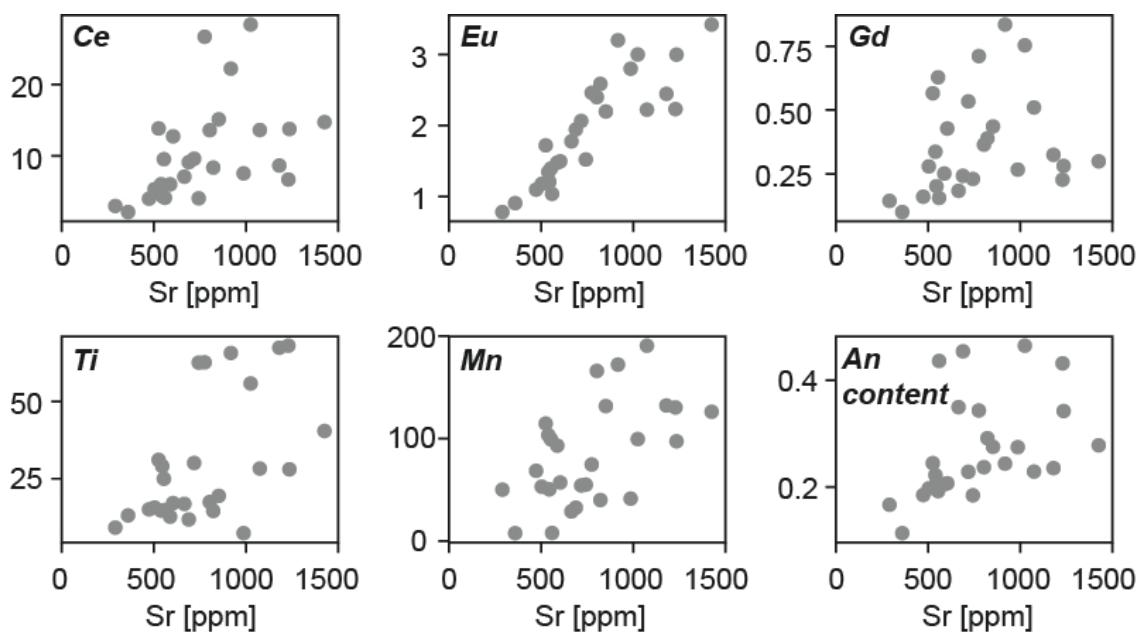


Fig. 3-9. Sr contents vs. Ce, Eu, Gd, Ti, Mn, and An contents of plagioclase grains.

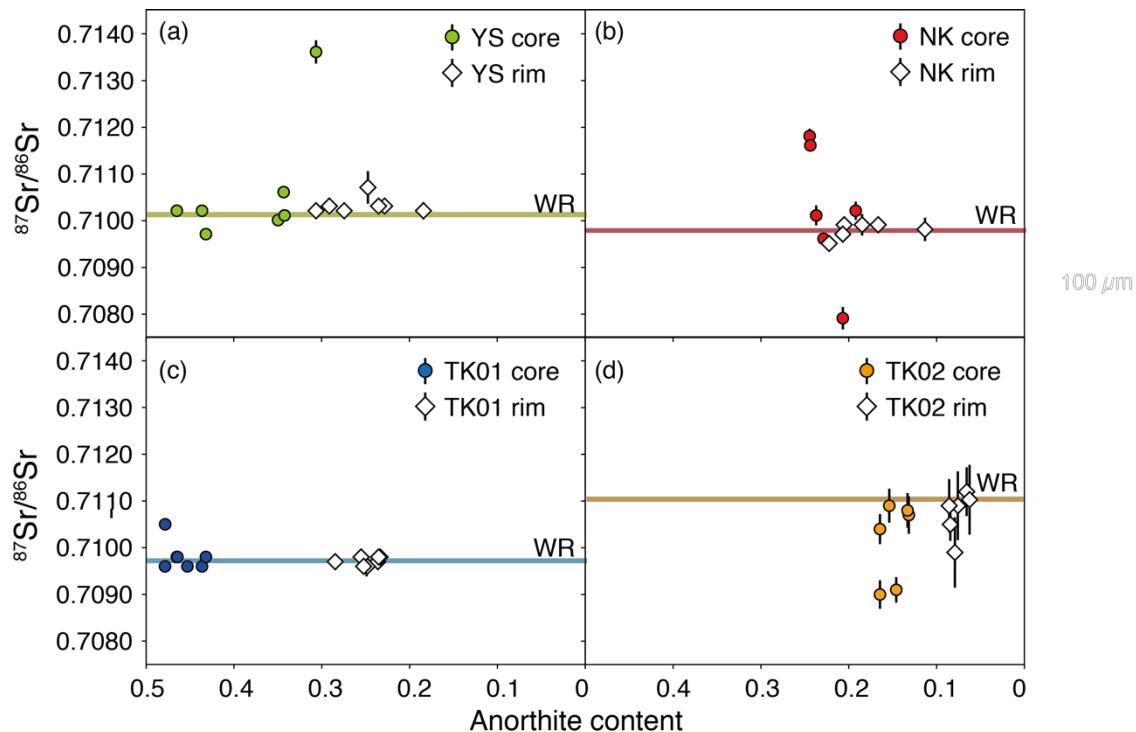


Fig. 3-10. Measured  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios of plagioclases for Samples (a) YS, (b) NK, (c) TK01, and (d) TK02. Solid lines represent the initial Sr isotope ratios of the whole-rock granite samples.

### 3.5.4 Whole-rock analyses

The correlations between major element contents were observed for the studied samples (Fig. 3-11, Appendix Table 3-6.), which is consistent with the reported data (Ishihara and Chappell, 2007). The SiO<sub>2</sub> content shows negative correlations with MgO, Fe<sub>2</sub>O<sub>3</sub>, and CaO. In contrast, the contents of SiO<sub>2</sub> and K<sub>2</sub>O display a weak positive correlation. For the Al<sub>2</sub>O<sub>3</sub>, P<sub>2</sub>O<sub>5</sub>, and Na<sub>2</sub>O contents, no obvious correlations with the SiO<sub>2</sub> contents were found. Regarding minor and trace elements, the spider diagram in Figure 3-12a is characterized by positive Pb and negative Ti anomalies. All of the granite samples have negative Eu anomalies (Fig. 3-12b). The REE contents decrease as the SiO<sub>2</sub> contents increase (Fig. 3-12c, d). Trace elements, which tend to concentrate into fewer minerals, are useful in formulating models for magmatic differentiation. When two trace elements are plotted in logarithmic space, the following relation is obtained:

$$\log(C^A) = \frac{D^A - 1}{D^B - 1} \log(C^B) + c$$

A, B: arbitrary elements, C<sup>X</sup>: the concentration of element, D<sup>X</sup>: bulk distribution coefficient, c: constant. The Sr and Ba contents show the strongest correlation and its linear approximation of the plots yielded (D<sub>Ba</sub>-1)/(D<sub>Sr</sub>-1) of ~1.03.

The whole-rock Sr-Nd isotopic compositions are presented in Tables 3-5 and -6. The [<sup>87</sup>Sr/<sup>86</sup>Sr]<sub>68Ma</sub> and [<sup>143</sup>Nd/<sup>144</sup>Nd]<sub>68Ma</sub> are also graphically shown in Fig. 13. The isotope compositions of the country rocks are also plotted for a reference. The Sr and Nd

isotopic compositions show a negative correlation (Fig. 3-13), which suggests the isotopically different multiple components were involved for the formation of the Busetsu granite. The  $[^{87}\text{Sr}/^{86}\text{Sr}]_{68\text{Ma}}$  values for the studied granitic samples range from  $0.70857 \pm 0.00003$  to  $0.71114 \pm 0.00005$  and the  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$  values range from  $0.51211 \pm 0.00001$  to  $0.51221 \pm 0.00001$ . Both the Sr and Nd isotope variations of whole rock samples are smaller than those recorded in plagioclase and monazite, respectively (Figs. 3-7, 10).

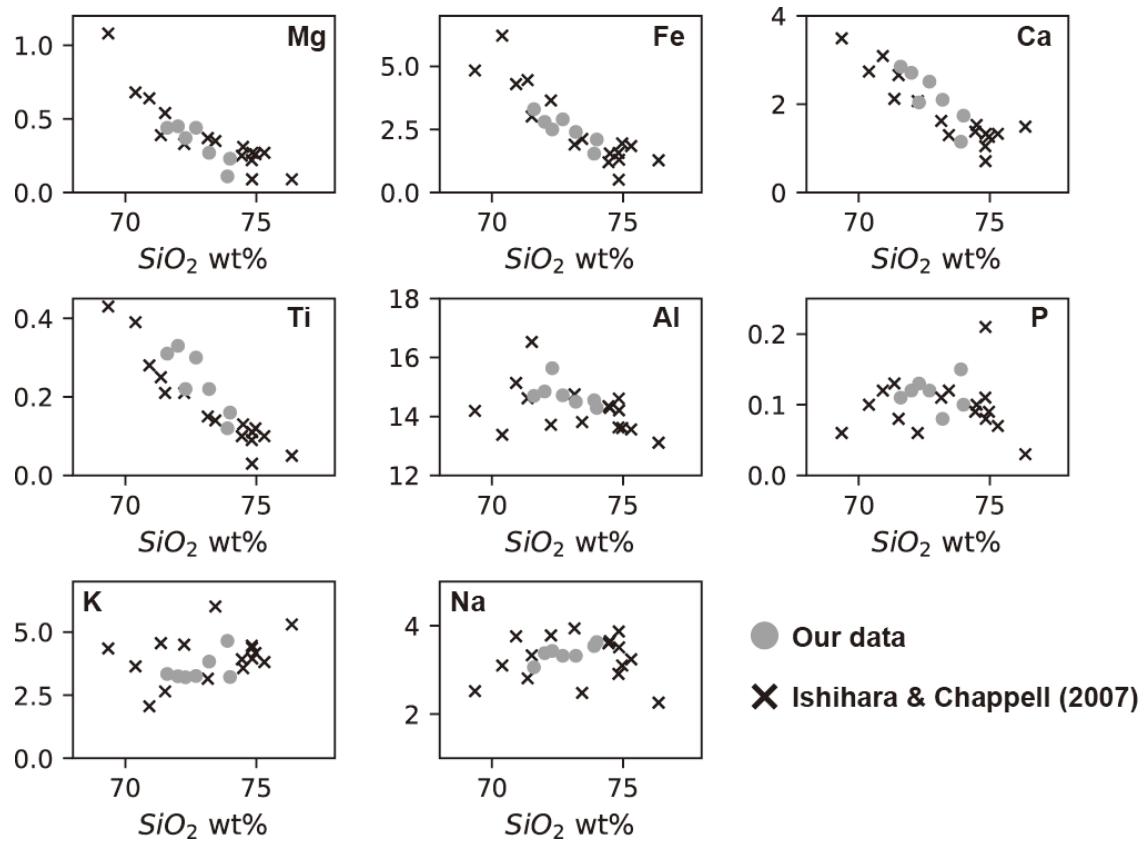


Fig. 3-11. Silica vs. Mg, Fe, Ca, Ti, Al, P, K, and Na of whole-rock granite samples. Circle and cross represent the obtained data in this study and the data of Ishihara and Chappell (2007), respectively. SiO<sub>2</sub> shows negative correlations with Mg, Fe, Ca, and Ti.

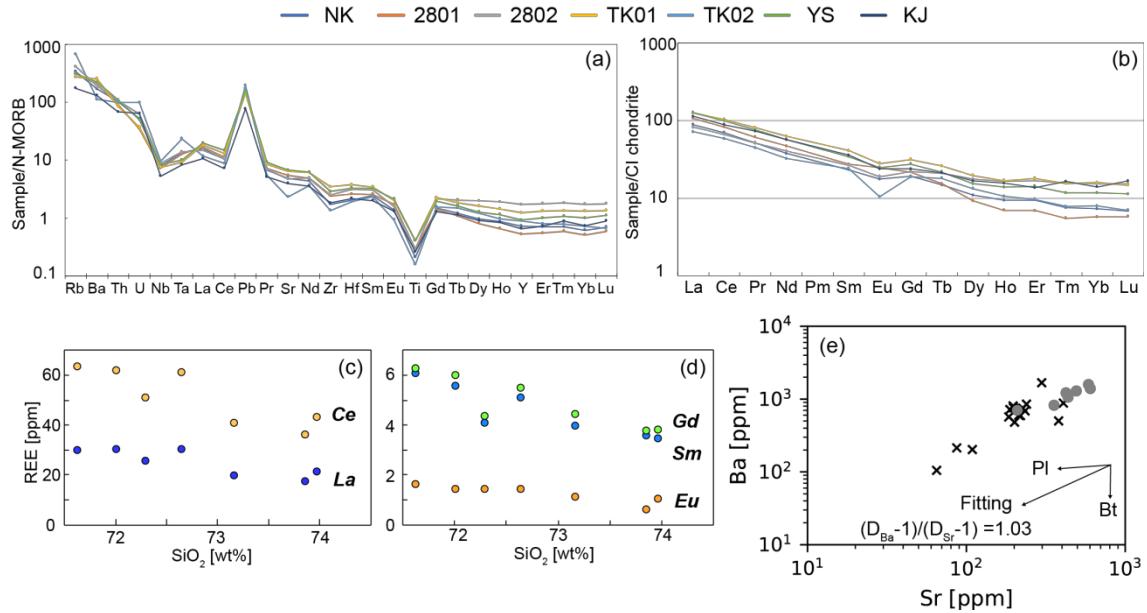


Fig. 3-12. Results of whole-rock trace element analyses of granite samples. (a) N-MORB normalized trace element patterns of the studied granites. (b) Chondrite normalized REE patterns of the granite samples. (c) and (d): SiO<sub>2</sub> vs La and Ce and SiO<sub>2</sub> vs. Sm, Eu, and Gd. (e) Plot of Sr and Ba contents. Symbols are the same as in Fig. 3-11.

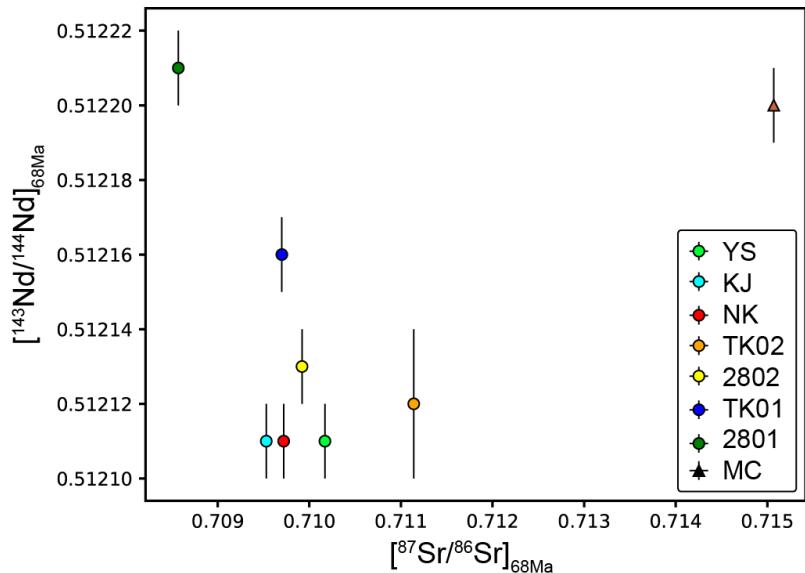


Fig. 3-13. Plots of initial Sr and Nd isotope compositions of bulk granitic and metasedimentary rocks.

Table 3-5. Sr isotopic data of Bussetsu granite samples

	$^{84}\text{Sr}/^{86}\text{Sr}$	2SE	$^{87}\text{Sr}/^{86}\text{Sr}$	2SE	$^{85}\text{Rb}/^{86}\text{Sr}$	$^{87}\text{Rb}/^{86}\text{Sr}$	2SE	$[^{87}\text{Sr}/^{86}\text{Sr}]_{\text{IS}_{\text{Ma}}}$	2 $\sigma$
NK	0.05639	0.00003	0.71136	0.00001	0.01447	1.7	$4.8 \times 10^{-4}$	0.70977	0.00001
YS	0.05650	0.00002	0.71111	0.00001	0.00732	0.95	$7.1 \times 10^{-4}$	0.71019	0.00001
TK01	0.05652	0.00002	0.71062	0.00001	0.00764	0.93	$1.1 \times 10^{-4}$	0.70973	0.00001
TK02	0.05655	0.00002	0.71552	0.00002	0.05764	4.4	$4.6 \times 10^{-4}$	0.71126	0.00005
2801	0.05632	0.00006	0.70949	0.00002	0.0004	0.92	$2.0 \times 10^{-2}$	0.7086	0.00003
2802	0.05640	0.00003	0.71132	0.00003	0.00176	1.4	$2.1 \times 10^{-2}$	0.70996	0.00003
KJ	0.05647	0.00003	0.71085	0.00002	0.00393	1.3	$1.5 \times 10^{-2}$	0.70957	0.00002
MC	0.05647	0.00001	0.71630	0.00001	0.00101	1.2	$1.0 \times 10^{-2}$	0.71511	0.00002

Table 3-6. Sr isotopic data of Bussetsu granite samples

	$^{143}\text{Nd}/^{144}\text{Nd}$	2SE	$^{145}\text{Nd}/^{144}\text{Nd}$	2SE	$^{147}\text{Sm}/^{144}\text{Nd}$	2SE	$^{151}\text{Nd}/^{144}\text{Nd}$	2SE	$[^{143}\text{Nd}/^{144}\text{Nd}]_{\text{IS}_{\text{Ma}}}$	2 $\sigma$
NK	1.14305	0.00004	0.51217	0.00001	0.34844	0.00001	0.1259	0.0043	0.241615	0.000012
YS	1.14299	0.00003	0.51216	0.00001	0.34843	0.00001	0.1223	0.0042	0.241611	0.000011
TK01	1.14309	0.00004	0.51222	0.00001	0.34841	0.00001	0.1339	0.0041	0.241552	0.000014
TK02	1.14221	0.00011	0.51219	0.00002	0.34843	0.00002	0.1508	0.0041	0.241614	0.000031
2801	1.14328	0.00003	0.51226	0.00001	0.34838	0.00001	0.1189	0.0047	0.241517	0.000011
2802	1.14339	0.00004	0.51219	0.00001	0.34843	0.00001	0.1356	0.0044	0.241599	0.000010
KJ	1.14272	0.00003	0.51217	0.00001	0.34842	0.00001	0.1272	0.0046	0.241579	0.000011
MC	1.14318	0.00003	0.51225	0.00001	0.34841	0.00001	0.1203	0.0052	0.241580	0.000011

### **3.6 Discussion**

#### **3.6.1 Geochronology of the Busetsu granite**

The monazite U–Pb age of 68.4 Ma ± 3.7 Ma is distinctly younger than the previously reported monazite CHIME ages of ~79–75 (Suzuki et al., 1994; Nakai and Suzuki, 2003), and consistent with the zircon U-Pb ages of 68–72 Ma (Takatsuka et al., 2018).

The age discrepancy might be due to the existence of multiple components of non-radiogenic Pb such as initial Pb and contaminant Pb, which results in inaccurate U–Pb age estimates in the CHIME dating methods. This problem is critical especially for young samples like the Busetsu granites. Indeed, even a high  $\mu$  value of 250,000 at the timing of crystallization, corresponding to U/Pb of ~37,000, causes ~10% discordant. If so, the monazite age obtained only from the concordant U-Pb data in this study should be more accurate, because the age estimate is insensitive to the presence of multiple non-radiogenic Pb components. There are two alternative possible explanations for the discrepancy. First, the presence of inherited monazite might account for the older CHIME ages of 79–75 Ma. Note, however, that the monazite CHIME ages do not correspond with the inherited zircon U-Pb ages of ~100 Ma, 95–84 Ma, and ~80 Ma (Takatsuka et al., 2018). In addition, no inherited monazite was observed in BSE images in this study, consistent with the higher solubility of monazite than zircon in crustal magmatic conditions (Yakymchuk and Brown, 2014). Second, the younger U-Pb ages

of monazite and zircon obtained by LA-ICP-MS may reflect secondary Pb-loss, given that analytical spatial resolution is much poorer in LA-ICP-MS than EPMA, leading to higher potential of analysis of altered domains in the former. I consider this unlikely, however, because the Busetsu granites have undergone little alteration and also because the BSE images show no significant alteration within the analyzed grains. Hence, I conclude that the LA-ICP-MS U-Pb ages are more representative of the crystallization age of the Busetsu pluton than the CHIME U-Pb ages.

Interestingly, the U-Pb age of monazite and zircon are also identical to a biotite K-Ar age of  $71.1 \pm 3.6$  Ma within uncertainty (Yamasaki, 2013). Given that the closure temperatures of the zircon and monazite U-Pb ages are  $\sim 800^\circ\text{C}$  (Lee et al., 1997; Cherniak et al., 2010), whereas that of and K-Ar age is  $\sim 350^\circ\text{C}$  (Harrison et al., 1985; Grove and Harrison, 1996), a minimum cooling rate in this range is estimated to be  $\sim 100^\circ\text{C/Ma}$  (Fig. 3-14). This estimated rapid cooling rate is consistent with the emplacement of the Busetsu granite magma in a shallow level at  $\sim 10$  km depth (Takatsuka et al., 2018).

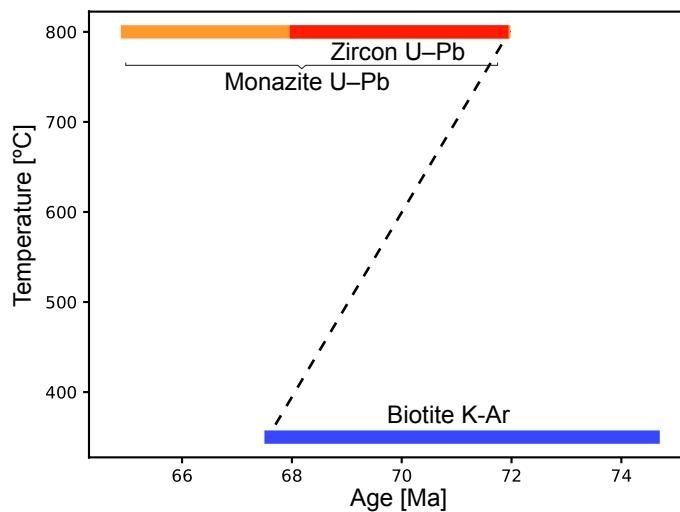


Fig. 3-14. Closure temperatures vs radiometric ages for the Busetsu granite samples. Minimum cooling rate was estimated using the monazite U–Pb age of 68.4 + 3.7 Ma and the youngest biotite K–Ar age of 71.1 – 3.6 Ma. The zircon and biotite age data are from Takatsuka et al. (2018) and Yamasaki (2013), respectively.

Table 3-7. Melt/plagioclase and melt/monazite distribution coefficient for REEs (Bédard et al., 2006; Stepanov et al., 2012)

	D(La)	D(Ce)	D(Pr)	D(Nd)	D(Sm)	D(Eu)	D(Gd)
Plagioclase	0.24	0.15	0.14	0.13	0.085	3.9	7.5
$\text{Log}f_{\text{O}_2} = -16 \quad \text{Log}f_{\text{O}_2} = -17 \quad \text{Log}f_{\text{O}_2} = -18$							
Monazite	2.8.E+03	2.5.E+03	2.8.E+03	3.1.E+03	2.5.E+03	1.8.E+03	1.9.E+03

### **3.6.2 Rare earth element behavior recorded in monazite**

The Eu anomaly, LREE/MREE, and Th/U ratios of the studied monazite grains are correlated with each other (Fig. 3-6), likely reflecting the evolution of granitic melt REE compositions during a series of its crystallization. The correlation between negative Eu anomalies and  $[Ce/Gd]_N$  of the monazite grains suggests that inter-element fractionation of REEs occurred as the granitic magma evolved. The negative Eu anomaly of monazite has been interpreted to reflect the crystal fractionation of plagioclase (Fisher et al., 2017). However, quantitative evaluation of the effect of plagioclase fractionation has not been made so far.

Here, I examine the impact of plagioclase fractionation on monazite REE composition based on element partitions. The monazite/melt partition coefficients for REEs experimentally determined by Stepanov et al. (2012) were adopted. On the other hand, the plagioclase/melt partition coefficients for REEs are well studied and parameterized by temperature and An molar content (Bédard et al., 2006). The  $D^{Pl/melt}$  values for REEs were calculated at 800°C and An content of 0.4. The used partition coefficients are summarized in Table 3-7.

The core domains of plagioclase are considered as restite from their Sr isotope compositions (details are discussed in the next section), whereas the rim domains of plagioclase would have been crystallized from melts because of textural features such as oscillatory zonings and overgrowth (Fig. 3-8). Monazite is also considered to have been

crystallized from melts, mainly because its U-Pb ages showed the lack of inherited core. As compared with the LREE contents of melts equilibrated with the monazite grains of samples YS and NK, those of melts equilibrated with the plagioclase are low (Fig. 3-15a). This difference suggests that the monazite grains crystallized after the growth of plagioclase rim domains.

The partition of Eu into minerals is subjective to a redox state of magma and the Eu<sup>2+</sup>/Eu<sup>3+</sup> ratio in the magma (Philpotts, 1970). Although it is difficult to accurately determine the degree of Eu anomaly in the equilibrium melt from the mineral compositions, the effect of plagioclase fractionation on changes in the Eu anomaly can be estimated. The  $D_{Eu}^{Pl/melt}$  can be expressed as a function of Log $f_{O_2}$  and temperature based on semi-empirical theory (Bédard et al. 2006). Considering that typical reduced-type granites have  $f_{O_2}$  below FMQ buffer and also that typical granitic magma temperatures are 700–800 °C (Frost and Frost, 1997), the partition coefficients of Eu were calculated at a temperature of 800°C for Log $f_{O_2}$  = −16, −17, and −18, respectively (Table 3-5). The melt composition equilibrated with the monazite showing the smallest Eu anomaly (sample NK-8) was assumed as an initial melt composition. The changes in the melt composition, caused by plagioclase fractionation, were calculated based on the Rayleigh fractionation model. Finally, the REE concentrations of monazite equilibrated with the residual melt were obtained. The change in the [Eu/Eu\*]<sub>N</sub> of the monazite are expressed as a ratio of the value of monazite crystallized from the fractionated melt to the initial value (Fig. 3-15b). The results indicated that 10–30 % plagioclase

fractionation can cause the decrease in the ratio by 0.75, which is the same as the observed variation for the studied monazite grains (Fig. 3-15b). This amount of removed fraction is consistent with the variation in modal abundance of plagioclase for the studied granite samples (Fig. 3-3g). Therefore, it is quantitatively reasonable that the change in Eu anomaly of monazite reflects the fractional crystallization of plagioclase.

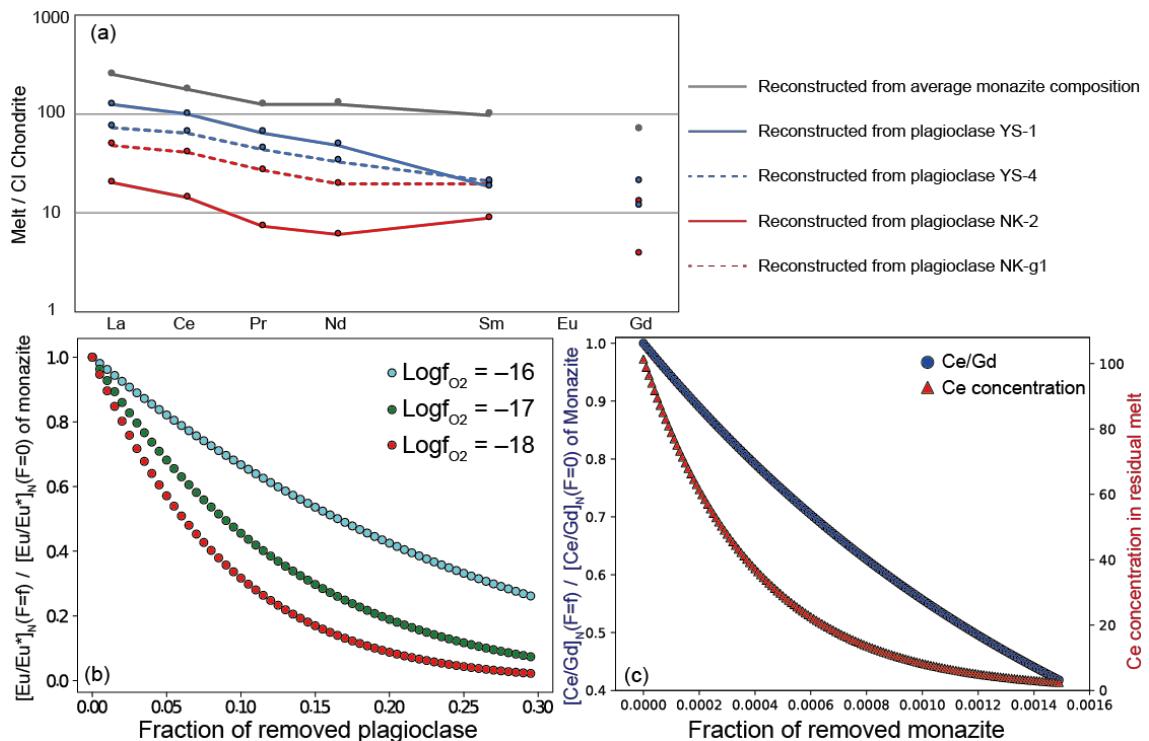


Fig. 3-15. (a) REE patterns of melts equilibrated with plagioclase and monazite. (b) Effect of plagioclase fractionation on REE composition of monazite crystalized from fractionated melt. The change in  $[\text{Eu}/\text{Eu}^*]_N$  is expressed as the ratio to initial value. Three different  $D_{\text{Eu}}^{\text{Pl/melt}}$  were used for  $\text{Log}f_{\text{O}_2} = -16$  (blue),  $-17$  (green), and  $-18$  (red). (c) Effect of monazite fractionation on REE composition of monazite crystalized from fractionated melt (circle) and Ce contents in residual melt (triangle).

The correlation between the  $[Eu/Eu^*]_N$  and  $[Ce/Gd]_N$  of monazite suggests that LREEs were more effectively removed from the melt as compared to MREEs during magma differentiation (Fig. 3-6). Regarding the whole-rock trace element compositions, both of the LREE and MREE contents decreased as  $SiO_2$  content increases (Fig. 3-12). The behaviors of lanthanides from La to Gd in peraluminous granites would be governed by monazite and apatite, given that these minerals account for >90% of the elements within the rocks (Bea, 1996). Monazite has higher partition coefficients for LREE than MREE; in contrast, apatite has lower partition coefficients for LREE than MREE (Fig. 3-16). Therefore, it is highly likely that the depletion of LREE relative to MREE in the parent melt resulted from the fractional crystallizations of monazite through its isolation from the melt as inclusions or settling in a magma chamber. When the Ce concentration in the residual melt and the  $[Ce/Gd]_N$  of crystallized monazite are calculated using the Rayleigh fractionation model in the same manner as the plagioclase fractionation, a removal of even small fraction (~0.02wt%) created the depletion of LREE in melt and monazite (Fig. 3-15c). When the decrease of  $[Ce/Gd]_N$  by 75 % is caused by the Rayleigh fractionation, the decrease of Ce content in melt reaches 100 ppm. This degree of decrease is larger than the observed change in whole-rock Ce concentrations ranging from ~60 to 40 ppm (Fig. 3-12c). Hence, the observed variation in the monazite  $[Ce/Gd]_N$  values of ~2–8 would be attributed to its isolation from the parent melt as inclusions rather than settling in a magma chamber. If that is the case, an elemental fraction between Th and U is expected at the same time due to the

significantly different partition coefficients of monazite for Th and U:  $D_{\text{Th}}/D_{\text{U}} > 7.5$  (Stepanov et al., 2012). Indeed, the observed positive correlation between  $[\text{Ce}/\text{Gd}]_{\text{N}}$  and  $[\text{Th}/\text{U}]_{\text{N}}$  supports the fractional crystallization of monazite (Fig. 3-6b).

The possible effect of xenotime fractionation was also considered using the above model. The distribution coefficients of xenotime for REEs were tentatively calculated using the REE concentrations of xenotime in the Busetsu granite (Suzuki et al., 1992) because the coefficients have not been determined experimentally. According to the modal abundances of monazite and xenotime in granitic rocks (Villaseca et al., 2003), the relative abundance of fractionated monazite and xenotime ( $F_{\text{Monazite}}/F_{\text{Xenotime}}$ ) were assumed to be 2.5–13. The calculated  $[\text{Ce}/\text{Gd}]_{\text{N}}$  values using the Rayleigh fractionation model were smaller than the original results by ~10–15%, indicating that the xenotime fractionation does not significantly affect the results of calculation. Hence, the  $[\text{Ce}/\text{Gd}]_{\text{N}}$  values of monazite are considered as recording primarily the fractionation of monazite.

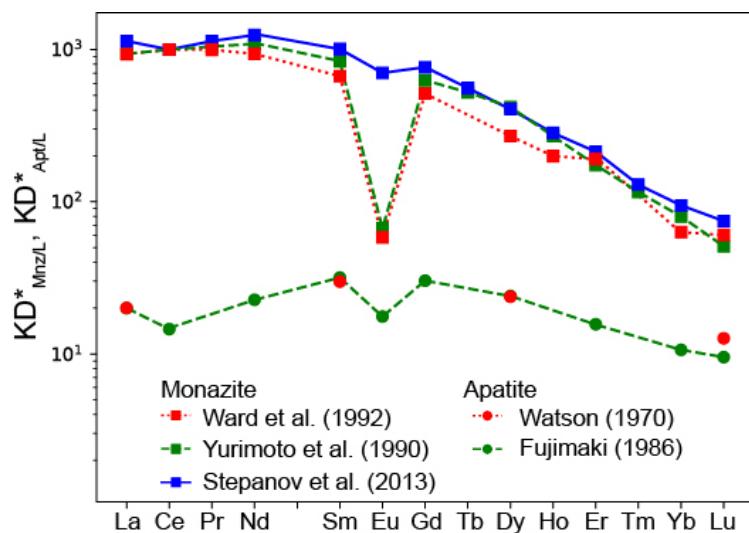


Fig. 3-16. Monazite/melt and Apatite/melt partitioning coefficients of trace elements. The plotted data are from Watson (1970), Fujimaki (1986), Yurimoto et al. (1990), Ward et al. (1992), and Stepanov et al. (2013).  $KD^*_{MnZ/L}$  and  $KD^*_{Apt/L}$  are monazite/melt partitioning coefficients normalized to  $D_{Ce} = 1000$  and apatite/melt partitioning coefficients normalized to  $D_{La} = 20$ , respectively.

### **3.6.3 Geochemical evolution of the Busetsu granitic magma: integration of whole-rock and mineral geochemistry**

The contents of  $\text{Fe}_2\text{O}_3$ ,  $\text{MgO}$ ,  $\text{CaO}$ , and  $\text{TiO}_2$  obviously decrease as the increase of  $\text{SiO}_2$  content for whole-rock compositions (Fig. 3-11). Considering that the mineral assemblages include Pl, Qtz, Kfs, and Bt (+ Ms) and that plagioclase and biotite are the major host phases of those major elements, the fractionation of plagioclase and biotite may be responsible for the major element variations in the Busetsu granites. This possibility can be tested using the trace element compositions. The concentrations of Sr and Ba, which are largely controlled by plagioclase and biotite, show a good correlation and the slope of the least squares regression line yielded the  $(D_{\text{Ba}}-1)/(D_{\text{Sr}}-1)$  of  $\sim 1.03$  (Fig. 3-12e). When the whole-rock partition coefficient is calculated for a major mineral assemblage of 80% plagioclase, 10% biotite, 5% quartz, and 5% K-feldspar, the  $(D_{\text{Ba}}-1)/(D_{\text{Sr}}-1)$  ratio was calculated to be  $\sim 1.04$ , which is in good agreement with the observed value. Therefore, the geochemical variation among the studied granites would be attributed to the fractional crystallization of plagioclase and biotite.

The negative correlation between the whole-rock Sr-Nd isotope compositions in Fig. 3-13 indicates open-system magmatic processes. Although the mineral isotopic data demonstrated that more than two components were involved in the formation of the Busetsu granites, a conventional two component mixing model is tentatively applied here to estimate the relative contributions of mantle-derived and crust-derived rocks. The isotope compositions of the mafic and metamorphic rocks associated with the

Ryoke Belt are plotted in the Sr-Nd isotope diagram (Fig. 3-17). The isotopic data of the Ryoke metamorphic rocks in the Ina area (Yuhara et al., 2000), which is located in the same Chubu district, were used due to the lack of the data in the Mikawa area. The isotopic data of mafic rocks are individually plotted for the Mikawa (Nakajima et al., 2004; Yuhara and Kagami, 2012) and other areas (Izumi et al., 2000). The end components were set on the basis of the maximum range observed in the mineral isotope compositions ( $[^{87}\text{Sr}/^{86}\text{Sr}]_{68\text{Ma}}$ : 0.71400–0.70814,  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$ : 0.51993–0.51228). Crust- and mantle-derived components were assumed to have 170 ppm Sr and 32 ppm Nd, and 295 ppm Sr and 13 ppm Nd, respectively based on the averages of the reported data (Yuhara et al., 2000; Nakajima et al., 2004; Yuhara and Kagami, 2012). The outcome of mixing model between the mantle-derived and crust-derived components implies that ~10–60% contribution of the crustal component roughly accounts for the Sr-Nd isotopic variations. The whole-rock geochemistry and isotope chemistry revealed that the reworking of sedimentary rocks in addition to magmatic fractionation of mantle-derived magma was responsible for the formation of the Busetsu granite. Moreover, the isotope compositions of plagioclase and monazite elucidated that the multiple components were involved in the evolution of granitic magma at various timings and levels.

The  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios in core domains of plagioclase show variation, whereas those in rim domains are relatively homogenous for each rock sample (Fig. 3-10). This finding is interpreted by integrating the textural and geochemical information. First, some

$^{87}\text{Sr}/^{86}\text{Sr}$  ratios of plagioclase core domains are obviously different from the average of  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios in the rim domains and whole-rock values for each sample. By contrast, the rim domains of the plagioclase grains yielded consistent isotope compositions with the whole-rock isotope compositions (Fig. 3-10). The variation in isotope composition of core domains demonstrated the isotopically different multiple sources for the formation of the Busetsu granite. Second, trace element chemistry of plagioclase and monazite suggested the monazite crystallization subsequent to the growth of plagioclase rim domains. This observation indicates that the Sr isotope compositions of the granitic magmas would have been homogenized during an intermediate stage in which the plagioclase rim domains crystallized but accessory phases including monazite have not crystallized. The  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$  ratios of monazite grains are generally homogeneous within each rock sample, as with the case of plagioclase rim domains (Fig. 3-7). However, the significant variation of  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$  ratios for monzogranite sample TK02 implies assimilations of radiogenic and unradiogenic sources. Indeed, the  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}}$  ratios range from  $0.51228 \pm 3$  to  $0.51193 \pm 3$  is consistent with the Nd isotopic compositions of mafic rocks and metasedimentary rocks in the Ryoke belt, respectively (Fig 3-13; Nakajima et al., 2004; Yuhara and Kagami, 2012). To verify whether the isotopic compositions of magma was changed by assimilation of country rocks at emplacement level, further measurements of  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios of plagioclase-outermost domains are necessary. The obtained mineral isotope

compositions have placed further constraints on the isotopic evolution of the Busetsu granite in addition to conventional whole-rock geochemistry and isotope chemistry.

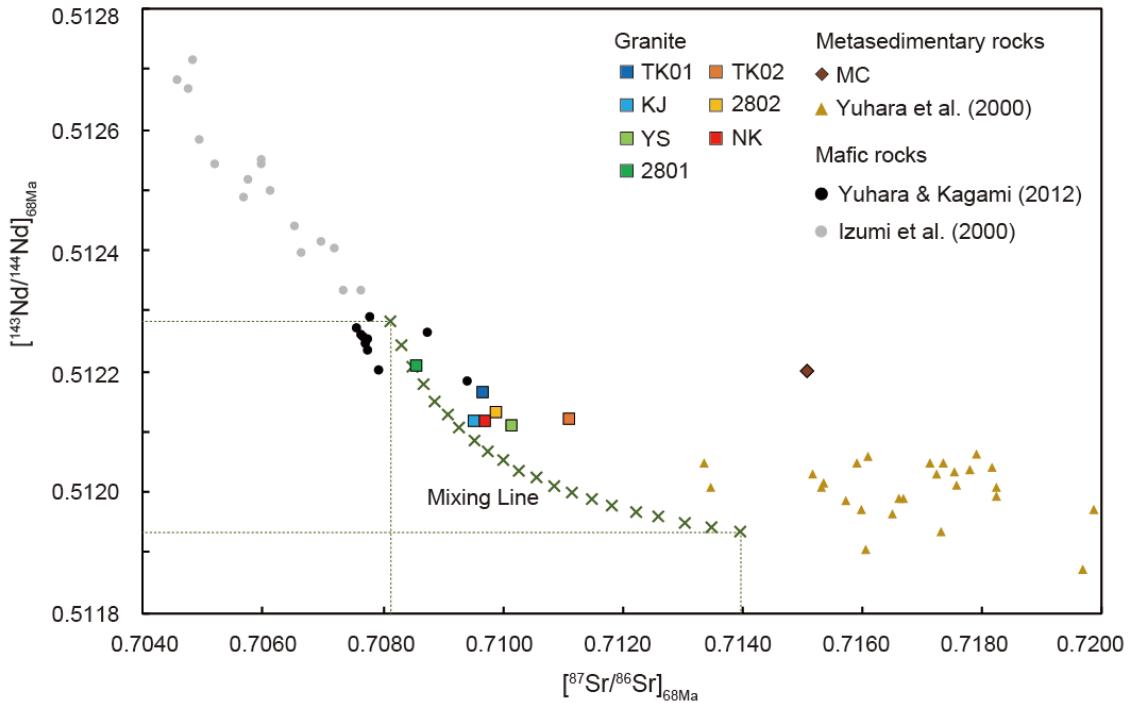


Fig. 3-17. Sr-Nd isotope systematics showing negative correlation associated with mantle-derived and sedimentary components. The mixing line is calculated using two components constrained by monazite and plagioclase isotope compositions:  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}} = 0.51227$  and  $[^{87}\text{Sr}/^{86}\text{Sr}]_{68\text{Ma}} = 0.7081$ ;  $[^{143}\text{Nd}/^{144}\text{Nd}]_{68\text{Ma}} = 0.51185$  and  $[^{87}\text{Sr}/^{86}\text{Sr}]_{68\text{Ma}} = 0.7140$ . The dotted lines indicate the range of observed variation in Sr and Nd isotope compositions of plagioclase and monazite, respectively. Data source for metasedimentary rocks in the Ryoke Belt: Yuhara et al. (2000); Data sources for mafic rocks in the Mikawa area; Iizumi et al. (2000), Yuhara and Kagami, (2012).

### **3.7 Conclusion**

The U–Pb dating of monazite indicated that the solidification age of the Busetsu is  $68.4 \pm 3.7$  Ma, which is consistent with the recently reported zircon U–Pb age. The systematic variation in the Eu anomaly and LREE depletion relative to MREE of monazite resulted from plagioclase and monazite fractionation. The Sr-Nd isotope compositions revealed the multiple source components for the formation of the Busetsu granite, which could not be constrained only by whole-rock isotope analysis. Moreover, the significant variations in plagioclase Sr and monazite Nd isotopic compositions for a monzogranite sample were taken as evidence of assimilation at emplacement level prior to the solidification. It is demonstrated that an integration of geochemical and isotope microanalyses of monazite and plagioclase enables tracking open system magmatic processes.

## **4. Trace element systematics of igneous monazite from granitic rocks in the Japan arc: Implication for its use as an indicator of source rock type**

### **4.1 Introduction**

Monazite is a common accessory mineral in low-Ca granitic rocks (Bea, 1996; Suzuki et al., 1992, 1990; Kelts et al., 2008; Fisher et al., 2017) and various metamorphic rocks (Overstreet, 1967; Parrish, 1990; Catlos, 2013). Monazite can be precisely dated by the U–Th–Pb system (Rasmussen et al., 2001; Harrison et al., 2002; Williams et al., 2007; Iizuka et al., 2010b), and is well-suited for rare earth element (REE) geochemistry and Nd isotope tracer studies (Evans and Zalasiewicz, 1996; Tomascak et al., 1998; Iizuka et al., 2010b; Stepanov et al., 2012; Fisher et al., 2017). Because monazite, especially from metamorphic rocks, often shows complex chemical zoning (Zhu and O’Nions, 1999; Spear and Pyle, 2002; Kawakami and Suzuki, 2011; Seydoux-Guillaume et al., 2002; Mahan et al., 2006), microanalysis techniques are essential for obtaining reliable data about the compositional variation of monazite grains. The major element composition of monazite has traditionally been determined using an electron probe micro analyzer (EPMA) with a spatial resolution of ~1 µm (Williams et al., 2007). The Th–U–total Pb data collected by the EPMA for monazite have also been used to estimate the timing of its (re)growth by assuming no common Pb and no Pb-loss or by constructing a chemical isochron (Suzuki and Adachi, 1998, 1991; Jercinovic and Williams, 2005; Suzuki and Kato, 2008; Montel et al., 1996; Cocherie et al., 1998; Pyle et al., 2005). In situ monazite U–Th–Pb dating has been also performed with secondary

ion mass spectrometry (SIMS) (Ireland and Gibson, 1998; Stern and Berman, 2001; DeWolf et al., 1993; Vry et al., 1996; Zhu et al., 1997; Rasmussen et al., 2001; Sano et al., 2006; Jantos et al., 2012; Kusiak et al., 2014) and laser ablation-inductively coupled plasma-mass spectrometry (LA-ICP-MS) (Machado and Gauthier, 1996; Kosler et al., 2001; Foster et al., 2002; Horstwood et al., 2003). Although the analytical resolutions achievable with the SIMS and LA-ICP-MS techniques ( $\geq 5 \mu\text{m}$ ) are lower than the resolutions possible with EPMA, these techniques have lower detection limits and provide Pb isotope data that allow us to evaluate the concordance of the U–Th–Pb system. The LA-ICP-MS technique is also capable of precise analysis of trace element abundances and Sm–Nd isotopic ratios in monazite with a spatial resolution of  $\geq 5 \mu\text{m}$  (Mcfarlane and Mcculloch, 2007; Bea et al., 1994; Poitrasson et al., 2000; Yang et al., 2008; Iizuka et al., 2011a, b).

In the last decade, these *in situ* analytical techniques have been applied to many metamorphic monazite studies (Rubatto et al., 2006; Buick et al., 2010; Iizuka et al., 2010b; Mottram et al., 2014; Štípká et al., 2015). These studies revealed that the geochemical composition of monazite is largely controlled by co-existing mineral assemblages and it can provide strong constraints on its crystallization conditions. For instance, monazite grains in granulite-facies metasedimentary rocks commonly exhibit significant heavy-REE (HREE) depletion, relative to light-REE (LREE) abundance, and elevated Th/U ratios, reflecting the effects of the co-existence with garnet and zircon on the HREE and Th/U fractionations, respectively (e.g., Rubatto et al., 2006).

Accordingly, the combined use of monazite geochronology and geochemistry is a powerful tool for tracing metamorphic history. In addition, Nd isotope microanalysis of monazites in metasedimentary rocks has indicated that detrital monazites are the primary source of LREE during metamorphic monazite formation (Iizuka et al., 2011b).

Despite its potential as a tracer of magmatic evolution, reports of igneous monazite geochemistry, and particularly trace element geochemistry, are far less common than metamorphic monazite studies. Hoshino et al. (2012) determined the major element compositions of monazites from a wide range of granitic rocks in the Japan arc, which revealed systematic variations in La/Sm ratios between granites and pegmatites and between magnetite-series (oxidized type) and ilmenite-series (reduced type) rocks. They interpreted the systematic LREE variations to indicate differences in the magmatic protolith for magnetite- and ilmenite-series granitic rocks. Recently, Fisher et al. (2017) demonstrated correlated changes in Ce/Gd, Eu/Eu\* (i.e., the degree of Eu anomaly), and Nd isotopic ratios in igneous monazites from the Sweetwater Wash pluton in southeastern California, which was interpreted to reflect the fractionation of monazite and plagioclase associated with the assimilation of country rocks.

In this study, to link monazite composition with various igneous petrogenesis and different processes, we conducted comprehensive measurements of REE-Th-U abundances and Sm-Nd isotope ratios in monazites from magnetite- and ilmenite-series granitic rocks in the Japan arc. The data revealed variations in trace elements among monazites from different types of rocks and within a single monazite grain. Based on

the results, we discussed the behavior of REE-Th-U in granitic magmas under different oxidation conditions. Finally, we explored the potential of REE-Th-U geochemistry as an indicator of the provenance of detrital monazites by comparing our results with previously reported data for monazite grains of different origins.

#### 4.2 Geologic outline and samples

The Japan arc sits along the eastern margin of the East Asian continent, having mainly developed through subduction-related orogeny since the Permian, followed by the formation of a back-arc basin ca. 20 Ma (e.g., Maruyama et al., 1997; Taira, 2001; Isozaki et al., 2010). In the southwestern part of the Japan arc, granitic rocks are distributed along the inward side of the Median Tectonic Line and can be divided into the Ryoke, Sanyo, and San-in belts (Fig. 4-1). In the northeastern part of the arc, granitic rocks are exposed mainly in the Abukuma and Kitakami belts. These Japanese granitic rocks have been classified into magnetite-series and ilmenite-series, reflecting distinctive redox states during their formation (Ishihara, 1977). Magnetite-series granites are characterized by a mineral assemblage of magnetite (0.1–2%), ilmenite, hematite, pyrite, titanite, epidote, and high-Mg/Fe biotite. Ilmenite-series granites contain ilmenite (<0.1%), pyrrhotite, graphite, muscovite, and low-Mg/Fe biotite. Both series contain zircon, monazite, and xenotime as accessory phases (Hoshino et al., 2012). The magnetite-series is dominant in the Kitakami and San-in belts, whereas ilmenite-series is dominant in the Abukuma, Ryoke, and Sanyo belts (Ishihara, 1977). The formation of magnetite-series granitic rocks mainly occurred at 120–110 Ma in the

Kitakami belt and 70–55 Ma in the San-in belt, whereas that of ilmenite-series granitic rocks was significant at 110–100 Ma in the Abukuma belt and 95–70 Ma in the Ryoke and Sanyo belts (Seno and Maruyama, 1984; Kinoshita, 1995; Takagi, 2004). The younger (100–55 Ma) Japanese granitic rocks generally have lower initial  $^{143}\text{Nd}/^{144}\text{Nd}$  and higher initial  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios than older rocks (Takagi, 2004). Furthermore, ilmenite-series granitic rocks tend to yield lower  $^{34}\text{S}/^{32}\text{S}$  and higher  $^{18}\text{O}/^{16}\text{O}$  ratios than contemporaneous magnetite-series rocks, suggesting greater contributions from sedimentary components to ilmenite-series granitic magmas (Ishihara and Sasaki, 1989; Takagi, 2004).

In this study, monazite grains from four magnetite-series granitic rocks of the San-in and Kitakami belts and nine ilmenite-series granitic rocks of the Ryoke, Sanyo, and Abukuma belts were investigated. Sample locations are illustrated in Fig. 4-1. The studied rock samples include four granites and nine granitic pegmatites. The sample locations, ages, mineral assemblages, and monazite morphologies of the granitic rocks are summarized in Table 1. Note that the Shirakawa and Nakano magnetite-series and Busetsu ilmenite-series granites were classified as S-type based on the geochemical classification proposed by Chappell and White (1974) (Ishihara, 1971; Hoshino and Ishihara, 2007). On the other hand, the Naegi ilmenite-series granite is I-type (Ishihara and Murakami, 2006). All of these rock samples and monazite grains analyzed in this study were used in the previous EPMA-based major element geochemical study by Hoshino et al. (2012). Geochemical analysis was carried out on monazites mounted in

epoxy resin for all samples except the Naegi and Busetsu ilmenite-series granites, which were studied in thin section.

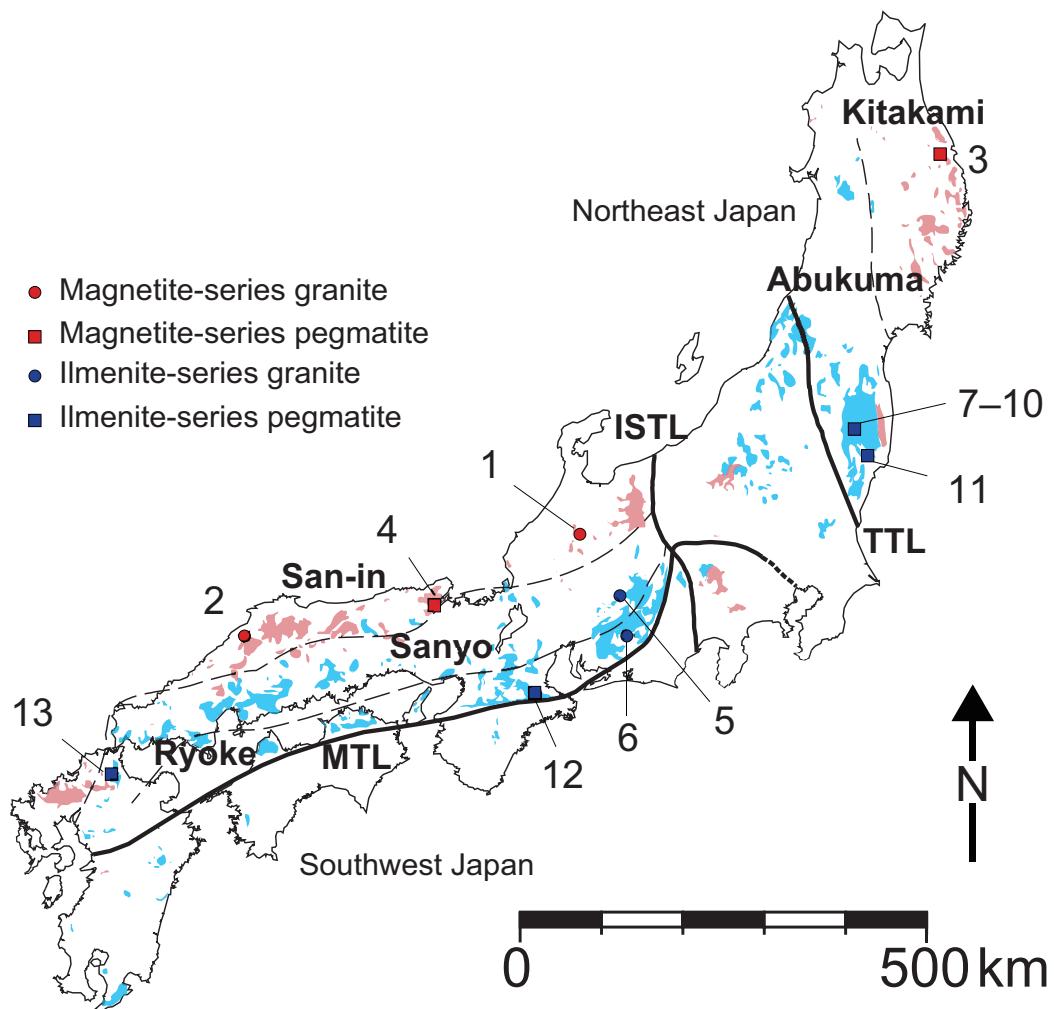


Figure 4-1: Distribution of magnetite (red)- and ilmenite (blue)-series granitic rocks in the Japan arc (modified after Takagi, 2004, Hoshino et al., 2012). Locations of samples are indicated (Table 1). Abbreviations are as follows: TTL, Tanakura tectonic line; ISTL, Itoigawa-Shizuoka tectonic line; MTL, Median tectonic line.

**Table 1**  
Summary of age and mineralogical data for studied granitic smapies.

Sample Number	Location	Granite belt	Age of granitic rocks [Ma]	Major mineral assemblage	Accessory mineral assemblage	Monazite morphology and grain size	Ref.
<b>Magnetite-series granites</b>							
1	Shirakawa, Gifu	San-in	65.6 ± 1.8	Qtz, Pl, Kfs, Hbl, Bt	allanite, titanite, Nb-Ta minerals, magnetite, zircon	Subhedral-anhedral, 20–100 µm	1–4
2	Nakano, Shimane	San-in	61	Qtz, Pl, Kfs, Hbl, Bt	allanite, titanite, Nb-Ta minerals, allamagnete, zircon	Subhedral-anhedral, 20–100 µm	1, 3–6
<b>Magnetite-series pegmatites</b>							
3	Kamiotomo, Iwate	Kitakami	135 ± 25	Qtz, Pl, Kfs, Hbl, Bt, Ms	zircon, thorite, xenotime, magnetite	Prismatic crystals, 2–5 mm	1, 7, 8
4	Oro, Kyoto	San-in	61.9 ± 0.9	Qtz, Pl, Kfs, Hbl, Bt, Ms	zircon, xenotime, allanite, magnetite, fergusonite	Platy-crystals, 3–8 mm	1, 3, 9
<b>Ilmenite-series granites</b>							
5	Naegi, Gifu	Sanyo	67.2 ± 3.2	Qtz, Pl, Kfs, Bt, Ms	fluorite, xenotime, Nb-Ta minerals, thorite, ilmenite, zircon	Subhedral-anhedral, 20–100 µm	1, 10, 11
6	Busetsu, Aichi	Ryoke	82.5 ± 3.9	Qtz, Pl, Kfs, Bt, Ms	apatite, xenotime, almandine, ilmenite, zircon	Subhedral-anhedral, 20–100 µm	1, 2
<b>Ilmenite-series pegmatites</b>							
7	Utsumine, Fukushima	Abukuma	106 ± 16	Qtz, Pl, Kfs, Hbl, Bt, Ms	allanite, zircon, xenotime	Long prismatic crystals, 5–8 mm	3, 4, 12
8	Morita, Fukushima	Abukuma	106 ± 16	Qtz, Pl, Kfs, Hbl, Bt, Ms	thorite	Prismatic crystals, 5–10 mm	12
9	Shichironai, Fukushima	Abukuma	106 ± 16	Qtz, Pl, Kfs, Hbl, Bt, Ms	zircon, xenotime	Long prismatic crystals, 5–8 mm	1, 12
10	Shiozawa, Fukushima	Abukuma	106 ± 16	Qtz, Pl, Kfs, Hbl, Bt, Ms	zircon, columbitite xenotime	Prismatic crystals, 2–5 mm	12
11	Shimo-ono, Ibaraki	Abukuma	115	Qtz, Pl, Kfs, Hbl, Bt, Ms	zircon, allanite, ilmenite, magnetite, hematite, fergusonite, xenotime	Long prismatic crystals, 10–12 mm	1, 3–5, 13
12	Misugi, Mie	Ryoke	75	Qtz, Pl, Kfs, Hbl, Bt, Ms	xenotime	Long prismatic crystals, 5–8 mm	5, 14
13	Masaki, Fukuoka	Sanyo	100 ± 5	Qtz, Pl, Kfs, Hbl, Bt, Ms	zircon, thorite	Long prismatic crystals, 5–8 mm	1, 5, 15, 16

Mineral abbreviations are from Kretz (1983); References: data sources are (1) Hoshino et al. (2010), (2) Shibata and Ishihara, (1979), (3) Hoshino et al., (2006), (4) Hoshino et al., (2007), (5) Kagami et al., (1989), (6) Yakuishi et al., (1982), (7) Fujimaki et al., (2012), (8) Takagi (2004), (9) Terakado and Nohda, (1993), (10) Suzuki and Adachi, (1998), (11) Suzuki et al., (1990), (12) Shibata and Tanaka, (1987), (13) Tanaka, (1977), (14) Hayama et al., (1982), (15) Yuhara et al., (2014), and (16) this study

### **4.3. Analytical method**

All monazite grains were investigated by back-scattered electron (BSE) imaging to check their internal zoning structures and the presence of inclusions before geochemical and Sm–Nd isotope analysis. The BSE images were obtained using a JEOL JSM-7000F Field Emission Scanning Electron Microscope at the Department of Earth and Planetary Science, The University of Tokyo. An accelerating voltage of 10 kV, an emission current of 20 nA, and a working distance of 10 mm were used.

Determination of REE-Th-U abundances were carried out using an iCAPQ ICP-MS attached to a Nd:YAG laser ablation (LA) system at the Department of Earth and Planetary Science, The University of Tokyo. The analytical parameters of the LA system and the ICP-MS are shown in Table 4-2. The LA system was equipped with the active two-volume HelExTM Cell (Eggins et al., 2005), in which the effective cell volume is so small ( $\sim 2.5 \text{ cm}^3$ ) that aerosol deposition around the ablation pit is minimized. Unknown samples were analyzed in runs of eight samples, bracketed by three measurements of a standard NIST SRM 610 glass (Pearce et al., 1997). The Ce concentrations were used as an internal standard, which were determined by a Field Emission-EPMA JEOL JXA 8530F HyperProbe at The University of Tokyo and at the National Institute of Advanced Industrial Science and Technology (AIST) in Tsukuba, Japan. The operating conditions of the EPMA analysis were: a beam diameter of  $5 \mu\text{m}$ , a probe current of 30 nA, and an accelerating voltage of 25 kV. Other element concentrations were determined using the Ce concentration measured by EPMA and

their abundances relative to Ce determined by LA-ICP-MS. To correlate EPMA data with LA-ICP-MS data, the ~15- $\mu\text{m}$  laser ablation pit was placed in a portion of the sample with a similar internal structure revealed by BSE imaging, and close to the pit created previously during EPMA analysis. Spectral interference from middle-REE (MREE) oxide ions on HREE ions can be problematic for accurate REE determinations in monazites during LA-ICP-MS analysis, but we corrected for them by determining oxide formation rates using synthetic MREE phosphates during each analytical session (Itano and Iizuka, 2017).

The Sm–Nd isotope analysis was performed on a Neptune Plus multi-collector-ICP-MS (MC-ICP-MS), coupled with the Nd:YAG LA system. The operating parameters for this analysis are presented in Table 3. A 30- $\mu\text{m}$  laser beam with a ~20-J/cm<sup>2</sup> laser fluence and 4-Hz repetition rate was used. The laser ablation site was located next to the pit created during REE analysis and within a similar domain, based on BSE imaging. To correct for the different responses of the Faraday amplifiers to signal intensity change, isobaric interference, and mass bias, we followed the protocols used by Iizuka et al. (2011a): after calibrating the amplifier response using the empirically determined inter amplifier calibration factors (Table 4-3), the contribution of <sup>144</sup>Sm interference on <sup>144</sup>Nd was corrected by measuring <sup>149</sup>Sm and the exponential mass bias factors for Sm and Nd were calculated by normalizing the measured <sup>147</sup>Sm/<sup>149</sup>Sm and <sup>146</sup>Nd/<sup>144</sup>Nd to 1.08507 and 0.7219, respectively (O’Nions et al., 1977; Hidaka et al., 1995). To ensure accurate comparison with literature results, the

$^{143}\text{Nd}/^{144}\text{Nd}$  ratios that had been corrected for the interference and mass bias were further normalized to the thermal ionization mass spectrometry (TIMS) values  $^{143}\text{Nd}/^{144}\text{Nd} = 0.511600$  and  $^{145}\text{Nd}/^{144}\text{Nd} = 0.348415$ . This normalization was completed based on the La–Ce–Nd synthetic monazite (Synth-LCN) standard, using the mean value obtained for this standard during each analytical session (Iizuka et al., 2011a). The instrument fractionation of  $^{147}\text{Sm}/^{144}\text{Nd}$  was corrected for using Nd–Sm–Pr synthetic monazite (Synth-NSP) standard. To evaluate the accuracy and precision of the data obtained during the course of this study, we analyzed an in-house monazite standard EDR (from the Western Alps; Paquette and Tiepolo, 2007) in all analytical sessions. These results ( $0.512254 \pm 25$  with two standard deviations, Appendix Table S4-1) were consistent with the TIMS results ( $0.512265 \pm 05$ , Iizuka et al., 2011a). The accuracy of the  $^{144}\text{Sm}$  interference correction was further monitored by checking whether the obtained  $^{145}\text{Nd}/^{144}\text{Nd}$  ratios for sample analyses were consistent with those for analyses of the Sm-free synthetic monazite Synth-LCN.

Table 4-2. Operational parameters of the LA-ICP-MS for REE analysis.

<b>Laser ablation system</b>		CETAC LSX-213 G2+ (CETAC, USA) equipped with the active two-volume HelEx™ Cell
Laser source		213 nm
Pulse width		5 ns
Beam diameter		15 µm
Repetition rate		3 Hz
Laser fluence		20.54 J cm <sup>-2</sup>
Laser He carrier gas		0.95 L min <sup>-1</sup>
<b>ICP-MS</b>		iCAP Qc (Thermo Fisher Scientific)
RF-power		1550 W
Sampling depth		5.1 mm
Cool gas (Ar)		14 L min <sup>-1</sup>
Auxiliary gas (Ar)		0.78 L min <sup>-1</sup>
Sample gas (Ar)		0.80 L min <sup>-1</sup>
Measured isotopes	<sup>89</sup> Y, <sup>139</sup> , <sup>138</sup> La, <sup>140</sup> Ce, <sup>141</sup> Pr, <sup>143</sup> Nd, <sup>147</sup> Sm, <sup>151</sup> , <sup>153</sup> Eu, <sup>157</sup> Gd, <sup>159</sup> Tb, <sup>163</sup> Dy, <sup>165</sup> Ho, <sup>167</sup> Er, <sup>169</sup> Tm, <sup>173</sup> Yb, <sup>175</sup> Lu, <sup>232</sup> Th, <sup>238</sup> U, <sup>232</sup> Th <sup>16</sup> O, <sup>238</sup> U <sup>16</sup> O	

Table 4-3. Operational parameters of the LA-MC-ICP-MS for Sm-Nd isotope analysis.

<b>Laser ablation</b>	CETAC LSX-213 G2+ (CETAC, USA) equipped with the active two-volume system										
Laser source	213 nm										
Pulse width	5 ns										
Beam diameter	30 $\mu\text{m}$										
Repetition rate	4 Hz										
Laser fluence	20.54 J $\text{cm}^{-2}$										
<b>MC-ICP-MS</b>	Neptune plus (Thermo Fisher Scientific)										
RF-power	1200 W										
Gas flow rates	Ar cooling: 14 L $\text{min}^{-1}$			Ar auxiliary: 1.0 L $\text{min}^{-1}$							
	Ar sample: 1.0–1.1 L $\text{min}^{-1}$			He carrier: 0.95 L $\text{min}^{-1}$							
N <sub>2</sub> carrier: ~4 mL $\text{min}^{-1}$											
Mass resolution	Low										
Data acquisition mode	Time resolved analysis										
Integration time	1.02 sec										
Analysis mode	static										
Collector arrangement											
Faraday cups	L3	L2	L1	Axial	H1	H2	H3				
Amplifier calibration factor*	0.0021	-0.00097	0.0021		-0.0013	-0.0011	0.0041				
Analyte isotopes	<sup>142</sup> (Nd+Ce)	<sup>143</sup> Nd	<sup>144</sup> (Sm+Nd)	<sup>145</sup> Nd	<sup>146</sup> Nd	<sup>147</sup> Sm	<sup>149</sup> Sm				

\*The calibration factors were determined by analyzing the synthetic monazite Synth-LCN for Nd isotopes using LA-MC-ICPMS, following Iizuka et al. (2011a).

## 4.4 Results

### 4.3.1 Monazite from granitic rocks

Monazite grains from our pegmatite samples were generally prismatic, with the long axis ranging from 2 mm to  $\geq$ 10 mm, whereas those from the granites were subhedral to anhedral with lengths of 20–100  $\mu\text{m}$  (Table 4-1). Representative BSE images of the monazite grains are shown in Fig. 2. The monazite grains from the Oro, Shichironai, Shiozawa, and Masaki pegmatites showed core/rim structure and sector zoning (Figs. 4-2d, i, j, and m). On the other hand, monazite grains from the Kamiotomo, Morita, and Misugi pegmatites exhibited patchy and oscillatory zoning patterns (Figs. 4-2c, h, and l). The monazite grains from the Naegi granite showed complex compositional structures whose spatial scale was smaller than that of the LA-ICP-MS resolution ( $\sim$ 15  $\mu\text{m}$ ). The brightness contrasts in these BSE images are mainly attributed to the difference in Th contents within the monazite grains (Hoshino et al., 2012; Swain et al., 2005). All analyzed monazite grains from the Shirakawa, Nakano, and Busetsu granites, and the Utsumine and Shimo-ono pegmatites were homogeneous in BSE images. The BSE images further revealed that some of monazite grains from pegmatites contained altered domains that were characterized by exsolution of xenotime and Th-U silicate minerals and by compositional changes along cracks (Figs. 4-2g, h, l, m). Because fluid-mediated alteration processes can significantly modify monazite composition (e.g., Bea, 1996; Budzyń et al., 2017), and because the primary purpose of this study was to link

monazite composition with igneous processes, these altered domains were avoided in our analysis.

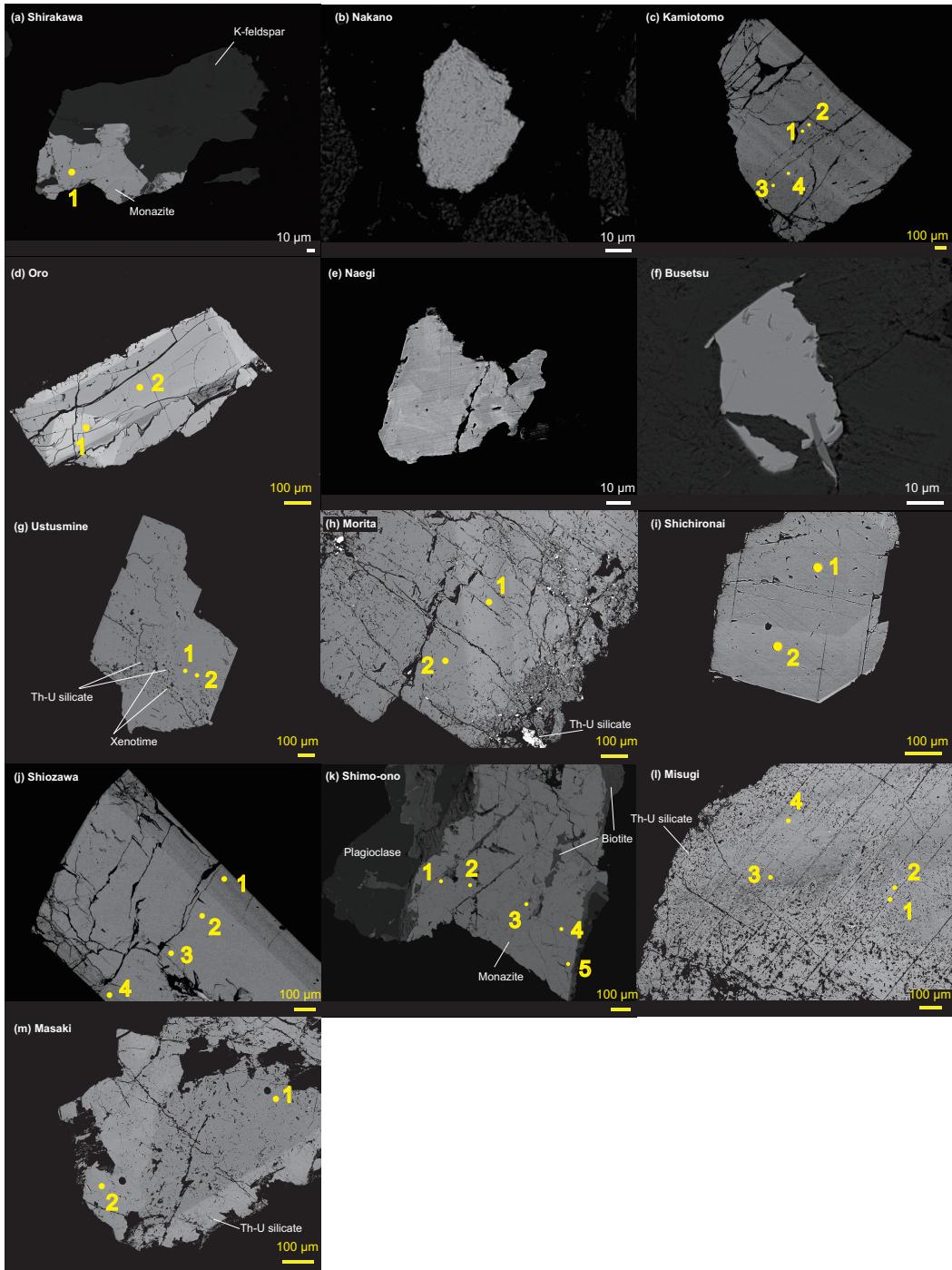


Fig. 4-2. Representative back-scattered electron images of the analyzed monazite samples. Magnetite-series: (a) Shirakawa granite, (b) Nakano granite, (c) Kamiotomo pegmatite, and (d) Oropegmatite. Ilmenite-series: (e) Naegi granite, (f) Busetsu granite, (g) Utsumine pegmatite, (h) Morita pegmatite, (i) Shichironai pegmatite, (j) Shiozawa pegmatite, (k) Shimo-ono pegmatite, (l) Misugi pegmatite, and (m) Masaki pegmatite. Yellow dots represent analytical spots for REE analyses of monazites from the pegmatites (Table S2, ESI).

Thirty-nine REE-Th-U abundance determinations were conducted on twenty-one monazite grains from the 13 granitic rocks. The monazite REE-Th-U data obtained by LA-ICP-MS and EPMA are listed in Appendix Table S4-2, together with the internal structures of the analytical spots. The chondrite-normalized REE patterns of monazites from the ilmenite-series granites and pegmatites and the magnetite-series granites and pegmatites are respectively shown in Fig. 4-3.

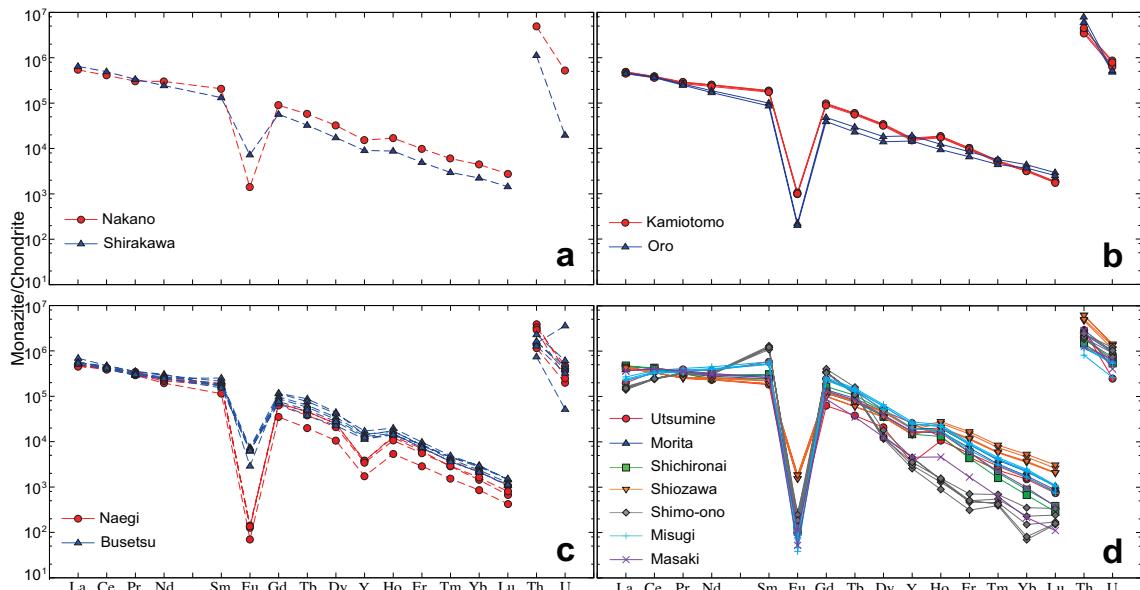


Fig. 4-3. Chondrite-normalized REE and Th-U patterns of monazites from magnetite-series (a) granites and (b) pegmatites and from ilmenite-series (c) granites and (d) pegmatites. Chondrite concentrations are taken from McDonough and Sun, (1995).

The normalized REE abundances generally formed smooth patterns from La to Lu with negative Eu anomalies, except for the Samples Utsumine, Shimo-ono, and Misugi showing the convex patterns with a peak of Sm (4-3d). The chondrite-normalized values

of  $[Eu/Eu^*]_N$ ,  $[Ce/Gd]_N$ ,  $[Gd/Lu]_N$ , and  $[Th/U]_N$  are graphed in Fig. 4-4, allowing a quantitative evaluation of the REE-Th-U systematics. All analyzed monazites exhibited REE patterns with negative Eu anomalies (i.e.,  $[Eu/Eu^*]_N < 1$ ) and HREE depletion relative to MREE ( $[Gd/Lu]_N > 1$ ). There were no significant differences in REE composition within all analyzed single grains except for the Masaki monazite grain, in which the core had a less fractionated REE pattern with more prominent negative Eu anomaly and a lower Th/U ratio than the rim (Fig. 4-5). We detected systematic differences between granites and pegmatites and between magnetite- and ilmenite-series samples (Figs. 4-3 and 4-4). Within magnetite- or ilmenite-series, monazites from the pegmatites basically exhibited more significant negative Eu anomalies than those from the granites. Monazites from the ilmenite-series granites and pegmatites tended to have stronger negative Eu anomalies ( $[Eu/Eu^*]_N$  ranging from  $\sim 10^{-2}$  to  $\sim 10^{-4}$ ) than their magnetite-series counterparts ( $[Eu/Eu^*]_N$  ranging from  $\sim 10^{-2}$  to  $\sim 10^{-3}$ ). The monazites with the most prominent negative Eu negative anomalies, which were from the ilmenite-series pegmatites, exhibited weak LREE enrichment, and sometimes even depletion, relative to MREE (i.e., lower  $[Ce/Gd]_N$  as low as 0.6); significant HREE depletion relative to MREE (i.e., high  $[Gd/Lu]_N$  of up to  $\sim 2400$ ); and less variable Th/U ratios (0.4 to 14). The results of L-MREE analyses are consistent with those obtained using EPMA by Hoshino et al. (2012).

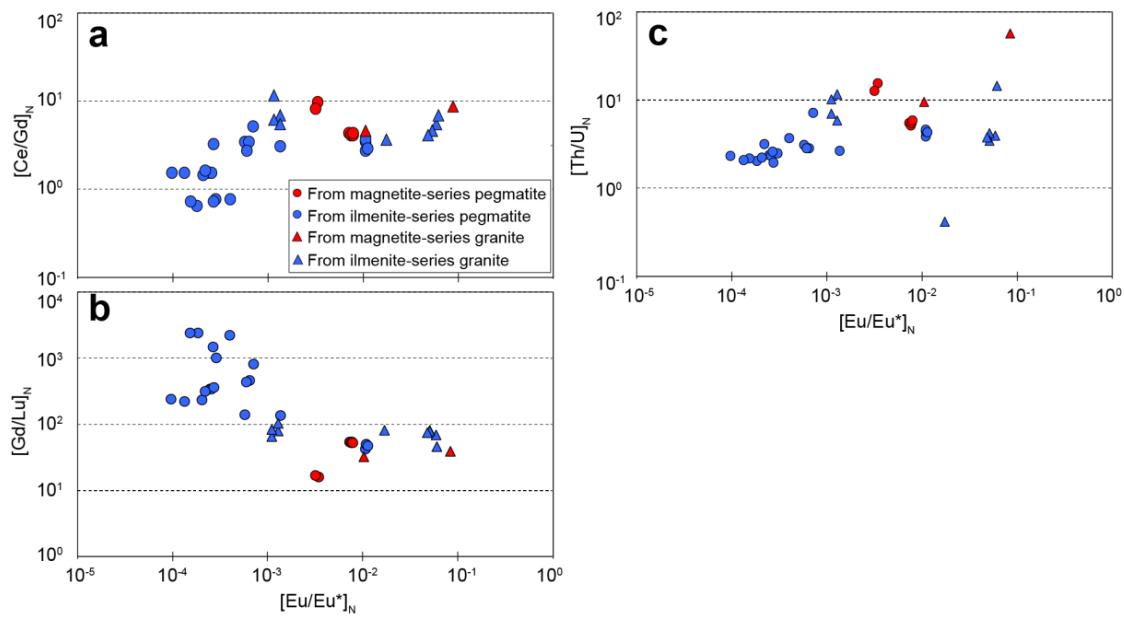


Fig. 4-4. Plots of (a)  $[Ce/Gd]_N$ , (b)  $[Gd/Lu]_N$ , and (c)  $[Th/U]_N$  vs.  $[Eu/Eu^*]_N$  for monazites from magnetite-series granites (red circle) and pegmatites (red triangle) and ilmenite-series granites (blue circle) and pegmatites (blue triangle).

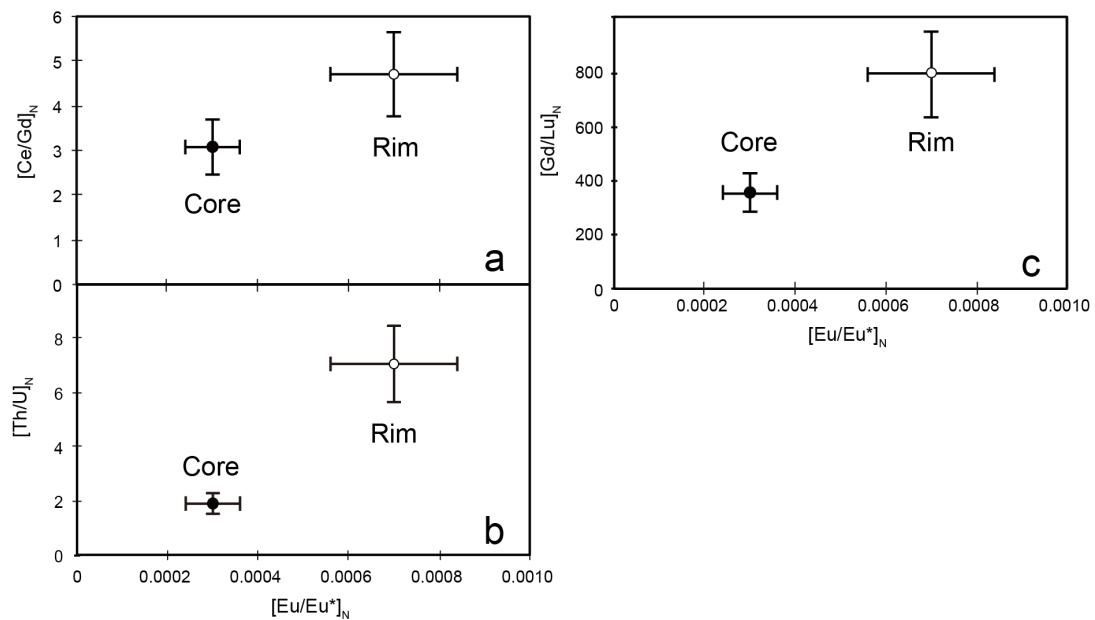


Fig. 4-5. Plots of (a)  $[Ce/Gd]_N$ , (b)  $[Th/U]_N$  and (c)  $[Gd/Lu]_N$  vs.  $[Eu/Eu^*]_N$  for the core and rim of the monazite from the Masaki pegmatite.

Ninety-four Sm–Nd isotope measurements were carried out on nine monazite grains from the two magnetite-series and seven ilmenite-series pegmatites (Appendix Table 4-3). Among these, two measurements of the rim in the Masaki pegmatite monazite yielded  $^{145}\text{Nd}/^{144}\text{Nd}$  ratios that were distinctively higher than those from the synthetic monazite Synth-LCN, indicating the inaccuracy in the  $^{147}\text{Sm}$  interference correction. This inaccuracy could be attributed to imperfect correction of the different amplifier responses when Sm and Nd signal intensities changed significantly and independently due to the sample’s chemical heterogeneity. Hence, we did not include the data from those two measurements in our discussion. The initial  $^{143}\text{Nd}/^{144}\text{Nd}$  ratios were calculated using the previously reported ages for each sample (Table 4-1) and  $\varepsilon\text{Nd}(t)$  values are graphically presented in Fig. 6a. The initial  $^{143}\text{Nd}/^{144}\text{Nd}$  ratios ranged from 0.512125 to 0.512525, corresponding to an  $\varepsilon\text{Nd}(t)$  of –8.3 to 0.6. All monazite grains showed no resolvable intra-grain variation in initial  $^{143}\text{Nd}/^{144}\text{Nd}$ , irrespective of their internal structures. There is no systematic difference between the magnetite- and ilmenite-series samples. Samples from the magnetite- and ilmenite-series exhibited the  $\varepsilon\text{Nd}(t)$  values ranging from –8.3 to 0.1 and from ca. –5.7 to 0.6, respectively. When the monazite  $\varepsilon\text{Nd}(t)$  values for each pegmatite are plotted against the crystallization age (Fig. 6b), on the other hand, it is evident that the monazites younger than 100 Ma ( $\varepsilon\text{Nd}(t)$  of –8.9 to –5.6) have lower  $\varepsilon\text{Nd}(t)$  values than older ones ( $\varepsilon\text{Nd}(t)$  of –0.7 to 1.0). Such a secular change in Nd isotopic composition has been recognizable in the whole-rock Nd isotopic data from granitic rocks in the Japan arc.

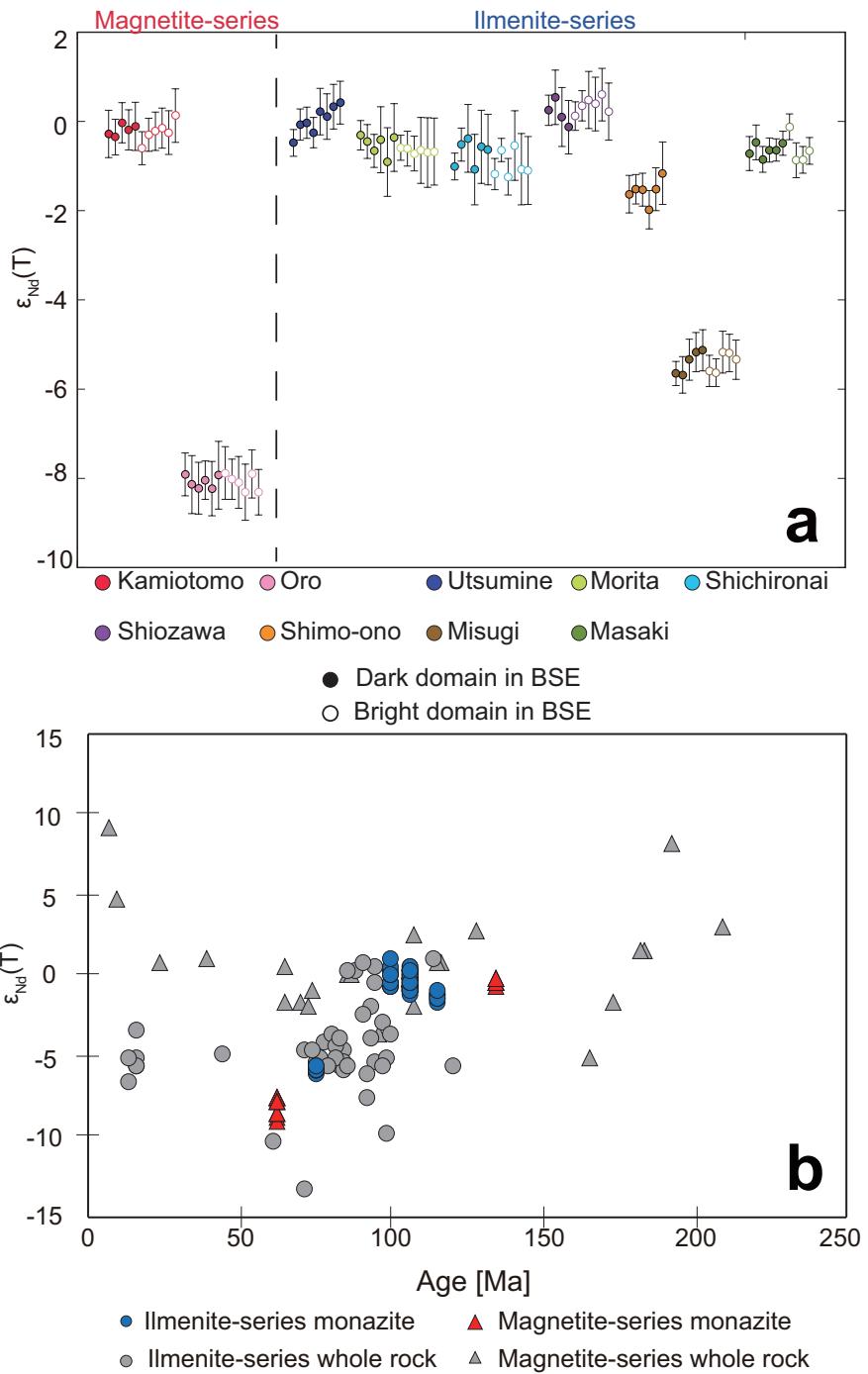


Fig. 4-6. (a)  $\epsilon_{\text{Nd}}(t)$  of pegmatite monazites. Filled and open circles represent data for dark and bright domains in back-scattered electron images, respectively. Error bars represent  $2\sigma$ . (b) Plot of  $\epsilon_{\text{Nd}}(t)$  vs. age for the monazites analyzed in this study (red and blue symbols). Whole-rock Nd isotopic data for granitic rocks in the Japan arc (gray symbols; Takagi, 2004) are also shown for comparison. Error bars represent  $2\sigma_m$ .

## 4.4 Discussions

### 4.4.1 Differentiation of granitic magmas recorded by monazite geochemistry

Major and trace element compositions in a monazite reflect the geochemical features of the magma from which it crystallized and its crystallization conditions. Our data reveal notable variations in monazite REE-Th-U abundances between pegmatites and granites, and between ilmenite-series and magnetite-series rocks. Given that granitic pegmatites are crystallized from highly differentiated volatile-rich granitic magmas (e.g., Yurimoto et al., 1990; London, 2005; Linnen et al., 2012), the results allow us to obtain a general picture of monazite REE-Th-U systematics during differentiation in granitic magmas under different redox states.

Monazites from pegmatites showed larger Eu negative anomalies than those from granites for each series (Figs. 4-3 and 4-4). Further, monazites from ilmenite-series rocks exhibited more prominent anomalies than those from the magnetite-series. Europium can exist in both divalent ( $\text{Eu}^{2+}$ ) and trivalent ( $\text{Eu}^{3+}$ ) state in crustal conditions, and  $\text{Eu}^{2+}$  is more strongly partitioned into feldspars than  $\text{Eu}^{3+}$  (Fujimaki and Tatsumoto, 1984; Bea et al., 1994; Bea, 1996). Thus, the larger Eu negative anomalies in pegmatitic monazites can be attributed to more significant crystal fractionation of feldspars in their parental magmas, which was discussed in Chapter 3. Moreover, considering that the ratio of  $\text{Eu}^{2+}/\text{Eu}^{3+}$  in magma changes with the redox state of the magma (Philpotts, 1970), the larger Eu negative anomalies in ilmenite-series monazites than magnetite-series monazites are interpreted to reflect more efficient incorporation of

Eu into fractionating feldspars in more reduced magmas. This is consistent with the fact that ilmenite, rather than magnetite, is the common opaque phase in reduced granites. Hence, monazite Eu negative anomalies are considered to be a measure of the degree of feldspar fractionation and the redox state in its parental granitic magma.

The larger Eu negative anomalies in the monazites from ilmenite-series pegmatites are accompanied by smaller Ce/Gd ratios and elevated Gd/Lu ratios (Fig. 4-4a). The variations in Ce/Gd and Gd/Lu ratios that we observed cannot be explained by changing the monazite/melt partition coefficients for MREE and HREE with the temperature and the water content of the melt (Stepanov et al., 2012). Instead, the fractionations between LREE and MREE would be attributed to fractional crystallization of LREE-rich mineral phases (e.g., monazite or allanite) during magma differentiation as discussed in the previous chapter. Similarly, the significant depletion of HREE relative to MREE in parent magma can be caused only by HREE-enriched mineral phases (e.g., titanite, xenotime, and garnet) with higher partition coefficients for HREE as compared to those for MREE (Bea et al., 1994). The lack of large variations in the Ce/Gd and Gd/Lu ratios in monazites from magnetite-series granitic rocks would reflect that such fractional crystallization is more vigorous in ilmenite-series granitic magmas relative to magnetite-series magmas. Fisher et al. (2017) also demonstrated a decrease in Ce/Gd ratios associated with the increase of Eu/Eu\* ratios in monazites from a peraluminous pluton in southeastern California, which was interpreted to result from the fractionation of monazite, which is a major host phase for LREE.

Experimental studies have shown that LREE solubility in a granitic melt is negatively correlated with its aluminum saturation index (ASI;  $\text{Al}_2\text{O}_3/[\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{CaO}]$ ), although the correlation is insignificant when the ASI is higher than ~1 (Rapp et al., 1987; Montel, 1993; Engi, 2017). Hence, monazites can be more readily saturated in peraluminous granitic melts than metaluminous or peralkaline granitic melts. Ilmenite-series granites are commonly related to sedimentary source rocks containing reducing agents such as organic carbon and, therefore, to peraluminous and S-type granites (Whalen and Chappell, 1988; Ishihara and Sasaki, 1989). By contrast, magnetite-series granites generally correspond to metaluminous or peralkaline I-type granites. Accordingly, the decrease in the Ce/Gd ratio that we observed only in monazites from ilmenite-series pegmatites can be attributed to significant fractionation of earlier-crystallized monazite in more peraluminous granitic magmas. In addition to monazite, crystallization of other LREE-rich minerals such as allanite and titanite, can potentially cause Ce/Gd fractionation, especially in high-Ca granitic magmas. Yet, the tendency of lower Th/U ratios in the monazites from ilmenite-series pegmatites (Fig. 4-4c) suggests that monazite played a more critical role in the host granitic magmas, because fractional crystallization of monazite, rather than allanite and titanite, should decrease the Th/U ratio in the residual melt due to its high partition coefficient of Th relative to U (Bea, 1996; Stepanov et al., 2012). In addition, fractional crystallization of other LREE- and Th-rich minerals, such as thorite-huttonite, could contribute to

produce the observed geochemical trends of ilmenite-series monazite, even though these minerals are generally much less abundant in granitic rocks than monazite (Bea, 1996).

The convex REE pattern with a peak of Sm for the ilmenite-series samples is linked with the fractionation of monazite and extremely differentiated silicic magma (Fig. 4-4d). The REE pattern with a discontinuity between LREE and MREE for highly evolved rocks such as granitic pegmatite has been reported since early 1980s (Goad and Čenrý, 1981; Christiansen et al., 1983; Mittlefeldt and Miller, 1983; Masuda et al., 1987). The large fractionation between LREE and MREE happens during the crystal fractionation of LREE-rich minerals (Mittlefeldt and Miller, 1983; Yurimoto et al., 1990) or loss of CO<sub>2</sub> vapor phase showing strong partition of LREE as compared to MREE (Wendlandt and Harrison, 1979). Yurimoto et al. (1990) quantitatively demonstrated that Rayleigh fractionation of monazite resulted in the LREE depletion relative to MREE of parent melt, assuming a peraluminous granite as an initial melt composition and that REE fractionation is governed only by monazite. The possible effect of CO<sub>2</sub> vapor phase might have been less significant for the studied samples, because carbonate minerals were not reported as co-existing mineral assemblages (Hoshino et al., 2010). Therefore, the unique REE pattern of pegmatic monazite was formed by the fractionation of monazite itself.

Similarly, the peraluminosity of granitic magma can greatly affect the HREE behavior, which would be reflected in the REE pattern of crystallized monazites. The peraluminous composition would result in efficient saturation and fractionation of

HREE-rich minerals of xenotime and garnet, as evidenced by their common occurrences in peraluminous granites (Bea, 1996; Casillas et al., 1995; Wark and Miller, 1993). This can account for the depletion of HREE in the monazites from ilmenite-series pegmatites (Fig. 4-4b). In addition to xenotime and garnet, zircon fractionation could contribute to HREE depletion in the residual melt due to its high partition coefficients for HREE relative to MREE. It is considered that the effects of zircon fractionation to be relatively minor in the studied samples, however, because there is little variation in La/Ce ratios in the analyzed monazites, despite the highly contrasting partitioning of these elements into zircon (Hanchar and van Westrenen, 2007). Hence, the fractionation from MREE to HREE in igneous monazites are interpreted to reflect the significance of the fractionation of monazite, xenotime, and garnet in the parental magma, which in turn provides constraints on the degree of differentiation and the source of the parental magma.

#### **4.4.2 Intra-grain compositional variations in the Masaki pegmatite monazite**

The monazite grain from the Masaki pegmatite consists of a dark core and a bright rim in the BSE image (Fig. 4-2m), and these domains have different chemical compositions. The rim had a more highly fractionated REE pattern (higher Gd/Lu) with a smaller Eu negative anomaly (higher Eu/Eu\*) and a higher Th/U ratio (Fig. 4-5). Despite these geochemical differences, the core and the rim had identical  $\epsilon_{\text{Nd}}$  (100 Ma) values within analytical uncertainty (Fig. 4-6). Note that our preliminary U-Pb dating showed no resolvable difference (>5 Ma) between the core and the rim, indicating that

the rim formed during a later stage of the pegmatite crystallization or, alternatively, during a hydrothermal alteration shortly after the crystallization. Yet, we consider the latter unlikely because hydrothermal monazites typically have very low Th contents (<~2 wt%) (Bea, 1996; Cabella et al., 2001; Rasmussen and Muhling, 2007; Williams et al., 2007).

The LREE-Th-U change from the core to the rim is essentially the opposite of the compositional evolution of ilmenite-series magma via fractionation of monazite and feldspars (Fig. 4-4). This may reflect that the magma from which the core crystallized underwent more fractionation of monazite and feldspars than the magma that formed the rim. The only exception is that the HREE depletion was more significant in the rim than in the core (Fig. 4-5c), corresponding to a differentiation trend between the ilmenite-series pegmatites and granites (Fig. 4-4b). The HREE trend suggests that the source magma of the rim suffered more fractionation of HREE-rich minerals, such as garnet, than did the core. Given that garnet can crystallize earlier from granitic magmas under high pressure and, to some extent, lower  $H_2O$  partial-pressure conditions (Clemens and Wall, 1981; Stern and Wyllie, 1981), the decoupling between the LREE-Th-U and HREE trends can be interpreted to reflect that the core crystallized from a magma differentiated at relatively shallow and/or  $H_2O$ -rich conditions, where garnet fractionation was suppressed. In addition, the homogeneous Nd isotope composition of the Masaki monazite grain indicates that the source magma of the core and rim had magmatic protoliths with similar crustal residence times. Overall, the

observed intra-grain compositional variations in the Masaki pegmatite monazite can be explained if the core and rim crystallized from distinct magmas that were originally derived from similar crustal rocks, which were subsequently differentiated through fractional crystallization under different conditions.

#### **4.4.3 Geochemical fingerprints of monazites in various rock types for a provenance indicator**

The study of detrital monazite has become more important for the investigation of continental crust evolution, as an effective complement to detrital zircon study. Detrital monazite can provide a more thorough record of tectonic events occurred in source terranes due to its crystallization during a wide range of metamorphic conditions and low-Ca granitic magmatism (Kusiak et al., 2006; Hietpas et al., 2010; Itano et al., 2016; Iizuka et al., 2017; Liu et al., 2017). As a result of its wide occurrence, however, interpretation of detrital monazite age spectra remains difficult. In the case of zircons, trace element data have been accumulated for grains of various origins, and their distinctive characteristics have been used as constraints on the provenance of detrital zircons (Hoskin and Ireland, 2000; Rubatto, 2002; Belousova et al., 2002). Given that the trace element composition of monazites reflects their formation environments such as co-existing minerals and the oxidation state, a geochemical indicator of source rock type can potentially be established for monazites.

In Fig. 4-7, a compilation of previously reported REE-Th-U data for monazites of various origins is shown to compare the data in this study. The acquisition of the trace element data for igneous monazite in this study has enabled the quantitative comparison of various origins for the first time.

The data obtained in this study reveal that the Eu negative anomalies of monazites from granitic rocks are generally larger than those of metamorphic and hydrothermal monazites, likely reflecting significant crystal fractionation of feldspars in their parental melts (Fig. 4-7a). Monazites from leucogranitic portions of migmatites also exhibit prominent negative Eu anomalies (Fig. 4-7a). Moreover, the granitic monazites tend to yield lower  $[Ce/Gd]_N$  values than metamorphic, hydrothermal, and migmatitic monazites, likely due to the significant fractionation of monazite in their parental melts.

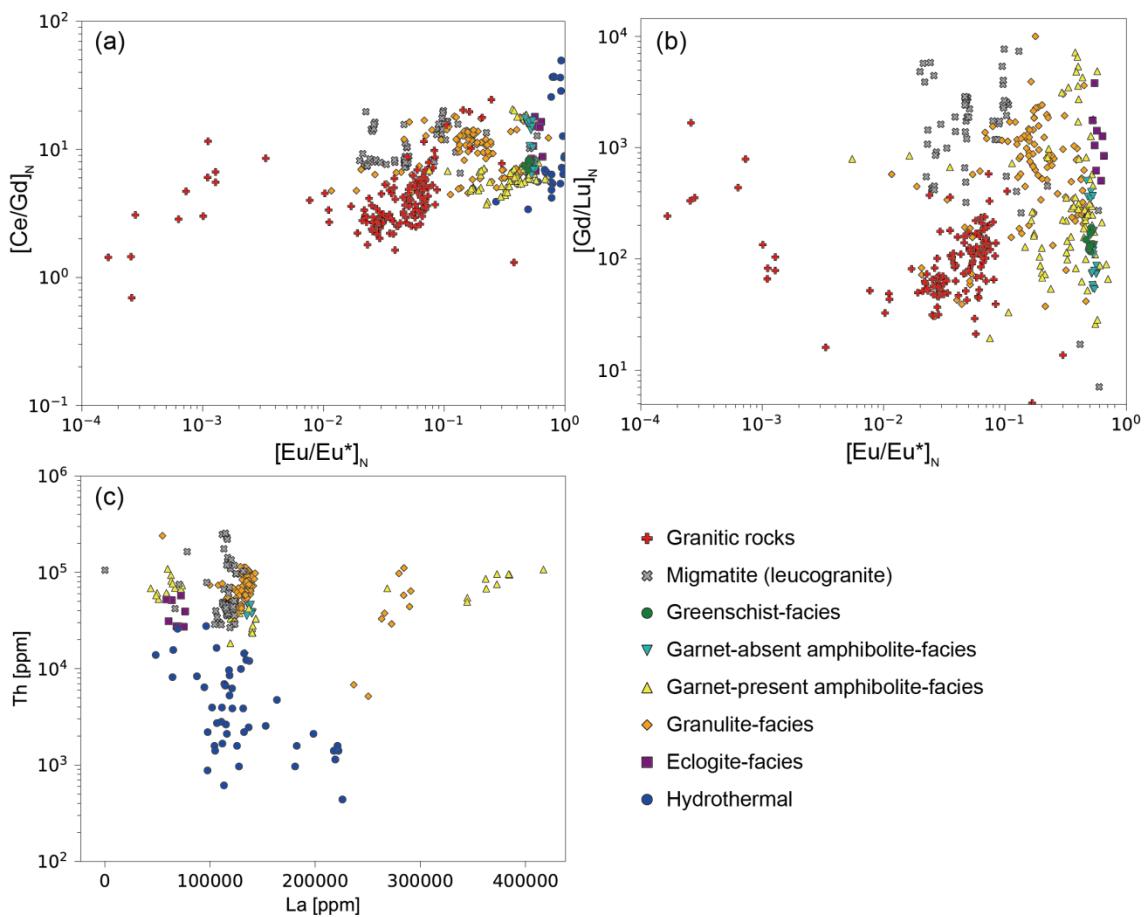


Fig. 4-7. Compilation of REE-Th-U systematics between igneous and metamorphic monazites by plotting (a)  $[Ce/Gd]_N$  and (b)  $[Gd/Lu]_N$  against  $[Eu/Eu^*]_N$  and (c) Th concentrations against La concentrations. The data for igneous monazites are from this study and Fisher et al. (2017), whereas those for metamorphic monazites are from Rubatto et al. (2006), Buick et al. (2010), Iizuka et al. (2010), Mottram et al. (2014), Holder et al. (2015), and Štípká et al. (2015); those for hydrothermal monazites are from Cebella et al. (2001), Poitrasson et al. (2000), Condon et al. (2005), and Rasmussen and Muhling (2007).

The compiled dataset for metamorphic monazites reinforces the view that its geochemical composition is highly dependent on the presence/absence of co-existing garnet. The monazites from low-grade metamorphic rocks display restricted H-REE depletion and negative Eu anomaly (Fig. 4-7b). During low-grade metamorphism, the growth of garnet is limited (e.g., Spandler et al., 2003) and, therefore, this limitation is responsible for the monazites with restricted H-REE depletion. In contrast, the monazite grains from garnet-bearing high-grade metamorphic rocks have more variable and generally higher  $[Gd/Lu]_N$  values than those from garnet-absent metamorphic rocks (Fig. 4-7b), as a result of the strong partitioning of HREE into garnet (Rubatto et al., 2006). Most detrital monazites with elevated  $[Gd/Lu]_N$  ( $>400$ ) display large Eu negative anomalies ( $[Eu/Eu^*]_N < 0.3$ ), suggesting their co-existence with not only garnet but also feldspar(s) (Fig. 4-7b). Some HREE-depleted grains with restricted Eu negative anomalies ( $[Eu/Eu^*]_N$  of  $\sim 0.5$ ) may be derived either from high-temperature metamorphic rocks where feldspars were destabilized during prograde metamorphism (Holder et al., 2015) or from mafic metamorphic rocks which contain feldspars with relatively low abundances.

Hydrothermal monazite shows unique geochemical features: a restricted negative Eu anomaly as compared to other origins (Fig. 4-7a) and a low-Th content (Fig. 4-7c). In terms of texture, hydrothermal monazites are characterized by the presence of many inclusions and anhedral shapes (Rasmussen and Muhling, 2007, 2009; Rasmussen et al., 2007).

Overall, the monazites from different source rock type show different geochemical signatures characterized by  $[Eu/Eu]_N$ ,  $[Gd/Lu]_N$ , and Th contents (Fig. 4-8). On the basis of the distribution of trace element data, I construct classification criteria for a provenance indicator of detrital monazite on the basis of distribution of the trace element data (Table 4-4). The distribution of  $[Eu/Eu^*]_N$  for igneous monazite are different from metamorphic monazites (Fig. 4-8a). Approximately 85% of monazite grains with  $[Eu/Eu^*]_N$  of  $<0.1$  are of igneous origin. Furthermore, the monazite grains with  $[Gd/Lu]_N$  of  $>400$  are mostly derived from garnet-present metamorphic rocks: the data for garnet-present metamorphic origin account for  $>90\%$  of the grain with  $[Gd/Lu]_N$  of  $>400$  and  $[Eu/Eu]_N$  of  $>0.1$  (Fig. 4-8b). It is therefore reasonable to expect that the detrital monazites highly depleted in HREEs ( $[Gd/Lu]_N$  of  $>400$ ) with restricted negative Eu anomalies are originated mainly from garnet-rich metamorphic rocks with typically medium- to high-grades. On the other hand, the grains, showing  $[Gd/Lu]_N$  of  $<400$  and  $[Eu/Eu]_N$  of  $<0.1$  is likely to be derived from garnet-poor or garnet-rich metamorphic rocks. Indeed, the geochemical features of monazites from garnet-absent metamorphic rocks, such as low-grade metamorphic rocks, are restricted HREE-depletion and Eu anomaly (Fig. 4-8a, b). Hydrothermal monazite is characterized by so low-Th content as compared to other origins (Fig. 4-8c). The grains with low-Th content ( $<10,000$  ppm) can be identified as hydrothermal origins. Thus, the combination of a significant negative Eu anomaly, depletion of HREE relative to MREE, and depletion in Th can be regarded as a strong indicator of source rock types.

Table 4-4. Geochemical criteria for an indicator of source rock type

Igneous	Metamorphic		Hydrothermal
	Garnet-poor to Garnet-rich	Garnet-rich	
$[\text{Eu/Eu}^*]_N < 0.1$ $[\text{Gd/Lu}]_N < 400$	$[\text{Eu/Eu}^*]_N > 0.1$ $[\text{Gd/Lu}]_N < 400$	$[\text{Gd/Lu}]_N > 400$	Th content $< 10^4 \text{ ppm}$

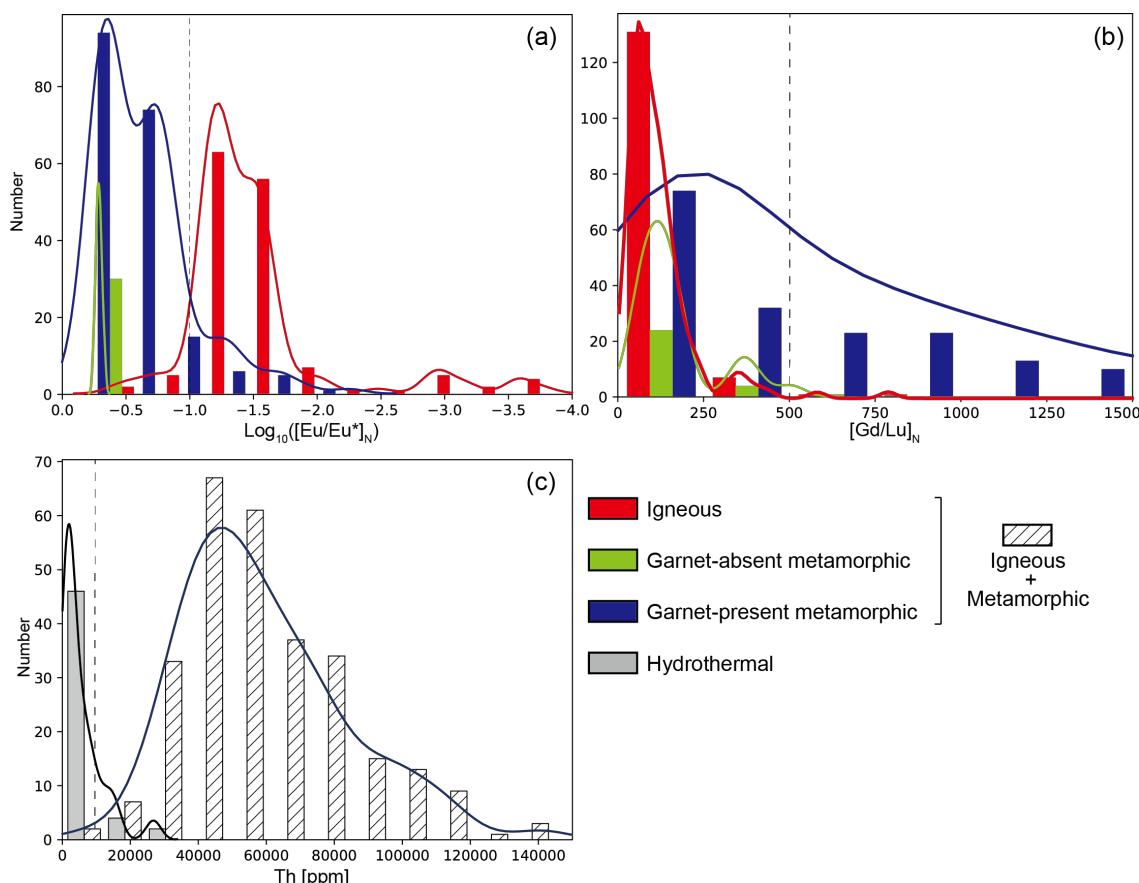


Fig. 4-8. Histograms of geochemical variations in (a)  $[\text{Eu/Eu}^*]_N$ , (b)  $[\text{Gd/Lu}]_N$ , and (c) Th contents for igneous, garnet-absent and garnet-present metamorphic, and hydrothermal monazites. The bold lines represent relative probability density for each genesis.

## 4.5 Conclusion

The REE-Th-U systematics of monazites reflects different differentiation schemes between magnetite-series and ilmenite-series granitic rocks. In ilmenite-series granitic rocks, monazites from pegmatites show larger negative Eu anomalies, lower LREE/MREE ratios, and higher MREE/HREE ratios than those from granites. These geochemical variations are related to significant fractional crystallization of feldspars, monazite, xenotime, and garnet during differentiation in relatively reduced and peraluminous magmas. In magnetite-series granitic rocks, by contrast, monazites from pegmatites and granites exhibit similarly fractionated REE patterns with variable negative Eu anomalies. The negative Eu anomalies are smaller compared to those observed in ilmenite-series samples. These features are considered to reflect the suppression of monazite, xenotime, and garnet fractionation and of Eu incorporation into fractionating feldspars in relatively oxidized and non-peraluminous magmas. The monazite from the Masaki ilmenite-series pegmatite shows intra-grain variations in REE–Th–U compositions, but homogeneous Nd isotope composition. The finding may reflect that the core and rim crystallized from magmas that were derived from similar crustal source rocks, but underwent fractional crystallizations under distinctive differentiation conditions. The REE-Th-U compositions of these igneous monazites are variable, but distinct from those of metamorphic and hydrothermal monazites, most notably in their larger negative Eu anomalies. Thus, REE-Th-U systematics shows promise as a provenance indicator for detrital monazites.

## **5. Application to detrital monazites from major African rivers: Constraints on the timing and nature of the Pan-African Orogeny**

### **5.1 Introduction**

The African continent comprises five Archean to Paleoproterozoic cratons: the West African Craton, Saharan Metacraton, Congo Craton, Tanzania Craton and Kalahari Craton (Fig. 1). These cratons are surrounded by orogenic belts, formed during the late Neoproterozoic to the earliest Paleozoic time (e.g., Meert, 2003; Kröner and Stern, 2004; Begg et al., 2009). The so-called Pan-African Orogeny is considered to have resulted from collisions of the cratons to form the Gondwana supercontinent and the African continent was located in the central area of the supercontinent (Gray et al., 2008; Meert and Lieberman, 2008). Thus, knowledge of the Pan-African Orogeny is a key to understanding the development of the African continent and Gondwana supercontinent. To better constrain the timing and nature of the Pan-African Orogeny, we have determined U–Pb ages and chemical compositions of ca. 500 detrital grains of monazite, a light rare earth element (LREE) phosphate, from the five largest African rivers: the Nile, Niger, Congo, Zambezi and Orange Rivers.

Detrital minerals in sediments of large rivers have been eroded from exposed crust in extensive drainage basins, which consists of sedimentary, igneous and metamorphic rocks. Since zircon and monazite are mechanically robust and can survive sedimentary recycling (Veizer and Jansen, 1979, 1985; Goldstein et al., 1984; Campbell et al., 2005),

these minerals in river sands would be partially derived from the basements that are currently overlain by a sedimentary cover or have been eroded away and thus inaccessible. Moreover, single grains of zircon and monazite can be precisely dated by U–Pb isotopes due to their high U–Th and low non-radiogenic Pb contents. Thus, detrital zircons and monazites from large rivers offer a unique means of elucidating the timing of major magmatic and/or metamorphic events in a continental scale that might have been overlooked by the researches based on the exposed geology (e.g., Ross et al., 1991; Goldstein et al., 1997; Yokoyama and Zhou, 2002; Rino et al., 2004; Hietpas et al., 2010; Moecher et al., 2011). This is particularly relevant for the African continent where a large area of the basement is covered by Phanerozoic sediments (Fig. 5-1).

Recently, Iizuka et al. (2013) reported U–Pb age and Hf–O isotope data for detrital zircons from the five largest African rivers. The data revealed that granitoid magmatism during the Pan-African Orogeny played a significant role in the formation and reworking of the African continental crust. Yet, because they analyzed detrital zircons of magmatic origin rather than metamorphic origin, the timing of metamorphism related to the Pan-African Orogeny is poorly constrained by the detrital zircon record.

Monazite commonly occurs as an igneous mineral in low-Ca peraluminous felsic rocks (e.g., Overstreet, 1967; Parrish, 1990; Xie et al., 2006; Kelts et al., 2008; Chattopadhyay et al., 2015) and a metamorphic mineral in a wide range of metamorphic rocks (e.g., Overstreet, 1967; Smith and Barreiro, 1990; Buick et al., 2010; Štípká et al., 2015). This is in contrast to zircon which is commonly crystallized from a wide range of

felsic magmas (Chappell and White, 2001; Kelts et al., 2008) and secondarily formed during high-grade metamorphism (Rubatto et al., 2001; Rubatto, 2002; Harley and Nandakumar, 2014). The restricted occurrence of igneous monazite in high-Ca felsic rocks is due to the decomposition of monazite to form allanite (Lee and Bastron, 1967; Watt and Harley, 1993; Kelts et al., 2008). Importantly, the Pb diffusion in monazite crystal is as slow as that in zircon, so the monazite U–Th–Pb system can potentially be preserved thorough high-grade metamorphism (Schaltegger et al., 1999; Rubatto et al., 2001; Cherniak et al., 2004; Schmitz and Bowring, 2004; McFarlane and Harrison, 2006). Thus, monazite is suitable for high-precision dating of peraluminous magmatic and low- to high-grade metamorphic events and detrital monazite is a powerful tool in provenance characterization (Ross et al., 1991; Yokoyama and Zhou, 2002). Indeed, Hietpas et al. (2010, 2011) compared the U–Pb age distributions of detrital monazites and zircons from the French Broad River for which the geology in the drainage basin is well known and demonstrated that the detrital monazites record metamorphic events more accurately than the detrital zircons. In addition, the genesis of monazite can be constrained by geochemical analysis because some elemental abundances in monazite highly depend on the assemblages of co-existing minerals via element partitioning (Hermann and Rubatto, 2003; Rubatto et al., 2006; Mottram et al., 2014; Štípská et al., 2015).

In this contribution, we explore the geological significance of detrital monazite in the five African river systems by combining the U–Pb ages and geochemical data and also

by comparing with previously obtained detrital zircon U–Pb age peaks (Iizuka et al., 2013). Along with the results obtained in the present and previous studies, we discuss new insights into the Pan-African Orogeny and implications for the development of the African continent.

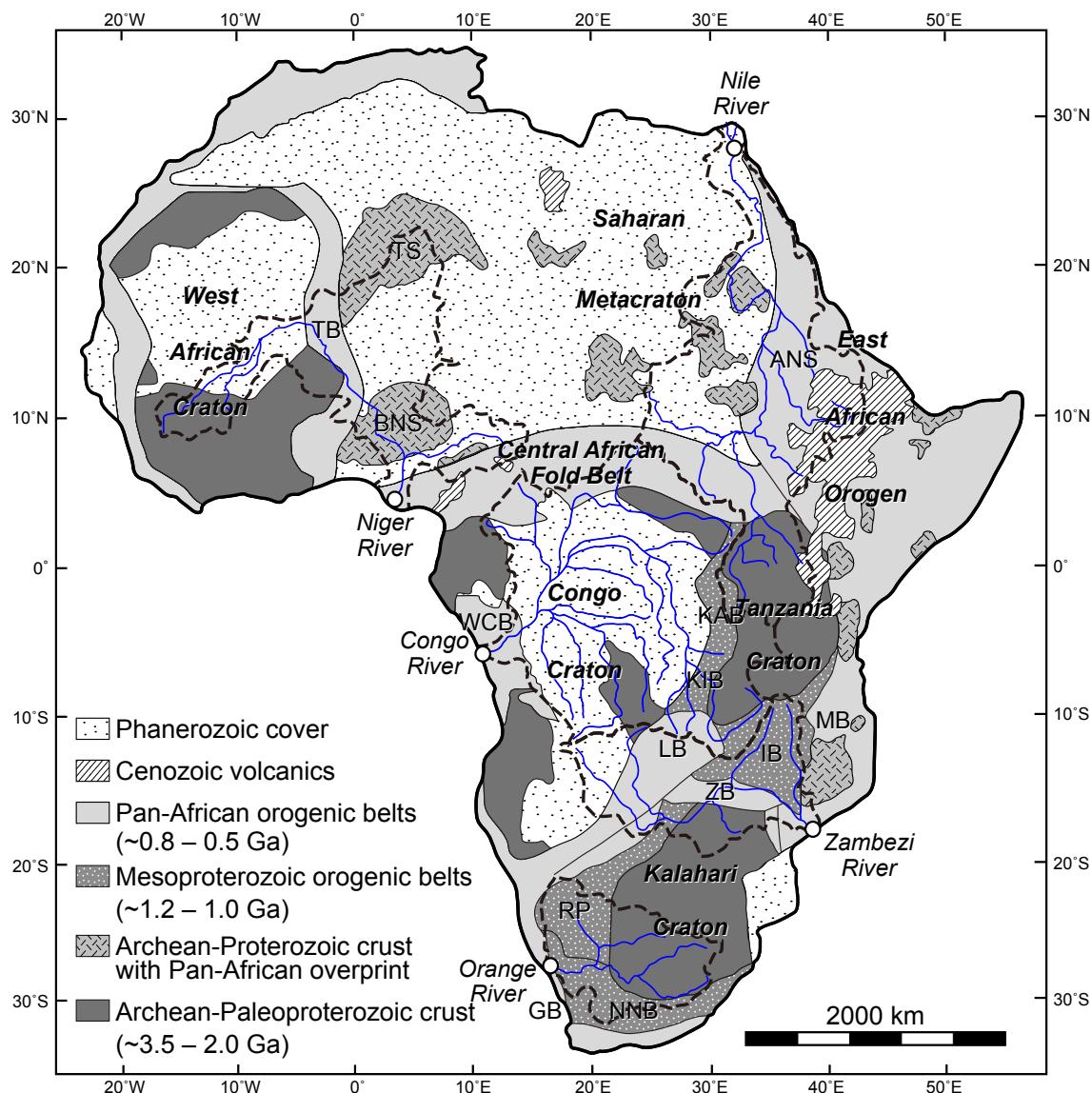


Fig. 5-1. Simplified geological map of Africa (modified after Kampunzu and Popoff, 1991, Tack et al., 2001, Begg et al., 2009 and Iizuka et al., 2013), with sampling locations (dots), major tributaries (blue lines), and drainage basin limits (broken lines) of the Niger, Nile, Congo, Zambezi and Orange Rivers. Abbreviations are as follows: TS, Tuareg Shield; TB, Trans-Saharan Belt; BNS, Benin-Nigeria Shield; ANS, Arabian-Nubian Shield; WCB, West Congo Belt; KAB, Karagwe-Ankole Belt; KIB, Kibaran Belt; Irumide Belt; MB, Mozambique Belt; ZB, Zambezi Belt; LB, Lufilian Belt; RP, Rehobothian Province; NNB, Namaqua-Natal Belt; GB, Gariep Belt.

## 5.2 Regional geology

The drainage basins of the Nile, Niger, Congo, Zambezi and Orange Rivers collectively occupy 40% surface area of the continental Africa (Fig. 1). The regional geology of the African continent is summarized below, with emphasis on the timing and nature of major orogenic events that occurred in the basement terranes of the drainage basin areas.

The Nile River runs through the north of the Tanzania Craton, the east of the Saharan Metacraton and the north of the East African Orogen. The Tanzania Craton mainly comprises Archean granitoids and greenstone belts (Leggo, 1974; Manya and Maboko, 2003; Manya et al., 2006; De Waele et al., 2008). U–Pb dating of zircons and monazites from the granitoids and metamorphic rocks in the north and east of the Tanzania Craton indicated that the basement rocks underwent multiple metamorphic events at ~2.6 Ga (Johnson et al., 2003), ~640–620 Ma (Sommer et al., 2003; Appel et al., 2005) and ~550 Ma (Cutten et al., 2006). The Saharan Metacraton was highly remobilized during the Pan- African Orogeny but is recognized as a craton by geological, geochronological and isotopic characteristics (Abdelsalam et al., 2002; Liégeois et al., 2013). Because the Saharan Metacraton is widely covered by Phanerozoic sediments, its basement geology especially in the central part is not known in detail. Yet, granitoids, high-grade metamorphic rocks and migmatites are patchily exposed (Abdelsalam et al., 2002; Küster et al., 2008; Shang et al., 2010; Liégeois et al., 2013) and zircon U–Pb dating indicated that the granitoid magmatism and high-grade metamorphism occurred at ca.

600 Ma (Shang et al., 2010). Furthermore, detrital zircon from sandstones from Libya show age peaks at 2750– 2500, 2200–1750, 1060–920 and 720–530 Ma (Meinhold et al., 2011). The Saharan Metacraton faces the East African Orogen that was formed by Neoproterozoic accretionary and collision type orogeny within the Gondwana (e.g., Stern, 1994; Meert, 2003; Johnson et al., 2011; Fritz et al., 2013; Johnson, 2014). The Arabian–Nubian Shield in the northern part of the East Africa Orogen is characterized by Neoproterozoic ophiolites and juvenile arcs that emplaced between 850 and 620 Ma and underwent granulite-facies metamorphism at 650–620 Ma, followed by granitic magmatism related to post-orogenic extension between 610 and 560 Ma (Johnson et al., 2011; Fritz et al., 2013; Johnson, 2014).

The Niger River flows through the west of the Saharan Metacraton and parts of the West African Craton, Trans-Saharan Belt and Central African Fold Belt. In the west area of the Saharan Metacraton, the Tuareg and Benin-Nigeria Shields are exposed. The Tuareg Shield mainly consists of Archean and Paleoproterozoic terranes and Neoproterozoic oceanic terranes, and was intruded by syn-tectonic Pan-African granites between 870 and 580Ma (e.g., Bertrand and Caby, 1978; Bertrand et al., 1986; Black et al., 1994; Paquette et al., 1998; Caby, 2003). The Benin-Nigeria Shield hosts Archean and Paleoproterozoic granitoids that were intruded by 670–580 Ma granites (Dada, 1998; Ferré et al., 2002; Ekwueme and Kröner, 2006). The West African Craton is made up of the northern and southern Archean–Paleoproterozoic domains separated by sedimentary basin. The Niger River basin covers the southern domain comprising an

Archean terrane in the west and Paleoproterozoic (2200–2100 Ma) terrane in the east (Boher et al., 1992; Potrel et al., 1996; Lombo, 2010; Parra-Avila et al., 2015). The Trans-Saharan Belt is a collision zone between the West African Craton and Saharan Metacraton and comprises Neoproterozoic (~650–600 Ma) granitoids and high-grade metamorphic rocks (Ferré et al., 2002; Caby, 2003; Liégeois et al., 2003; Agbossoumondé et al., 2007; Goodenough et al., 2014). The initiation of this collision was diachronous (Kröner and Stern, 2004). The central African Fold Belt is a collisional belt between the Saharan Metacraton and Congo Craton and comprises Paleoproterozoic basement gneiss and Neoproterozoic granitoids and gneiss (Toteu et al., 2004; Van Schmus et al., 2008; Bouyo Houketchang et al., 2013, 2015). In the western part of the orogenic belt, where the Niger River runs, granulites and granitoids are exposed (Ferré et al., 2002; Toteu et al., 2004; Bouyo Houketchang et al., 2013, 2015) and these rocks are considered to have been formed by crustal thickening during the collision between the two cratons at ~640–600 Ma (Toteu et al., 2004; Van Schmus et al., 2008; Bouyo Houketchang et al., 2009, 2013, 2015).

The Congo River drains through two cratons (the Congo Craton and southern Tanzania Craton) and four orogenic belts (the southern Central African Fold Belt, Karagwe–Ankole Belt, Kibaran Belt and West African Belt). The Congo Craton is widely covered by Phanerozoic sediments and its basement outcrops only in four shields along the edge of the craton. The basement rocks mainly consist of Paleo- and Neoarchean tonalites-trondhjemites-grano diorites (TTGs) and metamorphic rocks

intruded by Neoarchean and Paleoproterozoic felsic rocks (Walraven and Rumvegeri, 1993; Tchameni et al., 2000, 2001; Shang et al., 2004; Batumike et al., 2009). U–Pb dating of detrital zircons from the central part of the craton (Batumike et al., 2009) revealed a significant Pan- African age peak (850–580 Ma), but little Pan-African granitoids have been found within the craton. The southern part of the Tanzania Craton includes Paleoproterozoic high-grade metamorphic rocks (ca. 2.0 Ga), syn- and post-tectonic granitoids (2.0–1.8 Ga) and metasedimentary rocks (Reddy et al., 2003; Collins et al., 2004; Sommer et al., 2005). The Congo Craton is mantled by the Central African Fold Belt to the north. The southern part of the belt, covered by the Congo River, is characterized by continuous granulite belts with ages of ~640–600 Ma (Pin and Poidevin, 1987; Toteu et al., 2004; Owona et al., 2012; Bouyo Houketchang et al., 2013). The Karagwe–Ankole and Kibaran Belts are located between the Congo and Tanzania Cratons (Tack et al., 2010). The Karagwe–Ankole Belt consists of Archean–Paleoproterozoic basements, whereas the Kibaran Belt consists of Mesoproterozoic basements (Kokonyangi et al., 2004; Tack et al., 2010; Fernandez-Alonso et al., 2012; Koegelenberg et al., 2015). The West Congo Belt is located in the west of the Congo Craton and near the Congo River mouth. The Paleoproterozoic basement of the West Congo Belt is unconformably overlain by Neoproterozoic rocks (Tack et al., 2001). Zircon U–Pb dating indicated the main deformation and metamorphism of the basement rocks occurred at ~570–540 Ma, likely resulting from the collision between the Congo

Craton and São Francisco Craton of Brazil (Tack et al., 2001; Frimmel et al., 2006; Monié et al., 2012).

The Zambezi River runs through orogenic belts among the Tanzania, Congo and Kalahari Cratons: the Irumide, Lufilian, Zambezi and Mozambique Belts. The Irumide Belt consists of 2.7 and 2.0– 1.9 Ga basement rocks and overlying 1.85 Ga supracrustal rocks, which were intruded by various granitoids and underwent peak metamorphism at 1.1–1.0 Ga (Johnson et al., 2005, 2007a; De Waele et al., 2006, 2009). The Lufilian Belt has a complex mass of Paleoproterozoic granitoids and gneisses, which was overlain by Neoproterozoic to Cambrian volcano-sedimentary sequences and intruded by younger granites (Key et al., 2001; John et al., 2004; Johnson et al., 2005). Zircons from the granite intrusions and metamorphic monazites from the gneisses yielded U–Pb ages of ~565– 530 Ma (Hanson et al., 1993) and ~530 Ma (John et al., 2004), respectively. The ~530 Ma magmatism and metamorphism are considered as results of the collision between the Kalahari and Congo Cratons (Hanson et al., 1993; Porada and Berhorst, 2000; John et al., 2004; Johnson et al., 2005; Eglinger et al., 2014). The Zambezi Belt has Mesoproterozoic basements, comprising gneisses and granites that are unconformably overlain by metamorphic and sedimentary rocks. The supracrustal rocks were intruded by granitic rocks at ~820 Ma, possibly related to the continental rifting causing the dispersal of Rodinia (Johnson et al., 2007b). In the Zambezi and Lufilian Belts, eclogites are exposed and yielded Sm–Nd ages of ~600 Ma (John et al., 2003). Besides, metamorphic zircon ages of ca. 530Ma have been reported from these belts

(Goscombe et al., 2000; John et al., 2004). Moreover, metasedimentary and metabasic rocks in the Lufilian belt yielded a garnet-whole rock Lu–Hf isochrones age of ca. 530 Ma (Eglinger et al., 2016). These results suggest the collision between the Congo and Kalahari Cratons at ca. 530 Ma (John et al., 2004). The southern Mozambique Belt continues eastward from the Zambezi Belt. In the Mozambique Belt, the two distinct age groups are recognized. The northeastern domain is characterized by granulite-facies metamorphism and coeval deformation around ~650–610 Ma; by contrast, the southeastern domain is characterized by granulite- to amphibolite-facies metamorphism and granitic magmatism later than ~550 Ma (Bingen et al., 2009; Thomas et al., 2010; Ueda et al., 2012a,b; Fritz et al., 2013). The former is attributed to the collision between the southeast Africa and the fragments of East Gondwana (Meert, 2003; Collins and Pisarevsky, 2005). The latter is considered to represent the timing of crustal thickening and subsequent post-collisional extension possibly due to asthenosphere upwelling following lithospheric delamination (Jacobs et al., 2008a; Viola et al., 2008; Ueda et al., 2012a,b).

The Orange River runs through the Kalahari Craton, Rehobothian Province and Namaqua–Natal Provinces and Gariep Belt. The Kalahari Craton mainly comprises granitoids with intrusion ages between 3.67 and 2.55 Ga (Poujol et al., 2003; Singletary et al., 2003; Schmitz and Bowring, 2004; Jacobs et al., 2008b), which are overlain by Archean Witwatersrand sedimentary rocks (2.95– 2.71Ga), Paleoproterozoic sedimentary rocks of the Transvaal Supergroup (2.55–2.08 Ga) and Karoo sedimentary

rocks (300– 120 Ma) (Catuneanu et al., 2005; Eriksson et al., 2006; Zeh et al., 2016).

The northern and western margins of the craton are further covered by Phanerozoic sediments (Jacobs et al., 2008b). The Namaqua–Natal Province contains 1.2–1.0 Ga gneisses and granitic intrusions (Thomas et al., 1994; Robb et al., 1999; Cornell and Pettersson, 2007) and is considered to be accreted to the Kalahari Craton by ~1100–1050Ma during the Rodinia supercontinent amalgamation (Jacobs et al., 2003, 2008b; McCourt et al., 2006). The Rehoboth Province includes late Paleoproterozoic (~1.9–1.7 Ga) gneisses and granites (Van Schijndel et al., 2014) and Mesoproterozoic (~1.2–1.0Ga) meta-sedimentary, volcanic and intrusive rocks (Becker et al., 2005, 2006). The late Paleoproterozoic thermal events are interpreted as resulting from the collision between the Rehoboth Province and Kalahari Craton (Jacobs et al., 2008b). The Rehobothian Province is rimmed by the Gariep Belt which consists mainly of Neoproterozoic sedimentary sequences and oceanic crust. The Gariep Belt further underwent low-grade metamorphism, followed by ~500Ma granite intrusions (Frimmel, 2000). The timing of the metamorphism is determined to be 570–540Ma by hornblende Ar–Ar dating (Frimmel and Frank, 1998; Gray et al., 2008), suggesting the collision between the Kalahari Craton and South America at that time (Frimmel and Frank, 1998; Kröner and Stern, 2004; Gray et al., 2008).

Overall, two main periods are recognizable for Pan-African orogenic events in the African continent (Fig. 5-2): the older one at 640–600 Ma corresponding to the period

of the East African Orogeny and the younger one at 570–530 Ma often referred to as the Kuungan Orogeny (e.g., Meert, 2003; Fritz et al., 2013) (Fig. 5-2).

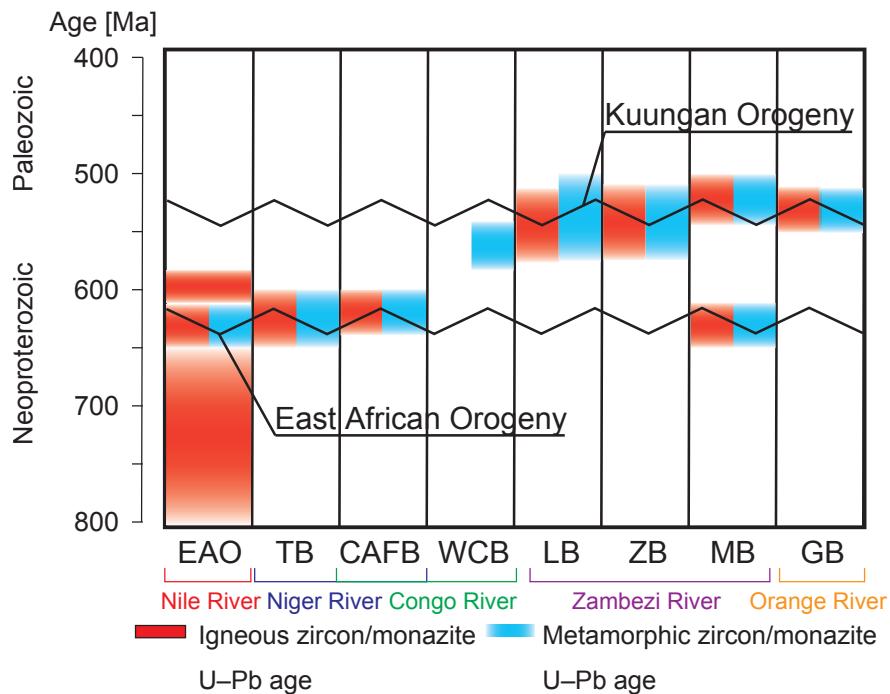


Fig. 5-2. Simplified summary of the major magmatic and metamorphic events/periods in the different orogenic belts related to the Pan-African Orogeny, ranging from 800 Ma to 400 Ma discussed in this paper. Abbreviations and references are as follow: EAO = East African Orogen, Stern (1994); Meert (2003); Stern et al. (2010); Johnson et al. (2011); Fritz et al. (2013); TB = Trans-Saharan Belt, Ferré et al. (2002); Caby et al. (2003); Liégeois et al., 2003; Agbossoumondé et al. (2007); Goodenough et al. (2014); CAFB = Central African Fold Belt, Pin and Poidevin (1987); Toteu et al. (2004); Owona et al. (2012); Bouyo Houketchang et al. (2013); WCB = West Congo Belt, tack et al. (2001); Frimmel et al. (2006); Monié et al. (2012); LB = Lufilian Belt, Hanson et al. (1993); Porada and Berhorst (2000); John et al. (2004); Eglinger et al., 2016; ZB = Zambezi Belt, Goscombe et al. (2000); John et al. (2002); John et al. (2004); Johnson et al. (2005); Grantham et al. (2013); MB = Mozambique Belt, Bingen et al. (2009); Fritz et al. (2013); GB = Gariep Belt, Frimmel and Frank (1998); Frimmel (2000); Gray et al. (2008).

### **5.3 Samples and analytical methods**

The river sand samples were collected at or near the mouths of the Nile (NIL), Niger (NGR1), Congo (CNG2), Zambezi (ZMB2) and Orange (ORG2) Rivers (Fig. 5-1, Table 5-1). These samples were previously used for zircon U–Pb dating and Hf (–O) isotopic studies (Iizuka et al., 2010a, 2013). For this study, monazite grains were newly concentrated from the river sand samples using the conventional magnetic and heavy liquid separation techniques. The analyzed monazite grains were randomly hand-picked from the aliquots of monazite concentrates to reduce sampling bias and mounted in epoxy resin. Before analysis, each grain was investigated by back-scattered electron (BSE) imaging to check internal zoning structures and the presence of inclusions. The BSE images were obtained using a JEOL JSM-7000F at the Department of Earth and Planetary Science, The University of Tokyo. An accelerating voltage of 10 kV, an emission current of 20 nA, and a working distance of 10 mm were used.

Table 5-1. Summary of the samples used in this study

Sample	River	Sample locality
NGR1	Niger	6°08'05"S, 6°45'19"E
NIL	Nile	Near Cairo City
CNG2	Congo	40°17'08"S, 15°27'29"E
ZMB2	Zambezi	17°48'33"S, 35°23'39"E
ORG2	Orange	28°30'25"S, 16°37'07"E

The U–Pb ages were measured with an ELEMENT-XR inductively coupled plasma-mass spectrometer (ICP-MS) attached to a 200 nm femtosecond laser ablation (LA) system at the Department of Solid Earth Geochemistry, Japan Agency for Marine-Earth Science and Technology. We measured  $^{202}\text{Hg}$ ,  $^{204}(\text{Pb} + \text{Hg})$ ,  $^{206}\text{Pb}$ ,  $^{207}\text{Pb}$ ,  $^{208}\text{Pb}$ ,  $^{232}\text{Th}$  and  $^{238}\text{U}$  using secondary multiplier with the counting mode. Because signals of  $^{232}\text{Th}$  as well as  $^{208}\text{Pb}$  during most sample analyses exceeded the maximum dynamic range of the counting mode, however, we report only U–Pb ages in this paper. For laser ablation sampling, a rotation raster ablation protocol (Kimura et al., 2011, 2015) was used to reduce elemental fractionation due to defocusing. In this protocol, a 10  $\mu\text{m}$  laser beam with  $\sim$ 6 J/cm was rotated along a circumference of a circle with a 5  $\mu\text{m}$  radius at a velocity of 2.1  $\mu\text{m s}^{-1}$  and 3 Hz, resulting in a  $\sim$ 20  $\mu\text{m}$  ablation pit. Each analysis consists of 30 s gas blank measurements and following ca. 90 s with laser ablation sampling. These measurements were carried out using a time resolved analytical (TRA) procedure, in which a temporal change in signal intensity is monitored to check a possible age variation within the ablation pit. Unknown samples were analyzed in runs of 6–10 analyses, bracketed by 2 analyses of the reference monazite 44069. The results of the bracketing analyses of the reference monazite 44069 ( $424.9 \pm 0.4$  Ma; Aleinikoff et al., 2006) were used to correct for the instrumental mass bias and Pb/U fractionation. Analytical uncertainties of each spot analysis combine the internal precisions (2 SE: two-standard error) and the reproducibility (2 SD: two-standard deviation) of the bracketing standard analyses, added in quadrature. Common Pb was

corrected using the  $^{204}\text{Pb}$  method (Williams, 1998) when the  $^{204}\text{Pb}$  signal was detectable (higher than 3 SD of background measurements); otherwise no common Pb correction has been made. The precision and accuracy of our monazite U–Pb dating were evaluated by analyzing the monazite Manangotry and 16-F-6 standards, whose U–Pb ages were previously determined by the isotope dilution-thermal ionization mass spectrometry ( $555 \pm 2$  Ma for Manangotry and  $2842.9 \pm 0.3$  Ma for 16-F-6, Horstwood et al., 2003; Simonetti et al., 2006). The mean U–Pb ages obtained for these standards during the course of this study are consistent with the previously reported values:  $^{206}\text{Pb}/^{238}\text{U}$  age of  $557 \pm 4$  Ma and  $^{207}\text{Pb}/^{206}\text{Pb}$  age of  $568 \pm 24$  Ma for Manangotry ( $n=18$ ) and  $^{206}\text{Pb}/^{238}\text{U}$  age of  $2823 \pm 20$  Ma and  $^{207}\text{Pb}/^{206}\text{Pb}$  age of  $2849 \pm 7$  Ma for 16-F-6 ( $n = 40$ ) (Appendix Table S5-1).

Abundances of Si, P, Ca, Y, La, Ce, Nd, Pr, Sm, Th and U of monazite grains were determined by field emission-electron probe microanalysis (FE-EPMA) using a JEOL JXA-8530F at the Department of Earth and Planetary Science, The University of Tokyo. We used the following standard materials: MnSiO<sub>4</sub> (Si, Ka), CaSiO<sub>3</sub> (Ca, Ka), LaPO<sub>4</sub> (P, Ka; La La) Y<sub>3</sub>Al<sub>5</sub>O<sub>12</sub> (Y, La), CePO<sub>4</sub> (Ce, La), PrPO<sub>4</sub> (Pr, L $\beta$ ), NdPO<sub>4</sub> (Nd, La), SmPO<sub>4</sub> (Sm, L $\beta$ ), ThO<sub>2</sub> (Th, Ma) and UO<sub>2</sub> (U, M $\beta$ ). The operating conditions were a beam diameter of 5  $\mu\text{m}$ , a probe current of 30 nA, and an accelerating voltage of 25 kV. The FE-EPMA analytical pits were placed within the same zones dated by LA-ICP-MS, closed to the produced laser pits.

Geochemical analyses of monazite grains were further performed using a CETAC LSX-213 G2+ Nd:YAG LA system coupled with an iCAP Q ICP-MS at the Department of Earth and Planetary Science, The University of Tokyo. The following elements were measured: Ca, Sr, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Th and U. The LA system is equipped with the active two- volume HelExTM Cell (Eggins et al., 2005), in which the effective cell volume is so small ( $\sim 2.5$  cm $^3$ ) that aerosol deposition around the ablation pit is minimized. The data were obtained from  $\sim 20$  lm ablation pits with a laser fluence of 20 J/m $^2$ , repetition rate of 3 Hz, cooling gas of 14 L/min and sampling gas of  $\sim 0.9$  L/min. The laser ablation site was placed on the pit for the FE-EPMA analysis next to the pit for the U–Pb dating. All analyses consist of 20 s baseline measurements and following ca. 40 s measurements with laser ablation sampling, using the TRA procedure. Six to eight analyses of unknown samples were bracketed by four analyses of NIST SRM 610 glass. The Ce concentrations determined by FE-EPMA were used as an internal standard. The other element concentrations were determined using the sensitivity factors relative to Ce obtained from the results of NIST SRM 610 glass analyses (Pearce et al., 1997). Because monazite is highly enriched in LREEs relative to heavy-REEs (HREEs), oxide interferences of L-REEs on H-REEs can be problematic. We corrected for the oxide interferences by monitoring  $^{232}\text{Th}^{16}\text{O}^+/\text{Th}^+$  (typically  $\sim 0.5\%$ ) and  $^{238}\text{U}^{16}\text{O}^+/\text{U}^+$  (typically  $\sim 0.4\%$ ) and using the correlation between oxide production rate and oxide bond energy (Aries et al., 2000; Kent and Ungerer, 2005).

## 5.4 Results

The internal structures (e.g., chemical zoning) and U–Pb isotopic data of detrital monazites are provided in Appendix Table S5-2. The internal structures were classified into the following four types (the number in parenthesis indicates the proportion of each type of grains): (1) core/rim structure (10%), (2) patchy structure (30%), (3) mosaic structure (16%) and (4) unzoned (44%) (Fig. 5-3). We defined the mosaic structure as a fine-scale zoning structure relative to analytical spatial resolution of U–Pb dating. Among 523 detrital monazites analyzed for U–Pb isotopes, a grain (CNG2-mnz-70) cannot be dated because of the extremely low U content and this is not included in the discussion below. The U–Pb data for detrital monazites of each river (Used data in Table S5-2) are plotted on Tera-Wasserburg concordia diagram (Fig. 5-4). Some grains yielded highly discordant data, which could be attributed to partial Pb-loss during metamorphism and to mixing analysis of multiple age domains. Because, inaccurate U–Pb ages would result in artificial age peaks, we have used only the data that yield either concordant U–Pb ages within analytical uncertainty or those with <5% discordance for the following discussion. For all studied rivers, approximately 70% of the dated monazite grains satisfy the criteria and the results are shown in the  $^{206}\text{Pb}/^{238}\text{U}$  age histograms (Fig. 5-5).

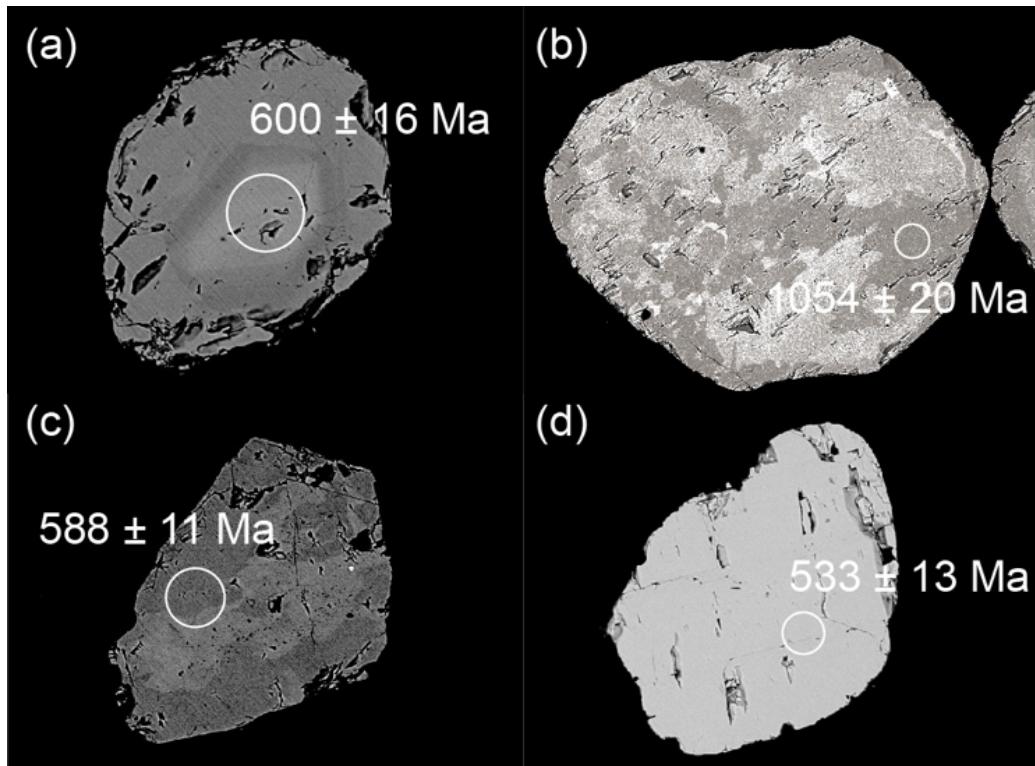


Fig. 5-3. Representative BSE images of detrital monazites from the African studied rivers. (a) NGR1-mnz-4; core-rim structure, (b) ORG2-mnz-105; patchy structure, (c) NGR1-mnz-89; mosaic structure and (d) NGR1-mnz-8; unzoned. White circles represent analytical sites with  $\sim 20 \mu\text{m}$  diameters for U–Pb dating.

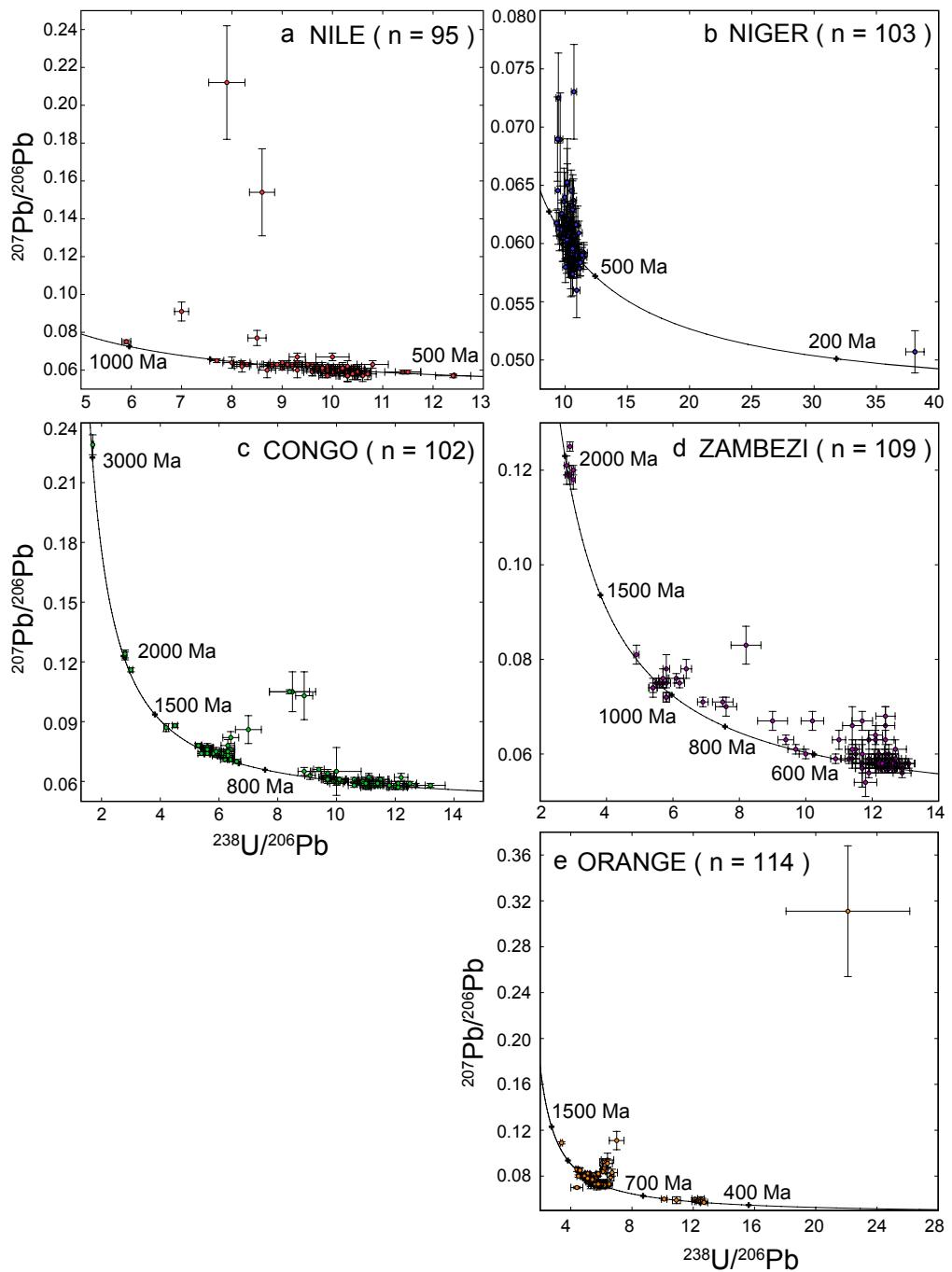


Fig. 5-4. Tera-Wasserburg concordia plots of U-Pb isotopic data for detrital monazites from the (a) Nile River, (b) Niger River, (c) Congo River, (d) Zambezi River and (e) Orange River, respectively. Error bars represent  $2\sigma$  quoted in Appendix Table S5-2.

For all studied rivers except for the Orange, the detrital monazites are dominated by grains having U–Pb ages between 800 and 400 Ma (Fig. 5-5a–e). However, the monazite populations of the four rivers show distinct Pan-African age peaks (Fig. 5-5f). We see no clear difference in age distribution among grains having different types of internal structure. The detrital monazites from the Nile River ( $n = 78$ ) have U–Pb ages ranging from 780 to 530 Ma with a peak at ~600 Ma (Fig. 5-5a, f). The U–Pb ages of detrital monazites from the Niger River ( $n = 73$ ) are mainly between 650 to 530 Ma and exhibited a broad peak around ~580 Ma (Fig. 5-5b, f). The Congo River monazites ( $n = 71$ ) include the oldest (3015 Ma) grain and define two small age peaks at ~630 and ~610 Ma and a main peak at ~560 Ma (Fig. 5-5c, f). The monazite age population of the Zambezi River ( $n = 68$ ) spans the range from 1960 to 480 Ma and a prominent peak at ~500 Ma (Fig. 5-5d, f). The U–Pb ages of the

Orange River monazites ( $n = 88$ ) are characterized by a dominant population between 1200 and 900 Ma, corresponding to the timing of the Grenville Orogeny (Fig. 5-5e).

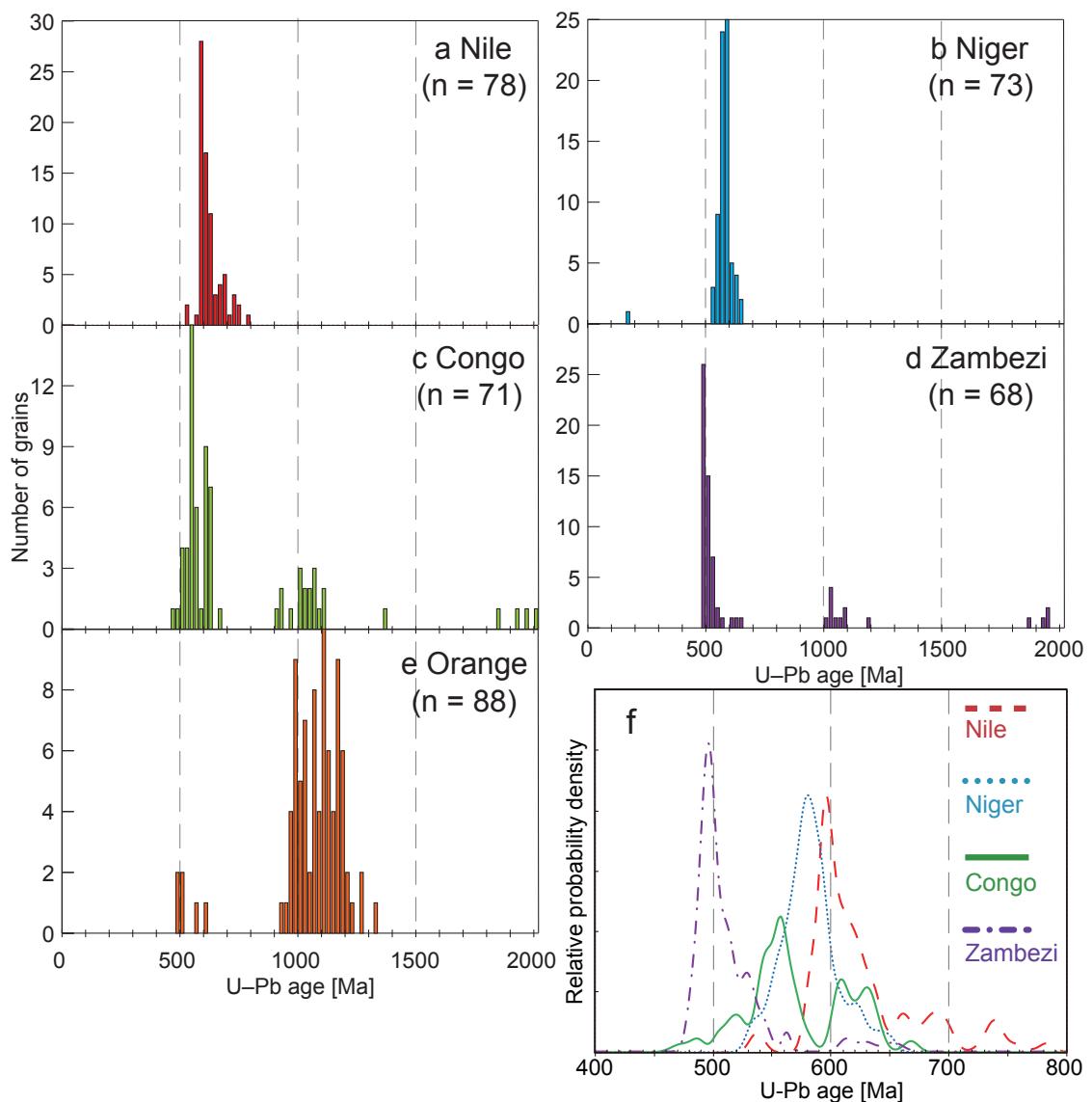


Fig. 5-5.  $^{206}\text{Pb}/^{238}\text{U}$  age histograms of the detrital monazites from the (a) Nile River, (b) Niger River, (c) Congo River, (d) Zambezi River and (e) Orange River, respectively. (f) Probability density graph of the four rivers except for the Orange River during 800–400 Ma. The concordant or <5% discordant data are included.

The elemental abundances determined by FE-EPMA and LA-ICP- MS are listed in Appendix Table S5-3. For the elements analyzed by both FE-EPMA and LA-ICP-MS, we compare the resultant abundances in Appendix Fig. S5-1. While the two data sets generally show good agreement, substantial discrepancies are observed in some grains especially for Ca and U. The discrepancies may be attributed to the different spatial resolution of the two analytical techniques. For the following discussion, we have essentially used the LA-ICP-MS results, because the analytical spatial resolution of LA-ICP-MS U-Pb dating is similar to that of LA-ICP-MS elemental analysis. In Fig. 5-6, however, the FE-EPMA data are shown as a cation plot, normalized to 16 oxygens. The LA-ICP-MS data are graphically presented for each river in Figs. 5-7 and 5-8 as plots of CI-chondrite normalized (McDonough and Sun, 1995)  $[Gd/Lu]_N$  vs.  $[Eu/Eu^*]_N$  and Th contents, respectively (brackets with subscript N denote CI-chondrite normalized value,  $[Eu/Eu^*]_N$  defined by  $[Eu]_N / ([Sm]_N * [Gd]_N)^{0.5}$ ). The  $[Gd/Lu]_N$  and  $[Eu/Eu^*]_N$  reflect the degrees of HREE fractionation and Eu anomaly, respectively. When the  $[Gd/Lu]_N$ ,  $[Eu/Eu^*]_N$  and Th contents are displayed together in a 3D plot (each ratio on one axis) (Fig. 5-9), it is evident that the detrital monazites from the five rivers plot preferentially along the x- and y-axes and show a variation in these parameters. The proportion of the grains having  $[Gd/Lu]_N$  of >400 varies substantially from river to river as well as among the different age populations (Fig. 5-7). For instance, the proportions are approximately one-third for the 500–480 Ma Zambezi and 640–600 Ma Congo populations, but less than one-tenth for the 600–560 Ma Niger

population. The grains containing <1wt% Th are yielded by the Nile, Congo, Zambezi, and Orange Rivers (Figs. 5-8). The low-Th monazite grains account for ~2% of all grains.

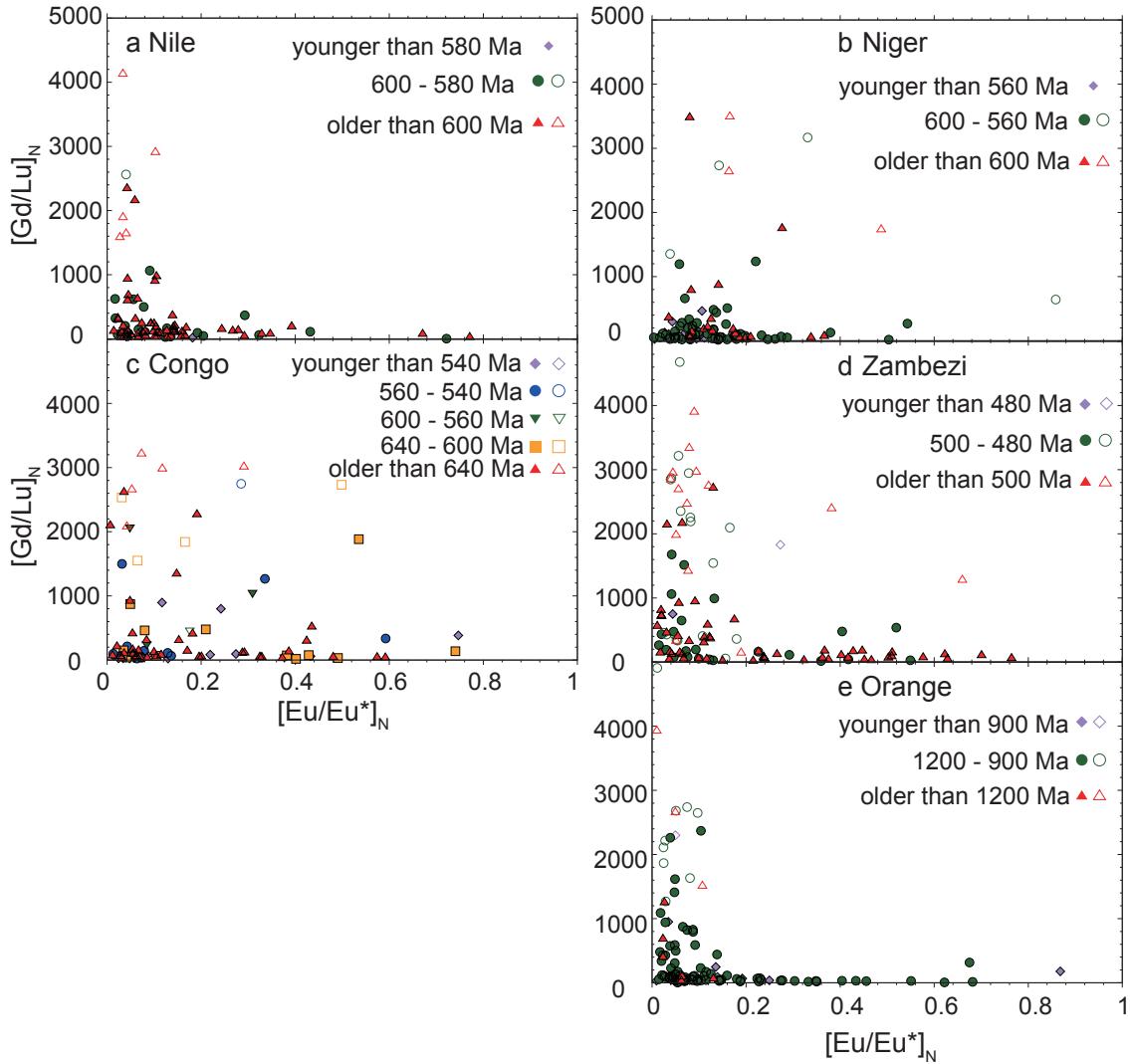


Fig. 5-7. Plots of  $[Gd/Lu]_N$  vs.  $[Eu/Eu^*]_N$  for detrital monazites from the (a) Nile River, (b) Niger River, (c) Congo River, (d) Zambezi River and (e) Orange River, respectively. The plots are coded according to U–Pb ages. The open symbols represent monazites having Lu contents below the analytical detection limit and therefore, should be regarded as lower limits of the actual values.

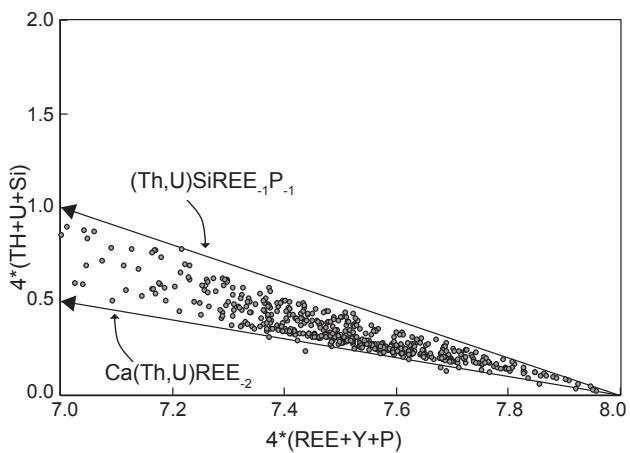


Fig. 5-6. Monazite cation plot, normalized to 16 oxygens. Huttonite exchange  $[(\text{Th}, \text{U})\text{SiREE}_{-1}\text{P}_{-1}]$  and Cheralite exchange  $[\text{Ca}(\text{Th}, \text{U})\text{REE}_{-2}]$  are represented as vectors.

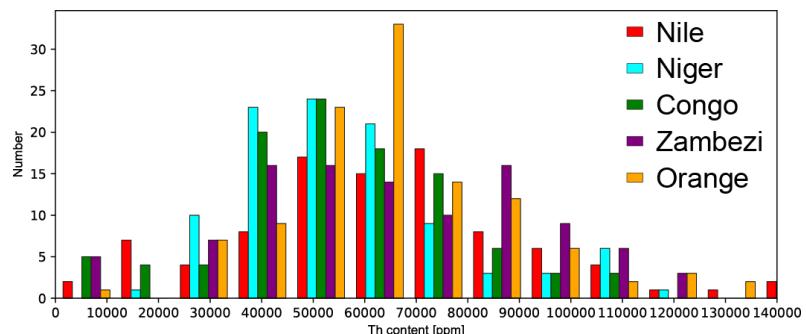


Fig. 5-8. Histogram of Th contents for detrital monazites from the five largest African rivers. Most grains contain more than 1wt% Th.

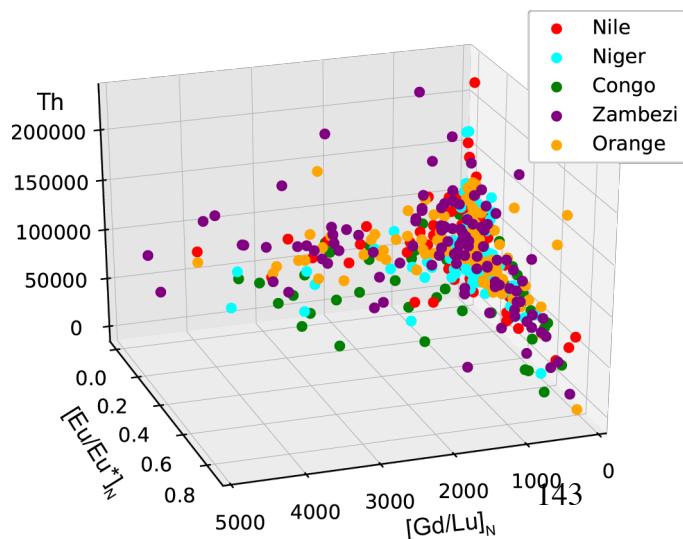


Fig. 5-9. A 3D plots of  $[\text{Eu}/\text{Eu}^*]_{\text{N}}$  vs.  $[\text{Gd}/\text{Lu}]_{\text{N}}$  vs. Th content for detrital monazites from the five largest African rivers.

## 5.5 Discussion

### 5.5.1 Geochemistry and genesis of the detrital monazites

The significant variation in major element compositions of the detrital monazites from the African rivers (Fig. 6) can be well explained by substitutions of  $\text{Ca}^{2+} + (\text{Th}, \text{U})^{4+} = 2\text{REE}^{3+}$  (cheralite substitution) and  $(\text{Th}, \text{U})^{4+} + \text{Si}^{4+} = \text{REE}^{3+} + \text{P}^{5+}$  (buttonite substitution) (Zhu and O’Nions, 1999, Pyle et al., 2001; Spear and Pyle 2002, Hoshino et al., 2012). Moreover, the detrital monazites plot preferentially along the cheralite substitution line rather than the buttonite substitution line. It has been indicated that the cheralite substitution is predominant in metamorphic monazites (Spear and Pyle 2002), whereas buttonite substitution is more common in magmatic monazite (Broska et al., 2000; Hoshino et al., 2012). The major element geochemistry, therefore, suggests that the detrital monazites were derived from various source rocks and a substantial proportion of them are metamorphic origin.

The significant variations in  $[\text{Gd/Lu}]_N$  with the increasing negative Eu anomaly ( $[\text{Eu/Eu}^*]_N$ ) are the most striking geochemical feature of the detrital monazites from the African rivers (Figs. 5-7). The understanding of the geochemical systematics can help in constraining the genesis of the detrital monazites. Below, I briefly review possible mechanism controlling the variation in geochemical compositions of monazite again, as discussed in the previous chapter.

Europium is unique among REEs in that it can exist as both trivalent and divalent under crustal condition and Eu<sup>2+</sup> is incorporated into plagioclase and K-feldspars more selectively than other trivalent REEs (Fujimaki and Tatsumoto, 1984; Bea et al. 1994; Bea, 1996). Both plagioclase and K-feldspar are major phases in low-Ca felsic rocks and the former would be residual during the parental magma genesis and also fractionate during the magma differentiation. As result, igneous monazites commonly have pronounced Eu negative anomalies ( $[Eu/Eu^*]_N$  of <0.1; shown in Chapter 3). Feldspars are common also in metamorphic rocks and their abundances change with metamorphic grade and chemical composition, leading to variable degrees of Eu negative anomaly in metamorphic monazites. For instance, higher-grade meta-sedimentary rocks contain more abundant K-feldspar due to the continuous breakdown of micas during prograde metamorphism, resulting in more prominent Eu negative anomalies in the monazites (Rubatto et al., 2006).

The observed  $[Gd/Lu]_N$  variation spanning three orders of magnitude (Fig. 5-7) is primary due to variable Lu concentrations down to ~1 ppm. The strong depletion in H-REEs including Lu would require the co-existence of minerals having much higher distribution coefficients for HREEs than for LREEs such as garnet and zircon (e.g., Schnetzler and Philpotts, 1970; Fujimaki and Tatsumoto, 1984; Hinton and Upton, 1991; Bea, 1994). In particular, garnet is so abundant in high-grade and sometimes medium-grade metamorphic rocks that the co-existing monazites can exhibit significant

H-REE depletion (Rubatto et al., 2006; Buick et al., 2010; Iizuka et al., 2010b; Mottram et al., 2014; Štípská et al., 2015).

The Th content in monazite are basically governed by cheralite and huttonite substitutions as a solid solution. Although igneous and metamorphic monazites show wide range of Th contents due to the change in co-existing assemblages (e.g., Xie et al., 2006; Kelts et al., 2008; Breiter et al., 2009; Rubatto et al., 2006), it has been reported that hydrothermal monazite is characterized by extremely low Th abundances beyond the range which igneous and metamorphic monazites form (shown in Chapter 3).

Overall, the observed variation in the REE and Th compositions of detrital monazites can be linked to their source rock type. The criteria determined in the previous chapter are used to interpret the genesis of the detrital monazite (Table 5-2). The studied detrital monazites are interpreted to be igneous ( $[Eu/Eu^*]_N < 0.1$  and  $[Gd/Lu]_N < 400$ ), garnet-rich metamorphic ( $[Gd/Lu]_N > 400$ ), garnet-poor or garnet-rich metamorphic ( $[Eu/Eu^*]_N > 0.1$  and  $[Gd/Lu]_N < 400$ ), and hydrothermal origins (Th content  $< 1\text{wt\%}$ ). The interpreted grains as a hydrothermal origin is minor (~2%) in the studied river sands because of the low-Th contents ( $< 1\text{wt\%}$ ) and scarce inclusions. Therefore, these hydrothermal monazites are not included in the following discussion.

Table 5-2. Geochemical criteria for an indicator of source rock type

Igneous	Metamorphic		Hydrothermal
	Garnet-poor	Garnet-rich	
[Eu/Eu*] <sub>N</sub> < 0.1 [Gd/Lu] <sub>N</sub> < 400	[Eu/Eu*] <sub>N</sub> > 0.1 [Gd/Lu] <sub>N</sub> < 400	[Gd/Lu] <sub>N</sub> > 400	Th content < 10 <sup>4</sup> ppm

### 5.5.2 Comparison with detrital zircon

The U-Pb age populations of detrital monazites and zircons combined for the five African rivers, where concordant or <5% discordant data are included, are compared in Fig. 5-10. Two distinct features are evident. First, the detrital monazite ages are rarely distributed in Archean and Paleoproterozoic times, whereas the detrital zircon ages indicate clear peaks in the period. Second, the peak at ~800 Ma is recognized in the zircon ages but not in the monazite ages.

Although the Pb diffusions in monazite and zircon are comparably slow, monazite can be more readily dissolved during metamorphism and abraded through weathering processes (Townsend et al., 2000; Williams, 2001; Rasmussen and Muhling, 2007, 2009; Harlov and Hetherington, 2010; Erickson et al., 2015). In addition, high-Ca felsic rocks commonly yield zircons but few monazites (Lee and Bastron, 1967; Montel, 1993; Kelts et al., 2008). Therefore, the discrepancy between the detrital monazite and zircon age distributions can be interpreted to reflect either the preferential disappearance of monazite through metamorphic/sedimentary processes or little occurrence of

monazite in high-Ca felsic magmatism. We consider the latter unlikely as a main reason for the lack of >1500 Ma detrital monazites because Hf–O isotopic data for the detrital zircons indicate their crystallization from geochemically evolved, and by extension, low-Ca felsic magmas (Iizuka et al., 2013). Instead, the scarcity of ancient monazites can be explained by dissolution-reprecipitation during the Pan-African as well as Grenville Orogeny and prolonged sedimentary recycling. In contrast, the unique detrital zircon age peak at ~800 Ma would be attributed to significant high-Ca magmatism, considering the juvenile Hf–O isotopic features of the 800–700 Ma detrital zircons (Iizuka et al., 2013) and the preservation of the Mesoproterozoic detrital monazites.

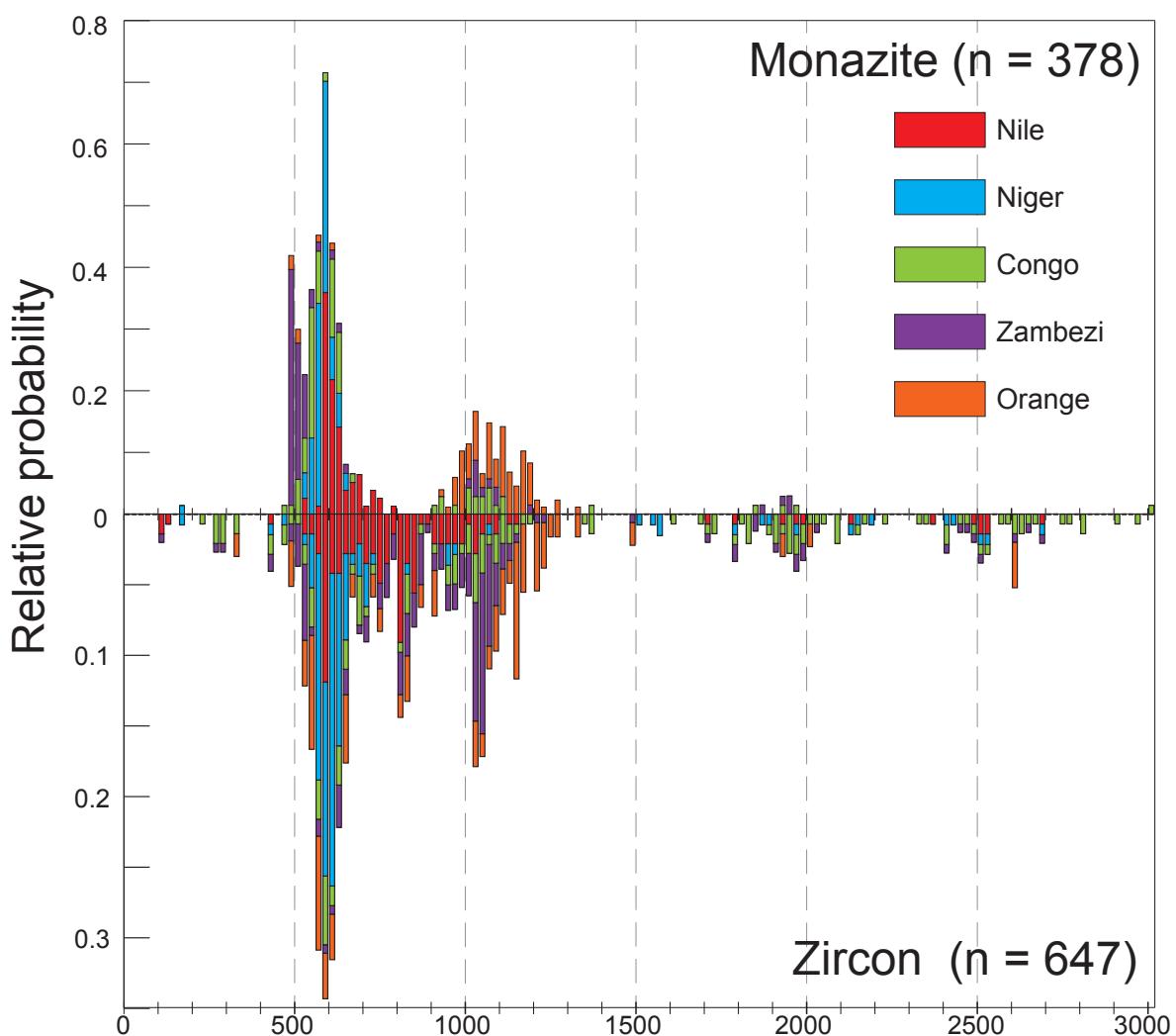


Fig. 5-10. Combined U–Pb age populations of detrital monazites (upper) and zircons (lower) from the five African rivers. The relative frequencies were calculated by accumulating the age proportions rather than the number of detrital grains from the five rivers. The detrital zircon data are from Iizuka et al. (2013).

When examining the Pan-African age distributions of the detrital monazites and zircons (concordant or <5% discordant) for each river (Fig. 5-11), we see that the monazite age peaks of the Niger, Congo and Zambezi Rivers are younger than the detrital zircon age peaks by 20–40 Myr, whereas in the Nile River the monazite and zircon age peaks are indistinguishable. These age gaps are larger than the analytical uncertainties (2SD:  $\pm \sim 15$  Ma and 2SE:  $\pm \sim 4$  Ma). Zircon and monazite U–Pb age gaps of  $\sim 20$  Myr were previously identified for single metamorphic rocks (e.g., Fitzsimons et al., 1997; Zeh et al., 2003). Yet the detrital zircons analyzed by Iizuka et al. (2013) mostly have oscillatory zoning and Th/U ratios higher than 0.1, which are common characteristics of magmatic zircon (Shore and Fowler, 1996; Hoskin and Ireland, 2000; Rubatto, 2002). Thus, the detrital zircon age peaks should reflect the timing of major felsic magmatic events in the river basins. In contrast, the Pan-African monazite populations of the studied rivers consist of grains interpreted as metamorphic origins as well as igneous origin (Table 5-3 and Fig. 11). High proportions of grains derived from garnet-rich metamorphic rocks are prominent feature of the age population at 640–600Ma for the Congo river and the age population around  $\sim 500$  Ma for the Zambezi river (Table 5-3). The discordance between the monazite and zircon age peaks is consistent with the large contributions from metamorphic monazites which are independently estimated by the geochemical indicator. Monazite age peaks, therefore, correspond to the timing of major metamorphic events as well as magmatic events. The implication is that during the early stages of the Pan-African Orogeny, felsic

magmatism was important probably due to pre- to syn-collisional subduction of oceanic plate (e.g., Hargrove et al., 2006) and, 20–40 Myr later, syn-collisional and post-collisional metamorphism became dominant. The metamorphism may result from crustal heating caused by asthenosphere upwelling associated with lithospheric delamination (Kay and Kay, 1993; Sacks and Secor, 1990; Massonne, 2005) or radioactive self-heating in a thickened crust (Beaumont et al., 2004; Jamieson et al., 2004). Indeed, numerical models for delamination (Krystopowicz and Currie, 2013) and radioactive self-heating (Jamieson et al., 2004) suggested that regional metamorphism occurs in several tens of million years after a continental collision, which is consistent with the gap between detrital zircon and monazite age peaks.

Table 5-3. Summary of geochemical data for detrital monazites from the five African rivers.

Age [ Ma]		Nile River		Niger River		Congo River			Zambezi River		Orange River		
		All	600–580	All	600–560	All	640–620	620–600	560–540	All	500–480	All	900–1200
Igneous		42%	43%	44%	42%	42%	29%	42%	60%	10%	12%	34%	36%
Metamorphic	Garnet-rich	17%	13%	19%	20%	30%	71%	42%	7%	49%	58%	31%	29%
	Garnet-poor	40%	43%	36%	38%	24%	0%	42%	20%	35%	31%	34%	36%
Hydrothermal		1%	0%	0%	0%	4%	0%	0%	13%	6%	0%	1%	0%

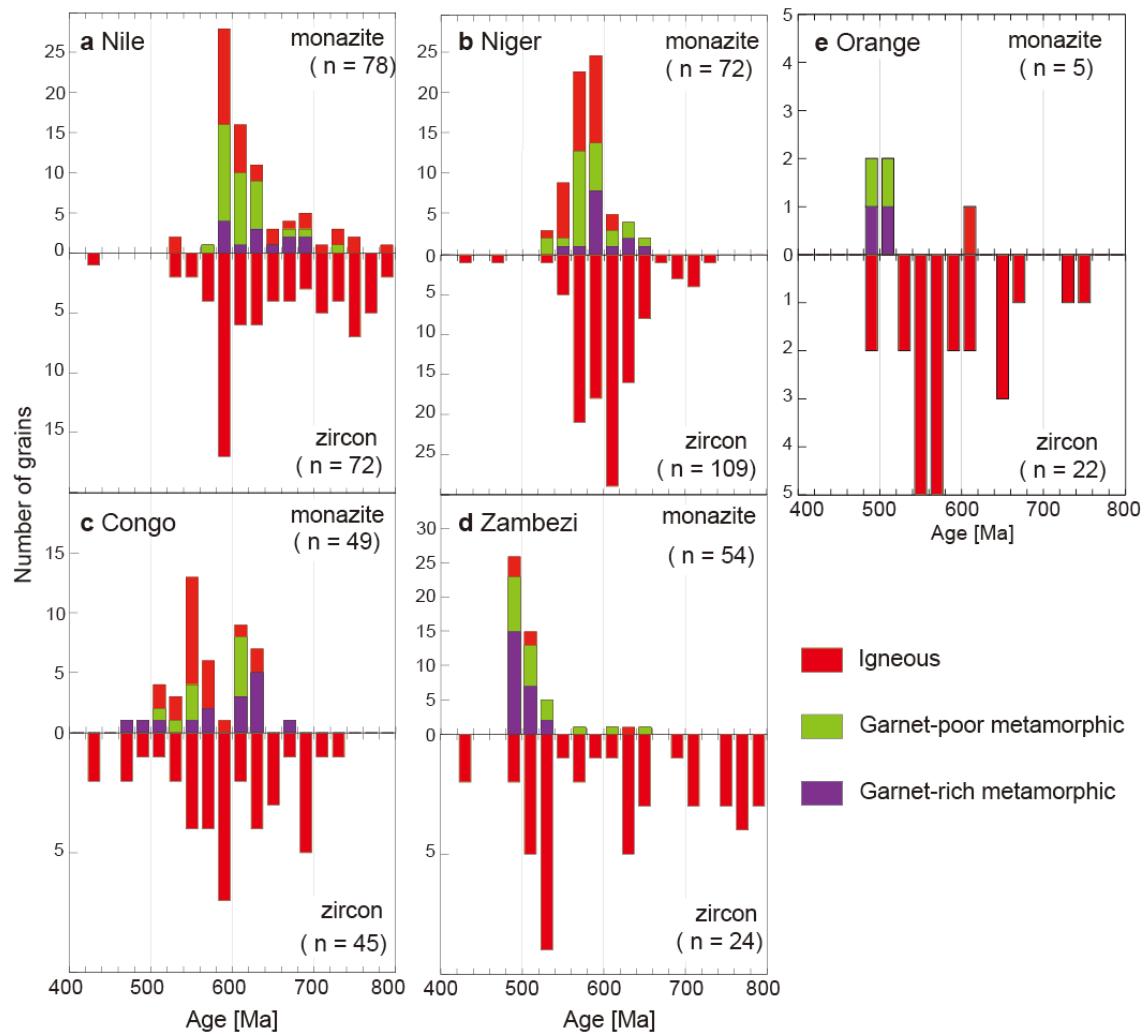


Fig. 5-11. Histograms of U–Pb ages between 800 and 400 Ma of concordant or <5% discordant monazites (upper) and zircons (lower) from the (a) Nile River, (b) Niger River, (c) Congo River, (d) Zambezi River and (e) Orange River, respectively. For the detrital monazites, columns are corded according to potential genesis inferred from the geochemical systematics (Section 5.1 and Table 2). All detrital zircons are interpreted as igneous origin on the basis of the internal structure and Th/U (Iizuka et al., 2013).

### **5.5.3 Constraints on the timing and nature of the Pan-African Orogeny**

The detrital monazites from the Congo, Nile, Niger and Zambezi Rivers are dominated by grains of Pan-African U–Pb ages. Moreover, the Pan-African populations of the four rivers indicate distinctive age peaks (Fig. 5-5), highlighting the asynchrony of major orogenic events in the source terranes of the detrital monazites. Considering that detritus of the large African rivers would have been substantially recycled from older sedimentary rocks, there is a controversy on whether the source terranes properly correspond with the basements of the modern river basins. Nevertheless, because the areas drained by the older riverine systems would be, at least partly, overlapping with the modern river basins, it is reasonable to assume that the basements in the modern drainage basins are important sources of the detrital monazites. Accordingly, we discuss the implications of our results for the timing and nature of orogenic events in the drainage basins.

The oldest Pan-African age peak was obtained from the Congo River (Fig. 5-5f), whose drainage area covers parts of two Neoproterozoic orogenic belts, the Central African Fold Belt and the West Congo Belt (Figs. 5-1 and 5-2). The former belt includes 640–610 Ma granites and granulites with zircon U–Pb ages of ~650–600 Ma (Pin and Poidevin, 1987; Toteu et al., 2001; Owona et al., 2012), overlapping the Congo detrital monazite age peak at ~630 and ~610 Ma. The granulite-facies metamorphism has been interpreted as a result of the collision between the Congo Craton and the Saharan Metacraton (Kröner and Stern, 2004; Bouyo Houketchanget al., 2013). Our

geochemical data indicate that the 640–600 Ma detrital monazites are essentially of metamorphic origin and, in more detail, that the 640–620 Ma grains are derived mainly from garnet-bearing metamorphic rocks, whereas younger ones are sourced from garnet-absent metamorphic rocks (Fig. 5-11c). These results are consistent with the collision between the Congo Craton and Saharan Metacraton occurred between ~640 and 600 Ma and further suggest that the major mode of the metamorphism changed from higher-grade around ~630 Ma to lower-grade around ~610 Ma.

The U–Pb ages of the Congo River monazites indicate another significant peak at 550 Ma (Fig. 5-5f), which is delayed by 40 Myr from the Pan-African detrital zircon age peak (Fig. 5-11c). Similar ages (~570–540 Ma) were yielded by monazite U–Pb dating and clinopyroxene Ar–Ar dating of metamorphic rocks from the West Congo Belt, which have been interpreted as the timing of the collision between the Congo Craton and the São Francisco Craton of Brazil (Tack et al., 2001; Frimmel et al., 2006; Monié et al., 2012). Moreover, zircon U–Pb dating of the orogenic belt indicated zircon crystallization at >580 Ma and a significant zircon Pb-loss event at ~540 Ma (Monié et al., 2012), consistent with the detrital monazite and zircon records. Given that the detrital zircons are essentially of magmatic origin, whereas the ~550 Ma detrital monazites were mainly from granitic rocks (Fig. 5-11c), these detrital records may reflect significant high-Ca felsic magmatism at ~600 Ma due to the subduction of oceanic plates to the assembling the Congo and São Francisco Craton, followed by low-Ca felsic magmatism at ~550 Ma resulting from the collision of these cratons. If

that is the case, the igneous detrital monazite recorded the timing of S-type granite formation associated with continental collision.

The Nile River basin area largely covers the northern part of the East African Orogen, in particular the Arabian-Nubian Shield (Fig. 5-1). Integration of geological, geochronological and geochemical studies revealed that the shield was formed by accretion of juvenile arcs between 850 and 620 Ma, followed by emplacement of low-Ca felsic magmas during post-collisional extension at 610–560 Ma (Meert, 2003; Breitkreuz et al., 2010; Johnson et al., 2011; Fritz et al., 2013; Eliwa et al., 2014). Moreover, the northern part of the East African Orogen underwent peak metamorphic conditions up to granulite-facies at 640–620 Ma (Fritz et al., 2013). The crust formation of the accreted juvenile arcs is recorded by the Nile River zircons of ~840–700 Ma, but not by the monazites, reflecting the limited monazite crystallization during high-Ca felsic magmatism (Figs. 5-5f and 5-11a). Note that the monazite and zircon age peaks of the Nile are coincident, unlike the Niger, Congo and Zambezi. The correspondence of the monazite and zircon age peaks, together with the monazite geochemical features, can be well explained by major low-Ca felsic magmatism associated with metamorphism around 600 Ma in the drainage basins. This interpretation is consistent with the occurrence of low-Ca felsic rocks formed at 610–560 Ma (Johnson et al., 2011; Eliwa et al., 2014).

The Niger River runs through the central and southern parts of the Trans-Saharan Belt (Fig. 5-1). The ~650–600 Ma peraluminous granites and ~600–580 Ma metaluminous

granitoids were reported in the southern area (Ferré et al., 2002; Agbossoumondé et al., 2007; Key et al., 2012; Goodenough et al., 2014). The older granitic magmatism is attributed to the collision of the West African Craton and Saharan Metacraton accompanied by high-temperature metamorphism at ~640 Ma (Ferré et al., 2002; Caby et al., 2003; Liégeois et al., 2013), whereas the younger granitoid magmatism is ascribed to post-collisional extension with the emplacement of mantle-derived magma (Goodenough et al., 2014). In the central area, the continental collisions forming the Trans-Sharan Belt are marked by the voluminous granitic rocks of around 620–580 Ma (Boullier et al., 1986; Liégeois et al., 1987). The detrital monazite ages of the Niger River show a predominant distribution between 640 and 550 Ma with a peak at ~580 Ma (Fig. 5-5f). The age population ranging from 640 to 580 Ma, together with the zircon age population, show generally good agreement with the timing of the continental collisions. However, the age peak of monazite is ~40 Myr younger than that of the detrital zircons. This skewed age distribution towards younger ages was possibly attributed to igneous or metamorphic events relating to the post-collisional extension. Indeed, the thermal history of the Eastern Nigeria terrane in the Trans-Saharan Belt constrained by dating granites and migmatites with multiple isotopic chronometers having different closure temperatures suggests a secular cooling from amphibolite- to granulite-facies condition at ~590 Ma to upper greenschist- to lower amphibolite-facies conditions at ~550 Ma (Ferré et al., 2002), which is consistent with the garnet-poor origins of the monazite grains yielding 600–560 Ma (Fig., 5-11b).

The Zambezi River detrital monazites define a remarkably sharp U–Pb age peak at ~500 Ma (Fig. 5-5f) that is ~30 Myr younger than that of the detrital zircons. The 520–480 Ma monazite population includes grains of igneous and garnet-bearing metamorphic origins with high populations (Fig. 5-11d), indicating that low-Ca felsic magmatism and high-grade metamorphism were significant at that time in the basin area. The river basin covers large parts of the Lufilian and Zambezi Belts and the southernmost part of the Mozambique Belt. These orogenic belts were subjected to thermal events related to the final stage of the Gondwana assembly. The collision between the Congo and Kalahari Cratons resulted in metamorphism and syn-tectonic granitic magmatism in the Lufilian and Zambezi Belts at 550–520 Ma (e.g., John et al., 2004; Johnson et al., 2005; Eglinger et al. 2016), while the collision between the Congo and East Antarctica Cratons caused high-grade metamorphism and migmatization in the Mozambique Belt between 570–530 Ma (Bingen et al., 2009; Fritz et al., 2013). The detrital zircon age peak corresponds to the timing these continental collisions. Moreover, voluminous low-Ca granite emplacement and high-grade metamorphism took place in the southern part of the Mozambique Belt at ca. 500 Ma (Bingen et al., 2009; Thomas et al., 2010; Ueda et al., 2012a, b), coincident with the detrital monazite age peak. Note, however, that the Mozambique Belt is only marginally overlapped by the current Zambezi River basin. Besides, ~500 Ma granites and high-grade metamorphic rocks have been found from the East Antarctica that was adjacent to the Mozambique Belt at that time (e.g., Grantham et al., 2003, 2013). These contemporaneous post-collisional

magmatism and metamorphism are considered to result from asthenosphere upwelling following lithospheric delamination (Jacobs et al., 2008a; Ueda et al., 2012a, b). The detrital monazite peak at ~500 Ma may indicate that post-collisional granitic magmatism and high-grade metamorphism caused by the asthenosphere upwelling were spread into the major source terrane(s) of the Zambezi River sediments, even though a limited number of ~500 Ma granites and metamorphic rocks are currently recognized (Goscombe et al., 2000; Johnson et al., 2006).

## 5.6 Conclusion

The U–Pb ages and geochemical compositions of the detrital monazites from the Nile, Niger, Congo, Zambezi and Orange Rivers were determined. The U–Pb age data showed peaks at ~600 Ma for the Nile, ~580 Ma for the Niger, ~630, ~610 and ~560 Ma for the Congo and ~500 Ma for the Zambezi and ~1200–900 Ma for the Orange. The Pan-African age peaks of the Niger, Congo and Zambezi are younger by 20–40 Myr than those defined by the detrital zircons from the same rivers, but the detrital monazites and zircons of the Nile show nearly identical Pan-African age peaks. The monazite geochemical data revealed the significant variation in negative Eu anomaly, depletion of HREE, and Th content. The geneses of detrital monazites were estimated based on the geochemical features, which revealed the monazite age spectra consisted of metamorphic monazites essentially reflect the timing of major low- to high-grade metamorphism as well as low-Ca felsic magmatism in the source terranes. The age gaps of 20–40 Myr between the zircon and monazite peaks would reflect the time intervals

between major felsic magmatism in pre-to syn-collisional stages and metamorphism in syn-to post-collisional stages during Pan-African orogenic events. The obtained results highlight the utility of the combined use of detrital monazite and zircon data for investigating the timescales and nature of major orogenic events in the source terranes.

## **6. Synthesis**

This work has focused on a key aspect of monazite as geochemical tracer for various crustal processes and discussed the linkage between the geochemical characteristics of monazite and its crystallization condition.

To elucidate the linkage, accurate determination of monazite trace element composition is essential. Synthetic REE phosphates were analyzed to assess the effect of oxide interference during REE analysis of natural monazite by LA-ICP-MS. The results have demonstrated the necessity of the oxide interference correction even when the plasma condition was optimized to minimize the oxide production rates. The determined oxide production rates using the synthetic REE phosphates allow more accurate quantification of REE concentrations in monazite.

The chapters 3 and 4 set out with the aim of assessing the variation in trace element compositions of igneous monazites from a single granite pluton and various granitic rocks across the Japan arc, respectively. The trace element analysis has shown the systematic variation in Eu anomaly, LREE-depletion, and Th-depletion of monazites in individual granite samples from the single pluton. On the basis of Rayleigh model calculations, it has been quantitatively confirmed that these parameters are controlled by the fractionation of plagioclase and monazite. This work enhances the utility of monazite trace element composition as an indicator of magma differentiation. Furthermore, the comprehensive analyses of monazites in granitic rocks from the Japan

arc have elucidated that Eu anomaly in monazite reflects not only the degree of magma differentiation but also the redox state of magma, the latter of which is controlled by the nature of source rocks.

The accumulation of trace element data of igneous monazite in this study permitted, for the first time, the quantitative comparison of chemical signatures between igneous and metamorphic monazites. The most obvious finding is that the magnitude of Eu anomaly of igneous monazite is much larger than that of metamorphic monazite. Moreover, it has been unraveled, based on the comprehensive compilation, that the chemical compositions of monazite derived from garnet-rich metamorphic rocks are characterized by the significant HREE-depletion. These findings provide a new geochemical indicator for source rock type of detrital monazite. Further accumulation of accurate monazite trace element data by follow-up studies would make the indicator more robust.

One of the novelties of this work is the integration of U–Pb dating and trace element analysis of detrital monazite for obtaining a high-resolution archive of orogenic events. To verify the geochemical indicator for source rock type, the age spectra of detrital monazite and zircon from the same river sands were compared. The comparison has corroborated the utility of the geochemical indicator. Furthermore, the age gaps of 20–40 Myr between the zircon and monazite peaks have been found. This finding implies that detrital zircon age spectra are the archive of major felsic magmatism in pre- to syn-collisional stages (Pacific-type orogeny); in contrast, detrital monazite age spectra

are the archive of major magmatism and metamorphism in syn- to post-collisional stages (collision-type orogeny). It would be beneficial to apply this approach to the detrital monazites in old sedimentary rocks for elucidating the timing and nature of orogenic events through the Earth's history and possibly gaining new insights into the supercontinent cycle.

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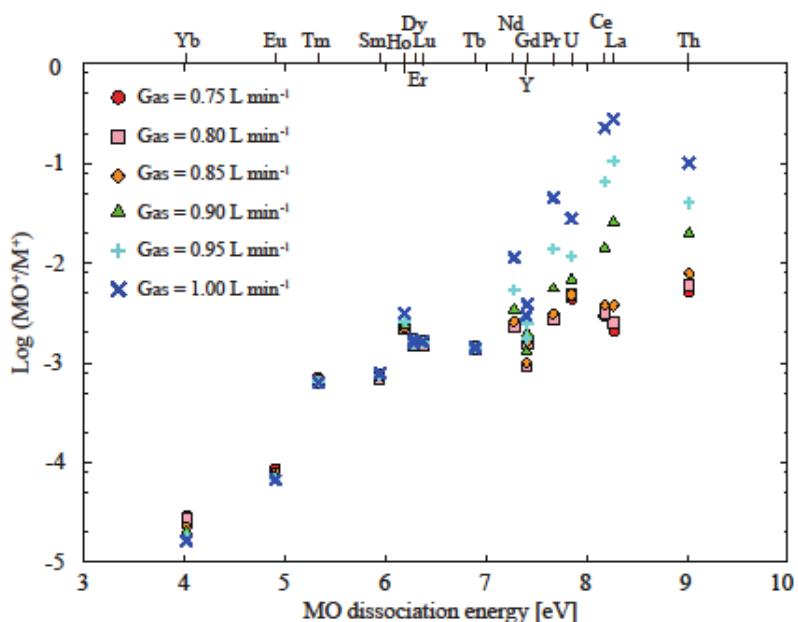
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## Appendix

Appendix Figure 2-1

### Electronic Supplementary Information

**ESI Figure S-1.** Plot of  $\log (\text{MO}^+/\text{M}^+)$  of phosphate versus MO dissociation energy for REE, Th, and U with various sample gas flow rates.



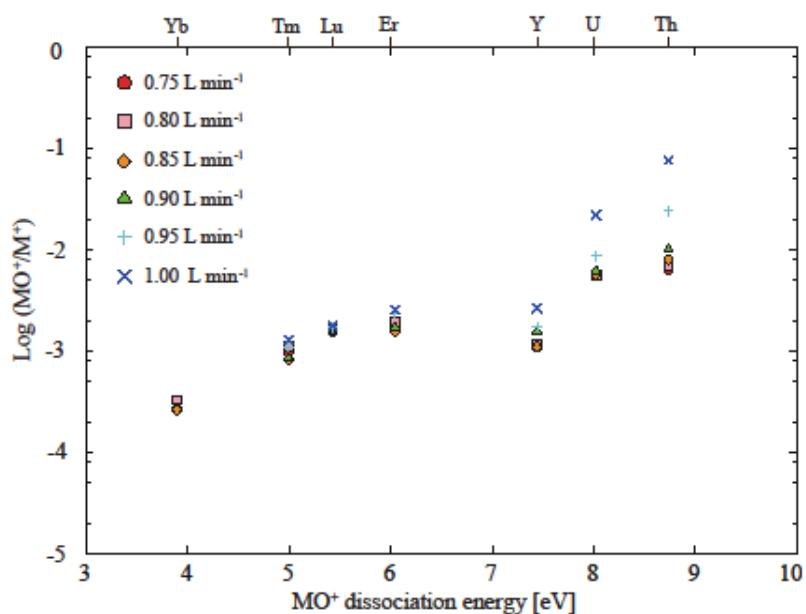
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Appendix Figure 2-2

**Electronic Supplementary Information**

**ESI Figure S-2.** Plot of log ( $\text{MO}^+/\text{M}^+$ ) of zircon versus  $\text{MO}^+$  dissociation energy for REE, Th, and U with various sample gas flow rates.



**References**

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S-1

**Appendix Table 2-1. Results of LA-ICP-MS analyses of synthetic REE phosphates**

Appendix Table 2-1 Results of solution nebulization-ICP-MS analysis of standard solutions											
Signal Intensity - elemental ions											
Gas flow rate [L/min]	88Sr	138Ba	139La	140Ce	141Pr	143Nd	149Sm	151Eu	159Tb	232Th	238U
0.95	5.4.E+05	4.7.E+05	7.1.E+05	6.5.E+05	7.6.E+05	8.7.E+04	9.4.E+04	3.3.E+05	6.9.E+05	4.1.E+05	4.0.E+05
0.975	1.7.E+06	1.5.E+06	9.7.E+05	8.8.E+05	1.1.E+06	1.4.E+05	1.6.E+05	5.5.E+05	1.0.E+06	1.4.E+06	1.4.E+06
1	1.2.E+06	1.1.E+06	1.6.E+06	1.5.E+06	1.8.E+06	2.1.E+05	2.4.E+05	8.3.E+05	1.7.E+06	1.1.E+06	1.1.E+06
1.025	1.6.E+06	1.4.E+06	2.0.E+06	1.9.E+06	2.3.E+06	2.8.E+05	3.2.E+05	1.1.E+06	2.3.E+06	1.6.E+06	1.7.E+06
1.05	1.0.E+06	9.6.E+05	2.1.E+06	2.0.E+06	2.7.E+06	3.3.E+05	3.9.E+05	1.4.E+06	2.8.E+06	9.8.E+05	1.1.E+06
1.075	4.1.E+06	3.6.E+06	1.4.E+06	1.5.E+06	2.3.E+06	3.5.E+05	4.4.E+05	1.6.E+06	2.6.E+06	2.7.E+06	3.6.E+06
2SD											
Gas flow rate [L/min]	88Sr	138Ba	139La	140Ce	141Pr	143Nd	149Sm	151Eu	159Tb	232Th	238U
0.95	4.2.E+04	5.2.E+04	5.5.E+04	7.2.E+04	1.0.E+05	7.4.E+03	7.9.E+03	2.1.E+04	1.7.E+04	7.2.E+04	6.2.E+04
0.975	1.5.E+05	2.1.E+05	5.5.E+04	1.2.E+05	1.4.E+05	1.5.E+04	1.1.E+04	7.9.E+04	5.8.E+04	2.2.E+05	1.9.E+05
1	9.1.E+04	1.2.E+05	8.9.E+04	1.6.E+05	1.6.E+05	1.8.E+04	1.0.E+04	6.0.E+04	1.6.E+05	8.8.E+04	1.6.E+05
1.025	6.8.E+04	8.0.E+04	2.0.E+05	9.8.E+04	1.8.E+05	2.5.E+04	1.1.E+04	5.6.E+04	9.8.E+04	9.4.E+04	1.5.E+05
1.05	1.0.E+06	9.3.E+05	8.9.E+04	1.1.E+05	7.1.E+04	1.2.E+04	1.1.E+04	6.5.E+04	5.3.E+04	9.1.E+05	1.0.E+06
1.075	7.8.E+04	2.0.E+05	1.3.E+05	2.0.E+05	2.1.E+05	2.0.E+04	4.8.E+03	4.1.E+04	1.0.E+05	2.5.E+05	3.3.E+05
Signal Intensity - oxide ions											
Gas flow rate [L/min]	88Sr	138Ba	139La	140Ce	141Pr	143Nd	149Sm	151Eu	159Tb	232Th	238U
0.95	1.5.E+02	2.8.E+02	9.4.E+03	1.2.E+04	1.5.E+04	1.7.E+03	3.8.E+02	1.8.E+02	7.4.E+03	1.5.E+04	1.4.E+04
0.975	3.5.E+02	8.7.E+02	1.3.E+04	1.6.E+04	2.0.E+04	2.4.E+03	6.5.E+02	2.9.E+02	9.8.E+03	4.8.E+04	4.4.E+04
1	2.8.E+02	6.9.E+02	2.4.E+04	2.9.E+04	3.4.E+04	3.8.E+03	9.6.E+02	4.8.E+02	1.8.E+04	4.5.E+04	3.8.E+04
1.025	3.6.E+02	1.0.E+03	5.8.E+04	5.5.E+04	4.9.E+04	5.2.E+03	1.3.E+03	7.1.E+02	2.5.E+04	1.2.E+05	6.6.E+04
1.05	3.0.E+02	9.3.E+02	3.1.E+05	2.5.E+05	1.2.E+05	9.0.E+03	1.7.E+03	8.3.E+02	3.9.E+04	2.6.E+05	1.1.E+05
1.075	2.0.E+03	8.8.E+03	8.7.E+05	7.0.E+05	4.3.E+05	3.1.E+04	2.3.E+03	1.0.E+03	8.3.E+04	1.9.E+06	1.0.E+06
2SD											
Gas flow rate [L/min]	88Sr	138Ba	139La	140Ce	141Pr	143Nd	149Sm	151Eu	159Tb	232Th	238U
0.95	7.4.E+01	1.2.E+02	1.2.E+03	1.0.E+03	8.8.E+02	3.6.E+02	1.6.E+02	6.9.E+01	2.9.E+02	1.8.E+03	8.1.E+02
0.975	1.2.E+02	2.2.E+02	1.1.E+03	1.7.E+03	1.5.E+03	3.4.E+02	2.3.E+02	8.4.E+01	1.4.E+03	5.4.E+03	4.6.E+03
1	7.5.E+01	1.7.E+02	1.9.E+03	2.5.E+03	2.8.E+03	5.5.E+02	2.0.E+02	1.0.E+02	1.2.E+03	4.3.E+03	3.0.E+03
1.025	1.3.E+02	2.9.E+02	1.4.E+04	1.6.E+04	2.3.E+03	4.6.E+02	2.0.E+02	2.0.E+02	1.6.E+03	3.0.E+04	9.2.E+03
1.05	3.9.E+02	8.4.E+02	1.1.E+05	7.7.E+04	3.7.E+04	2.0.E+03	2.4.E+02	2.9.E+02	9.1.E+03	2.5.E+05	1.0.E+05
1.075	4.4.E+02	1.0.E+03	1.1.E+05	9.1.E+04	7.5.E+04	4.0.E+03	3.7.E+02	2.2.E+02	1.4.E+04	1.3.E+05	2.0.E+05
Oxide Production Rates											
Gas flow rate [L/min]	SrO/Sr	BaO/Ba	LaO/La	CeO/Ce	PrO/Pr	NdO/Nd	SmO/Sm	EuO/Eu	TbO/Tb	ThO/Th	UO/U
0.95	0.02%	0.06%	1.32%	1.86%	2.00%	1.97%	0.41%	0.06%	1.06%	3.56%	3.50%
0.975	0.02%	0.06%	1.31%	1.81%	1.85%	1.68%	0.41%	0.05%	0.99%	3.47%	3.24%
1	0.02%	0.06%	1.49%	1.95%	1.93%	1.79%	0.40%	0.06%	1.01%	4.05%	3.43%
1.025	0.02%	0.07%	2.93%	2.93%	2.14%	1.83%	0.42%	0.06%	1.07%	7.19%	3.88%
1.05	0.03%	0.09%	15.23%	12.32%	4.49%	2.70%	0.43%	0.06%	1.38%	25.59%	9.50%
1.075	0.05%	0.25%	60.71%	47.69%	18.61%	8.82%	0.53%	0.07%	3.13%	71.16%	28.25%
2SD											
Gas flow rate [L/min]	SrO/Sr	BaO/Ba	LaO/La	CeO/Ce	PrO/Pr	NdO/Nd	SmO/Sm	EuO/Eu	TbO/Tb	ThO/Th	UO/U
0.95	0.010%	0.019%	0.138%	0.174%	0.199%	0.317%	0.117%	0.015%	0.031%	0.534%	0.401%
0.975	0.005%	0.013%	0.092%	0.219%	0.192%	0.605%	0.109%	0.012%	0.109%	0.473%	0.389%
1	0.005%	0.011%	0.104%	0.176%	0.166%	0.212%	0.060%	0.009%	0.081%	0.326%	0.384%
1.025	0.006%	0.015%	0.548%	0.592%	0.131%	0.163%	0.044%	0.013%	0.058%	1.307%	0.454%
1.05	0.111%	0.250%	3.628%	2.707%	0.890%	0.405%	0.044%	0.015%	0.203%	78.843%	29.981%
1.075	0.007%	0.022%	6.708%	6.408%	2.454%	0.881%	0.059%	0.010%	0.371%	5.676%	4.253%

Appendix Table2.2. Concentrations of REE, Th and U of the natural monazites in pegmatite and granite from Japan and meta-conglomerate from the Yilgarn Craton.

**Uncorrected values for oxide interferences**

Concentration	[ppm]	La	Ce†	Pr	Nd	Sm	Eu	Gd	Tb
Monazite from Pegmatite	1.1E+05	2.6.E+05	3.2.E+04	1.4.E+05	4.5.E+04	7.8.E+00	3.3.E+04	3.8.E+03	
Monazite from Granite	1.7E+05	3.1.E+05	3.5.E+04	1.5.E+05	2.6.E+04	3.8.E+02	1.5.E+04	1.5.E+03	
Monazite from Meta-conglomerate	1.7E+05	2.8.E+05	2.6.E+04	9.4.E+04	2.4.E+04	4.2.E+03	2.0.E+04	2.2.E+03	
	[ppm]	Dy	Ho	Er	Tm	Yb	Lu	Th	U
Monazite from Pegmatite	1.2.E+04	9.4.E+02	9.1.E+02	5.0.E+01	2.0.E+02	1.8.E+01	5.4.E+04	4.9.E+03	
Monazite from Granite	5.9.E+03	8.2.E+02	1.5.E+03	1.2.E+02	5.0.E+02	4.3.E+01	2.3.E+04	4.1.E+02	
Monazite from Meta-conglomerate	6.8.E+03	5.5.E+02	5.3.E+02	2.6.E+01	9.8.E+01	9.6.E+00	7.4.E+04	1.3.E+04	

**Corrected for oxide interferences using REEO+/REEP+ determined for the synthetic phosphates**

Concentration	[ppm]	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb
Monazite from Pegmatite	-	-	-	-	-	-	-	3.3.E+04	3.7.E+03
Monazite from Granite	-	-	-	-	-	-	-	1.4.E+04	1.4.E+03
Monazite from Meta-conglomerate	-	-	-	-	-	-	-	1.9.E+04	2.2.E+03
	[ppm]	Dy	Ho	Er	Tm	Yb	Lu	Th	U
Monazite from Pegmatite	1.2.E+04	9.3.E+02	9.1.E+02	5.0.E+01	1.4.E+02	9.5.E+00	-	-	
Monazite from Granite	5.9.E+03	8.2.E+02	1.5.E+03	1.2.E+02	4.7.E+02	3.9.E+01	-	-	
Monazite from Meta-conglomerate	6.7.E+03	5.5.E+02	5.2.E+02	2.6.E+01	5.7.E+01	3.4.E+00	-	-	

**Considered interferences**

from	143Nd/Pr	143Sm/Sm	147SmO/Sm	151EuO/Eu	153EuO/Eu	157GdO/Gd	159TbO/Tb
to	157Gd	159Tb	163Dy	165Ho	167Er	169Tm	173Yb

Appendix Table 3-11. Results of U-Pb dating of monazite in the Busetsu granite

sample name	$^{206}\text{Pb}/^{238}\text{U}$	$2\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$	$2\sigma$	$^{207}\text{Pb}/^{235}\text{U}$	$2\sigma$	$^{206}\text{Pb}/^{204}\text{Pb}$	Age 206/238	$2\sigma$	Age 207/206	$2\sigma$	Age 207/235	$2\sigma$	%discordance	$^{204}\text{Pb}/^{206}\text{Pb}$
2801_mnz-01	0.011	0.0004	0.108	0.016	0.166	0.021	UDL	72	3	1765	135	156	19	54%	0.01
2801_mnz-02	0.011	0.0004	0.074	0.010	0.126	0.019	UDL	70	3	1041	136	121	17	42%	0.00
2801_mnz-06	0.011	0.0004	0.069	0.010	0.110	0.016	UDL	73	2	898	150	106	14	31%	0.02
2801_mnz-07	0.023	0.0008	0.048	0.025	1.481	0.208	UDL	144	5	3974.4	45	923	85	85%	0.03
2801_mnz-08	0.011	0.0003	0.057	0.007	0.084	0.009	UDL	69	2	491	135	82	9	16%	(0.01)
2801_mnz-09	0.011	0.0004	0.075	0.009	0.118	0.012	UDL	73	3	1068	121	113	11	36%	0.01
2801_mnz-10	0.012	0.0004	0.098	0.015	0.173	0.029	UDL	74	3	1586	143	162	25	54%	(0.03)
2801_mnz-11	0.011	0.0004	0.077	0.009	0.123	0.013	UDL	70	2	1120	117	118	12	41%	0.02
2801_mnz-12	0.011	0.0003	0.069	0.012	0.107	0.017	UDL	71	2	898	179	103	15	31%	(0.02)
2801_mnz-13	0.013	0.0011	0.222	0.027	0.410	0.045	UDL	82	7	2994.5	97.9	349	32	77%	0.01
2801_mnz-14	0.011	0.0003	0.067	0.008	0.106	0.012	UDL	69	2	837	124	102	11	32%	0.01
2801_mnz-15	0.013	0.0003	0.177	0.006	0.319	0.018	UDL	85	2	2624.2	28.4	281	14	70%	0.01
2801_mnz-16	0.013	0.0004	0.135	0.015	0.251	0.027	UDL	83	3	2163.2	96.9	227	22	64%	0.00
2801_mnz-17	0.011	0.0004	0.090	0.011	0.160	0.020	UDL	72	3	1425	117	151	17	52%	(0.01)
2801_mnz-18	0.012	0.0004	0.100	0.014	0.169	0.022	UDL	77	3	1623	130	158	19	51%	0.00
2801_mnz-19	0.011	0.0003	0.061	0.005	0.090	0.008	UDL	68	2	638.3	88.2	87	7	22%	0.01
2801_mnz-20	0.011	0.0002	0.051	0.002	0.071	0.005	UDL	69	2	239.8	45.3	69	5	0%	0.00
2801_mnz-22	0.014	0.0005	0.236	0.014	0.493	0.050	UDL	92	3	3092.5	47.5	407	34	75%	0.01
2801_mnz-23	0.011	0.0003	0.080	0.010	0.118	0.012	UDL	73	2	1196	123	114	10	36%	(0.01)
2801_mnz-24	0.011	0.0003	0.072	0.008	0.105	0.011	UDL	71	2	985	113	101	10	30%	(0.02)
2801_mnz-25	0.011	0.0003	0.065	0.007	0.096	0.010	UDL	70	2	773	113	93	9	25%	(0.00)
2801_mnz-26	0.012	0.0004	0.120	0.012	0.203	0.015	UDL	76	3	1955.4	89.3	188	13	60%	0.01
2801_mnz-28	0.018	0.0005	0.049	0.025	1.041	0.067	UDL	118	3	3941.8	46	725	33	84%	0.03
2801_mnz-29	0.011	0.0003	0.062	0.006	0.092	0.008	UDL	69	2	673	104	90	7	24%	(0.00)
2801_mnz-30	0.014	0.0005	0.251	0.019	0.496	0.036	UDL	91	3	3190.4	60	409	24	75%	0.02
2801_mnz-31	0.011	0.0006	0.107	0.031	0.182	0.052	UDL	73	4	1748	265	170	45	58%	0.00
2801_mnz-32	0.012	0.0004	0.076	0.012	0.131	0.016	UDL	76	3	1094	158	125	15	39%	0.02
2801_mnz-33	0.012	0.0009	0.135	0.028	0.254	0.049	UDL	79	6	2163	181	230	40	66%	0.01
2801_mnz-34	0.015	0.0005	0.269	0.023	0.585	0.057	UDL	94	3	3299.5	67.2	467	36	80%	0.02
2801_mnz-35	0.011	0.0003	0.075	0.008	0.111	0.009	UDL	70	2	1068	107	107	8	35%	0.00
2801_mnz-36	0.011	0.0003	0.072	0.009	0.108	0.010	UDL	71	2	985	127	104	9	32%	(0.00)
2801_mnz-37	0.011	0.0004	0.070	0.009	0.097	0.008	UDL	68	2	927	132	94	8	28%	(0.00)
2801_mnz-39	0.013	0.0005	0.192	0.019	0.378	0.043	UDL	81	3	2758.6	81.3	325	32	75%	0.02
2801_mnz-40	0.011	0.0004	0.074	0.009	0.110	0.011	UDL	71	3	1041	123	106	10	34%	0.01
2801_mnz-41	0.013	0.0005	0.217	0.024	0.462	0.058	UDL	85	3	2957.8	89.3	386	40	75%	(0.00)
2801_mnz-42	0.011	0.0004	0.084	0.011	0.131	0.013	UDL	70	2	1292	127	125	12	45%	0.01
2801_mnz-43	0.011	0.0003	0.075	0.007	0.112	0.008	UDL	69	2	1067.6	93.8	108	8	36%	0.00
2801_mnz-44	0.011	0.0004	0.101	0.016	0.181	0.027	UDL	71	3	1642	147	169	23	58%	0.01
2801_mnz-45	0.013	0.0005	0.163	0.023	0.335	0.044	UDL	84	3	2486	119	293	33	72%	(0.00)
2801_mnz-46	0.012	0.0003	0.121	0.010	0.217	0.018	UDL	75	2	1970.2	73.7	199	15	63%	0.00
2801_mnz-47	0.012	0.0003	0.093	0.009	0.144	0.010	UDL	75	2	1487	91.7	137	8	45%	0.00
2801_mnz-48	0.014	0.0004	0.275	0.043	0.490	0.052	UDL	89	3	3334	122	405	35	75%	0.01
2801_mnz-50	0.011	0.0004	0.076	0.012	0.116	0.016	UDL	69	2	1094	158	112	15	38%	0.01
2801_mnz-51	0.011	0.0003	0.067	0.006	0.111	0.010	UDL	71	2	836.8	93.3	107	9	34%	0.01
2801_mnz-52	0.011	0.0003	0.080	0.007	0.129	0.009	UDL	73	5	1196.1	86.3	123	8	41%	0.01
2801_mnz-53	0.021	0.0011	0.444	0.030	1.478	0.185	4096401938	133	7	4064.6	50.5	922	76	86%	0.08
2801_mnz-54	0.011	0.0003	0.073	0.008	0.111	0.012	UDL	69	2	1013	111	106	11	35%	(0.02)
2801_mnz-55	0.011	0.0004	0.071	0.009	0.109	0.011	UDL	68	2	956	130	105	10	36%	0.01
2801_mnz-56	0.011	0.0002	0.054	0.005	0.077	0.005	UDL	67	2	370	104	76	5	11%	0.00
2801_mnz-57	0.012	0.0006	0.111	0.019	0.170	0.019	UDL	80	4	1815	155	160	16	50%	(0.00)
2801_mnz-58	0.012	0.0006	0.111	0.029	0.198	0.047	UDL	77	4	1815	237	183	40	58%	(0.01)
2801_mnz-59	0.011	0.0003	0.101	0.015	0.178	0.019	UDL	72	2	1642	138	166	17	57%	0.01
2801_mnz-61	0.011	0.0004	0.079	0.009	0.136	0.014	UDL	72	2	1171	113	129	12	45%	0.00
2801_mnz-62	0.011	0.0003	0.072	0.009	0.112	0.011	UDL	72	2	985	127	107	10	33%	(0.01)
2802_mnz-01	0.011	0.0004	0.063	0.005	0.096	0.007	UDL	70	3	707.3	84.4	93	7	24%	(0.00)
2802_mnz-03	0.011	0.0008	0.085	0.006	0.132	0.026	UDL	73	5	1314.7	68.5	126	23	43%	0.00
2802_mnz-04	0.011	0.0007	0.054	0.003	0.081	0.015	UDL	72	5	370	62.6	79	14	9%	0.00
2802_mnz-05	0.013	0.0006	0.143	0.010	0.262	0.021	UDL	83	4	2263.1	60.4	236	17	65%	0.01
2802_mnz-06	0.010	0.0007	0.055	0.004	0.076	0.015	UDL	67	4	411.2	81.3	74	14	10.4%	0.01
2802_mnz-08	0.011	0.0007	0.054	0.003	0.079	0.015	UDL	68	5	370	62.6	77	14	12%	(0.00)
2802_mnz-09	0.010	0.0007	0.052	0.004	0.072	0.014	UDL	67	4	284.4	88	71	13	5%	(0.00)
2802_mnz-10	0.011	0.0007	0.061	0.005	0.088	0.017	UDL	68	5	638.3	88.2	85	16	20%	0.01
2802_mnz-12	0.011	0.0004	0.070	0.005	0.117	0.012	UDL	73	3	927.4	73.4	113	11	35%	0.00
2802_mnz-13	0.013	0.0005	0.176	0.010	0.298	0.020	UDL	83	3	2614.8	47.4	265	16	69%	0.00
2802_mnz-14	0.011	0.0004	0.064	0.005	0.094	0.008	UDL	72	3	740.7	82.7	91	7	22%	0.01
2802_mnz-15	0.011	0.0007	0.051	0.004	0.073	0.014	UDL	68	5	239.8	90.4	71	13	5%	0.01
2802_mnz-16	0.010	0.0002	0.050	0.003	0.072	0.003	UDL	66	1	193.9	94.5	108	8	36%	(0.00)
2802_mnz-17	0.011	0.0003	0.065	0.006	0.092	0.006	UDL	67	2	773.4	97.1	90	5	25%	0.00
2802_mnz-18	0.011	0.0003	0.074	0.007	0.112	0.009	UDL	69	2	1040.6	90.4	70	4	7%	0.00
2802_mnz-24	0.011	0.0003	0.059	0.006	0.089	0.007	UDL	68	2	566	111	87	6	22%	0.00
2802_mnz-26	0.011	0.													

sample name	206Pb/238U	2 σ	207Pb/206Pb	2 σ	207Pb/235U	2 σ	206Pb/204Pb	Age 206/238	2 σ	Age 207/206	2 σ	Age 207/235	2 σ	%discordance	204Pb/206Pb
2802_mnz-50	0.011	0.0002	0.073	0.004	0.109	0.006	UDL	69	1	1013	55.6	105	5	34%	0.00
2802_mnz-51	0.011	0.0002	0.054	0.004	0.080	0.004	UDL	69	1	370	83.5	79	4	13%	(0.00)
2802_mnz-52	0.011	0.0002	0.053	0.004	0.077	0.005	UDL	68	1	327.8	85.7	75	4	10%	(0.00)
2802_mnz-53	0.011	0.0002	0.070	0.006	0.101	0.008	UDL	70	1	927.4	88	98	7	29%	0.00
2802_mnz-54	0.011	0.0003	0.070	0.007	0.107	0.008	UDL	73	2	927	103	104	7	30%	(0.00)
2802_mnz-55	0.011	0.0003	0.089	0.006	0.145	0.008	UDL	72	2	1403.3	64.6	137	7	48%	0.00
2802_mnz-56	0.012	0.0003	0.065	0.004	0.104	0.006	UDL	74	2	773.4	64.8	101	5	26%	0.00
2802_mnz-57	0.011	0.0002	0.051	0.004	0.077	0.005	UDL	68	2	239.8	90.4	75	5	10%	0.00
KJ_mnz-01	0.010	0.0004	0.052	0.003	0.073	0.004	UDL	67	2	284.4	66	71	3	7%	(0.00)
KJ_mnz-02	0.011	0.0003	0.050	0.002	0.072	0.003	UDL	69	2	193.9	46.5	71	3	3%	0.00
KJ_mnz-04	0.011	0.0003	0.066	0.004	0.091	0.004	UDL	68	2	805.4	63.5	88	3	23%	(0.00)
KJ_mnz-05	0.011	0.0003	0.058	0.003	0.084	0.004	UDL	68	2	528.8	56.7	82	4	17%	0.00
KJ_mnz-06	0.010	0.0003	0.048	0.002	0.068	0.003	UDL	67	2	98.2	49.3	67	3	0%	(0.00)
KJ_mnz-07	0.011	0.0002	0.049	0.002	0.070	0.002	UDL	69	1	146.8	47.9	69	2	0%	0.00
KJ_mnz-08	0.013	0.0003	0.071	0.005	0.119	0.006	UDL	80	2	956.5	72	114	6	30%	0.00
KJ_mnz-09	0.011	0.0002	0.048	0.002	0.072	0.003	UDL	70	1	98.2	49.3	70	3	0%	0.00
KJ_mnz-10	0.012	0.0003	0.060	0.002	0.095	0.004	UDL	75	2	602.6	36.1	92	4	18%	0.00
KJ_mnz-11	0.011	0.0003	0.064	0.008	0.099	0.010	UDL	70	2	741	132	96	9	27%	0.01
KJ_mnz-12	0.011	0.0002	0.055	0.003	0.082	0.004	UDL	73	1	411.2	61	80	3	9%	0.00
KJ_mnz-13	0.011	0.0003	0.062	0.004	0.097	0.005	UDL	73	2	673.1	69	94	5	22%	0.00
KJ_mnz-14	0.011	0.0002	0.052	0.002	0.075	0.003	UDL	71	1	284.4	44	73	3	4%	(0.00)
KJ_mnz-15	0.011	0.0002	0.054	0.003	0.076	0.004	UDL	68	1	370	62.6	74	3	9%	0.00
KJ_mnz-16	0.011	0.0004	0.072	0.005	0.112	0.007	UDL	74	2	985	70.7	108	7	32%	0.00
KJ_mnz-17	0.012	0.0004	0.074	0.011	0.117	0.011	UDL	75	3	1041	150	112	10	33%	0.00
KJ_mnz-18	0.011	0.0003	0.049	0.002	0.070	0.003	UDL	69	2	146.8	47.9	68	3	-1%	(0.00)
KJ_mnz-19	0.011	0.0003	0.054	0.002	0.075	0.003	UDL	69	2	370	41.8	74	3	7%	0.00
NK_mnz-02	0.010	0.0004	0.057	0.004	0.083	0.007	UDL	67	3	490.5	77.4	81	6	17%	0.00
NK_mnz-03	0.011	0.0004	0.058	0.006	0.084	0.008	UDL	70	2	529	113	82	8	15%	(0.00)
NK_mnz-04	0.011	0.0003	0.056	0.005	0.084	0.008	UDL	71	2	451.4	99.2	82	8	13%	(0.00)
NK_mnz-05	0.011	0.0004	0.053	0.003	0.074	0.005	UDL	68	3	327.8	64.3	73	5	7%	0.00
NK_mnz-07	0.011	0.0004	0.085	0.014	0.142	0.022	UDL	70	3	1315	160	135	19	48%	(0.03)
NK_mnz-08	0.011	0.0005	0.112	0.024	0.208	0.043	UDL	70	3	1831	194	192	36	64%	0.01
NK_mnz-09	0.011	0.0004	0.060	0.006	0.089	0.009	UDL	70	2	603	108	86	8	18%	(0.01)
NK_mnz-10	0.011	0.0004	0.055	0.004	0.082	0.007	UDL	73	2	411.2	81.3	80	6	8%	0.00
NK_mnz-11	0.016	0.0019	0.131	0.025	0.261	0.045	UDL	101	12	2111	167	236	36	58%	0.01
NK_mnz-12	0.011	0.0006	0.103	0.020	0.258	0.064	UDL	69	4	1678	179	233	52	71%	(0.02)
NK_mnz-13	0.011	0.0004	0.063	0.006	0.092	0.008	UDL	69	3	707	101	90	8	23%	(0.00)
NK_mnz-14	0.011	0.0005	0.056	0.005	0.084	0.007	UDL	70	3	451.4	99.2	82	7	14%	(0.01)
NK_mnz-16	0.011	0.0003	0.052	0.003	0.073	0.006	UDL	68	2	284.4	66	71	5	4%	(0.00)
NK_mnz-17	0.012	0.0004	0.068	0.007	0.119	0.011	UDL	76	2	868	107	114	10	34%	0.01
NK_mnz-18	0.011	0.0003	0.059	0.005	0.088	0.007	UDL	71	2	566.1	92.3	85	6	17%	(0.01)
NK_mnz-19	0.012	0.0003	0.059	0.006	0.096	0.009	UDL	75	2	566	111	93	8	20%	0.00
NK_mnz-20	0.011	0.0002	0.054	0.004	0.076	0.005	UDL	68	2	370	83.5	74	5	8%	0.00
NK_mnz-21	0.011	0.0005	0.099	0.020	0.182	0.027	UDL	74	3	1605	188	170	23	57%	(0.01)
NK_mnz-22	0.011	0.0002	0.055	0.005	0.081	0.005	UDL	68	1	411	102	79	5	14%	0.00
NK_mnz-23	0.011	0.0002	0.058	0.003	0.086	0.004	UDL	71	2	528.8	56.7	83	4	15%	0.00
NK_mnz-24	0.011	0.0004	0.078	0.013	0.135	0.015	UDL	74	2	1146	166	128	14	43%	0.00
NK_mnz-25	0.012	0.0004	0.061	0.005	0.097	0.006	UDL	75	2	638.3	88.2	94	5	20%	0.00
NK_mnz-28	0.011	0.0002	0.066	0.004	0.096	0.005	UDL	70	2	805.4	63.5	94	4	25%	0.00
NK_mnz-30	0.012	0.0004	0.088	0.006	0.140	0.008	UDL	75	2	1381.6	65.6	133	7	44%	(0.00)
NK_mnz-31	0.011	0.0003	0.075	0.008	0.123	0.012	UDL	73	2	1068	107	118	11	39%	0.01
NK_mnz-33	0.011	0.0003	0.055	0.005	0.078	0.005	UDL	68	2	411	102	76	5	11%	(0.00)
NK_mnz-34	0.012	0.0004	0.109	0.015	0.180	0.019	UDL	74	3	1782	125	168	16	56%	0.01
NK_mnz-36	0.011	0.0002	0.062	0.003	0.089	0.004	UDL	70	2	673.1	51.8	87	4	19%	0.00
TK02_mnz-01	0.013	0.0004	0.095	0.018	0.192	0.029	UDL	80	3	1527	178	178	25	55%	0.03
TK02_mnz-02	0.012	0.0002	0.066	0.006	0.107	0.011	UDL	75	1	805.4	95.2	103	10	28%	0.00
TK02_mnz-03	0.013	0.0007	0.137	0.032	0.297	0.079	UDL	85	5	2189	203	264	62	68%	(0.01)
TK02_mnz-04	0.012	0.0004	0.083	0.009	0.146	0.015	UDL	76	2	1268	106	138	13	45%	(0.01)
TK02_mnz-05	0.011	0.0003	0.047	0.003	0.067	0.003	UDL	69	2	48.2	76.3	66	3	-4%	(0.00)
TK02_mnz-06	0.012	0.0005	0.098	0.017	0.166	0.023	UDL	76	3	1586	162	156	20	51%	0.01
TK02_mnz-09	0.011	0.0004	0.068	0.007	0.110	0.009	UDL	73	3	868	107	106	8	31%	(0.00)
TK02_mnz-10	0.011	0.0003	0.054	0.006	0.083	0.007	UDL	72	2	370	125	81	7	12%	0.00
TK02_mnz-11	0.013	0.0006	0.100	0.019	0.176	0.019	UDL	85	4	1623	177	165	16	49%	0.00
TK02_mnz-12	0.011	0.0004	0.059	0.005	0.087	0.006	UDL	70	3	566.1	92.3	84	6	17%	(0.00)
TK02_mnz-13	0.011	0.0003	0.062	0.009	0.087	0.010	UDL	70	2	673	155	85	9	18%	(0.00)
TK02_mnz-15	0.012	0.0004	0.089	0.013	0.173	0.025	UDL	77	2	1403	140	162	22	53%	(0.01)
TK02_mnz-18	0.011	0.0004	0.051	0.002	0.071	0.003	UDL	67	2	239.8	45.3	69	2	3%	(0.00)
TK02_mnz-19	0.011	0.0005	0.067	0.010	0.103	0.012	UDL	72	3	837	155	100	11	28%	0.01
TK02_mnz-20	0.014	0.0006	0.270	0.022	0.598	0.060	UDL	90	4	3305.3	64	476	38	82%	(0.03)
YS_mnz-02	0.015	0.0011	0.229	0.019	0.460	0.096	UDL	98	7	3044.4	66.5	384	67	75%	0.01
YS_mnz-03	0.011	0.0008	0.065	0.008	0.097	0.021	UDL	69	5	773	130	94	19	27%	(0.01)
YS_mnz-06	0.012	0.0008	0.099	0.011	0.152	0.032	UDL	74	5	1605	104	143	28	49%	0.01
YS_mnz-07	0.011	0.0005	0.064	0.008	0.098	0.011	UDL	70	3	741	132	95	10	26%	0.00
YS_mnz-08	0.017	0.0013	0.322	0.023	0.784	0.160	UDL	107							

Appendix Table 3-2. Results of trace element analysis of monazite in the Busetsu granite

Concentration [ppm]	Y	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Th	U
2801_mnz-06	11643	121537	253259	25431	111445	16911	290	11279	1165	4597	537	819	52	172	12	23349	407
2801_mnz-10	4943	144351	274901	26502	104072	10768	134	6041	509	1847	213	302	20	59	4	26044	181
2801_mnz-12	5489	142683	270555	25204	101574	10179	178	5857	537	2044	241	342	22	75	5	20326	313
2801_mnz-15	9203	130646	262046	23780	102417	12283	232	8033	818	3484	448	724	51	189	16	24321	500
2801_mnz-17	3952	143695	271275	24702	98119	9995	185	5532	411	1451	172	225	18	68	5	28585	176
2801_mnz-18	6330	128205	249849	26115	106027	13473	242	8163	675	2550	293	429	27	93	7	43647	330
2801_mnz-19	10007	121568	250111	25488	108081	16831	349	11617	1186	4114	373	413	19	45	3	35934	1241
2801_mnz-20	14984	104785	240563	24714	110588	20370	214	15447	1844	7175	862	1189	82	263	18	23418	4295
2801_mnz-23	8601	123708	248941	27838	102126	13922	261	9139	866	3357	397	624	41	142	11	33908	422
2801_mnz-24	8558	128441	256440	25731	109505	14707	271	9314	868	3444	403	587	41	125	10	19831	299
2801_mnz-25	8070	131448	255396	22230	100106	12229	207	8294	840	3317	383	586	38	123	9	16546	334
2801_mnz-26	8552	129955	258378	26411	99723	13281	248	8846	860	3143	383	561	40	113	8	16731	343
2801_mnz-28	10749	108545	244402	26958	106290	20007	345	12637	1074	3722	463	653	47	153	15	28023	391
2801_mnz-29	10518	123450	256440	26599	102586	14100	217	9885	947	3522	471	725	52	163	13	13812	631
2801_mnz-30	8086	110813	241705	26454	103563	17722	294	10612	869	2941	357	506	34	93	8	24257	251
2801_mnz-33	7967	146642	267966	26181	95873	10944	180	7516	758	2849	344	500	37	107	8	11894	453
2801_mnz-34	7444	123296	251728	26633	101326	159937	273	9435	799	2736	330	464	28	72	6	23145	236
2801_mnz-35	6939	133810	262988	26240	99491	13151	253	8354	722	2597	329	484	33	98	8	24186	357
2801_mnz-36	8838	129273	252265	26386	102063	14479	242	9762	931	3334	402	546	35	100	7	18424	366
2801_mnz-38	11109	119237	250140	27197	102895	17394	316	11509	1095	4005	492	694	51	165	16	21394	475
2801_mnz-40	5414	134321	264013	26813	101874	12901	202	7529	567	2013	237	350	31	132	18	18372	436
2801_mnz-42	6247	137792	260265	25770	98197	11806	205	8178	749	2534	332	436	36	112	9	28417	395
2801_mnz-43	3388	150061	271441	26052	91750	9710	159	5964	458	1389	147	198	14	33	3	14921	244
2801_mnz-44	8363	120840	245606	25882	96833	13091	235	8718	819	3052	377	547	38	108	8	17904	419
2801_mnz-45	5907	122418	244727	26548	100177	14350	273	8767	692	2370	278	381	26	63	5	21302	178
2801_mnz-46	7574	130706	258506	25813	99577	14341	312	8898	799	2829	342	479	35	105	8	22805	401
2801_mnz-47	8838	129828	253520	26428	100434	13893	259	9420	860	3213	400	561	40	119	10	18921	327
2801_mnz-48	11683	111359	245657	26548	103923	18515	294	12804	1255	4567	534	724	48	134	10	19794	562
2801_mnz-49	4174	142916	265029	25685	96448	10909	222	6550	469	1638	194	270	19	48	4	17841	168
2801_mnz-50	6401	134048	260274	26232	101840	13479	241	8370	677	2469	291	402	26	70	5	20920	208
2801_mnz-51	7760	127031	257678	27522	102672	14730	277	9897	901	3257	383	520	39	104	8	23794	325
2801_mnz-52	12296	118751	235128	25283	102963	16299	293	11706	1097	4304	556	852	66	212	18	23845	564
2801_mnz-53	6784	135728	264739	26710	98402	11685	217	7590	676	2499	301	443	29	86	7	16900	243
2801_mnz-54	7920	125462	245452	26189	99825	14566	276	9813	821	2979	369	530	39	113	9	30119	362
2801_mnz-55	6801	127986	255527	26873	101789	13151	236	8600	726	2633	299	422	27	72	6	18919	228
2802_mnz-01	14400	113004	235074	25802	111631	19542	118	14055	1520	5766	694	1128	85	342	27	35003	851
2802_mnz-03	15995	118311	246056	26346	113551	19317	171	13101	1530	5743	704	1135	86	323	24	17280	727
2802_mnz-05	13554	104526	220509	26050	114860	20919	133	14650	1563	5673	667	1080	79	313	25	64250	1111
2802_mnz-06	20794	120954	250251	27101	109719	17667	179	15504	1886	8035	1009	1570	134	536	43	26565	1199
2802_mnz-10	16942	108343	226050	24981	112157	21617	162	14433	1653	7420	826	1350	105	401	31	26886	1216
2802_mnz-12	14345	116188	237063	23925	108847	18375	118	13667	1537	5713	684	1116	85	348	28	37514	1088
2802_mnz-13	15269	109277	231297	25676	115180	20237	144	14797	1618	6049	736	1161	89	344	27	37933	1043
2802_mnz-14	16359	119382	249226	28126	108404	16081	140	11755	1370	5642	694	1172	93	359	28	18660	860
2802_mnz-15	11124	119442	250433	23794	110082	17780	157	12848	1309	4669	505	814	68	289	28	31274	605
2802_mnz-16	15875	108895	232202	25104	103795	19766	168	15560	1938	8043	1001	1601	138	524	42	20328	1681
2802_mnz-17	17776	113976	245265	27026	105715	18783	174	12724	1533	6276	771	1171	91	349	26	19041	932
2802_mnz-18	16703	116415	230898	26155	99414	18981	156	13436	1509	6025	724	1120	93	337	27	16263	988
2802_mnz-19	19049	117012	242055	27291	100143	17627	154	13300	1644	6890	876	1360	112	415	33	14348	978
2802_mnz-20	18029	113687	238588	26146	104601	19404	157	14160	1636	6442	823	1166	95	350	27	16842	954
2802_mnz-28	18404	116620	239630	25984	102492	18722	157	14099	1731	7173	915	1445	127	472	38	15014	1035
2802_mnz-29	16678	115938	248236	27052	103941	18127	154	13688	1563	6220	757	1126	92	327	25	13994	852
2802_mnz-32	17196	113959	239442	26300	102723	18395	155	14185	1706	6841	875	1300	108	407	32	17805	1316
2802_mnz-34	15112	115452	240210	25839	101763	18110	164	11276	1362	5270	664	1004	91	369	29	27372	2180
2802_mnz-38	21649	113183	235284	26240	105295	21051	185	14804	1811	7460	954	1486	126	491	39	14329	982
2802_mnz-39	15251	115417	236027	24711	97545	16817	131	11817	1384	5528	710	1068	92	345	27	18365	1081
2802_mnz-40	19466	104367	221684	25531	103752	22836	153	17015	1996	8096	1025	1653	134	528	42	28366	1553
2802_mnz-41	19502	114966	239792	25796	103615	19205	154	14086	1734	7070	903	1429	121	476	38	16710	1005
2802_mnz-43	15262	114326	235207	26283	105098	19809	145	13652	1543	5815	712	1024	77	285	21	13801	606
2802_mnz-48	19780	103949	224980	26967	104738	21974	160	15404	1951	7558	952	1451	116	449	35	23302	1328
2802_mnz-50	17884	113516	228548	27505	101789	20680	184	16916	1874	8412	976	1396	103	420	32	24090	1044
2802_mnz-52	18197	115682	236847	27642	102329	19343	162	15394	1770	6918	907	1437	112	442	34	19400	1197
2802_mnz-55	11797	120994	251668	27069	103040	17118	209	9918	1165	4350	478	804	76	357	29	31361	4161
2802_mnz-56</																	

Concentration [ppm]	Y	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Th	U
KJ_mnz-01	15892	109526	238836	28026	108733	23267	146	15829	1694	6308	690	1087	98	422	37	23683	3769
KJ_mnz-05	13481	116091	239049	27838	110122	22112	120	15838	1725	6517	753	1217	110	462	40	31839	2892
KJ_mnz-06	16129	118521	246187	27129	104609	20240	150	15022	1660	6500	740	1216	108	466	39	15949	3311
KJ_mnz-08	11926	117438	240706	27924	105518	18972	155	14250	1713	7144	860	1518	143	647	57	17242	4804
KJ_mnz-09	11809	115264	242526	27573	103975	19783	158	14783	1741	7029	826	1416	131	579	50	16735	4186
KJ_mnz-10	12643	117788	247903	27411	103666	20568	149	15264	1890	7698	918	1585	152	694	61	20430	5632
KJ_mnz-11	5262	127312	251113	26898	97219	13634	128	8698	715	2251	217	271	15	35	3	38346	510
KJ_mnz-12	12807	114599	239101	26821	104198	20378	153	15371	1828	7362	918	1535	152	711	63	20106	5175
KJ_mnz-14	18000	112382	232117	27343	108459	21517	333	14600	1618	6275	722	1192	132	626	63	37688	1625
KJ_mnz-19	11686	117643	235464	27488	102843	21672	126	15011	1613	5833	672	1063	95	423	36	22246	3684
NK_mnz-02	11311	125414	240193	26995	98825	11521	199	7780	890	3769	521	889	69	258	22	11849	753
NK_mnz-03	7749	123802	240680	27021	108250	13631	216	8908	874	3054	325	414	25	73	5	27270	903
NK_mnz-06	9712	127665	250749	25582	105824	14672	112	9840	1154	4096	422	617	49	176	13	17140	5319
NK_mnz-09	7243	131283	253936	23828	104241	11639	238	7357	714	2547	300	428	30	99	8	23555	299
NK_mnz-11	10225	114892	231499	23287	101826	15287	154	11162	1264	4464	458	574	37	115	8	19326	979
NK_mnz-12	5546	113721	243808	27160	104861	18633	351	10544	827	2495	244	285	16	52	4	43299	232
NK_mnz-13	6954	131871	263139	24828	103403	11734	200	7368	719	2628	288	427	29	123	13	26239	1660
NK_mnz-14	11178	129447	257189	24409	105821	13554	181	9704	1144	4483	557	885	71	283	24	12640	667
NK_mnz-18	16245	122623	252772	26784	106704	17917	216	14513	1524	6086	728	1047	75	263	20	30576	754
NK_mnz-20	10981	121125	257500	26001	109933	18849	270	13129	1280	4661	475	585	43	134	10	30901	1341
NK_mnz-22	9302	118027	229547	22575	86983	10625	175	8829	950	3782	423	611	44	145	12	12936	879
NK_mnz-23	9889	120337	236719	24514	96688	16998	309	11100	928	3530	420	624	43	121	10	23345	315
NK_mnz-25	11242	125104	255569	26941	103280	15014	282	11255	1235	4336	493	685	46	120	9	22249	822
NK_mnz-30	11766	123936	244069	24976	94047	14471	126	11528	1272	4613	486	666	44	118	8	16503	1308
NK_mnz-31	18423	107283	232996	26420	102895	20706	250	18378	1932	8308	1109	1789	127	441	36	16425	707
NK_mnz-33	13254	125377	249636	26539	100143	14781	181	12358	1370	5150	561	780	51	139	10	17337	979
TK02_mnz-01	7082	130291	260015	24739	108430	14313	261	8255	744	2751	316	453	30	101	8	27941	202
TK02_mnz-03	12689	123592	253981	24335	101712	15756	231	11673	1384	5213	573	750	51	172	12	24652	1493
TK02_mnz-05	18658	118984	245925	26895	106355	17728	221	14595	1700	7230	978	1574	125	470	39	14079	682
TK02_mnz-06	8448	124854	250506	24959	103812	11797	127	7738	889	3466	375	528	41	149	11	13754	466
TK02_mnz-09	6334	114494	252960	26437	107341	17360	111	11355	984	2913	277	311	18	49	4	42943	571
TK02_mnz-11	6471	125388	245966	24922	101537	14149	301	8389	743	2547	286	403	31	142	16	36577	510
TK02_mnz-12	13393	108650	229348	23968	97662	13936	160	11831	1399	5339	616	832	57	182	14	27168	1513
TK02_mnz-13	5895	127899	261859	24520	105775	13240	213	7061	621	2204	260	363	28	100	9	30504	521
TK02_mnz-18	10265	118982	234029	25095	90824	14419	244	10882	1169	4180	464	652	44	142	11	28389	3259
TK02_mnz-19	6786	141236	265029	27240	92967	10995	152	6973	666	2388	295	434	29	92	7	13104	179
YS_mnz-02	11969	119209	243995	27898	112437	16736	254	10860	1174	4504	550	819	56	176	13	28482	501
YS_mnz-03	7174	127690	258421	24406	105069	14390	300	7962	710	2663	322	494	35	116	10	27396	321
YS_mnz-05	10650	121560	248808	27656	110130	14563	150	9221	982	3877	467	689	50	178	14	26483	440
YS_mnz-06	9108	124072	259312	25460	108973	15325	223	9953	959	3633	432	630	46	143	11	21710	339
YS_mnz-07	9823	126320	250996	26386	107761	15267	159	10211	991	3764	468	684	49	164	12	32333	376
YS_mnz-09	3417	126715	254659	25272	108859	15842	249	7138	495	1531	147	204	14	47	4	47083	414
YS_mnz-10	7399	140205	270368	26069	103895	9480	102	6436	696	2731	334	515	36	124	10	14592	354
YS_mnz-11	13271	115255	240726	25272	110573	17253	225	12231	1312	5202	683	1051	77	268	21	29652	539
YS_mnz-13	7144	126843	258962	25204	111914	14229	183	8122	695	2531	324	480	33	102	8	27020	315
YS_mnz-14	10111	114931	244308	26736	101737	17196	217	11461	1060	3869	474	666	43	126	9	15468	229
YS_mnz-15	11389	114897	228395	26369	104729	18283	230	12070	1181	4428	542	786	52	151	11	17235	247
YS_mnz-17	15752	107419	239109	27889	111219	21887	240	16260	1505	5891	728	1100	73	238	18	35490	577
YS_mnz-18	3119	111248	221667	25488	105158	19921	466	8649	476	1352	129	181	20	23	3	59412	148
YS_mnz-19	6975	148416	260094	25514	91271	8822	130	6473	623	2397	304	472	30	87	6	10354	204
YS_mnz-20	11743	126775	252931	26027	98299	14919	188	11332	1091	4158	542	803	55	169	13	17990	383
YS_mnz-25	23282	126204	261050	27958	103812	16170	271	12158	1405	6723	1006	1925	183	804	71	29426	5716

Appendix Table 3-3. Results of Nd isotope analysis of monazite in the Busetsu granite

Results	147Sm/144Nd	2s	143Nd/144Nd	2s	145Nd/144Nd	2s	150Nd/144Nd	2SE	Age (Ma)	**143Nd/144Nd(t)	2s
2801_mnz-06	0.1083	0.0024	0.512255	0.000048	0.348409	0.000028	0.236424	0.000035	68	0.512207	0.000048
2801_mnz-10	0.0778	0.0018	0.512287	0.000044	0.348412	0.000029	0.236390	0.000026	68	0.512252	0.000044
2801_mnz-15	0.0945	0.0023	0.512217	0.000044	0.348423	0.000031	0.236435	0.000036	68	0.512175	0.000044
2801_mnz-18	0.0924	0.0021	0.512226	0.000042	0.348440	0.000024	0.236429	0.000021	68	0.512185	0.000042
2801_mnz-23	0.1016	0.0023	0.512247	0.000048	0.348403	0.000026	0.236421	0.000029	68	0.512202	0.000048
2801_mnz-37	0.0831	0.0016	0.512242	0.000030	0.348421	0.000019	0.236395	0.000024	68	0.512205	0.000030
2801_mnz-42	0.0731	0.0012	0.512275	0.000032	0.348408	0.000017	0.236381	0.000023	68	0.512242	0.000032
2801_mnz-44	0.1013	0.0021	0.512229	0.000031	0.348431	0.000019	0.236375	0.000021	68	0.512184	0.000031
2801_mnz-45	0.1329	0.0022	0.512243	0.000033	0.348414	0.000021	0.236360	0.000029	68	0.512184	0.000033
2801_mnz-50	0.1043	0.0017	0.512258	0.000028	0.348432	0.000022	0.236373	0.000029	68	0.512211	0.000028
YS_mnz-7	0.1073	0.0025	0.512168	0.000047	0.348406	0.000029	0.236422	0.000035	68	0.512120	0.000047
YS_mnz-9	0.0847	0.0043	0.512123	0.000048	0.348411	0.000029	0.236416	0.000035	68	0.512085	0.000048
YS_mnz-10	0.1184	0.0027	0.512201	0.000046	0.348430	0.000027	0.236395	0.000031	68	0.512148	0.000046
YS_mnz-11	0.1135	0.0026	0.512178	0.000041	0.348415	0.000025	0.236387	0.000021	68	0.512128	0.000041
YS_mnz-12	0.0929	0.0023	0.512137	0.000062	0.348427	0.000031	0.236463	0.000056	68	0.512096	0.000062
YS_mnz-14	0.0999	0.0016	0.512108	0.000054	0.348467	0.000027	0.236546	0.000064	68	0.512063	0.000054
YS_mnz-15	0.1019	0.0017	0.512128	0.000038	0.348449	0.000022	0.236454	0.000051	68	0.512083	0.000038
YS_mnz-20	0.1104	0.0019	0.512167	0.000031	0.348419	0.000018	0.236398	0.000031	68	0.512118	0.000031
NK_mnz-02	0.0960	0.0022	0.512201	0.000050	0.348420	0.000028	0.236406	0.000035	68	0.512158	0.000050
NK_mnz-06	0.1105	0.0038	0.512265	0.000049	0.348449	0.000034	0.236572	0.000060	68	0.512216	0.000049
NK_mnz-12	0.1148	0.0034	0.512185	0.000052	0.348449	0.000039	0.236566	0.000084	68	0.512134	0.000052
NK_mnz-13	0.0863	0.0021	0.512187	0.000042	0.348429	0.000025	0.236439	0.000022	68	0.512149	0.000043
NK_mnz-14	0.1218	0.0028	0.512232	0.000048	0.348418	0.000028	0.236406	0.000035	68	0.512178	0.000048
TK02_mnz-03	0.1233	0.0028	0.512087	0.000073	0.348487	0.000047	0.236648	0.000065	68	0.512032	0.000073
TK02_mnz-05	0.1192	0.0027	0.512200	0.000045	0.348430	0.000027	0.236441	0.000029	68	0.512147	0.000045
TK02_mnz-09	0.1200	0.0027	0.512152	0.000046	0.348407	0.000027	0.236407	0.000030	68	0.512099	0.000046
TK02_mnz-11	0.0992	0.0041	0.512140	0.000041	0.348376	0.000026	0.236404	0.000015	68	0.512096	0.000041
TK02_mnz-12	0.1342	0.0031	0.512185	0.000049	0.348419	0.000028	0.236406	0.000031	68	0.512125	0.000049
TK02_mnz-13	0.0956	0.0015	0.512190	0.000035	0.348433	0.000026	0.236473	0.000038	68	0.512147	0.000035
TK02_mnz-18	0.1144	0.0021	0.511986	0.000035	0.348408	0.000018	0.236371	0.000037	68	0.511935	0.000035
TK02_mnz-19	0.1003	0.0016	0.512326	0.000033	0.348425	0.000020	0.236390	0.000021	68	0.512282	0.000033
2802_mnz-03	0.1216	0.0019	0.512208	0.000038	0.348440	0.000018	0.236391	0.000026	68	0.512153	0.000038
2802_mnz-05	0.1420	0.0032	0.512235	0.000044	0.348419	0.000026	0.236417	0.000025	68	0.512172	0.000044
2802_mnz-06	0.1170	0.0018	0.512194	0.000031	0.348426	0.000019	0.236391	0.000027	68	0.512142	0.000031
2802_mnz-10	0.1461	0.0023	0.512223	0.000030	0.348435	0.000018	0.236387	0.000027	68	0.512158	0.000030
2802_mnz-12	0.1244	0.0028	0.512225	0.000043	0.348401	0.000028	0.236412	0.000031	68	0.512169	0.000043
2802_mnz-13	0.1316	0.0030	0.512220	0.000042	0.348423	0.000029	0.236441	0.000027	68	0.512161	0.000042
2802_mnz-14	0.1137	0.0026	0.512187	0.000041	0.348410	0.000024	0.236418	0.000018	68	0.512136	0.000041
2802_mnz-15	0.1248	0.0029	0.512228	0.000044	0.348430	0.000027	0.236407	0.000031	68	0.512172	0.000044
2802_mnz-20	0.1252	0.0020	0.512205	0.000032	0.348425	0.000019	0.236367	0.000026	68	0.512149	0.000032
2802_mnz-52	0.1208	0.0020	0.512222	0.000033	0.348418	0.000018	0.236362	0.000025	68	0.512168	0.000033
KJ-4	0.1243	0.0020	0.512199	0.000030	0.348421	0.000019	0.236374	0.000026	68	0.512143	0.000030
KJ-5	0.1308	0.0021	0.512194	0.000030	0.348421	0.000018	0.236376	0.000017	68	0.512136	0.000030
KJ-6	0.1199	0.0019	0.512174	0.000033	0.348422	0.000019	0.236390	0.000026	68	0.512121	0.000033
KJ-8	0.1173	0.0018	0.512208	0.000033	0.348417	0.000019	0.236347	0.000023	68	0.512156	0.000033
KJ-9	0.1393	0.0022	0.512199	0.000030	0.348425	0.000017	0.236364	0.000022	68	0.512137	0.000030
KJ-10	0.1352	0.0021	0.512189	0.000033	0.348424	0.000018	0.236379	0.000023	68	0.512129	0.000033
KJ-11	0.0974	0.0018	0.512097	0.000037	0.348424	0.000024	0.236388	0.000026	68	0.512054	0.000037
KJ-12	0.1284	0.0020	0.512199	0.000028	0.348403	0.000019	0.236388	0.000019	68	0.512142	0.000028
KJ-14	0.1333	0.0021	0.512210	0.000031	0.348427	0.000018	0.236377	0.000032	68	0.512151	0.000031
KJ-19	0.1337	0.0021	0.512187	0.000029	0.348420	0.000020	0.236388	0.000019	68	0.512128	0.000029

Appendix Table 3-4. Geochemical composition of plagioclase in the Busetsu granite

Sample	Major elements [w%]						SrO	FeO	CaO	K2O	SiO2	Al2O3	MgO	Na2O	Total	An content	
	Na2O	MgO	Al2O3	SiO2	K2O	CaO											
YS-1_core	6.2	0.0	26.4	54.6	0.1	9.4	0.0	0.0	0.1	96.9						45.4%	
YS-1_rim	8.8	0.0	24.0	58.8	0.2	6.2	0.0	0.0	0.1	98.0						27.8%	
YS-3_core	6.9	0.0	27.6	55.0	0.2	9.7	0.0	0.0	0.0	99.5						43.2%	
YS-3_rim	8.5	0.0	24.5	59.2	0.2	6.3	0.0	0.0	0.0	98.8						29.1%	
YS-4_core	6.1	0.1	27.8	54.0	0.3	9.8	0.2	0.0	0.0	98.2						46.5%	
YS-4_rim	9.6	0.0	23.6	60.6	0.3	5.2	0.0	0.1	0.1	99.5						22.8%	
YS-5_core	7.7	0.0	25.2	57.1	0.2	7.6	0.0	0.1	0.1	98.0						35.0%	
YS-5_rim	10.7	0.0	22.4	61.4	0.2	4.3	0.0	0.1	0.1	99.1						18.4%	
YS-91_core	6.6	0.0	27.1	55.3	0.2	9.3	0.0	0.0	0.0	98.6						43.6%	
YS-91_rim	9.7	0.0	24.0	60.6	0.3	5.5	0.0	0.0	0.0	100.1						23.5%	
YS-g2_core	8.1	0.0	25.5	57.0	0.2	7.7	0.0	0.1	0.1	98.7						34.2%	
YS-g2_rim	8.8	0.0	24.0	58.6	0.3	6.2	0.0	0.1	0.1	98.1						27.5%	
YS-g5_core	8.0	0.0	25.2	56.7	0.2	7.7	0.0	0.1	0.1	97.9						34.3%	
YS-95_rim	8.8	0.0	23.1	60.1	0.3	5.3	0.0	0.1	0.1	97.8						24.7%	
YS-g3_core	7.4	0.0	26.9	54.7	0.1	9.7	0.0	0.1	0.1	99.0						41.8%	
YS-g3_rim	9.6	0.0	23.1	60.5	0.4	5.1	0.0	0.1	0.1	98.8						22.6%	
NK-1_core	8.6	0.0	23.1	57.8	0.2	5.9	0.0	0.1	0.1	95.9						27.5%	
NK-1_rim	10.2	0.0	22.7	60.8	0.3	4.6	0.0	0.1	0.1	98.6						19.7%	
NK-2_core	9.9	0.0	23.4	59.6	0.3	5.4	0.0	0.1	0.1	98.8						22.9%	
NK-2_rim	11.9	0.0	21.1	63.2	0.3	2.8	0.0	0.0	0.0	99.3						11.3%	
NK-4_core	10.3	0.0	24.6	60.7	0.2	5.9	0.0	0.0	0.0	101.7						23.7%	
NK-4_rim	11.2	0.0	23.6	62.5	0.3	4.7	0.0	0.0	0.0	102.3						18.5%	
NK-5_core	9.6	0.0	23.6	59.5	0.3	5.7	0.0	0.1	0.1	98.8						24.5%	
NK-5_rim	9.7	0.0	21.7	61.9	0.4	3.6	0.0	0.0	0.0	97.3						16.7%	
NK-g1_core	10.5	0.0	22.7	60.8	0.3	4.6	0.0	0.1	0.1	99.1						19.2%	
NK-g1_rim	9.8	0.0	23.1	60.0	0.3	5.1	0.0	0.0	0.0	98.4						22.2%	
NK-g2_core	10.2	0.0	24.7	60.3	0.2	6.0	0.0	0.0	0.0	101.4						24.4%	
NK-g2_rim	10.8	0.0	23.8	61.5	0.3	5.1	0.0	0.0	0.0	101.7						20.5%	
NK-g3_core	10.1	0.0	23.1	60.2	0.2	5.2	0.0	0.0	0.0	98.8						22.1%	
NK-g3_rim	10.0	0.0	22.0	61.5	0.3	3.9	0.0	0.1	0.1	97.8						17.4%	

Appendix Table 3-4. Geochemical composition of plagioclase in the Busetsu granite

Trace elements [ppm]		Li	Mg	Ti	Mn	Zn	Rb	Sr	Y	Ba
3	3	26	12	33	6	1	689	0	104	
4	4	519	40	126	13	3	1425	0	207	
6	6	234	68	130	27	8	1229	0	190	
3	3	21	15	40	12	1	822	0	156	
7	7	874	56	99	51	15	1025	2	194	
4	4	19	30	54	12	1	718	1	68	
2	2	7	17	29	7	1	665	1	113	
3	3	220	63	55	11	3	743	0	48	
2	2	58	15	8	11	2	560	0	40	
3	3	34	68	132	12	0	1180	0	129	
1	1	13	28	97	5	0	1235	1	197	
1	1	14	7	41	7	1	986	0	198	
1	1	18	63	75	11	1	775	1	309	
3	3	45	47	105	15	1	1475	2	408	
3	3	268	31	52	12	4	614	0	98	
2	2	27	19	132	10	0	852	0	236	
2	2	10	16	53	4	0	504	0	62	
18	18	678	28	191	7	1	1074	1	321	
3	3	19	13	8	13	1	360	0	46	
2	2	14	17	166	6	0	803	1	239	
3	3	19	15	69	11	1	473	0	25	
3	3	19	31	115	6	1	525	1	92	
2	2	11	9	50	4	0	290	0	41	
3	3	13	25	99	10	1	554	1	103	
1	1	13	15	103	6	1	539	1	74	
2	2	10	66	172	5	1	917	1	238	
3	3	16	13	93	10	1	588	0	69	
4	4	48	17	57	5	5	604	1	83	
4	4	26	29	50	12	2	545	0	37	

La	Ce	Pr	Nd	Sm	Eu	Gd	Pb
6	9	0.8	2.7	0.4	1.9	0.2	22
10	15	1.3	4.3	0.4	3.4	0.3	17
5	7	0.7	2.0	0.3	2.2	0.2	8
6	8	0.7	2.5	0.5	2.6	0.4	22
19	28	2.6	8.2	1.4	3.0	0.8	32
6	10	0.9	2.9	0.5	2.1	0.5	34
4	7	0.6	2.2	0.4	1.8	0.2	24
3	4	0.4	1.5	0.3	1.5	0.2	15
2	4	0.4	1.1	0.2	1.0	0.2	17
6	9	0.8	2.6	0.4	2.4	0.3	8
9	14	1.2	4.3	0.4	3.0	0.3	8
6	8	0.7	2.4	0.3	2.8	0.3	18
17	27	3.1	9.2	1.3	2.5	0.7	18
18	27	2.4	8.0	1.2	4.2	0.9	22
6	10	0.9	2.3	0.3	1.6	0.2	13
10	15	1.2	3.8	0.5	2.2	0.4	27
4	5	0.5	1.6	0.3	1.2	0.3	26
8	14	1.2	4.4	0.6	2.2	0.5	31
2	2	0.1	0.5	0.2	0.9	0.1	21
9	14	1.1	3.7	0.4	2.4	0.4	33
3	4	0.3	0.9	0.2	1.1	0.2	34
8	14	1.3	4.4	0.7	1.7	0.6	27
2	3	0.3	1.0	0.2	0.8	0.1	16
6	10	0.9	2.8	0.7	1.4	0.6	36
4	6	0.5	1.7	0.4	1.3	0.3	30
14	22	2.1	7.1	1.2	3.2	0.8	32
4	6	0.6	1.8	0.3	1.5	0.3	33
8	13	1.2	3.0	0.6	1.5	0.4	31
3	4	0.3	0.8	0.3	1.2	0.2	44

Appendix Table 3-5. Results of Sr isotope analysis of plagioclase in the Busetsu granite

Sample	87Sr/86Sr	2SE	84Sr/86Sr	2SE	85Rb/86Sr
NK_Pl-1_core	0.70989	0.00006	0.05701	0.00021	0.00964
NK_Pl-1_rim	0.70930	0.00009	0.05700	0.00031	0.01099
NK_Pl-2_core	0.70963	0.00005	0.05710	0.00020	0.00589
NK_Pl-2_rim	0.70980	0.00017	0.05733	0.00074	0.02034
NK_Pl-4_core	0.71007	0.00014	0.05679	0.00050	0.01169
NK_Pl-4_rim	0.70990	0.00015	0.05685	0.00046	0.02791
NK_Pl-5_core	0.71177	0.00010	0.05709	0.00030	0.03203
NK_Pl-5_rim	0.70995	0.00012	0.05698	0.00042	0.00919
NK_Pl-g1_core	0.71021	0.00013	0.05657	0.00037	0.01210
NK_Pl-g1_rim	0.70952	0.00010	0.05761	0.00034	0.01622
NK_Pl-g2_core	0.71159	0.00008	0.05715	0.00012	0.02895
NK_Pl-g2_rim	0.70992	0.00011	0.05685	0.00042	0.01573
NK_Pl-g3_core	0.70788	0.00016	0.05646	0.00041	0.39300
NK_Pl-g3_rim	0.70968	0.00011	0.05684	0.00037	0.04678
YS_Pl-1_core	0.71007	0.00006	0.05658	0.00021	0.13917
YS_Pl-1_rim	0.71023	0.00008	0.05609	0.00024	0.00432
YS_Pl-2_core	0.71359	0.00017	0.05586	0.00051	0.03096
YS_Pl-2_rim	0.71016	0.00009	0.05609	0.00051	0.00689
YS_Pl-3_core	0.70965	0.00006	0.05613	0.00019	0.06845
YS_Pl-3_rim	0.71032	0.00008	0.05617	0.00026	0.01932
YS_Pl-4_core	0.71023	0.00008	0.05588	0.00019	0.01668
YS_Pl-4_rim	0.71029	0.00009	0.05607	0.00030	0.00534
YS_Pl-5_core	0.71004	0.00006	0.05629	0.00024	0.19730
YS_Pl-5_rim	0.71016	0.00012	0.05615	0.00057	0.01615
YS_Pl-g1_core	0.71016	0.00005	0.05640	0.00012	0.03182
YS_Pl-g1_rim	0.71034	0.00009	0.05728	0.00028	0.00312
YS_Pl-g2_core	0.71012	0.00007	0.05633	0.00020	0.00346
YS_Pl-g2_rim	0.71020	0.00009	0.05669	0.00034	0.00678
YS_Pl-g5_core	0.71057	0.00007	0.05586	0.00025	0.00565
YS_Pl-g5_rim	0.71073	0.00023	0.05256	0.00145	0.00515
TK01_Pl-1_core	0.70978	0.00007	0.05626	0.00020	0.02099
TK01_Pl-1_rim	0.70965	0.00013	0.05666	0.00019	0.22427
TK01_Pl-10_core	0.70982	0.00007	0.05610	0.00026	0.13448
TK01_Pl-10_rim	0.70974	0.00007	0.05660	0.00026	0.00745
TK01_Pl-3_core	0.71047	0.00007	0.05614	0.00016	0.01736
TK01_Pl-3_rim	0.70978	0.00009	0.05570	0.00038	0.00466
TK01_Pl-4_core	0.70963	0.00009	0.05646	0.00027	0.02732
TK01_Pl-4_rim	0.70958	0.00007	0.05653	0.00031	0.00444
TK01_Pl-5_core	0.70976	0.00007	0.05614	0.00017	0.01092
TK01_Pl-5_rim	0.70976	0.00008	0.05653	0.00029	0.22090
TK01_Pl-7_core	0.70960	0.00008	0.05658	0.00022	0.02590
TK01_Pl-7_rim	0.70977	0.00012	0.05583	0.00035	0.00713
TK01_Pl-8_core	0.70963	0.00006	0.05626	0.00021	0.02995
TK01_Pl-8_rim	0.70959	0.00009	0.05626	0.00021	0.00439
TK01_Pl-g1_core	0.70979	0.00008	0.05614	0.00026	0.01973
TK01_Pl-g1_rim	0.70974	0.00010	0.05621	0.00023	0.00943
TK02_Pl-3_core	0.71032	0.00017	0.05673	0.00103	0.08343
TK02_Pl-3_rim	0.70981	0.00033	0.05988	0.00124	0.04869
TK02_Pl-4_core	0.71041	0.00021	0.05616	0.00117	0.03629
TK02_Pl-4_rim	0.71049	0.00024	0.05408	0.00096	0.04695
TK02_Pl-7_core	0.70914	0.00018	0.05656	0.00066	0.10392
TK02_Pl-7_rim	0.70990	0.00050	0.05666	0.00235	0.08549
TK02_Pl-9_core	0.70903	0.00020	0.05639	0.00067	0.06751
TK02_Pl-9_rim	0.71087	0.00049	0.05487	0.00294	0.14441
TK02_Pl-g1_core	0.71069	0.00027	0.05678	0.00104	0.11028
TK02_Pl-g1_rim	0.71119	0.00035	0.05093	0.00189	0.06126
TK02_Pl-g4_core	0.71095	0.00024	0.05450	0.00111	0.24733
TK02_Pl-g4_rim	0.71092	0.00038	0.05546	0.00179	0.05090
TK02_Pl-g5_core	0.71084	0.00025	0.05434	0.00113	0.08871
TK02_Pl-g5_rim	0.71103	0.00050	0.05450	0.00295	0.33805

Appendix Table 3-6. Results of whole-rock geochemical analyses of the Busetsu granites

Major elements [wt%]		NK	YS	TK01	TK02	2801	2802	KJ	MC
SiO <sub>2</sub>	74.0	72.7	71.6	73.9	72.3	73.2	72.01	74.78	0.49
TiO <sub>2</sub>	0.16	0.3	0.31	0.12	0.22	0.22	0.33	0.33	12.81
Al <sub>2</sub> O <sub>3</sub>	14.29	14.72	14.7	14.55	15.64	14.5	14.85	14.85	4.31
Fe <sub>2</sub> O <sub>3</sub>	2.1	2.9	3.3	1.54	2.5	2.4	2.8	2.8	0.1
MnO	0.04	0.05	0.05	0.07	0.05	0.05	0.05	0.05	0.1
MgO	0.23	0.44	0.44	0.11	0.37	0.27	0.45	0.45	1.32
CaO	1.74	2.51	2.85	1.15	2.04	2.1	2.71	2.71	0.42
Na <sub>2</sub> O	3.63	3.32	3.06	3.54	3.43	3.32	3.38	3.38	1.3
K <sub>2</sub> O	3.22	3.26	3.34	4.65	3.21	3.83	3.25	3.25	3.36
P <sub>2</sub> O <sub>5</sub>	0.10	0.12	0.11	0.15	0.13	0.08	0.12	0.12	0.07
Total	99.519	100.316	99.775	99.736	99.85	99.906	99.921	98.941	
Trace elements [ppm]		Rb	175	152	377	173	230	98	102
Ba	1065	1396	1588	703	1296	1221	824	824	442
Th	13	13	10	12	11	13	8	8	10
U	2.4	2.4	1.7	4.7	1.6	3.0	3	3	3
Nb	19	20	18	22	17	21	12	12	10
Ta	1.8	1.3	1.2	3.1	1.7	1.8	1	1	1
La	88	125	125	71	107	82	126	126	21
Ce	70	99	103	58	82	66	101	101	49
Pb	47	47	43	58	52	51	23	23	20
Pr	51	75	81	44	61	51	76	76	5
Sr	437	603	587	209	492	423	356	356	81
Nd	37	57	62	32	47	40	60	60	18
Zr	129	214	257	100	179	185	135	135	100
Hf	4.1	6.7	7.7	4.0	5.3	6.4	4	4	3
Sm	23	34	41	24	27	27	37	37	3
Eu	17	24	28	10	25	19	25	25	1
Ti	1600	3100	3100	1200	2200	2200	1978	1978	2937
Gd	19	27	31	19	22	22	30	30	4
Tb	15	22	26	18	15	21	23	23	1
Dy	11	15	20	13	9	18	15	15	3
Ho	9	14	17	11	7	16	13	13	1
Y	14	19	24	16	9	24	18	18	16
Er	9	14	18	10	7	17	13	13	2
Tm	7	12	15	8	5	16	11	11	0.3
Yb	7	12	16	8	6	15	12	12	2
Lu	7	11	15	7	6	15	11	11	0.3

Appendix table 4-1. Externally calibrated Sm-Nd isotopic data for EDR4

Results	147Sm/144Nd	2σ	143Nd/144Nd	2σ	145Nd/144Nd	2σ	150Nd/144Nd	2SE
20160511_EDR4-01	0.1102	0.0623	0.512255	0.000021	0.348413	0.000009	0.236462	0.000019
20160511_EDR4-02	0.1105	0.0625	0.512252	0.000015	0.348414	0.000009	0.236476	0.000017
20160511_EDR4-03	0.1105	0.0625	0.512255	0.000013	0.348411	0.000010	0.236493	0.000023
20160511_EDR4-04	0.1103	0.0624	0.512256	0.000015	0.348417	0.000010	0.236461	0.000018
20160511_EDR4-05	0.1108	0.0627	0.512255	0.000014	0.348420	0.000010	0.236486	0.000013
20160511_EDR4-06	0.1103	0.0624	0.512267	0.000015	0.348428	0.000012	0.236475	0.000020
20160512_EDR4-01	0.1014	0.0016	0.512259	0.000012	0.348421	0.000008	0.236425	0.000017
20160512_EDR4-02	0.1013	0.0016	0.512277	0.000020	0.348424	0.000011	0.236404	0.000026
20160512_EDR4-03	0.1010	0.0016	0.512248	0.000016	0.348416	0.000009	0.236431	0.000023
20160512_EDR4-04	0.1015	0.0016	0.512235	0.000015	0.348423	0.000011	0.236433	0.000020
20160512_EDR4-05	0.1017	0.0016	0.512253	0.000013	0.348423	0.000010	0.236416	0.000018
20160512_EDR4-06	0.1014	0.0016	0.512261	0.000020	0.348424	0.000011	0.236404	0.000022
20160512_EDR4-07	0.1016	0.0016	0.512239	0.000021	0.348416	0.000011	0.236392	0.000030
20160512_EDR4-08	0.1010	0.0016	0.512244	0.000019	0.348416	0.000010	0.236387	0.000025
20170418_EDR-01	0.1012	0.0016	0.512229	0.000026	0.348407	0.000017	0.236519	0.000026
20170418_EDR-02	0.1039	0.0016	0.512266	0.000021	0.348416	0.000016	0.236479	0.000027
20170418_EDR-03	0.1030	0.0016	0.512252	0.000028	0.348428	0.000018	0.236478	0.000033
20170418_EDR-04	0.1028	0.0016	0.512253	0.000030	0.348428	0.000016	0.236494	0.000027
20170418_EDR-05	0.1024	0.0016	0.512253	0.000034	0.348411	0.000014	0.236497	0.000024
20170418_EDR-06	0.1016	0.0016	0.512239	0.000028	0.348404	0.000016	0.236489	0.000024
20170418_EDR-07	0.0908	0.0014	0.512250	0.000021	0.348412	0.000019	0.236506	0.000026
20170418_EDR-08	0.0972	0.0015	0.512258	0.000023	0.348405	0.000014	0.236501	0.000033
20170418_EDR-09	0.1009	0.0016	0.512260	0.000025	0.348413	0.000016	0.236490	0.000021
20170418_EDR-10	0.1016	0.0016	0.512237	0.000029	0.348410	0.000017	0.236507	0.000033
20170418_EDR-11	0.1011	0.0016	0.512252	0.000024	0.348408	0.000014	0.236470	0.000024
20170418_EDR-12	0.1002	0.0016	0.512247	0.000020	0.348412	0.000014	0.236484	0.000025
20170704a_EDR-07	0.1009	0.0007	0.512265	0.000019	0.348414	0.000013	0.236381	0.000019
20170704a_EDR-08	0.1009	0.0007	0.512262	0.000020	0.348420	0.000012	0.236377	0.000018
20170704a_EDR-09	0.1000	0.0007	0.512274	0.000021	0.348412	0.000011	0.236404	0.000022
20170704a_EDR-10	0.1011	0.0007	0.512275	0.000029	0.348439	0.000010	0.236271	0.000017
20170704a_EDR-11	0.1008	0.0007	0.512248	0.000020	0.348437	0.000012	0.236312	0.000015
20170704a_EDR-12	0.1009	0.0007	0.512241	0.000024	0.348434	0.000012	0.236284	0.000024
20170704a_EDR-13	0.1005	0.0007	0.512250	0.000022	0.348422	0.000014	0.269686	0.000033
20170704a_EDR-14	0.1000	0.0007	0.512257	0.000020	0.348413	0.000013	0.269663	0.000029
20170704a_EDR-15	0.1009	0.0007	0.512260	0.000022	0.348419	0.000015	0.269664	0.000028
20170704a_EDR-16	0.1014	0.0009	0.512257	0.000026	0.348421	0.000016	0.236347	0.000020
20170704a_EDR-17	0.1001	0.0009	0.512242	0.000023	0.348416	0.000017	0.236351	0.000017
20170704a_EDR-18	0.1011	0.0009	0.512242	0.000025	0.348423	0.000016	0.236348	0.000022
20170704b_EDR-1	0.1018	0.0020	0.512260	0.000017	0.348419	0.000012	0.236295	0.000020
20170704b_EDR-2	0.1020	0.0020	0.512277	0.000016	0.348416	0.000012	0.236306	0.000021
20170704b_EDR-3	0.1013	0.0020	0.512263	0.000017	0.348419	0.000012	0.236316	0.000020
20170704b_EDR-4	0.1019	0.0020	0.512256	0.000024	0.348447	0.000012	0.236246	0.000015
20170704b_EDR-5	0.1016	0.0020	0.512250	0.000027	0.348430	0.000012	0.236264	0.000017
20170704b_EDR-6	0.1009	0.0020	0.512271	0.000026	0.348447	0.000013	0.236255	0.000020
20170704b_EDR-7	0.1026	0.0024	0.512256	0.000019	0.348443	0.000012	0.269639	0.000029
20170704b_EDR-8	0.1026	0.0024	0.512235	0.000020	0.348437	0.000013	0.236290	0.000017
20170704b_EDR-9	0.1026	0.0024	0.512256	0.000021	0.348441	0.000012	0.236289	0.000020
20170704b_EDR-10	0.1013	0.0021	0.512230	0.000031	0.348433	0.000013	0.236309	0.000021
20170704b_EDR-11	0.1014	0.0021	0.512267	0.000027	0.348439	0.000011	0.236303	0.000031
20170704b_EDR-12	0.1015	0.0021	0.512232	0.000028	0.348448	0.000012	0.236305	0.000021
20170704c_EDR-1	0.1019	0.0020	0.512265	0.000036	0.348440	0.000021	0.270293	0.000033
20170704c_EDR-2	0.1020	0.0020	0.512286	0.000035	0.348436	0.000022	0.270226	0.000036
20170704c_EDR-3	0.1018	0.0020	0.512274	0.000035	0.348437	0.000021	0.270120	0.000025
20170704c_EDR-4	0.1007	0.0020	0.512256	0.000035	0.348419	0.000012	0.269323	0.000022
20170704c_EDR-5	0.1013	0.0020	0.512252	0.000034	0.348404	0.000013	0.269354	0.000022
20170704c_EDR-6	0.1009	0.0020	0.512266	0.000034	0.348421	0.000014	0.269319	0.000022
20170704c_EDR-7	0.1014	0.0020	0.512252	0.000035	0.348437	0.000012	0.269049	0.000029
20170704c_EDR-8	0.1011	0.0020	0.512252	0.000037	0.348442	0.000011	0.268986	0.000023
20170704c_EDR-9	0.1011	0.0020	0.512229	0.000035	0.348447	0.000013	0.269021	0.000030
Mean	0.1021		<b>0.512254</b>		0.348424		0.243706	
2SD	0.0065		<b>0.000025</b>		0.000024		0.027733	
2SE	0.0009		0.000003		0.000003		0.003611	
%2SE	0.8%		0.001%		0.001%		1.482%	

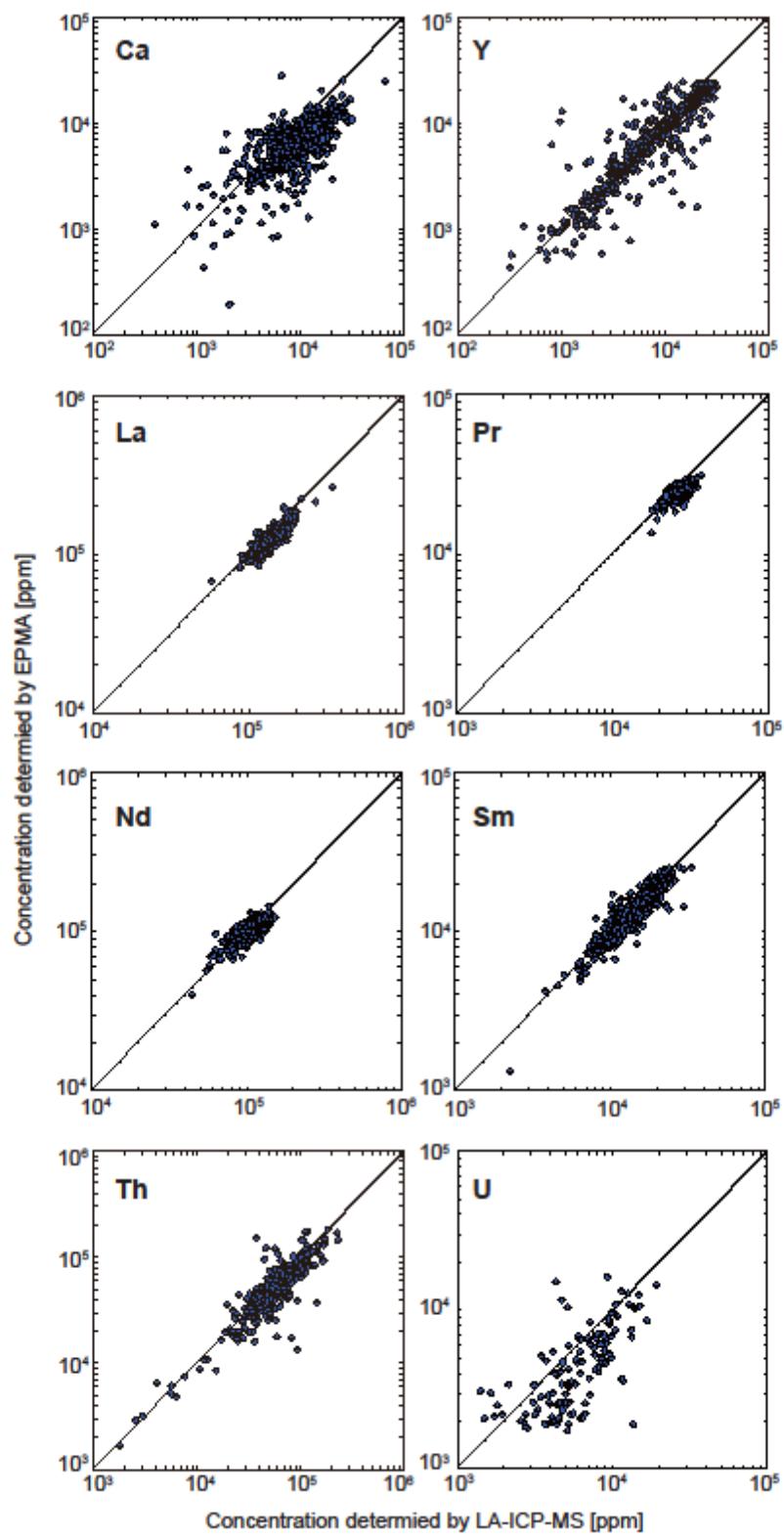
Sm/Nd and Nd isotopic ratios were calibrated based on the results of Synth-NSP and Synth-LCN, respectively.

Appendix Table 4-2. Results of trace element analysis of monzonite in granitic rocks in the Iwate arc

Sample #	Rock-type	Measurement	Content in RSE	Y	La	2.9865	250912	Pr	28631	Nd	1373.50	30721	Sm	1116.55	19647	Tb	17440	Dy	79.50	Tm	1514	Lu	149	Th	142790	3856		
1	From mafic-sens. granite	Nakano-1		1.2437	151.034	302.291																						
2	From mafic-sens. granite																											
3	From mafic-sens. granite	Kamikosono-1	Briphier	24854	15896	26744	27830		5.5	1147.50	1124.79	26056	1124.79	26660	5.6	187.25	2099	80.83	999	1521	121	599	44	108486	5144			
4	From mafic-sens. granite	Kamikosono-2	Briphier	25414	101.052	224.464	22660		5.6	187.74	211.9	1151.13	1151.13	27669	60	195.57	84.87	1024	982.3	1636	130	534	46	125295	6192			
5	From mafic-sens. granite	Kamikosono-3	Briphier	26340	231.766	26001	27250		5.6	171.88	199.5	1058.58	28537	1058.58	5.6	171.88	199.5	7724	945	1526	124	505	43	130240	4696			
6	From mafic-sens. granite	Kamikosono-4	Briphier	23.905	1660.04	220859	24814		5.6	171.88	199.5	122.760	120.131	220859	122.760	220859	5.6	171.88	199.5	7387	83.3	1526	124	505	43	126037	5839	
7	From mafic-sens. granite	Ono-1	Briphier	22814	160935	226760	22984		5.6	171.88	199.5	231.194	24451	231.194	5.6	171.88	199.5	12006	4493	678	13.57	140	704	72	171818	3588		
8	From mafic-sens. granite	Dakkar		23693	109791																							
9	From mafic-sens. granite	Nagai-1		5436	106625	256935	30641		8	124.99	137.1	269531	101186	269531	8	124.99	137.1	5142	579	890	72	262	20	83982	1853			
10	From tholeiitic-sens. granite	Nagai-1		5993	116096	257337	31786		7	138.35	164.8	28134	113564	28134	7	138.35	164.8	6106	712	94	112.2	59	351	26	344681	3459		
11	From tholeiitic-sens. granite	Nagai-1		5922	111.272	235677	27896		8	132.80	162.0	9960.93	25735	9960.93	8	132.80	162.0	6079	678	99	71	211	16	34741	1469			
12	From tholeiitic-sens. granite	Nagai-1		2717	121.765	242461	28762		8	132.80	162.0	16960	872.62	16960	8	132.80	162.0	721	2591	292	46	10	110992	2764				
13	From tholeiitic-sens. granite	Bischoff-1		191.10	128.777	268833	27971		15.6	122.49	127.50	1105.88	1105.88	1105.88	15.6	122.49	127.50	68037	779	119	89	34.4	28	344	28	2046	2838	
14	From tholeiitic-sens. granite	Bischoff-2		20380	120.131	244773	26761		15.6	122.49	127.50	121.514	270835	121.514	15.6	122.49	127.50	7671	829	117.6	91	34.4	27	344	27	126037	47380	
15	From tholeiitic-sens. granite	Bischoff-3		22807	131.041	270835	29518		15.6	122.49	127.50	132.31	289327	132.31	15.6	122.49	127.50	2767	10151	1087	102	381	30	45537	3382			
16	From tholeiitic-sens. granite	Bischoff-4		26916	133.151	289327	32343		15.6	122.49	127.50	132.31	30647	132.31	15.6	122.49	127.50	2767	10151	1559	123	469	37	66642	4546			
17	From tholeiitic-sens. granite	Bischoff-5		17746	164.089	292.277	32734		15.6	122.49	127.50	132.31	32734	132.31	15.6	122.49	127.50	2767	10151	1755	1382	112	444	37	21570	3384		
18	From tholeiitic-sens. granite	Bischoff-6		22149	14571	258073	27459		15.6	122.49	127.50	132.31	32734	132.31	15.6	122.49	127.50	2767	10151	1755	1382	112	444	35	42941	26551		
19	From tholeiitic-sens. granite	Nagai-1		40627	48441	220546	36790		5	48.74	49.70	487.55	86375	487.55	5	48.74	49.70	120.12	706	71.8	59	244	18	41102	4521			
20	From tholeiitic-sens. granite	Urumae-1		40413	47.489	361.67	361.67		6	48.75	49.70	1680.67	86375	1680.67	6	48.75	49.70	119.18	707	73.1	58	243	18	40757	4597			
21	From tholeiitic-sens. granite	Moria-1	Briphier	25746	942.912	234017	29340		6	48.75	49.70	328.43	328.43	328.43	6	48.75	49.70	93.22	914	118.1	80	285	21	689864	6902			
22	From tholeiitic-sens. granite	Moria-2	Briphier	30768	942.913	234044	31747		16	22.69	22.69	1369.49	32729	1369.49	16	22.69	22.69	113.13	1088	140.7	97	354	108	26	689864	6902		
23	From tholeiitic-sens. granite	Dakkar		23173	114.057	249711	29292		7	25.68	30.68	36239	36239	36239	7	25.68	30.68	285.8	719	70.2	39	108	10	42968	3976			
24	From tholeiitic-sens. granite	Dakkar		30518	108.567	249711	32265		8	31.06	32.3	1362.56	45831	1362.56	8	31.06	32.3	11602	937	93.14	51	143	10	57197	4912			
25	From tholeiitic-sens. granite	Dakkar		32223	101.915	249711	29366		10.4	33.54	34.87	211.85	25138	211.85	10.4	33.54	34.87	282.5	11673	13.50	299	299	13.50	143942	9769			
26	From tholeiitic-sens. granite	Briphier (core)		23418	109.243	267104	21532		8.7	28.96	28.96	1062.46	1062.46	1062.46	8.7	28.96	28.96	191.18	214.8	86.34	147	607	52	10591	10591			
27	From tholeiitic-sens. granite	Briphier (core)		22589	109.710	268930	23737		8.6	28.96	28.96	1067.03	24054	1067.03	8.6	28.96	28.96	192.7	239.7	11.97	142	740	64	136410	10155			
28	From tholeiitic-sens. granite	Dakkar (inner)		29520	100.223	264477	24054		10.5	11.26	11.26	11.26	2665	11.26	11.26	10.5	11.26	11.26	30.29	2665	11.26	11.26	11.26	11.26	8300	8300		
29	From tholeiitic-sens. granite	Suzhou-1		5572	36222	30065	31240		11	140.74	17.9826	17.9826	17.9826	17.9826	17.9826	11	140.74	17.9826	3957	866	11.4	57	8	8	7659	7659		
30	From tholeiitic-sens. granite	Suzhou-2		7100	39287	30254	31438		14	142.60	160.62	160.62	160.62	160.62	14	142.60	160.62	67704	4049	33.19	81	12	10	13	4	59966	4239	
31	From tholeiitic-sens. granite	Suzhou-3		4793	36679	15.4347	31348		8	15.98	15.98	15.98	15.98	15.98	8	15.98	15.98	5776	4449	75	75	75	75	8850	8850			
32	From tholeiitic-sens. granite	Suzhou-4		4793	34.343	149.80	29246		6	146.94	146.94	146.94	146.94	146.94	6	146.94	146.94	69161	4225	11.61	11	14	14	37	6	53363	5432	
33	From tholeiitic-sens. granite	Nissaki-1	Briphier	39415	53.640	199315	23235		3	45.78	45.78	179.16	177.53	179.16	3	45.78	45.78	69126	414.14	28.56	49	10	11	4	56445	6703		
34	From tholeiitic-sens. granite	Nissaki-2	Briphier	39083	181.697	199015	31329		4	181.697	181.697	76688	5151.2	76688	4	181.697	181.697	149.53	118.5	34.97	102	363	25	401	401			
35	From tholeiitic-sens. granite	Nissaki-3	Dakkar	30819	21.161	213612	35242		2	187.03	187.03	187.03	187.03	187.03	2	187.03	187.03	12762	12762	194.6	105.11	105.11	77	269	18	23477	1939	
36	From tholeiitic-sens. granite	Nissaki-4	Dakkar (core)	26119	86683	31246	32548		3	136.94	136.94	136.94	136.94	136.94	3	136.94	136.94	36589	31241	265.29	27	402	402	3595	3595			
37	From tholeiitic-sens. granite	Nissaki-5	Dakkar (inner)	7181	87038	34345	34345		6	146.94	146.94	146.94	146.94	146.94	6	146.94	146.94	30654	3141	26.51	35	3	3	82840	2993			

Appendix table 4-3. Sm-Nd isotopic data of monazite from granitic rocks in Japan	Sample #	Rock-type	Measurement syst	Counters in BSE	147Sm/144Nd	2 $\sigma$	143Nd/144Nd	2 $\sigma$	145Nd/144Nd	2 $\sigma$	150Nd/144Nd	2 $\sigma$	Age (Ma)	**143Nd/144Nd(0)	2 $\sigma$		
<b>3 From magnetic-series pegmatite</b>																	
Kamitomo, d-1	Darker	0.1488	0.0023	0.51257	0.00002	0.34842	0.00001	0.23643	0.00002	0.34842	0.00001	0.23640	0.00002	135	0.51244	0.00002	
Kamitomo, d-2	Darker	0.1493	0.0023	0.51257	0.00003	0.34842	0.00001	0.23642	0.00002	0.34841	0.00001	0.23640	0.00002	135	0.51244	0.00003	
Kamitomo, d-3	Darker	0.1483	0.0023	0.51257	0.00002	0.34841	0.00001	0.23640	0.00002	0.34843	0.00001	0.23637	0.00001	135	0.51244	0.00002	
Kamitomo, d-4	Darker	0.1476	0.0010	0.51259	0.00002	0.34842	0.00001	0.23637	0.00001	0.34843	0.00001	0.23637	0.00001	135	0.51245	0.00002	
Kamitomo, d-5	Darker	0.1476	0.0010	0.51258	0.00002	0.34843	0.00001	0.23637	0.00001	0.34843	0.00001	0.23637	0.00001	135	0.51245	0.00002	
Kamitomo, b-1	Brighter	0.1482	0.0023	0.51256	0.00002	0.34843	0.00001	0.23643	0.00003	0.34843	0.00001	0.23642	0.00002	135	0.51243	0.00002	
Kamitomo, b-2	Brighter	0.1474	0.0023	0.51257	0.00002	0.34841	0.00001	0.23642	0.00002	0.34842	0.00001	0.23641	0.00002	135	0.51244	0.00002	
Kamitomo, b-3	Brighter	0.1477	0.0023	0.51258	0.00002	0.34844	0.00001	0.23634	0.00002	0.34843	0.00001	0.23634	0.00002	135	0.51245	0.00002	
Kamitomo, b-4	Brighter	0.1480	0.0010	0.51258	0.00002	0.34844	0.00001	0.23634	0.00002	0.34843	0.00001	0.23634	0.00002	135	0.51244	0.00003	
Kamitomo, b-5	Brighter	0.1479	0.0010	0.51257	0.00003	0.34843	0.00001	0.23634	0.00002	0.34843	0.00001	0.23632	0.00002	135	0.51244	0.00003	
Kamitomo, b-6	Brighter	0.1473	0.0010	0.51259	0.00003	0.34842	0.00001	0.23632	0.00002	0.34842	0.00001	0.23632	0.00002	135	0.51246	0.00003	
<b>4 From magnetic-series pegmatite</b>																	
Oro, d-1	Darker	0.1030	0.0012	0.51219	0.00002	0.34841	0.00002	0.23641	0.00004	0.34841	0.00002	0.23641	0.00004	61.9	0.51215	0.00002	
Oro, d-2	Darker	0.1033	0.0012	0.51218	0.00003	0.34842	0.00002	0.23646	0.00003	0.34841	0.00003	0.23646	0.00003	61.9	0.51213	0.00003	
Oro, d-3	Darker	0.1034	0.0012	0.51217	0.00003	0.34841	0.00002	0.23644	0.00003	0.34843	0.00003	0.23644	0.00003	61.9	0.51213	0.00003	
Oro, d-4	Darker	0.1048	0.0008	0.51218	0.00002	0.34842	0.00002	0.23626	0.00002	0.34843	0.00002	0.23626	0.00002	61.9	0.51214	0.00002	
Oro, d-5	Darker	0.1048	0.0008	0.51217	0.00003	0.34843	0.00002	0.23626	0.00003	0.34843	0.00002	0.23626	0.00003	61.9	0.51213	0.00003	
Oro, d-6	Darker	0.1053	0.0008	0.51219	0.00004	0.34843	0.00002	0.23626	0.00003	0.34843	0.00002	0.23626	0.00003	61.9	0.51214	0.00004	
Oro, b-1	Brighter	0.1025	0.0015	0.51219	0.00003	0.34839	0.00002	0.23644	0.00001	0.34839	0.00002	0.23644	0.00001	61.9	0.51215	0.00003	
Oro, b-2	Brighter	0.0989	0.0013	0.51218	0.00002	0.34841	0.00001	0.23641	0.00001	0.34841	0.00001	0.23641	0.00001	61.9	0.51214	0.00002	
Oro, b-3	Brighter	0.1014	0.0010	0.51218	0.00003	0.34843	0.00002	0.23626	0.00002	0.34843	0.00002	0.23626	0.00002	61.9	0.51212	0.00003	
Oro, b-4	Brighter	0.1022	0.0010	0.51217	0.00003	0.34842	0.00002	0.23627	0.00002	0.34840	0.00001	0.23640	0.00004	61.9	0.51215	0.00003	
Oro, b-5	Brighter	0.1018	0.0013	0.51219	0.00003	0.34843	0.00001	0.23629	0.00001	0.34844	0.00001	0.23629	0.00001	61.9	0.51212	0.00002	
Oro, b-6	Brighter	0.1003	0.0010	0.51217	0.00003	0.34843	0.00002	0.23627	0.00002	0.34843	0.00001	0.23627	0.00002	61.9	0.51212	0.00003	
<b>7 From ilmenite-series pegmatite</b>																	
Utsunime-1			0.3166	0.0050	0.51269	0.00002	0.34842	0.00001	0.23641	0.00002	0.34842	0.00001	0.23641	0.00002	106	0.51247	0.00002
Utsunime-2			0.3165	0.0050	0.51271	0.00002	0.34842	0.00001	0.23639	0.00002	0.34842	0.00001	0.23639	0.00002	106	0.51249	0.00002
Utsunime-3			0.3165	0.0050	0.51271	0.00002	0.34842	0.00001	0.23639	0.00002	0.34842	0.00001	0.23639	0.00002	106	0.51249	0.00002
Utsunime-4			0.3160	0.0050	0.51270	0.00002	0.34842	0.00001	0.23640	0.00001	0.34842	0.00001	0.23640	0.00001	106	0.51248	0.00002
Utsunime-5			0.3148	0.0022	0.51272	0.00003	0.34843	0.00001	0.23632	0.00001	0.34843	0.00001	0.23632	0.00001	106	0.51251	0.00003
Utsunime-6			0.3153	0.0022	0.51272	0.00003	0.34843	0.00001	0.23630	0.00001	0.34843	0.00001	0.23630	0.00001	106	0.51250	0.00003
Utsunime-7			0.3147	0.0022	0.51273	0.00003	0.34844	0.00001	0.23629	0.00001	0.34844	0.00001	0.23629	0.00001	106	0.51251	0.00003
Utsunime-8			0.3150	0.0022	0.51273	0.00002	0.34844	0.00001	0.23629	0.00001	0.34844	0.00001	0.23629	0.00001	106	0.51252	0.00002
<b>8 From ilmenite-series pegmatite</b>																	
Morita, d-1	Darker	0.1833	0.0029	0.51261	0.00002	0.34842	0.00001	0.23638	0.00002	0.34842	0.00001	0.23638	0.00002	106	0.51248	0.00002	
Morita, d-2	Darker	0.1725	0.0027	0.51259	0.00002	0.34842	0.00001	0.23638	0.00002	0.34842	0.00001	0.23638	0.00002	106	0.51247	0.00002	
Morita, d-3	Darker	0.1811	0.0028	0.51259	0.00002	0.34842	0.00001	0.23642	0.00002	0.34842	0.00001	0.23642	0.00002	106	0.51246	0.00002	
Morita, d-4	Darker	0.1879	0.0037	0.51260	0.00004	0.34838	0.00002	0.23631	0.00001	0.34840	0.00002	0.23631	0.00001	106	0.51247	0.00004	
Morita, d-5	Darker	0.1838	0.0036	0.51258	0.00004	0.34840	0.00002	0.23631	0.00001	0.34840	0.00002	0.23631	0.00001	106	0.51245	0.00004	
Morita, d-6	Darker	0.1886	0.0037	0.51261	0.00004	0.34840	0.00002	0.23631	0.00001	0.34840	0.00002	0.23631	0.00001	106	0.51248	0.00004	
Morita, b-1	Brighter	0.1661	0.0026	0.51258	0.00001	0.34842	0.00001	0.23641	0.00002	0.34841	0.00001	0.23641	0.00002	106	0.51246	0.00001	
Morita, b-2	Brighter	0.1695	0.0027	0.51258	0.00002	0.34841	0.00001	0.23643	0.00002	0.34841	0.00001	0.23643	0.00002	106	0.51246	0.00002	
Morita, b-3	Brighter	0.1659	0.0026	0.51257	0.00002	0.34842	0.00001	0.23642	0.00002	0.34842	0.00001	0.23642	0.00002	106	0.51246	0.00002	
Morita, b-4	Brighter	0.1877	0.0037	0.51259	0.00004	0.34840	0.00002	0.23626	0.00002	0.34840	0.00002	0.23626	0.00002	106	0.51246	0.00004	
Morita, b-5	Brighter	0.1879	0.0037	0.51259	0.00004	0.34840	0.00002	0.23625	0.00002	0.34840	0.00002	0.23625	0.00002	106	0.51246	0.00004	
Morita, b-6	Brighter	0.1875	0.0037	0.51259	0.00004	0.34840	0.00002	0.23627	0.00002	0.34840	0.00002	0.23627	0.00002	106	0.51246	0.00004	
<b>9 From ilmenite-series pegmatite</b>																	
Shichirouai, b-1	Brighter	0.2070	0.0024	0.51258	0.00002	0.34842	0.00001	0.23648	0.00002	0.34842	0.00001	0.23648	0.00002	106	0.51243	0.00002	
Shichirouai, b-2	Brighter	0.2084	0.0022	0.51261	0.00003	0.34841	0.00002	0.23625	0.00002	0.34841	0.00002	0.23625	0.00002	106	0.51246	0.00001	
Shichirouai, c-3	Brighter	0.1429	0.0029	0.51262	0.00003	0.34843	0.00002	0.23625	0.00002	0.34843	0.00002	0.23625	0.00002	106	0.51252	0.00003	
Shichirouai, c-4	Brighter	0.1425	0.0029	0.51261	0.00003	0.34842	0.00002	0.23626	0.00003	0.34842	0.00002	0.23626	0.00003	106	0.51251	0.00003	
Shichirouai, c-5	Brighter	0.1433	0.0029	0.51262	0.00003	0.34843	0.00002	0.23625	0.00002	0.34843	0.00002	0.23625	0.00002	106	0.51251	0.00003	
Shichirouai, c-6	Brighter	0.1433	0.0029	0.51260	0.00003	0.34842	0.00002	0.23627	0.00002	0.34842	0.00002	0.23627	0.00002	106	0.51251	0.00003	
Shichirouai, rim-1	Darker	0.1548	0.0024	0.51261	0.00002	0.34841	0.00001	0.23650	0.00002	0.34841	0.00001	0.23650	0.				

Appendix Figure 5-1



Appendix Table 5-1: Results from U-Pb dating of the secondary monazite standards

	$^{206}\text{Pb}/^{238}\text{U}$	$2\text{L}_v$	$^{206}\text{Pb}/^{238}\text{U}$	$2\text{L}_v$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\text{L}_v$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\text{L}_v$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\text{L}_v$	Disc. <sup>†</sup> (%)
	ratio		age (Ma)		ratio		age (Ma)		ratio		
<b>Manangotry</b>											
Manangotry-1	0.090	0.002	557	9	0.060	0.001	618	36			10%
Manangotry-2	0.090	0.001	557	9	0.059	0.001	557	38			0%
Manangotry-3	0.089	0.002	552	12	0.059	0.001	566	34			2%
Manangotry-4	0.091	0.002	559	12	0.062	0.003	659	89			16%
Manangotry-5	0.091	0.002	561	13	0.060	0.001	619	46			10%
Manangotry-6	0.090	0.002	557	12	0.059	0.001	585	44			5%
Manangotry-7	0.091	0.002	563	10	0.058	0.001	535	45			-6%
Manangotry-8	0.090	0.002	558	9	0.060	0.001	592	40			6%
Manangotry-9	0.089	0.001	550	8	0.057	0.003	493	134			-12%
Manangotry-10	0.092	0.002	566	10	0.056	0.003	468	135			-22%
Manangotry-11	0.093	0.002	574	10	0.059	0.001	561	44			-3%
Manangotry-12	0.092	0.002	569	13	0.061	0.001	645	52			12%
Manangotry-13	0.089	0.003	549	17	0.059	0.003	574	119			5%
Manangotry-14	0.089	0.002	550	15	0.058	0.003	516	114			-7%
Manangotry-15	0.089	0.001	552	9	0.059	0.001	553	38			0%
Manangotry-16	0.089	0.001	551	7	0.057	0.001	502	39			-10%
Manangotry-17	0.088	0.002	545	12	0.059	0.002	585	57			7%
Manangotry-18	0.090	0.002	554	11	0.060	0.001	598	53			8%
<b>16-F-6</b>											
16-F-6-1	0.535	0.015	2761	62	0.204	0.004	2859	30			4%
16-F-6-2	0.545	0.011	2806	44	0.204	0.004	2859	33			2%
16-F-6-3	0.561	0.009	2871	39	0.199	0.002	2820	17			-2%
16-F-6-4	0.572	0.009	2915	37	0.203	0.002	2847	20			-3%
16-F-6-5	0.547	0.011	2814	45	0.205	0.002	2865	20			2%
16-F-6-6	0.560	0.012	2867	50	0.204	0.002	2862	17			0%
16-F-6-7	0.556	0.012	2850	49	0.202	0.002	2840	18			0%
16-F-6-8	0.557	0.011	2853	45	0.202	0.002	2846	18			0%
16-F-6-9	0.558	0.009	2857	36	0.204	0.002	2859	19			0%
16-F-6-10	0.543	0.022	2795	91	0.203	0.003	2852	20			3%
16-F-6-11	0.561	0.012	2870	49	0.202	0.003	2839	24			-1%
16-F-6-12	0.558	0.009	2858	39	0.203	0.003	2854	27			0%
16-F-6-13	0.521	0.019	2705	80	0.206	0.004	2873	34			7%
16-F-6-14	0.546	0.011	2810	46	0.203	0.003	2852	22			2%
16-F-6-15	0.533	0.018	2753	74	0.202	0.003	2845	26			4%
16-F-6-16	0.571	0.011	2914	45	0.199	0.003	2820	22			-4%
16-F-6-17	0.524	0.019	2718	81	0.206	0.006	2874	45			7%
16-F-6-18	0.552	0.013	2833	54	0.205	0.004	2863	35			1%
16-F-6-19	0.557	0.007	2856	30	0.199	0.002	2821	18			-2%
16-F-6-20	0.559	0.008	2862	33	0.199	0.002	2818	20			-2%
16-F-6-21	0.531	0.010	2747	44	0.204	0.003	2857	24			5%
16-F-6-22	0.550	0.011	2826	47	0.200	0.003	2823	25			0%
16-F-6-23	0.551	0.015	2829	61	0.206	0.004	2871	31			2%
16-F-6-24	0.532	0.026	2749	110	0.205	0.005	2862	42			5%
16-F-6-25	0.531	0.019	2748	78	0.210	0.007	2904	51			7%
16-F-6-26	0.512	0.015	2663	65	0.213	0.005	2930	37			11%
16-F-6-27	0.560	0.012	2865	49	0.204	0.004	2861	35			0%
16-F-6-28	0.550	0.011	2824	46	0.204	0.004	2860	33			2%
16-F-6-29	0.554	0.012	2843	48	0.200	0.002	2830	18			-1%
16-F-6-30	0.586	0.019	2973	77	0.205	0.003	2864	21			-5%
16-F-6-31	0.550	0.007	2826	31	0.204	0.003	2860	21			1%
16-F-6-32	0.543	0.008	2796	34	0.202	0.003	2842	20			2%
16-F-6-33	0.567	0.019	2897	78	0.200	0.012	2827	94			-3%
16-F-6-34	0.566	0.021	2891	88	0.200	0.011	2828	92			-3%
16-F-6-35	0.546	0.010	2807	40	0.201	0.002	2832	16			1%
16-F-6-36	0.542	0.008	2791	35	0.199	0.002	2816	21			1%
16-F-6-37	0.552	0.009	2832	38	0.201	0.003	2832	23			0%
16-F-6-38	0.560	0.010	2865	41	0.202	0.003	2841	24			-1%
16-F-6-39	0.540	0.009	2783	36	0.201	0.003	2835	24			2%
16-F-6-40	0.543	0.009	2795	38	0.201	0.003	2833	22			2%

<sup>†</sup>The % discordance was calculated as  $\{1 - [^{206}\text{Pb}/^{238}\text{U}]_{\text{measured}} / [^{206}\text{Pb}/^{238}\text{U}]_{\text{207Pb}^*/^{206}\text{Pb}^* \text{ age}}$

Nile River	$^{238}\text{U}/^{206}\text{Pb}$	2 $\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$	2 $\sigma$	$^{206}\text{Pb}/^{204}\text{Pb}$	$^{238}\text{U}/^{206}\text{Pb}^*$	2 $\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	2 $\sigma$	$^{206}\text{Pb}^*/^{238}\text{U}$	2 $\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	2 $\sigma$	Disc. <sup>†</sup>	internal structure	Used da	
	ratio	ratio	ratio	ratio	ratio	ratio	ratio	ratio	ratio	age (Ma)	age (Ma)	age (Ma)	age (Ma)	(%)			
NIL-mnz-1	10.00	0.244	0.0596	0.0012	10.00	0.244	0.0596	0.0012	614	14	590	45	-4% patchy-brighter	✓			
NIL-mnz-2	10.38	0.228	0.0599	0.0019	10.38	0.228	0.0599	0.0019	592	12	600	67	1% unzoned	✓			
NIL-mnz-3	10.26	0.217	0.0598	0.0013	10.26	0.217	0.0598	0.0013	599	12	596	46	-1% unzoned	✓			
NIL-mnz-4	9.67	0.201	0.0619	0.0016	9.67	0.201	0.0619	0.0016	633	13	671	54	6% unzoned	✓			
NIL-mnz-5	10.20	0.242	0.0606	0.0012	10.20	0.242	0.0606	0.0012	602	14	626	43	4% unzoned	✓			
NIL-mnz-6	10.51	0.227	0.0584	0.0013	10.51	0.227	0.0584	0.0013	586	12	543	50	8% unzoned	✓			
NIL-mnz-7	10.35	0.228	0.0601	0.0016	10.35	0.228	0.0601	0.0016	593	12	607	56	2% unzoned	✓			
NIL-mnz-8	8.51	0.203	0.0614	0.0013	8.53	0.204	0.0633	0.0013	731	17	717	45	2% unzoned	✓			
NIL-mnz-9	10.58	0.196	0.0587	0.0007	10.58	0.196	0.0587	0.0007	583	8	585	27	-5% core	✓			
NIL-mnz-10	2.58	0.140	0.4962	0.0252	2.58	0.140	0.4962	0.0252	2112	97	4200	77	58% unzoned				
NIL-mnz-11	10.26	0.179	0.0585	0.0010	10.26	0.179	0.0585	0.0010	599	10	549	38	10% unzoned				
NIL-mnz-12	10.36	0.136	0.0599	0.0010	10.36	0.136	0.0599	0.0010	593	7	601	36	1% patchy-dark	✓			
NIL-mnz-13	9.84	0.160	0.0610	0.0014	9.84	0.160	0.0610	0.0014	623	10	638	50	2% patchy-dark	✓			
NIL-mnz-14	9.64	0.200	0.0600	0.0026	9.64	0.203	0.0600	0.0026	636	13	604	94	-6% unzoned	✓			
NIL-mnz-15	8.52	0.177	0.0765	0.0039	8.52	0.177	0.0765	0.0039	703	14	1109	102	38% unzoned				
NIL-mnz-16	8.01	0.181	0.0645	0.0026	8.01	0.181	0.0645	0.0026	757	16	757	84	0% unzoned	✓			
NIL-mnz-17	10.35	0.154	0.0593	0.0011	10.35	0.154	0.0593	0.0011	594	8	578	40	-3% unzoned				
NIL-mnz-18	8.21	0.163	0.0637	0.0011	8.21	0.163	0.0637	0.0011	740	14	732	38	-1% mosaic	✓			
NIL-mnz-19	8.97	0.221	0.0615	0.0014	8.97	0.221	0.0615	0.0014	681	16	655	48	-4% unzoned	✓			
NIL-mnz-20	10.08	0.202	0.0586	0.0014	10.08	0.202	0.0586	0.0014	610	12	554	53	-11% unzoned	✓			
NIL-mnz-21	9.62	0.167	0.0592	0.0015	9.62	0.167	0.0592	0.0015	638	11	574	56	-12% unzoned	✓			
NIL-mnz-22	9.94	0.238	0.0601	0.0015	9.94	0.238	0.0601	0.0015	618	14	608	55	-2% core	✓			
NIL-mnz-23	10.31	0.191	0.0603	0.0015	10.31	0.191	0.0603	0.0015	596	11	613	55	3% unzoned	✓			
NIL-mnz-24	10.02	0.225	0.0614	0.0017	10.02	0.225	0.0614	0.0017	612	13	653	58	6% patchy-dark	✓			
NIL-mnz-25	10.09	0.200	0.0595	0.0014	10.09	0.206	0.0595	0.0014	609	12	586	52	-4% unzoned	✓			
NIL-mnz-26	9.90	0.247	0.0595	0.0013	9.90	0.247	0.0595	0.0013	620	15	584	48	-6% patchy-brighter	✓			
NIL-mnz-27	9.91	0.190	0.0600	0.0021	9.91	0.191	0.0600	0.0021	620	11	604	75	-3% mosaic	✓			
NIL-mnz-28	8.56	0.250	0.1538	0.0226	8.56	0.250	0.1538	0.0226	636	18	2389	250	74% unzoned				
NIL-mnz-29	10.77	0.313	0.0631	0.0016	10.77	0.313	0.0631	0.0016	570	16	713	52	21% unzoned				
NIL-mnz-30	9.30	0.158	0.0672	0.0024	9.30	0.158	0.0672	0.0024	653	11	844	74	23% patchy-dark				
NIL-mnz-31	10.33	0.162	0.0605	0.0012	10.33	0.162	0.0605	0.0012	595	9	623	42	5% patchy-dark	✓			
NIL-mnz-32	9.69	0.218	0.0617	0.0018	9.69	0.218	0.0617	0.0018	632	14	664	62	5% unzoned	✓			
NIL-mnz-33	10.07	0.194	0.0601	0.0010	10.07	0.194	0.0601	0.0010	613	11	612	39	0% unzoned	✓			
NIL-mnz-34	7.74	0.142	0.0651	0.0014	7.74	0.142	0.0651	0.0014	783	14	778	46	-1% patchy-brighter	✓			
NIL-mnz-35	9.97	0.217	0.0594	0.0011	9.97	0.217	0.0594	0.0011	616	13	611	40	-6% core	✓			
NIL-mnz-36	8.77	0.196	0.0629	0.0011	8.77	0.196	0.0629	0.0011	695	15	703	37	1% unzoned	✓			
NIL-mnz-37	6.98	0.145	0.0910	0.0048	6.98	0.145	0.0910	0.0048	839	16	1446	101	43% unzoned				
NIL-mnz-38	10.25	0.200	0.0607	0.0014	10.25	0.208	0.0607	0.0014	599	12	628	50	5% unzoned	✓			
NIL-mnz-39	10.26	0.245	0.0595	0.0009	10.26	0.245	0.0595	0.0009	599	14	586	32	-3% unzoned	✓			
NIL-mnz-40	9.90	0.194	0.0598	0.0013	9.90	0.196	0.0598	0.0013	620	12	595	47	-4% unzoned	✓			
NIL-mnz-41	10.04	0.225	0.0580	0.0011	10.04	0.225	0.0580	0.0011	612	13	531	41	-16% core				
NIL-mnz-42	10.37	0.178	0.0600	0.0009	10.37	0.178	0.0600	0.0009	593	10	603	32	2% patchy-dark	✓			
NIL-mnz-43	5.90	0.088	0.0748	0.0011	5.90	0.086	0.0748	0.0011	1005	21	1064	31	6% patchy-brighter				
NIL-mnz-44	11.43	0.360	0.0591	0.0014	11.43	0.360	0.0591	0.0014	540	12	571	51	5% core	✓			
NIL-mnz-45	9.87	0.226	0.0620	0.0009	9.87	0.226	0.0620	0.0009	620	14	675	32	8% unzoned				
NIL-mnz-46	9.58	0.187	0.0625	0.0011	9.58	0.187	0.0625	0.0011	638	16	692	39	8% core	✓			
NIL-mnz-47	10.26	0.254	0.0590	0.0008	10.26	0.254	0.0590	0.0008	599	14	567	31	-6% patchy-dark	✓			
NIL-mnz-48	10.27	0.238	0.0611	0.0008	10.27	0.238	0.0611	0.0008	598	12	642	29	7% unzoned				
NIL-mnz-49	9.22	0.195	0.0619	0.0021	9.22	0.195	0.0619	0.0021	663	14	672	74	1% mosaic	✓			
NIL-mnz-50	10.52	0.270	0.0599	0.0011	10.52	0.270	0.0599	0.0011	585	13	622	42	5% unzoned	✓			
NIL-mnz-51	10.16	0.210	0.0607	0.0014	10.16	0.210	0.0607	0.0014	604	13	628	48	4% core	✓			
NIL-mnz-52	9.53	0.199	0.0617	0.0009	9.53	0.199	0.0617	0.0009	642	14	664	31	3% unzoned	✓			
NIL-mnz-53	9.46	0.170	0.0628	0.0012	9.46	0.170	0.0628	0.0012	646	12	702	41	8% unzoned				
NIL-mnz-54	10.48	0.246	0.0590	0.0008	10.48	0.246	0.0590	0.0008	587	13	566	152	-4% unzoned	✓			
NIL-mnz-55	11.54	0.254	0.0587	0.0013	11.54	0.254	0.0587	0.0013	534	11	557	48	4% mosaic	✓			
NIL-mnz-56	9.99	0.327	0.0668	0.0019	9.99	0.327	0.0668	0.0019	609	13	833	60	27% mosaic				
NIL-mnz-57	10.24	0.215	0.0602	0.0010	10.24	0.215	0.0602	0.0010	599	11	612	37	2% patchy-dark	✓			
NIL-mnz-58*	10.23	0.243	0.0605	0.0012	10.23	0.243	0.0605	0.0012	590	12	622	35	6% unzoned	✓			
NIL-mnz-59	10.41	0.261	0.0605	0.0012	10.41	0.269	0.0605	0.0012	590	12	622	42	5% unzoned	✓			
NIL-mnz-60	10.24	0.214	0.0606	0.0019	10.24	0.214	0.0606	0.0019	599	15	626	47	4% unzoned	✓			
NIL-mnz-61	9.67	0.211	0.0611	0.0013	9.67	0.211	0.0611	0.0013	634	13	643	47	1% patchy-dark	✓			
NIL-mnz-62	10.04	0.220	0.0607	0.0014	10.04	0.220	0.0607	0.0014	611	13	630	49	3% mosaic	✓			
NIL-mnz-63	9.29	0.167	0.0636	0.0008	9.29	0.167	0.0636	0.0008	657	11	730	27	10% unzoned				
NIL-mnz-64	9.21	0.198	0.0632	0.0014	9.21	0.199	0.0632	0.0014	663	14	714	48	7% unzoned	✓			
NIL-mnz-65	8.24	0.181	0.0635	0.0014	8.24	0.181	0.0635	0.0014	738	15							

$^{238}\text{U}/^{206}\text{Pb}$	$2\sigma$	$^{207}\text{Pb}^{206}\text{Pb}$	$2\sigma$	$^{206}\text{Pb}^{204}\text{Pb}$	$^{238}\text{U}/^{206}\text{Pb}^*$	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\sigma$	$^{206}\text{Pb}^*/^{238}\text{U}$	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\sigma$	Disc. <sup>†</sup>	internal structure	Used data <sup>‡†</sup>	
ratio		ratio		ratio	ratio		ratio		age (Ma)	age (Ma)			(%)			
<b>Niger River</b>																
NGR1-mnz-1*	10.87	0.239	0.0691	0.0022	1123	11.02	0.260	0.0560	0.0024	560	13	452	93	-25% core		
NGR1-mnz-2	10.62	0.219	0.0586	0.0011		10.62	0.219	0.0586	0.0011	580	11	554	40	-5% patchy-brg	✓	
NGR1-mnz-3	10.27	0.221	0.0653	0.0037		10.27	0.221	0.0653	0.0037	599	12	784	119	25% mosaic		
NGR1-mnz-4 (	10.22	0.277	0.0611	0.0018		10.22	0.277	0.0611	0.0018	600	16	644	62	7% core	✓	
NGR1-mnz-4 (	10.11	0.173	0.0605	0.0020		10.11	0.173	0.0605	0.0020	600	10	596	73	-2% rim		
NGR1-mnz-5	10.01	0.242	0.0636	0.0029		10.01	0.242	0.0636	0.0029	611	14	728	96	16% unzoned	✓	
NGR1-mnz-6	10.37	0.259	0.0604	0.0019		10.37	0.259	0.0604	0.0019	592	14	617	67	4% unzoned	✓	
NGR1-mnz-7	11.22	0.225	0.0590	0.0010		11.22	0.225	0.0590	0.0010	549	11	568	37	3% unzoned	✓	
NGR1-mnz-8	10.52	0.253	0.0620	0.0013		10.52	0.253	0.0620	0.0013	583	13	673	46	14% unzoned		
NGR1-mnz-9	38.02	0.732	0.0507	0.0018		38.02	0.732	0.0507	0.0018	167	3	227	83	27% patchy-dar	✓	
NGR1-mnz-10	10.76	0.241	0.0637	0.0018		10.76	0.241	0.0637	0.0018	569	12	732	61	23% patchy-brighter		
NGR1-mnz-11	10.66	0.228	0.0609	0.0018		10.66	0.228	0.0609	0.0018	577	12	637	62	10% unzoned	✓	
NGR1-mnz-12	9.55	0.184	0.0725	0.0039		9.55	0.184	0.0725	0.0039	633	12	1000	108	38% unzoned		
NGR1-mnz-13	10.35	0.179	0.0597	0.0015		10.35	0.179	0.0597	0.0015	600	10	593	54	0% unzoned	✓	
NGR1-mnz-14	11.10	0.224	0.0615	0.0017		11.10	0.224	0.0615	0.0017	554	11	658	58	16% core		
NGR1-mnz-15	10.86	0.193	0.0605	0.0018		10.86	0.193	0.0605	0.0018	567	10	621	66	9% core	✓	
NGR1-mnz-16	10.52	0.204	0.0590	0.0014		10.52	0.204	0.0590	0.0014	585	11	569	51	-3% patchy-brg	✓	
NGR1-mnz-17	10.89	0.240	0.0596	0.0011		10.89	0.240	0.0596	0.0011	565	12	590	41	4% mosaic	✓	
NGR1-mnz-18	9.90	0.234	0.0599	0.0015		9.90	0.234	0.0599	0.0015	620	14	602	54	-3% unzoned	✓	
NGR1-mnz-19	11.17	0.222	0.0589	0.0010		11.17	0.222	0.0589	0.0010	552	10	565	38	2% patchy-dar	✓	
NGR1-mnz-20	10.91	0.197	0.0598	0.0016		10.91	0.197	0.0598	0.0016	564	10	596	58	5% unzoned		
NGR1-mnz-21	10.07	0.165	0.0640	0.0021		10.07	0.165	0.0640	0.0021	607	9	741	71	18% patchy-darker		
NGR1-mnz-22	10.58	0.231	0.0645	0.0018		10.58	0.231	0.0645	0.0018	578	12	759	58	24% unzoned		
NGR1-mnz-23	10.38	0.157	0.0591	0.0009		10.38	0.157	0.0591	0.0009	592	9	570	34	-4% patchy-brg	✓	
NGR1-mnz-24	10.81	0.225	0.0597	0.0009		10.81	0.225	0.0597	0.0009	569	11	593	34	4% patchy-dar	✓	
NGR1-mnz-25	11.05	0.265	0.0587	0.0010		11.05	0.265	0.0587	0.0010	558	13	557	37	0% patchy-brg	✓	
NGR1-mnz-26	10.93	0.242	0.0607	0.0010		10.93	0.242	0.0607	0.0010	563	12	628	36	11% unzoned		
NGR1-mnz-27	11.59	0.272	0.0592	0.0009		11.59	0.272	0.0592	0.0009	532	12	575	32	7% unzoned	✓	
NGR1-mnz-28	9.65	0.202	0.0607	0.0013		9.65	0.202	0.0607	0.0013	635	13	628	46	-1% patchy-brg	✓	
NGR1-mnz-29	11.19	0.272	0.0609	0.0010		11.19	0.272	0.0609	0.0010	550	13	636	36	14% patchy-darker		
NGR1-mnz-30	9.84	0.214	0.0604	0.0010		9.84	0.214	0.0604	0.0010	623	13	619	35	-1% patchy-dar	✓	
NGR1-mnz-31	10.33	0.243	0.0620	0.0016		10.33	0.243	0.0620	0.0016	593	13	674	57	12% core		
NGR1-mnz-32	10.70	0.235	0.0580	0.0010		10.70	0.235	0.0580	0.0010	576	12	528	37	-9% core	✓	
NGR1-mnz-33	10.45	0.203	0.0595	0.0008		10.45	0.203	0.0595	0.0008	588	11	584	31	-1% mosaic	✓	
NGR1-mnz-34	10.27	0.181	0.0650	0.0018		10.27	0.181	0.0650	0.0018	594	10	776	58	24% unzoned		
NGR1-mnz-35	10.21	0.199	0.0585	0.0011		10.21	0.199	0.0585	0.0011	602	11	550	41	-10% unzoned		
NGR1-mnz-36	11.07	0.198	0.0583	0.0009		11.07	0.198	0.0583	0.0009	557	10	540	35	-3% core	✓	
NGR1-mnz-37	10.37	0.205	0.0677	0.0029	2503	10.43	0.211	0.0618	0.0016	590	11	668	56	12% unzoned		
NGR1-mnz-38	9.80	0.223	0.0624	0.0011		9.80	0.223	0.0624	0.0011	624	14	686	39	9% unzoned		
NGR1-mnz-39	10.55	0.216	0.0574	0.0012		10.55	0.216	0.0574	0.0012	584	11	505	48	-16% mosaic		
NGR1-mnz-40	11.14	0.242	0.0591	0.0008		11.14	0.242	0.0591	0.0008	553	11	569	31	3% patchy-dar	✓	
NGR1-mnz-41	10.48	0.237	0.0590	0.0010		10.48	0.237	0.0590	0.0010	587	13	566	37	-4% patchy-brg	✓	
NGR1-mnz-42	10.79	0.211	0.0584	0.0008		10.79	0.211	0.0584	0.0008	571	11	546	29	-5% patchy-brg	✓	
NGR1-mnz-43	9.70	0.210	0.0689	0.0040		9.70	0.210	0.0689	0.0040	625	13	897	120	31% unzoned		
NGR1-mnz-44	10.79	0.246	0.0586	0.0014		10.79	0.246	0.0586	0.0014	571	12	550	53	4% unzoned		
NGR1-mnz-45	10.52	0.203	0.0594	0.0015		10.52	0.203	0.0594	0.0015	585	11	581	54	-1% unzoned		
NGR1-mnz-46	10.57	0.311	0.0584	0.0012		10.57	0.311	0.0584	0.0012	583	16	544	45	-8% patchy-dar	✓	
NGR1-mnz-47	10.71	0.163	0.0582	0.0009		10.71	0.163	0.0582	0.0009	575	8	539	32	-7% unzoned	✓	
NGR1-mnz-48	10.27	0.204	0.0597	0.0010		10.27	0.204	0.0597	0.0010	598	11	593	38	-1% unzoned	✓	
NGR1-mnz-49	10.10	0.215	0.0599	0.0012		10.10	0.215	0.0599	0.0012	608	12	598	42	-2% mosaic	✓	
NGR1-mnz-50	10.67	0.212	0.0586	0.0010		10.67	0.212	0.0586	0.0010	577	11	551	37	-5% patchy-dar	✓	
NGR1-mnz-51	10.67	0.220	0.0593	0.0010		10.67	0.220	0.0593	0.0010	577	11	577	35	0% mosaic	✓	
NGR1-mnz-52	10.35	0.217	0.0597	0.0009		10.35	0.217	0.0597	0.0009	594	12	591	34	-1% unzoned		
NGR1-mnz-53	9.49	0.216	0.0690	0.0036		9.49	0.216	0.0690	0.0036	639	14	898	108	29% mosaic		
NGR1-mnz-54	10.68	0.246	0.0633	0.0021		10.68	0.246	0.0633	0.0021	573	13	717	71	20% mosaic		
NGR1-mnz-55	10.54	0.208	0.0591	0.0011		10.54	0.208	0.0591	0.0011	584	11	570	39	-3% unzoned	✓	
NGR1-mnz-56	11.60	0.236	0.0588	0.0011		11.60	0.236	0.0588	0.0011	532	10	651	39	5% patchy-dar	✓	
NGR1-mnz-57	10.58	0.213	0.0597	0.0010		10.58	0.213	0.0597	0.0010	581	11	593	35	2% core	✓	
NGR1-mnz-58	10.84	0.229	0.0592	0.0010		10.84	0.229	0.0592	0.0010	568	11	575	35	1% unzoned	✓	
NGR1-mnz-59	10.62	0.238	0.0601	0.0011		10.62	0.238	0.0601	0.0011	579	12	609	41	5% mosaic	✓	
NGR1-mnz-60	10.61	0.194	0.0601	0.0011		10.61	0.194	0.0601	0.0011	579	10	608	40	5% patchy-brg	✓	
NGR1-mnz-61	10.80	0.211	0.0730	0.0041		10.80	0.211	0.0730	0.0041	561	10	1015	113	46% unzoned		
NGR1-mnz-62	11.04	0.221	0.0593	0.0010		11.04	0.221	0.0593	0.0010	558	11	579	37	4% unzoned	✓	
NGR1-mnz-63	9.82	0.201	0.0626	0.0012		9.82	0.201	0.0626	0.0012	623	12	694	39	10% unzoned		
NGR1-mnz-64	9.54	0.189	0.0613	0.0014		9.54	0.189	0.0613	0.0014	641	12	650				

	$^{238}\text{U}/^{206}\text{Pb}$ ratio	$2\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$ ratio	$2\sigma$	$^{206}\text{Pb}/^{204}\text{Pb}$ ratio	$2\sigma$	$^{238}\text{U}/^{206}\text{Pb}^*$ ratio	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$ ratio	$2\sigma$	$^{206}\text{Pb}^*/^{238}\text{U}$ age (Ma)	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$ age (Ma)	$2\sigma$	Disc. <sup>†</sup> (%)	internal structure	Used data <sup>‡</sup>		
<b>Congo River</b>																			
CNG2-mnz-1	5.85	0.104	0.0743	0.0013	5.85	0.104	0.0743	0.0013	1014	17	1049	35	3%	unzoned	✓				
CNG2-mnz-2	1.67	0.027	0.2289	0.0048	1.67	0.027	0.2289	0.0048	3015	39	3044	34	1%	patchy-dar	✓				
CNG2-mnz-3	6.42	0.127	0.0725	0.0014	6.42	0.127	0.0725	0.0014	929	17	1001	38	7%	patchy-brighter	✓				
CNG2-mnz-4	9.79	0.175	0.0608	0.0011	9.79	0.175	0.0608	0.0011	626	11	632	40	1%	patchy-bri	✓				
CNG2-mnz-5	6.38	0.118	0.0718	0.0013	6.38	0.118	0.0718	0.0013	935	16	981	37	5%	mosaic	✓				
CNG2-mnz-6	5.57	0.104	0.0748	0.0014	5.57	0.104	0.0748	0.0014	1064	18	1063	38	0%	patchy-bri	✓				
CNG2-mnz-7	12.29	0.258	0.0581	0.0015	12.29	0.258	0.0581	0.0015	503	10	535	57	6%	unzoned	✓				
CNG2-mnz-8	4.19	0.080	0.0873	0.0019	4.19	0.080	0.0873	0.0019	1379	24	1368	41	-1%	unzoned	✓				
CNG2-mnz-9	9.61	0.190	0.0622	0.0014	9.61	0.190	0.0622	0.0014	637	12	679	49	6%	mosaic	✓				
CNG2-mnz-10	9.67	0.212	0.0625	0.0013	9.67	0.212	0.0625	0.0013	633	13	693	43	9%	patchy-brighter	✓				
CNG2-mnz-11	9.14	0.180	0.0635	0.0016	9.14	0.180	0.0635	0.0016	668	12	725	53	8%	unzoned	✓				
CNG2-mnz-12	11.12	0.252	0.0600	0.0036	11.12	0.252	0.0600	0.0036	554	12	605	131	9%	unzoned	✓				
CNG2-mnz-13	5.54	0.098	0.0758	0.0012	5.54	0.098	0.0758	0.0012	1068	17	1090	32	2%	patchy-dar	✓				
CNG2-mnz-14	10.17	0.208	0.0599	0.0010	10.17	0.208	0.0599	0.0010	604	12	600	37	-1%	unzoned	✓				
CNG2-mnz-15	11.37	0.211	0.0579	0.0010	11.37	0.211	0.0579	0.0010	543	10	526	40	-3%	mosaic	✓				
CNG2-mnz-16	9.91	0.181	0.0604	0.0011	9.91	0.181	0.0604	0.0011	619	11	617	39	0%	mosaic	✓				
CNG2-mnz-17	11.03	0.247	0.0596	0.0021	11.03	0.247	0.0596	0.0021	559	12	588	75	5%	unzoned	✓				
CNG2-mnz-18	5.80	0.133	0.0735	0.0014	5.80	0.133	0.0735	0.0014	1024	22	1028	40	0%	patchy-dar	✓				
CNG2-mnz-19	9.37	0.185	0.0657	0.0012	9.37	0.185	0.0657	0.0012	650	12	797	37	19%	unzoned	✓				
CNG2-mnz-20	10.61	0.256	0.0608	0.0008	10.61	0.256	0.0608	0.0008	579	13	631	30	8%	patchy-darker	✓				
CNG2-mnz-21	6.21	0.151	0.0712	0.0009	6.21	0.151	0.0712	0.0009	962	22	962	26	0%	mosaic	✓				
CNG2-mnz-22	11.95	0.270	0.0570	0.0006	11.95	0.270	0.0570	0.0006	518	11	490	23	-6%	patchy-bri	✓				
CNG2-mnz-23	5.52	0.134	0.0769	0.0009	5.52	0.134	0.0769	0.0009	1070	24	1119	24	4%	patchy-bri	✓				
CNG2-mnz-24	11.08	0.260	0.0592	0.0008	11.08	0.260	0.0592	0.0008	556	13	574	29	3%	patchy-dar	✓				
CNG2-mnz-25	7.04	0.438	0.0858	0.0067	7.04	0.438	0.0858	0.0067	838	49	1334	150	38%	mosaic	✓				
CNG2-mnz-26	5.94	0.132	0.0733	0.0012	5.94	0.132	0.0733	0.0012	1001	21	1022	34	2%	mosaic	✓				
CNG2-mnz-27	11.12	0.362	0.0624	0.0012	11.12	0.362	0.0624	0.0012	552	17	688	41	20%	unzoned	✓				
CNG2-mnz-28	9.58	0.203	0.0619	0.0008	9.58	0.203	0.0619	0.0008	639	13	670	27	5%	patchy-dar	✓				
CNG2-mnz-29	11.06	0.194	0.0583	0.0009	11.06	0.194	0.0583	0.0009	558	9	541	32	-3%	patchy-dar	✓				
CNG2-mnz-30	10.94	0.216	0.0587	0.0007	10.94	0.216	0.0587	0.0007	563	11	555	26	-2%	patchy-dar	✓				
CNG2-mnz-31	10.79	0.201	0.0591	0.0009	10.79	0.201	0.0591	0.0009	571	10	571	31	0%	unzoned	✓				
CNG2-mnz-32	8.91	0.286	0.1031	0.0118	8.91	0.286	0.1031	0.0118	651	20	1681	211	62%	unzoned	✓				
CNG2-mnz-33	8.88	0.207	0.0652	0.0017	8.88	0.207	0.0652	0.0017	685	15	780	55	12%	patchy-darker	✓				
CNG2-mnz-34	10.96	0.171	0.0574	0.0007	10.96	0.171	0.0574	0.0007	563	8	505	26	-12%	core	✓				
CNG2-mnz-35	9.72	0.154	0.0608	0.0008	9.72	0.154	0.0608	0.0008	630	10	632	28	0%	patchy-bri	✓				
CNG2-mnz-36	6.21	0.165	0.0743	0.0019	6.21	0.165	0.0743	0.0019	958	24	1050	50	9%	unzoned	✓				
CNG2-mnz-37	5.97	0.150	0.0752	0.0012	5.97	0.150	0.0752	0.0012	995	23	1074	31	8%	patchy-darker	✓				
CNG2-mnz-38	11.43	0.195	0.0588	0.0011	11.43	0.195	0.0588	0.0011	540	9	561	42	4%	unzoned	✓				
CNG2-mnz-39	2.78	0.045	0.1231	0.0016	2.78	0.045	0.1231	0.0016	1978	28	2001	23	1%	mosaic	✓				
CNG2-mnz-40	2.83	0.044	0.1238	0.0018	2.83	0.044	0.1238	0.0018	1938	26	2012	25	4%	mosaic	✓				
CNG2-mnz-41	10.76	0.181	0.0593	0.0010	10.76	0.181	0.0593	0.0010	572	9	580	37	1%	core	✓				
CNG2-mnz-42	9.70	0.235	0.0643	0.0021	9.70	0.235	0.0643	0.0021	629	14	752	69	17%	patchy-brighter	✓				
CNG2-mnz-43	8.51	0.785	0.1047	0.0095	8.51	0.785	0.1047	0.0095	681	59	1709	167	61%	unzoned	✓				
CNG2-mnz-44	11.31	0.189	0.0577	0.0009	11.31	0.189	0.0577	0.0009	546	9	517	36	-6%	unzoned	✓				
CNG2-mnz-45	6.35	0.142	0.0757	0.0014	6.35	0.142	0.0757	0.0014	936	19	1086	38	14%	unzoned	✓				
CNG2-mnz-46	5.60	0.115	0.0762	0.0009	5.61	0.115	0.0762	0.0009	1058	20	1044	31	-1%	unzoned	✓				
CNG2-mnz-47	9.69	0.192	0.0611	0.0018	9.69	0.192	0.0611	0.0018	633	12	644	64	2%	unzoned	✓				
CNG2-mnz-48	10.62	0.198	0.0584	0.0012	10.62	0.198	0.0584	0.0012	580	10	546	46	-6%	unzoned	✓				
CNG2-mnz-49	11.09	0.185	0.0580	0.0009	11.09	0.185	0.0580	0.0009	557	9	531	33	-5%	unzoned	✓				
CNG2-mnz-50	6.36	0.269	0.0817	0.0027	6.36	0.269	0.0817	0.0027	928	36	1238	66	26%	unzoned	✓				
CNG2-mnz-51	10.11	0.185	0.0595	0.0010	10.11	0.185	0.0595	0.0010	608	11	586	36	-4%	unzoned	✓				
CNG2-mnz-52	10.06	0.168	0.0597	0.0010	10.06	0.168	0.0597	0.0010	611	10	594	37	-3%	patchy-dar	✓				
CNG2-mnz-53	5.72	0.106	0.0766	0.0010	5.72	0.106	0.0766	0.0010	1034	18	1112	27	7%	patchy-darker	✓				
CNG2-mnz-54	5.64	0.165	0.0774	0.0016	5.64	0.165	0.0774	0.0016	1048	28	1132	42	8%	unzoned	✓				
CNG2-mnz-55	10.18	0.181	0.0600	0.0006	10.18	0.181	0.0600	0.0006	603	10	604	20	0%	unzoned	✓				
CNG2-mnz-56	12.73	0.320	0.0577	0.0007	12.73	0.320	0.0577	0.0007	487	12	517	27	6%	unzoned	✓				
CNG2-mnz-57	4.50	0.095	0.0877	0.0011	4.50	0.095	0.0877	0.0011	1287	25	1376	25	7%	patchy-darker	✓				
CNG2-mnz-58	11.54	0.254	0.0609	0.0012	11.54	0.254	0.0609	0.0012	533	11	634	43	16%	mosaic	✓				
CNG2-mnz-59	5.35	0.115	0.0783	0.0008	5.35	0.115	0.0783	0.0008	1101	22	1156	20	5%	unzoned	✓				
CNG2-mnz-60	10.03	0.232	0.0594	0.0011	10.03	0.232	0.0594	0.0011	613	14	580	39	-6%	unzoned	✓				
CNG2-mnz-61	6.39	0.107	0.0706	0.0008	6.39	0.107	0.0706	0.0008	936	15	946	22	1%	unzoned	✓				
CNG2-mnz-62	11.10	0.217	0.0582	0.0009	11.10	0.217	0.0582	0.0009	556	10	536	33	-4%	unzoned	✓				
CNG2-mnz-63	6.30																		

$^{238}\text{U}/^{206}\text{Pb}$	$2\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$	$2\sigma$	$^{206}\text{Pb}/^{204}\text{Pb}$	$2\sigma$	$^{238}\text{U}/^{206}\text{Pb}^*$	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\sigma$	$^{206}\text{Pb}^*/^{238}\text{U}$	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\sigma$	Disc. <sup>†</sup>	internal structure	Used data <sup>‡</sup>	
ratio		ratio		ratio		ratio		ratio		age (Ma)		age (Ma)		(%)			
<b>Zambezi River</b>																	
ZMB2-mnz-1	2.96	0.057	0.1175	0.0023		2.96	0.057	0.1175	0.0023	1866	31	1919	34	3%	patchy-dar	✓	
ZMB2-mnz-2	5.48	0.101	0.0753	0.0013		5.48	0.101	0.0753	0.0013	1078	18	1076	35	0%	patchy-bric	✓	
ZMB2-mnz-3	12.71	0.248	0.0570	0.0010		12.71	0.248	0.0570	0.0010	487	9	493	39	1%	unzoned	✓	
ZMB2-mnz-4	12.42	0.252	0.0577	0.0012		12.42	0.252	0.0577	0.0012	498	10	519	46	4%	patchy-bric	✓	
ZMB2-mnz-5	6.44	0.159	0.0776	0.0022		6.44	0.159	0.0776	0.0022	920	21	1136	56	19%	unzoned		
ZMB2-mnz-6	12.53	0.243	0.0574	0.0012		12.53	0.243	0.0574	0.0012	494	9	508	45	3%	unzoned	✓	
ZMB2-mnz-7	11.90	0.198	0.0631	0.0016		11.90	0.198	0.0631	0.0016	516	8	711	54	28%	patchy-darker		
ZMB2-mnz-8	4.93	0.084	0.0814	0.0016		4.93	0.084	0.0814	0.0016	1185	18	1231	39	4%	mosaic	✓	
ZMB2-mnz-9	12.36	0.283	0.0658	0.0042		12.36	0.283	0.0658	0.0042	501	11	800	135	39%	core		
ZMB2-mnz-10	12.01	0.200	0.0578	0.0013		12.01	0.200	0.0578	0.0013	515	8	521	50	1%	unzoned	✓	
ZMB2-mnz-11	12.52	0.219	0.0583	0.0012		12.52	0.219	0.0583	0.0012	494	8	540	47	9%	unzoned	✓	
ZMB2-mnz-12	12.50	0.238	0.0580	0.0012		12.50	0.238	0.0580	0.0012	498	9	528	46	6%	patchy-dar	✓	
ZMB2-mnz-13	11.37	0.289	0.0655	0.0037		11.37	0.289	0.0655	0.0037	538	13	791	119	33%	unzoned		
ZMB2-mnz-14	12.60	0.228	0.0574	0.0013		12.60	0.228	0.0574	0.0013	491	9	507	50	3%	patchy-dar	✓	
ZMB2-mnz-15	2.86	0.045	0.1192	0.0025		2.86	0.045	0.1192	0.0025	1926	26	1944	37	1%	core	✓	
ZMB2-mnz-16	12.02	0.287	0.0672	0.0021	1868	12.12	0.303	0.0593	0.0025	511	12	580	91	12%	patchy-dar		
ZMB2-mnz-17	5.61	0.114	0.0754	0.0012		5.61	0.114	0.0754	0.0012	1055	20	1081	33	2%	mosaic	✓	
ZMB2-mnz-18	12.51	0.301	0.0584	0.0013		12.51	0.301	0.0584	0.0013	495	11	546	47	10%	unzoned	✓	
ZMB2-mnz-19	12.89	0.269	0.0563	0.0009		12.89	0.269	0.0563	0.0009	481	10	462	35	-4%	rim		
ZMB2-mnz-20	12.41	0.215	0.0585	0.0014		12.41	0.215	0.0585	0.0014	498	8	548	52	9%	unzoned	✓	
ZMB2-mnz-21	12.63	0.281	0.0585	0.0014		12.63	0.281	0.0585	0.0014	498	10	549	54	11%	unzoned	✓	
ZMB2-mnz-22	10.99	0.199	0.0631	0.0017		10.99	0.199	0.0631	0.0017	558	10	712	57	22%	unzoned		
ZMB2-mnz-23	12.47	0.236	0.0573	0.0010		12.47	0.236	0.0573	0.0010	497	9	503	38	1%	patchy-dar	✓	
ZMB2-mnz-24	11.94	0.247	0.0579	0.0009		11.94	0.247	0.0579	0.0009	518	10	525	36	1%	unzoned	✓	
ZMB2-mnz-25	11.51	0.196	0.0596	0.0019		11.51	0.196	0.0596	0.0019	535	9	590	70	9%	core		
ZMB2-mnz-26	12.39	0.198	0.0567	0.0010		12.39	0.198	0.0567	0.0010	500	8	480	41	-4%	unzoned	✓	
ZMB2-mnz-27	12.47	0.219	0.0579	0.0009		12.47	0.219	0.0579	0.0009	496	8	526	35	6%	mosaic	✓	
ZMB2-mnz-28	12.26	0.224	0.0577	0.0008		12.26	0.224	0.0577	0.0008	504	9	517	32	2%	unzoned	✓	
ZMB2-mnz-29	11.31	0.228	0.0585	0.0009		11.31	0.228	0.0585	0.0009	545	11	549	34	1%	unzoned	✓	
ZMB2-mnz-30	12.48	0.216	0.0572	0.0008		12.48	0.216	0.0572	0.0008	498	8	498	30	0%	core		
ZMB2-mnz-31	12.87	0.233	0.0571	0.0009		12.87	0.233	0.0571	0.0009	482	8	497	35	3%	mosaic	✓	
ZMB2-mnz-32	12.44	0.224	0.0586	0.0009		12.44	0.224	0.0586	0.0009	497	9	539	56	10%	mosaic		
ZMB2-mnz-33	12.23	0.226	0.0587	0.0009		12.23	0.226	0.0587	0.0009	505	9	555	34	9%	mosaic		
ZMB2-mnz-34	12.36	0.222	0.0579	0.0012		12.36	0.222	0.0579	0.0012	501	9	528	45	5%	patchy-bric	✓	
ZMB2-mnz-35	12.13	0.209	0.0586	0.0008		12.13	0.209	0.0586	0.0008	510	8	554	31	8%	unzoned		
ZMB2-mnz-36	5.71	0.117	0.0749	0.0012		5.71	0.117	0.0749	0.0012	1038	20	1067	32	3%	patchy-dar	✓	
ZMB2-mnz-37	11.88	0.165	0.0590	0.0010		11.88	0.165	0.0590	0.0010	520	7	569	36	9%	patchy-brighter		
ZMB2-mnz-38	12.50	0.248	0.0571	0.0010		12.50	0.248	0.0571	0.0010	498	9	496	39	0%	unzoned	✓	
ZMB2-mnz-39	12.69	0.215	0.0583	0.0013		12.69	0.215	0.0583	0.0013	488	8	540	47	10%	unzoned	✓	
ZMB2-mnz-40	12.56	0.228	0.0582	0.0015		12.56	0.228	0.0582	0.0015	493	9	539	56	9%	unzoned		
ZMB2-mnz-41	11.91	0.267	0.0582	0.0014		11.91	0.267	0.0582	0.0014	519	11	536	52	3%	patchy-dar	✓	
ZMB2-mnz-42	12.47	0.249	0.0573	0.0014		12.47	0.249	0.0573	0.0014	497	10	504	54	1%	unzoned	✓	
ZMB2-mnz-43	5.84	0.125	0.0748	0.0012		5.84	0.125	0.0748	0.0012	1016	20	1064	33	5%	unzoned	✓	
ZMB2-mnz-44	10.17	0.341	0.0670	0.0024		10.17	0.341	0.0670	0.0024	599	19	839	74	29%	unzoned		
ZMB2-mnz-45	10.03	0.199	0.0596	0.0011		10.03	0.199	0.0596	0.0011	613	12	588	40	-5%	unzoned	✓	
ZMB2-mnz-46	12.26	0.198	0.0584	0.0008		12.26	0.198	0.0584	0.0008	505	8	545	31	8%	patchy-dar	✓	
ZMB2-mnz-47	12.52	0.180	0.0585	0.0008		12.52	0.180	0.0585	0.0008	494	7	548	31	10%	patchy-brighter		
ZMB2-mnz-48	12.46	0.276	0.0597	0.0017		12.46	0.276	0.0597	0.0017	498	11	591	61	16%	patchy-dar		
ZMB2-mnz-49	12.68	0.334	0.0610	0.0017		12.68	0.334	0.0610	0.0017	487	12	639	59	24%	core		
ZMB2-mnz-50	6.09	0.221	0.0759	0.0010		6.09	0.221	0.0759	0.0010	976	33	1091	26	11%	patchy-brighter		
ZMB2-mnz-51	12.44	0.227	0.0599	0.0014		12.44	0.227	0.0599	0.0014	497	9	598	51	17%	patchy-brighter		
ZMB2-mnz-52	3.01	0.052	0.1202	0.0013		3.01	0.052	0.1202	0.0013	1838	28	1960	20	6%	core		
ZMB2-mnz-53	11.38	0.220	0.0593	0.0017		11.38	0.220	0.0593	0.0017	542	10	580	62	7%	unzoned	✓	
ZMB2-mnz-54	12.33	0.238	0.0579	0.0011		12.33	0.238	0.0579	0.0011	502	9	527	43	5%	mosaic	✓	
ZMB2-mnz-55	13.02	0.270	0.0581	0.0008		13.02	0.270	0.0581	0.0008	476	9	535	29	11%	patchy-brighter		
ZMB2-mnz-56	12.21	0.297	0.0602	0.0012		12.21	0.297	0.0602	0.0012	505	12	610	42	18%	patchy-brighter		
ZMB2-mnz-57	12.14	0.215	0.0578	0.0009		12.14	0.215	0.0578	0.0009	510	9	522	33	2%	patchy-dar	✓	
ZMB2-mnz-58	5.84	0.104	0.0780	0.0025		5.84	0.104	0.0780	0.0025	1013	17	1146	64	12%	core		
ZMB2-mnz-59	12.45	0.255	0.0631	0.0018		12.45	0.255	0.0631	0.0018	494	10	712	61	31%	core		
ZMB2-mnz-60	12.58	0.212	0.0582	0.0010		12.58	0.212	0.0582	0.0010	492	8	539	38	9%	patchy-dar		
ZMB2-mnz-61	13.15	0.193	0.0576	0.0008		13.15	0.193	0.0576	0.0008	472	7	514	32	8%	unzoned		
ZMB2-mnz-62	12.50	0.229	0.0586	0.0009		12.50	0.229	0.0586	0.0009	495	9	553	35	11%	patchy-brighter		
ZMB2-mnz-63	9.73	0.243	0.0612	0.0010		9.73	0.243	0.0612	0.0010	630	15	646	35	2%	mosaic	✓	
ZMB2-mnz-64	12.39	0.290	0.0681	0.0020		12.39	0.290	0.0681	0.0020	493	11	871	61	44%	patchy-dar		
ZMB2-mnz-65	12.15	0.278	0.0570	0.0013		12.15	0.278	0.0570	0.0013	509	11	493	52	-3%	core	✓	
ZMB2-mnz-6																	

$^{238}\text{U}/^{206}\text{Pb}$	$2\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$	$2\sigma$	$^{206}\text{Pb}^{204}\text{Pb}$	$^{238}\text{U}/^{206}\text{Pb}^*$	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\sigma$	$^{206}\text{Pb}^{238}\text{U}$	$2\sigma$	$^{207}\text{Pb}^*/^{206}\text{Pb}^*$	$2\sigma$	Disc. <sup>†</sup>	internal structure	Used data <sup>††</sup>		
ratio		ratio		ratio			ratio		age (Ma)		age (Ma)		(%)				
<b>Orange River</b>																	
ORG2-mnz-1	5.39	0.096	0.0758	0.0013	5.39	0.096	0.0758	0.0013	1095	18	1089	34	-1%	patchy-dar	✓		
ORG2-mnz-2	5.39	0.105	0.0768	0.0014	5.39	0.105	0.0768	0.0014	1094	20	1117	35	2%	unzone	✓		
ORG2-mnz-3	5.72	0.091	0.0736	0.0012	5.72	0.091	0.0736	0.0012	1038	15	1029	33	-1%	mosaic	✓		
ORG2-mnz-4	5.29	0.147	0.0730	0.0037	5.29	0.147	0.0730	0.0037	1116	28	1013	102	-11%	mosaic	✓		
ORG2-mnz-5	5.29	0.136	0.0752	0.0038	5.29	0.136	0.0752	0.0038	1116	26	1075	102	-4%	patchy-br	✓		
ORG2-mnz-6	6.27	0.159	0.0732	0.0037	6.27	0.159	0.0732	0.0037	949	22	1018	102	7%	patchy-br	✓		
ORG2-mnz-7	5.94	0.186	0.0729	0.0041	5.94	0.186	0.0729	0.0041	1001	29	1011	115	1%	patchy-dar	✓		
ORG2-mnz-8	5.37	0.152	0.0761	0.0043	5.37	0.152	0.0761	0.0043	1098	28	1098	114	0%	patchy-br	✓		
ORG2-mnz-9 <sup>*</sup>	5.28	0.162	0.0779	0.0045	4173		5.30	0.163	0.0744	0.0013	1114	31	1053	34	-6%	patchy-br	✓
ORG2-mnz-10	5.68	0.104	0.0740	0.0044	5.68	0.104	0.0740	0.0044	1045	28	973	111	-8%	mosaic	✓		
ORG2-mnz-11	5.58	0.186	0.0748	0.0035	5.58	0.186	0.0748	0.0035	1062	33	1060	94	0%	unzon	✓		
ORG2-mnz-12	2.09	4.039	0.3113	0.0571	2.09	4.039	0.3113	0.0571	2828	51	3527	233	94%	core			
ORG2-mnz-13	6.02	0.145	0.0728	0.0033	6.02	0.145	0.0728	0.0033	991	22	1009	92	2%	mosaic	✓		
ORG2-mnz-14	5.48	0.110	0.0747	0.0033	5.48	0.110	0.0747	0.0033	1080	20	1059	88	-2%	mosaic	✓		
ORG2-mnz-15	6.09	0.114	0.0730	0.0013	6.09	0.114	0.0730	0.0013	980	17	1015	36	4%	mosaic	✓		
ORG2-mnz-16	5.02	0.083	0.0801	0.0016	5.02	0.083	0.0801	0.0016	1172	18	1199	39	3%	core	✓		
ORG2-mnz-17	4.90	0.085	0.0793	0.0016	4.90	0.085	0.0793	0.0016	1197	19	1178	40	-2%	unzon	✓		
ORG2-mnz-18	6.56	0.221	0.0787	0.0025	6.56	0.221	0.0787	0.0025	915	29	1165	63	23%	mosaic			
ORG2-mnz-19	5.22	0.141	0.0760	0.0014	5.22	0.141	0.0760	0.0014	1129	28	1095	37	-3%	patchy-dar	✓		
ORG2-mnz-20 <sup>*</sup>	5.45	0.187	0.0898	0.0020	1083		5.52	0.191	0.0765	0.0013	1073	34	1108	35	3%	core	✓
ORG2-mnz-21	5.79	0.091	0.0732	0.0015	5.79	0.091	0.0732	0.0015	1027	15	1020	42	-1%	patchy-dar	✓		
ORG2-mnz-22	4.93	0.098	0.0790	0.0014	4.93	0.098	0.0790	0.0014	1190	22	1173	35	-2%	patchy-br	✓		
ORG2-mnz-23	6.41	0.190	0.0722	0.0014	6.41	0.190	0.0722	0.0014	935	26	992	40	6%	patchy-dar	✓		
ORG2-mnz-24	5.96	0.143	0.0732	0.0015	5.96	0.143	0.0732	0.0015	999	22	1019	42	2%	unzon	✓		
ORG2-mnz-25	5.07	0.140	0.0807	0.0015	5.07	0.140	0.0807	0.0015	1162	29	1215	36	5%	unzon	✓		
ORG2-mnz-26	5.21	0.142	0.0779	0.0015	5.21	0.142	0.0779	0.0015	1131	28	1144	39	1%	patchy-dar	✓		
ORG2-mnz-27	6.70	0.353	0.0831	0.0033	6.70	0.353	0.0831	0.0033	897	44	1272	78	32%	patchy-darker			
ORG2-mnz-28	5.07	0.125	0.0794	0.0015	5.07	0.125	0.0794	0.0015	1159	26	1183	36	2%	unzon	✓		
ORG2-mnz-29	5.20	0.134	0.0802	0.0016	5.20	0.134	0.0802	0.0016	1135	27	1203	40	6%	core			
ORG2-mnz-30	6.53	0.207	0.0731	0.0009	6.53	0.207	0.0731	0.0009	918	27	1017	25	10%	patchy-brighter			
ORG2-mnz-31	5.61	0.119	0.0778	0.0015	5.61	0.119	0.0778	0.0015	1058	21	1143	38	8%	rim			
ORG2-mnz-32	5.13	0.129	0.0768	0.0015	5.13	0.129	0.0768	0.0015	1148	26	1116	38	-3%	patchy-br	✓		
ORG2-mnz-33	4.94	0.131	0.0795	0.0012	4.94	0.131	0.0795	0.0012	1188	29	1183	30	0%	mosaic	✓		
ORG2-mnz-34	12.52	0.224	0.0604	0.0016	12.52	0.224	0.0604	0.0016	495	9	616	58	20%	patchy-brighter			
ORG2-mnz-35	5.73	0.124	0.0779	0.0014	5.73	0.124	0.0779	0.0014	1037	21	1144	35	10%	unzon	✓		
ORG2-mnz-36	5.10	0.091	0.0793	0.0012	5.10	0.091	0.0793	0.0012	1153	19	1180	31	2%	patchy-dar	✓		
ORG2-mnz-37	5.52	0.094	0.0751	0.0013	5.52	0.094	0.0751	0.0013	1073	17	1071	35	0%	patchy-br	✓		
ORG2-mnz-38	5.00	0.071	0.0794	0.0014	5.00	0.071	0.0794	0.0014	1174	15	1182	36	1%	patchy-br	✓		
ORG2-mnz-39	5.58	0.175	0.0739	0.0012	5.58	0.175	0.0739	0.0012	1063	31	1038	34	-3%	patchy-dar	✓		
ORG2-mnz-40	5.11	0.104	0.0754	0.0014	5.11	0.104	0.0754	0.0014	1153	22	1089	37	-7%	unzon			
ORG2-mnz-41	12.24	0.289	0.0707	0.0011	12.24	0.289	0.0707	0.0011	506	11	517	40	2%	core	✓		
ORG2-mnz-42	5.00	0.100	0.0775	0.0010	6.00	0.079	0.0725	0.0010	948	12	1001	29	1%	patchy-dar	✓		
ORG2-mnz-43	5.37	0.109	0.0753	0.0013	5.37	0.109	0.0753	0.0013	1101	21	1078	33	-2%	unzon	✓		
ORG2-mnz-44	6.11	0.175	0.0837	0.0025	6.11	0.175	0.0837	0.0025	977	26	1286	58	26%	patchy-darker			
ORG2-mnz-45	5.53	0.085	0.0768	0.0009	4552		5.55	0.087	0.0768	0.0016	1068	15	1030	43	5%	patchy-dar	✓
ORG2-mnz-46	5.05	0.114	0.0821	0.0016	10158		5.06	0.114	0.0807	0.0009	1163	24	1214	21	5%	mosaic	✓
ORG2-mnz-47	4.57	0.091	0.0810	0.0016	4.57	0.091	0.0810	0.0016	1277	23	1220	38	5%	mosaic	✓		
ORG2-mnz-48	5.14	0.130	0.0763	0.0015	5.14	0.130	0.0763	0.0015	1146	26	1103	39	-4%	patchy-br	✓		
ORG2-mnz-49	5.48	0.180	0.0806	0.0023	5.48	0.180	0.0806	0.0023	1081	33	1212	56	12%	patchy-brighter			
ORG2-mnz-50	6.42	0.044	0.0945	0.0056	6.42	0.044	0.0945	0.0056	933	56	1518	112	41%	mosaic			
ORG2-mnz-51	5.07	0.106	0.0806	0.0014	5.07	0.106	0.0806	0.0014	1161	22	1211	33	5%	patchy-br	✓		
ORG2-mnz-52	4.86	0.116	0.0799	0.0013	4.86	0.116	0.0799	0.0013	1207	26	1195	33	-1%	unzon	✓		
ORG2-mnz-53	5.57	0.107	0.0776	0.0014	5.57	0.107	0.0776	0.0014	1065	19	1137	35	7%	unzon			
ORG2-mnz-54	4.40	0.069	0.0864	0.0016	4.40	0.069	0.0864	0.0016	1320	19	1346	36	2%	unzon			
ORG2-mnz-55	5.21	0.124	0.0786	0.0014	5.21	0.124	0.0786	0.0014	1132	25	1161	36	3%	patchy-br	✓		
ORG2-mnz-56	4.87	0.110	0.0797	0.0015	4.87	0.110	0.0797	0.0015	1204	25	1189	38	-1%	patchy-br	✓		
ORG2-mnz-57	12.14	0.188	0.0600	0.0013	12.14	0.188	0.0600	0.0013	510	8	604	45	16%	unzon			
ORG2-mnz-58	4.95	0.098	0.0794	0.0014	4.95	0.098	0.0794	0.0014	1187	21	1183	34	0%	unzon	✓		
ORG2-mnz-59	5.05	0.111	0.0795	0.0017	5.05	0.111	0.0795	0.0017	1165	23	1185	42	2%	unzon	✓		
ORG2-mnz-60	5.20	0.113	0.0780	0.0014	5.20	0.113	0.0780	0.0014	1133	23	1146	36	1%	mosaic	✓		
ORG2-mnz-61 <sup>*</sup>	6.16	0.244	0.0883	0.0037	11244		6.17	0.244	0.0871	0.0006	968	35	1362	14	31%	patchy-dar	✓
ORG2-mnz-62	5.37	0.078	0.0808	0.0008	5.37	0.078	0.0778	0.0008	1102	15	1142	20	4%	patchy-dar	✓		
ORG2-mnz-63	5.78	0.128	0.0817	0.0019	5.78	0.128	0.0817	0.0019	1029	21	1238	45	18%	patchy-brighter			
ORG2-mnz-64	6.13	0.215	0.0919	0.0034	6.13	0.215	0.0919	0.0034	974	32	1464	71	36%	unzon			
ORG2-mnz-65	12.34	0.210	0.0582	0.0010	12.34	0.210	0.0582	0.0010	502	8	537	38	7%	patchy-br	✓		
ORG2-mnz-66	4.79	0.056	0.0800	0.0004	4.81	0.056	0.0800	0.0004	1172	19	1197	20	2%	mosaic			
ORG2-mnz-67	3																





La-ICP-MS DATA	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm
	Ca	Se	Y	La	Ce	Pr	Nd	Su	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb <sup>1</sup>	Lu <sup>11</sup>	Tb	U			
<b>Congo River</b>																						
CNG2-mmz-1	18473	24	12065	141498	225278	24276	80064	15584	143	9336	1034	4135	454	645	47	138	11	66545	8777			
CNG2-mmz-2	5500	22	7197	145104	260854	23649	89518	12054	171	7526	759	2968	341	508	36	141	10	60799	659			
CNG2-mmz-3	21064	12	22839	88103	199606	21755	85504	21872	72	16829	2113	8219	878	1339	117	417	31	103516	7518			
CNG2-mmz-4	14961	39	9046	121526	240998	23501	111132	21133	299	16229	1549	5060	389	467	21	44	2	56650	4432			
CNG2-mmz-5	6864	27	4571	129462	266446	24221	87713	11572	178	5698	529	1913	199	313	21	59	5	84301	2578			
CNG2-mmz-6	3988	40	6495	136975	257106	25707	114402	14206	692	8187	684	2886	372	657	59	260	22	43450	711			
CNG2-mmz-7	5532	25	4805	140241	256688	22522	82701	9997	188	4859	461	1725	207	318	33	102	8	72346	1067			
CNG2-mmz-8	7639	34	21874	128371	244163	25787	110156	16976	143	11518	1373	6510	803	1588	123	533	43	52684	1777			
CNG2-mmz-9	6185	20	1397	140753	272269	29076	117491	14523	94	5921	385	1066	57	57	3	<4	<0.3	42214	665			
CNG2-mmz-10	6437	46	4823	127644	253367	22278	101620	14411	624	9169	759	2543	233	238	11	<7	<0.6	37521	2419			
CNG2-mmz-11	5183	23	415	137088	269682	24459	98100	8432	83	2768	117	286	18	27	1	<2	<0.1	49819	327			
CNG2-mmz-12	2498	16	5753	149305	273268	26895	104695	18319	1711	13193	1047	3722	328	395	24	40	1	4019	98			
CNG2-mmz-13	8991	25	19165	121517	240390	24333	92266	18207	299	14514	1543	6414	755	1120	90	256	19	53944	5198			
CNG2-mmz-14	9241	34	7802	132485	254067	27241	105414	14017	322	10810	996	3474	314	368	21	44	3	57158	4461			
CNG2-mmz-15	7298	20	9053	131902	237248	21553	69033	10542	191	6201	658	2918	353	646	57	187	18	46178	2995			
CNG2-mmz-16	16885	39	21693	132978	235942	24240	88538	14884	182	17161	1183	6192	793	1447	130	480	36	57527	3675			
CNG2-mmz-17	2191	42	1584	219688	291205	21722	67607	5053	737	2849	184	620	68	94	7	18	1	2965	217			
CNG2-mmz-18	13173	93	20889	141587	233893	25951	105998	17700	1622	13097	1387	6132	762	1486	120	414	35	41218	2368			
CNG2-mmz-19	15456	77	12698	147836	245444	22226	11818	1275	9220	785	3469	447	891	88	483	44	66918	734				
CNG2-mmz-20	12278	85	17789	123269	234917	25482	95285	16936	1833	12979	1390	5357	752	1231	100	369	35	49800	12526			
CNG2-mmz-21	7489	26	19258	119568	230257	28264	110129	22114	392	22315	2369	7547	810	956	50	93	7	43565	3980			
CNG2-mmz-22	16595	15	31123	113065	21886	27816	92169	16266	172	14635	1682	7707	1195	2361	190	694	64	104456	3041			
CNG2-mmz-23	6283	123	3032	128030	254186	27853	105992	18084	1369	1183	631	1692	144	172	6	<6	<0.5	41293	4372			
CNG2-mmz-24	12622	85	2841	161225	263296	24221	75141	8136	168	5230	355	1283	156	231	16	37	4	54753	2262			
CNG2-mmz-25	18199	243	10887	111390	242806	27303	99898	16780	2003	14868	1501	5530	610	754	55	192	14	78403	3744			
CNG2-mmz-26	5008	39	3032	146965	267104	27192	10034	400	516	376	1115	235	22	52	5	52460	713					
CNG2-mmz-27	5384	102	1800	135289	280123	29816	112778	17902	1490	14189	977	1965	111	53	2	<7	<0.6	28002	5156			
CNG2-mmz-28	7862	86	5990	123360	254349	28983	110791	14385	271	11407	937	3095	283	262	13	<7	<0.9	55234	2070			
CNG2-mmz-29	10434	20	18273	125620	244701	28620	10227	9982	13742	1047	78	249	18	69158	3424							
CNG2-mmz-30	13925	94	17249	147363	253478	26418	76827	11531	204	9415	1014	4700	607	965	80	245	22	55372	5927			
CNG2-mmz-31	1492	21	6844	132781	259589	25407	93295	11213	113	7012	671	2754	296	452	42	120	10	58998	2009			
CNG2-mmz-32	19472	127	20716	133977	23592	28166	91444	15212	182	12656	1215	5839	817	1466	140	477	45	47945	362			
CNG2-mmz-33	12034	14	5917	118273	242217	27321	100710	18601	232	15231	1590	3381	253	212	12	<9	<0.9	64296	2420			
CNG2-mmz-34	19940	28	20968	119018	237035	24471	81858	1383	189	10963	1141	5589	768	1237	138	506	40	68370	11629			
CNG2-mmz-35	13877	26	249	137416	261520	27026	103755	16444	2171	9296	351	459	13	9	1	<6	<0.6	72920	3886			
CNG2-mmz-36	7997	42	1410	140277	265387	31064	111648	17980	326	10211	586	1246	75	69	6	<5	<0.4	45988	2841			
CNG2-mmz-37	11197	184	6443	12882	242396	26865	105216	110864	21320	2423	13074	143	8985	973	310	24	44883	2050				
CNG2-mmz-38	17220	20	10667	140662	254000	29081	97555	13376	114	9466	869	3664	449	708	58	175	16	69446	2617			
CNG2-mmz-39	9474	74	23330	119843	250805	28230	106769	13743	11603	1380	6813	1002	1471	127	493	36	40699	2563				
CNG2-mmz-40	9431	9	5051	133807	240373	30772	112166	17756	842	11238	850	2522	180	182	11	<7	3	53774	2135			
CNG2-mmz-41	14284	42	1107	122804	248022	32994	111098	14864	164	7117	325	678	38	56	2	<4	0.4	79577	3259			
CNG2-mmz-42	4635	12	5657	157359	274915	27719	9759	10016	262	5771	452	1666	207	338	32	118	11	42730	703			
CNG2-mmz-43	19575	41	20498	114778	217364	28237	98234	21182	386	16462	1795	6763	714	1141	119	435	35	53411	21942			
CNG2-mmz-44	6884	10	6176	151374	257534	27217	91653	1095	136	6183	543	2343	256	423	46	145	12	53572	2591			
CNG2-mmz-45	10901	62	187	134635	270373	30329	119198	15992	1048	14929	930	1801	73	54	4	<5	0.8	36371	13556			
CNG2-mmz-46	8099	16	8309	161848	241807	28729	106745	11639	106	7373	703	2725	347	450	33	93	6	35231	2176			
CNG2-mmz-47	21127	65	2384	145145	246836	25572	15321	465	9628	669	14686	87	92	2	<7	1	93353	3138				
CNG2-mmz-48	8286	61	6105	164858	263842	28919	95646	11244	153	6364	493	1860	237	411	32	92	7	66304	2014			
CNG2-mmz-49	9508	24	1999	119195	258267	29970	109940	15921	100	5806	309	801	61	85	5	5	0.5	64534	3509			
CNG2-mmz-50	23016	55	13645	162767	234892	23823	82896	14155	384	9203	822	3834	420	913	81	344	21	77352	7988			
CNG2-mmz-51	7260	151	2563	175477	283239	29825	107043	8974	1559	4587	278	928	93	187	10	37	4	27131	555			
CNG2-mmz-52	14282	172	21155	129526	244340	26707	9832	17121	2289	15197	1458	6384	886	1626	121	457	32	56786	2923			
CNG2-mmz-53	4952	14	3437	137418	269631	28762	101043	15845	614	10107	641	1771										

La-ICP-MS DATA	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm
	Ca	Sr	Y	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb†	Lu†‡	Tb	U	
Zambezi River																				
ZMB2-mmz-1	27304	62	4078	123036	240373	25291	96143	11736	62	7582	517	1831	178	219	10	18	1	95314	762	
ZMB2-mmz-2	10677	22	14781	120262	241414	25365	102525	15086	931	10434	1284	5576	585	813	41	118	8	57616	7120	
ZMB2-mmz-3	8842	58	4890	158542	277494	25929	103913	10547	233	4850	447	1632	229	393	36	142	11	50055	5557	
ZMB2-mmz-4†	20685	87	1029	153038	271714	27024	93650	10441	97	5403	307	703	43	50	4	<3	<0.2	82423	1411	
ZMB2-mmz-5	26687	60	3986	120257	234943	24720	93972	11471	60	7410	505	1790	174	214	10	18	1	93161	745	
ZMB2-mmz-6	10472	22	14498	117957	236787	24879	100560	14796	913	10234	1260	5469	573	798	40	116	8	56511	6984	
ZMB2-mmz-7	6534	43	3614	117155	205953	19160	76786	7794	172	3584	331	1209	170	291	26	105	8	36988	4106	
ZMB2-mmz-8†	20539	86	1022	151957	269793	26833	92988	10367	97	5365	304	698	43	50	4	<3	<0.2	81841	1401	
ZMB2-mmz-9†	15708	83	12003	145862	266045	28536	110325	17160	1465	13618	1437	5527	537	774	56	215	15	46552	3203	
ZMB2-mmz-10	8141	99	8668	181930	277664	28572	97907	13308	2243	7153	682	2466	347	537	38	113	9	24293	2068	
ZMB2-mmz-11†	17146	48	980	122590	250208	24558	96211	11865	196	4996	263	523	27	39	2	<3	<0.2	82693	2428	
ZMB2-mmz-12	7298	54	7998	120299	254630	25840	92093	9490	293	4959	478	2101	306	635	57	255	21	64127	3163	
ZMB2-mmz-13	2922	16	1718	181378	280883	20124	58021	4595	512	2898	173	562	78	114	10	43	2	39856	406	
ZMB2-mmz-14	8265	60	7101	176130	272772	24160	79452	7594	1096	4873	420	1704	249	613	54	254	20	32435	2876	
ZMB2-mmz-15	27437	46	3539	98163	191948	17912	68269	9267	79	6228	527	1599	136	132	6	18	0.4	183706	2799	
ZMB2-mmz-16	5624	53	1100	123761	278151	26245	103696	7983	256	4184	256	652	54	60	3	<3	1	60896	1222	
ZMB2-mmz-17	9362	81	18752	119387	252129	25221	97458	16119	1116	12518	1376	6200	717	1331	97	363	32	64646	5454	
ZMB2-mmz-18†	16948	67	792	111515	241730	25659	112063	17532	223	7594	299	526	27	42	1	<5	<0.2	85755	2531	
ZMB2-mmz-19	810	6	4608	102872	244635	23576	91657	9442	112	5786	445	1609	198	302	22	95	10	59333	1787	
ZMB2-mmz-20†	8722	18	506	144968	274164	26843	90093	9371	59	3621	152	315	19	31	2	3	<1	71836	382	
ZMB2-mmz-21	10573	43	13185	139658	256521	22859	83309	10742	1245	7700	772	3486	472	947	68	249	22	6680	952	
ZMB2-mmz-22	3303	217	799	172650	279811	21703	72675	3834	386	2227	109	312	35	53	4	18	3	29000	213	
ZMB2-mmz-23	3807	13	9377	115562	240962	28380	101480	16058	285	11944	4117	398	481	27	30	2	41192	2124		
ZMB2-mmz-24	4119	47	21181	115464	242490	23286	95273	16568	2501	13511	1619	6515	740	1101	77	303	19	32269	4108	
ZMB2-mmz-25	5062	34	10987	144082	285956	27333	101054	14236	2263	8470	793	3768	397	842	54	255	22	7572	506	
ZMB2-mmz-26†	9488	86	177	176184	291179	24626	75511	6295	203	1888	82	167	9	18	0	<2	<0.6	30350	1342	
ZMB2-mmz-27	12972	9	3573	112847	250200	22786	75006	10661	36	5309	401	1690	120	206	12	30	2	103946	1086	
ZMB2-mmz-28†	11073	84	3551	125670	259377	22577	13469	1533	11125	880	2254	114	152	3	<2	<0.1	68960	4198		
ZMB2-mmz-29†	1933	6	1648	135829	288456	29870	109541	14952	263	7330	357	1101	65	97	5	<4	<0.6	5537	5972	
ZMB2-mmz-30†	13521	55	1364	133413	262186	27084	93859	14470	278	7362	421	896	42	57	2	<5	<0.4	64413	4291	
ZMB2-mmz-31	3551	29	2224	156883	267112	22559	117984	9173	98	5644	391	1280	80	88	4	<3	0.4	78476	2296	
ZMB2-mmz-32	5699	19	5582	124906	265285	25226	80596	10653	203	6307	487	1954	212	438	29	119	10	80731	894	
ZMB2-mmz-33	5372	48	12513	127308	258648	20455	74502	9124	362	5918	600	3145	408	1113	86	359	41	89712	2506	
ZMB2-mmz-34†	10146	40	977	121922	252914	25321	94291	11131	119	5050	282	670	617	41	58	3	<1.4	101729	951	
ZMB2-mmz-35	7374	40	10286	147872	264679	25515	87613	11587	1259	6232	652	2737	321	579	51	245	23	29860	6265	
ZMB2-mmz-36	12614	37	2185	127682	254758	26865	99497	15239	189	8390	704	1502	87	78	2	<5	0.5	66301	3610	
ZMB2-mmz-37	18034	102	7099	141587	229137	19574	74955	8001	252	5439	551	2272	288	565	42	171	14	212073	2271	
ZMB2-mmz-38†	8454	54	647	141816	250656	24309	96965	14982	350	6586	337	565	28	34	1	<2	56660	2084		
ZMB2-mmz-39	15337	60	6860	131670	246938	24635	90372	11087	332	6228	582	997	268	546	53	196	17	83459	1919	
ZMB2-mmz-40	3801	40	605	152580	278603	25234	86465	9270	82	3209	168	386	23	39	2	<3	0.8	55455	795	
ZMB2-mmz-41†	6579	69	292	157733	276742	24568	91652	10183	269	4570	276	650	33	25	0	<3	<0.2	39860	1641	
ZMB2-mmz-42†	5469	31	808	148623	266002	28916	109358	13760	121	5885	311	731	30	57	3	<4	<0.3	73912	1050	
ZMB2-mmz-43	22670	57	3325	129793	256082	26888	101368	14464	142	4641	245	350	25	25	2	3	<1.7	103591	2179	
ZMB2-mmz-44†	11994	18	3327	16615	294517	29084	106081	13142	157	5040	355	1070	129	191	15	43	4	5598	967	
ZMB2-mmz-45	1242	45	483	148424	287954	26751	89502	9252	465	4837	351	1211	192	349	29	121	8	40963	1592	
ZMB2-mmz-46	18794	51	3231	127548	258822	20386	110326	14519	360	5944	486	1192	129	155	8	17	1	88082	3392	
ZMB2-mmz-47†	50841	50	3840	18220	271919	243549	84176	14155	405	6165	547	1518	158	213	10	29	0.8	227932	1920	
ZMB2-mmz-48†	13936	17	11558	163507	259855	27898	87429	8340	161	3330	219	534	62	52	4	9	2	88761	303	
ZMB2-mmz-49†	10680	33	414	145385	268350	30807	113296	13691	142	4641	245	350	25	25	2	3	<1.7	12661	433	
ZMB2-mmz-50†	19621	51	999	122663	272635	22175	85180	11078	119	4589	286	520	46	44	4	13	<1.7	103591	2179	
ZMB2-mmz-51	7030	51	1242	137387	253452	26908	97699	11811	65	3634	212	582	45	54	5	10	2	96576	759	
ZMB2-mmz-52	14635	33	1032	155440	264833	28189	108929	12597	161	5054	3241	401	479	28	49	0.5	91049	2492		
ZMB2-mmz-53	5667	104	2192	113226	251426	24533	109843	17859	2124	10790	1457	6698	979	1770	140	585	50	95294	4641	
ZMB2-mmz-54†	14463	42	124273	259388	26976	105442	9448	132	3092	164	369	26	33	2	<2	<0.2	90457	1067		
ZMB2-mmz-55†	15476	31	641	152787	263593	26976	105442	9448	132	3092	164	369	26	33	2	<2	<0.2	44834	321	
ZMB2-mmz-56†	14463	101	19008	130906	241995	25087	98787	14072	1162	8545	1009	4961	979	1272	72	54	2	1	62866	1042
ZMB2-mmz-57	12473	66	714	148723	2643															

\* The 14C/12C ratio was set at 12.8‰ for the modern reference material ( $\delta^{13}\text{C}_{\text{PDB}} = -25.0$  ‰) and the 14C/12C ratio of the sample was calculated relative to this value.

††: the detection limits were estimated 3 SD of background measurements by taking into account the oxide interferences of  $^{157}\text{Gd}^{16}\text{O}$  onto  $^{173}\text{Yb}$

\*: calculated with the following equations

$$X_{\text{moxoite}} = A/\bar{D} \quad A = \text{La, Ce, Pr, Nd, Sm}$$

$$X_{\text{cheralite}} = 2\text{Ca}/D$$

$$X_{\text{Bamboo}} = (\text{Th+U-Ca})/D$$

$$D = (\text{La} + \text{Ce} + \text{Pr} + \text{Nd} + \text{Sm}) + 2\text{Ca} + (\text{Th+U-Ca})$$