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# Finite Element Modeling of Electrochemical-Mechanical Response of Ionic Conducting Polymer-Metal Composite Plates

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### 1. Introduction

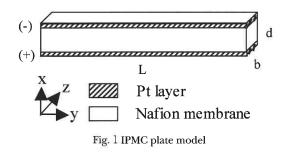
Much attention has been focused on ionic conducting polymermetal composites (abbreviated as IPMC) shown in Fig. 1 as an intelligent material of artificial muscles and MEMS. The electrochemical-mechanical behaviors of IPMC materials have been theoretically and experimentally studied for recent years<sup>1-5)</sup>.

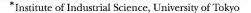
The purpose of the present study is to develop a prototype program of the computational tool for the design of IPMC actuators. The one-dimensional basic equations governing the electrochemical behavior of IPMC materials are discretized by the Galerkin finite element method<sup>6</sup>. The calculated results are compared with the finite difference solutions in order to check the validity of the present formulation. The three-dimensional finite element analysis for the deformation response of an IPMC plate is conducted by using the calculated water distribution as an eigenstrain.

## 2. Electrochemical Finite Element Formulation

### 2.1 Forward motion

When a Nafion membrane is under an electric field by the step voltage on platinum plates, hydrated cations are subjected to the electric force to the anode side. Hydrated cations are also subject-



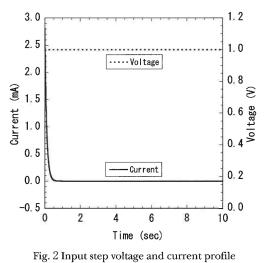


ed to viscous resistance force and diffusion force in the cathode direction. Based on the balance of these forces, the total charge Q(x,t) is expressed as follows<sup>5</sup>:

where *i* is the electric current as shown in Fig. 2. The initial and boundary conditions are given as follows<sup>4</sup>:

$$\left\{ Q^{(e)}(x,0) \right\} = N_a e S_x c_0 x \quad \dots \quad \dots \quad \dots \quad (2)$$
$$\left\{ Q^{(e)}(0,t) \right\} = 0$$
$$\left\{ Q^{(e)}(d,t) \right\} = N_a e S_x c_0 d \quad \dots \quad \dots \quad \dots \quad (3)$$

Table 1 shows the material constants and calculation conditions for eqs. (1)-(3) in which  $c_0 = c(x,0) = 1147 \text{ mol/m}^3$ .



The charge density is calculated from the total charge by the following equation<sup>4</sup>).

$$Q(x,t) = N_a e S_x \int_0^x c(\xi,t) d\xi \qquad (4)$$

Zero slopes are assumed as the boundary conditions.

The finite element formulation is conducted for eqs. (1) and (4). The total charge Q(x,t) is assumed to be a linear function of the element coordinate in each element as shown in the following equation:

where  $Q_i$  and  $Q_j$  are the nodal total charges. *h* is the element length. The following equation is obtained by the finite element formulation based on Galerkin method<sup>6</sup>:

$$\int_{0}^{n} [N]^{r} \left( \eta_{1} \frac{\partial Q(x,t)}{\partial t} - kT \frac{\partial^{2} Q(x,t)}{\partial x^{2}} + \frac{\partial Q(x,t)}{\partial x} \right)$$

$$\times \left\{ \frac{e}{\varepsilon_{e} S_{x}} \left[ \int_{0}^{t} i(\tau) d\tau + Q(x,t) - Q(x,0) \right] \right\} dx = 0 \quad \dots \quad (6)$$

Substituting eq. (5) into eq. (6) and integrating numerically by Euler method, the following form of equation<sup>7)</sup> is obtained:

$$\begin{bmatrix} A_{11}^{(e)} & A_{12}^{(e)} \\ A_{21}^{(e)} & A_{22}^{(e)} \end{bmatrix} \{ Q^{(e)}(x, t + \Delta t) \} = \begin{bmatrix} B_{11}^{(e)} & B_{12}^{(e)} \\ B_{21}^{(e)} & B_{22}^{(e)} \end{bmatrix} \{ Q^{(e)}(x, t) \} \cdots (7)$$

The following equation is obtained by the Galerkin finite element formulation<sup>6</sup> for eq. (4):

$$\int_{0}^{h} \left[ N \right]^{T} \left( \frac{\partial Q(x,t)}{\partial x} - N_{a} e S_{x} c(x,t) \right) dx = 0 \quad \cdots \cdots \cdots \cdots (8)$$

The substitution of eq. (5) into eq. (8) leads to the following form of equation<sup>7)</sup>:

$$\begin{bmatrix} C_{11}^{(e)} C_{12}^{(e)} \\ C_{21}^{(e)} C_{22}^{(e)} \end{bmatrix} \{ c^{(e)}(x,t) \} = \begin{bmatrix} D_{11}^{(e)} D_{12}^{(e)} \\ D_{21}^{(e)} D_{22}^{(e)} \end{bmatrix} \{ Q^{(e)}(x,t) \} \quad \dots \dots (9)$$

# 2.2 Backward motion

The second process is the slow diffusion of free water molecules in the anode direction. The density of water molecules w(x,t) is calculated by the following equation<sup>4)</sup>:

motion	
Coefficient of viscosity for hydrated $Na^+$ moving through membrane	$\eta_1 = 1.18 \times 10^{-11} (\text{N} \cdot \text{s/m}^2)$
Boltzman constant	$k = 1.380 \times 10^{-23} (\text{N} \cdot \text{m/K})$

Table 1 Material constants and calculation conditions for forward

Absolute temperature	T = 293(K)
Elemental charge	$e = 1.6 \times 10^{-19}$ (C)
Dielectric constant of hydrated Nafion membrane	$\epsilon_{e} = 2.8 \times 10^{-3} (C^{2}/N \cdot m^{2})$
Time interval	$\Delta t = 5 \times 10^{-4} \mathrm{s}$
Space interval	$h = 2 \mu m$
Cross section area	$S_x = 34.2 \times 10^{-6} (\text{m}^2)$
Avogadro number	$N_a = 6.02 \times 10^{-23} (/\text{mol})$

 Table 2 Material constants and calculation conditions for backward motion

Concentration of bound water to the fixed SO <sub>3</sub> groups	$w_1 = 4381(\text{mol/m}^3)$
Concentration of free water molecules	$w_2 = 4381 (mol/m^3)$
Time of ending of first stage and beginning of second stage	$t_1 = 0.11 \mathrm{s}$
Hydration number of $Na^+$ ions in swollen Nafion	<i>n</i> = 4.7
Coefficient of viscosity for free water molecules	$\eta_2 = 11.6 \times 10^{-11} (\text{N} \cdot \text{s/m}^2)$
Time interval	$\Delta t = 5 \times 10^{-3} \mathrm{s}$

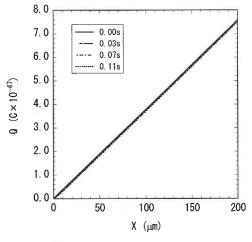


Fig. 3 Total charge distribution in membrane

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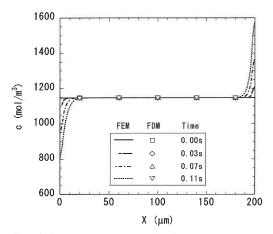


Fig. 4 Charge concentration distribution in membrane

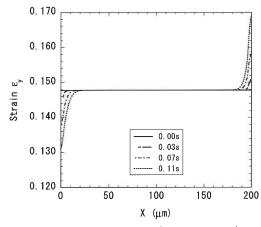


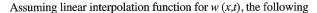
Fig. 6 Strain distribution (forward motion)

The initial and boundary conditions are given by eqs. (11) and (12), respectively<sup>5)</sup>.

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Table 2 shows the material parameters and calculation conditions for the backward motion.

The finite element formulation based on Galerkin method<sup>6</sup> for eq. (10) leads to the following equation:



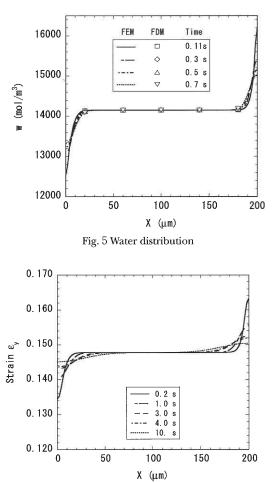


Fig. 7 Strain distribution (backward motion)

form of equation<sup>7</sup>) is obtained from eq. (13):

$$\begin{bmatrix} A_{11}^{(e)} A_{12}^{(e)} \\ A_{21}^{(e)} A_{22}^{(e)} \end{bmatrix} \{ w^{(e)}(x, t + \Delta t) \} = \begin{bmatrix} E_{11}^{(e)} E_{12}^{(e)} \\ E_{21}^{(e)} E_{22}^{(e)} \end{bmatrix} \{ w^{(e)}(x, t) \} \cdots (14)$$

### 3. Finite Element Analysis of Mechanical Response

The water mol  $W_i(x,t)$  is calculated as follows<sup>5)</sup>:

The water content  $m_i$  in each element is calculated by multiplying the weight of water per mol (=18g/mol) with the water mol  $M_w$  as below<sup>5</sup>).

The standard water content  $m_R$  is calculated by the following equation<sup>5)</sup>:

 $W = m_i / m_R m_R = S_x \cdot h \cdot \rho_w \quad \cdots \quad \cdots \quad \cdots \quad \cdots \quad (17)$ 

where  $\rho_w$  (=1379000g/m<sup>3</sup>) is the density of hydrated Nafion membrane. *W* is the water content of each element. The distribution of eigenstrain is evaluated by using the experimental relation between the strain and the water content, which is almost linear.

The three-dimensional finite element deformation analysis with eight-noded hexahedron element can be conducted by using the estimated eigenstrain  $\{\varepsilon_y(x,t)\}$  as the initial strain in the IPMC plate. The incremental stiffness equation for the initial strain increment  $\{\Delta \varepsilon_i\}$  is expressed as follows:

where

$$\left\{\Delta f_i\right\} = \int_{V_e} \left\{B\right\}^T \left\{\Delta\sigma_i\right\} dV \quad \dots \quad \dots \quad \dots \quad \dots \quad (19)$$

$$\{\Delta\sigma_i\} = [D]\{\Delta\varepsilon_i\}$$
 .....(20)

### 4. Numerical Example

The calculated results for the IPMC plate for the material properties of Table 3 are shown in Figs. 3–9. The results for the charge concentration distribution (Fig. 4) and the water distribution (Fig. 5) agree well with those by the finite difference modeling<sup>5)</sup>.

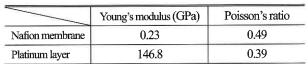
## 5. Concluding Remarks

The finite element formulation for the electrochemicalmechanical behaviors of IPMC materials has been conducted and its validity has been demonstrated by the numerical study for the IPMC plate. The multi-dimensional formulation for the electrochemical behavior and the application to other materials such as Flemion are the topics of future study.

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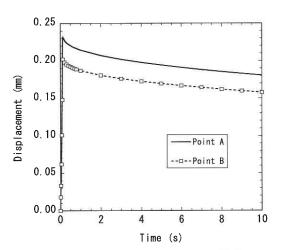


Fig. 8 Time-history of displacement for IPMC plate  $(L = 20 \text{ mm}, d = 200 \mu\text{m}, b = 20 \text{ mm})$ 

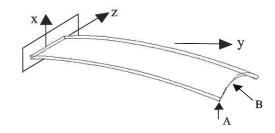


Fig. 9 Deformation of IPMC plate

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